Imperial College London

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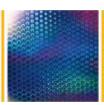
Micro-Structured Alumina Hollow Fibre Membranes – Potential Applications in Wastewater Treatment

















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Department of Chemical Engineering

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Membrane research at Imperial College

Membrane Formation

Flat sheet, hollow fibres and dual-layer hollow fibres

Membrane Separations

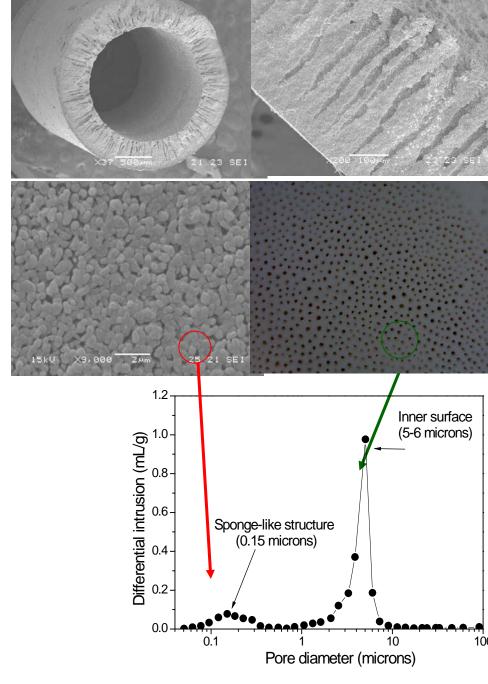
- Applications to solvent systems
- NH3 removal from water
- Removal of VOC from water
- Decolorization of dye-containing wastewater
- Forward osmosis for seawater desalination

Membrane Reactors

- Membrane reactors for energy applications
- Membrane reactors for dehydrogenation reactions
- Membrane Bioreactors/Artificial Organs
- Chlorophenol elimination
- Membrane reactors for water purification

Outline:

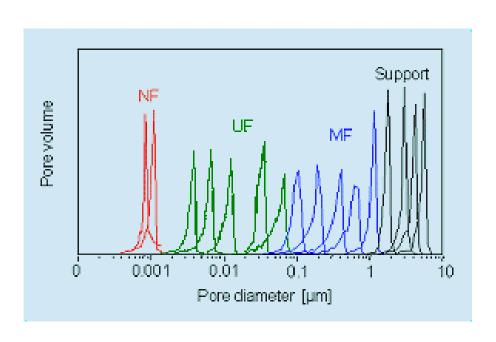
- (1) Micro-Structured
 Alumina Hollow Fibre
 Membranes
- (2) Possible applications
- (3) Graphene based ceramic composite membranes

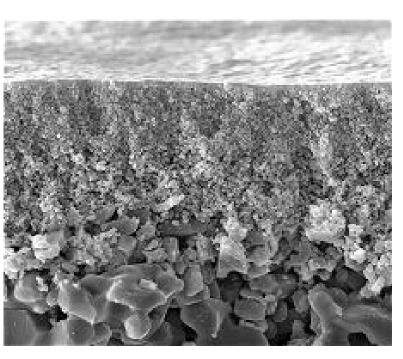


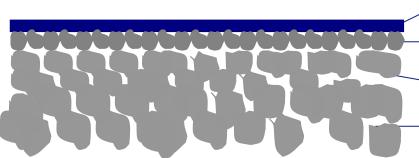
Bi-modal pore size distribution

1. Asymmetric ceramic hollow fibre membranes

Current commercial ceramic membranes







- 4. Separation layer (Dense or < 2 nm)
- 3. Separation Layer (3-100 nm)
- 2. Intermediate layer(s) (100-1500 nm)
- 1. Porous support (1-15 μm)

Asymmetric Structure

2. Separation layer

1. Porous support (1-15 μm)

Randomly packed **Sponge-like structure** particles **Micro-channels** Properties determined

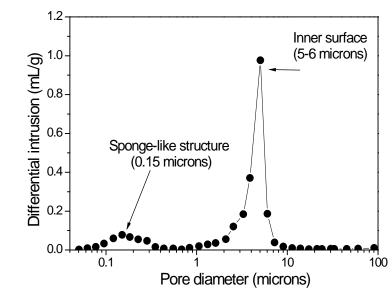
- by suspension composition

Inner surface

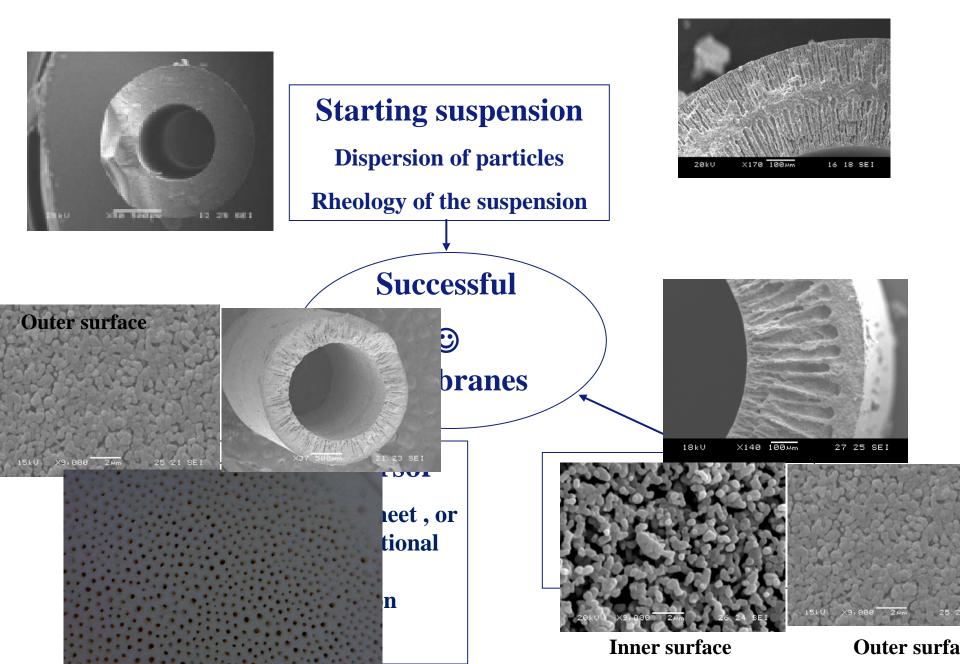
Outer surface

- Micro-channels result from hydrodynamically unstable viscous fingering
- Micro-channel structure is retained during heat treatment

Bi-modal pore size distribution

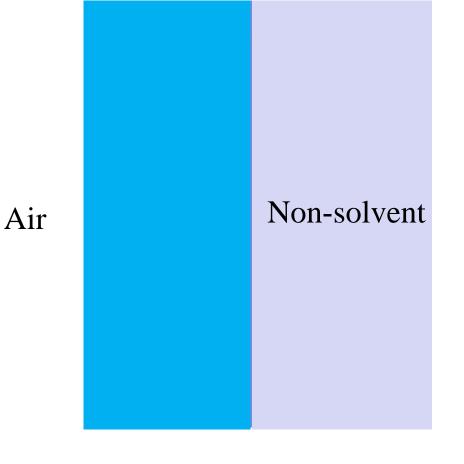


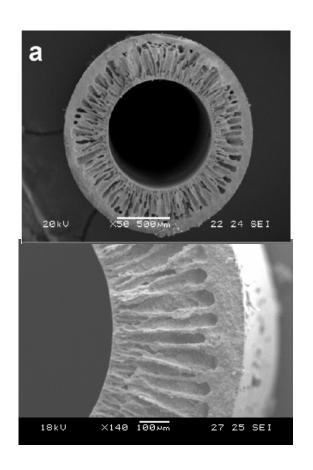
Preparation method



Micro channel formation mechanism (viscous fingering)

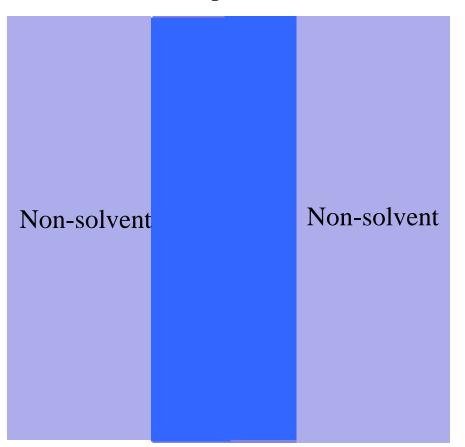
Suspension

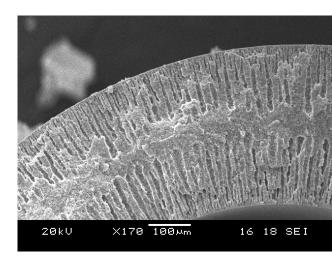




Micro channel formation mechanism

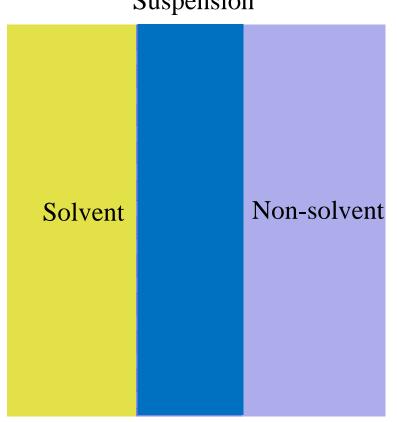
Suspension

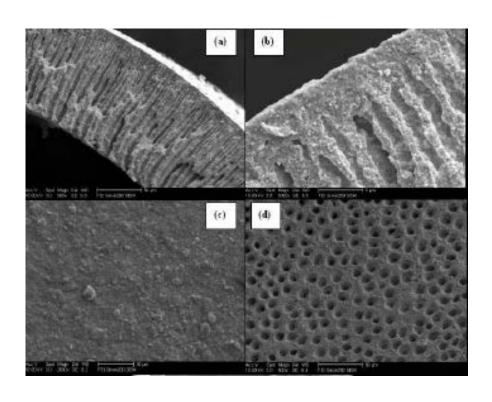


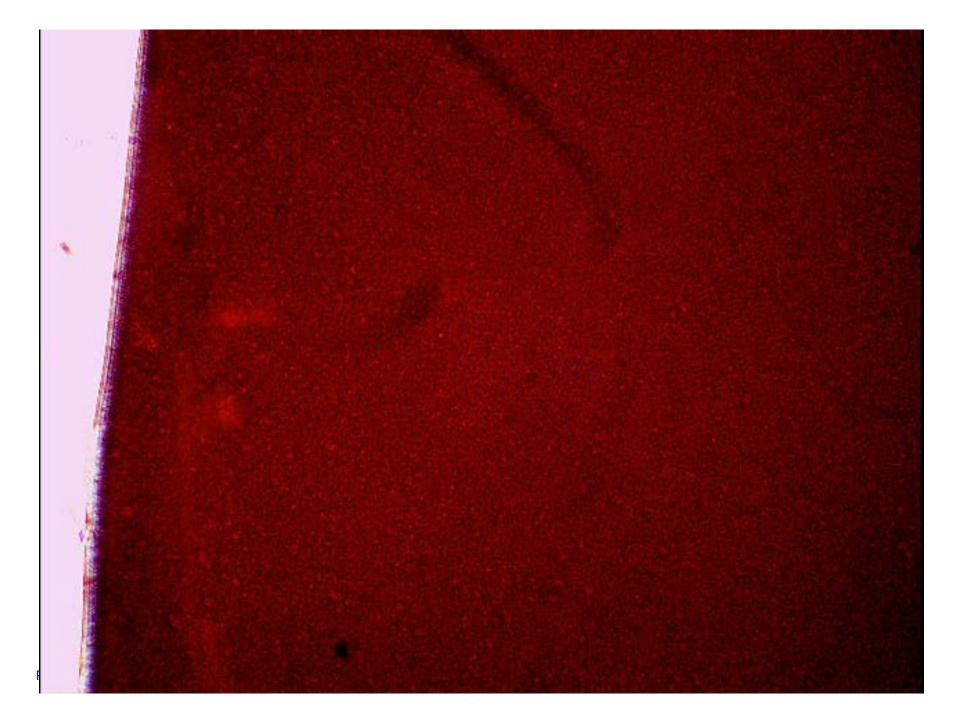


Micro channel formation mechanism

Suspension

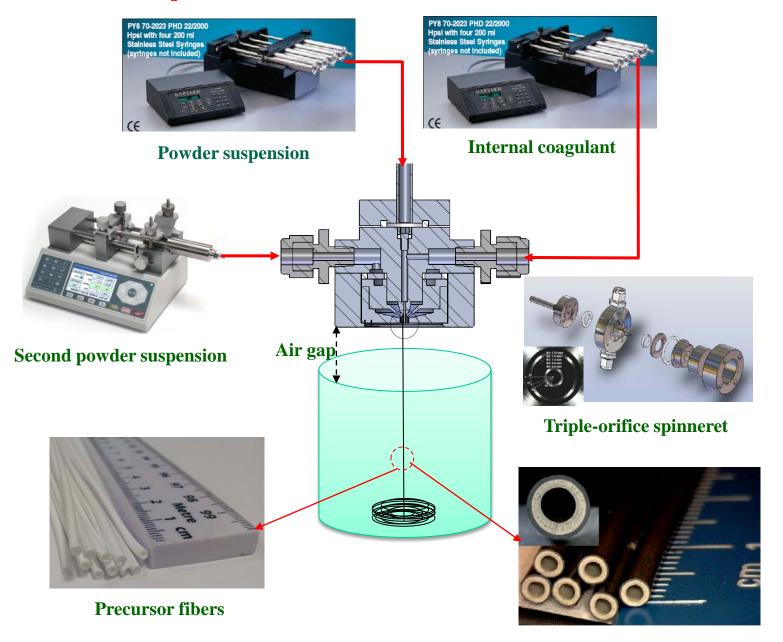






Membrane precursors

Phase inversion technique/Viscous fingering



Precursor dual-layer fibers

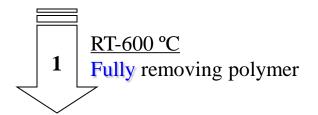
Sintering

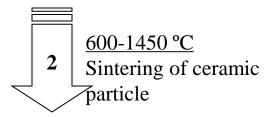
Normal sintering

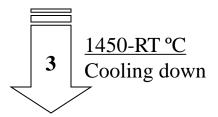
(in static air)

Precursor HF

(ceramic particle & polymer)







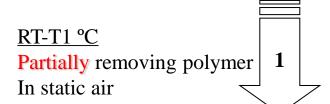
Ceramic HF

Controllable sintering

(in controlled atmosphere)

Precursor HF

(ceramic particle & polymer)

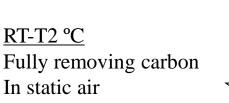


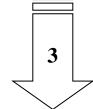
T1-1450 °C

Sintering of membrane In pure N_2

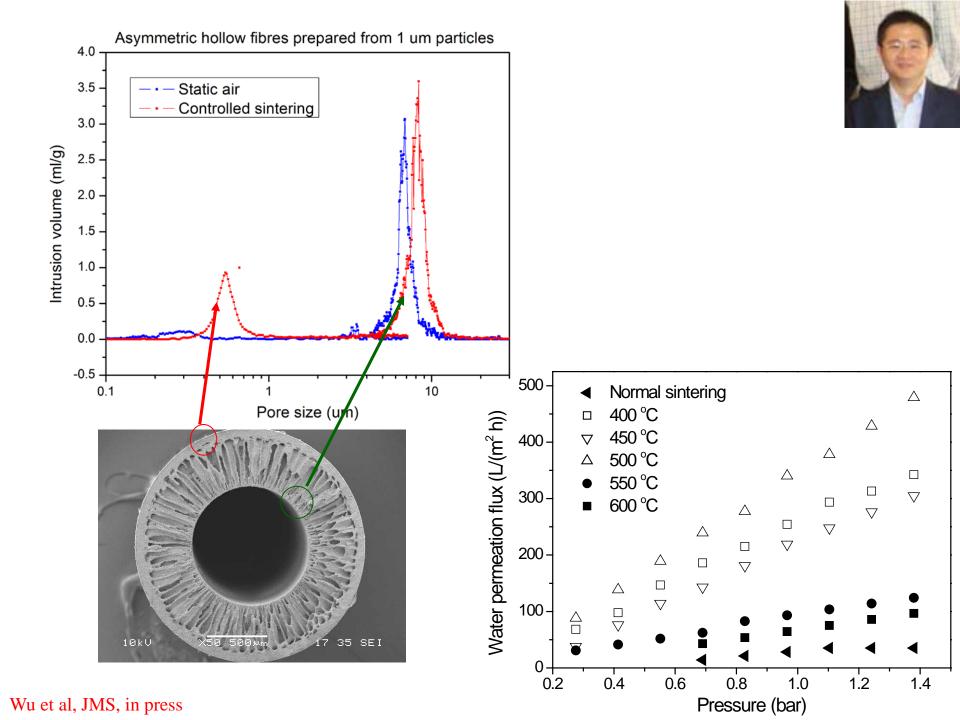
1450-RT °C Cooling down

In pure N₂



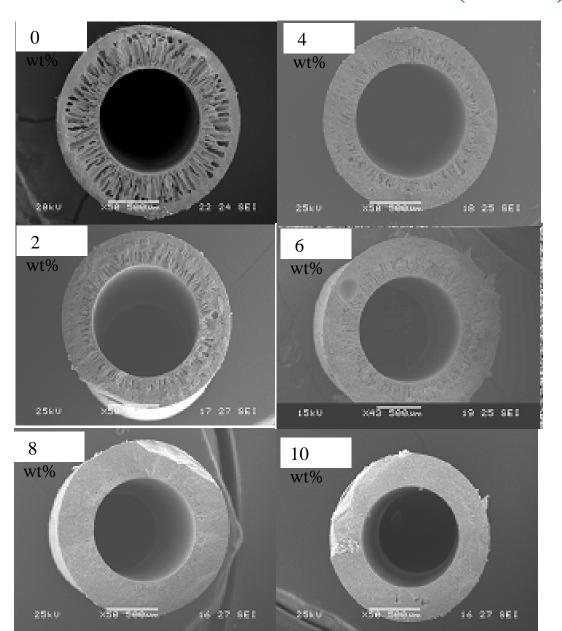


Ceramic HF



Fibre morphology:

Effect of Non-solvent additive (water)

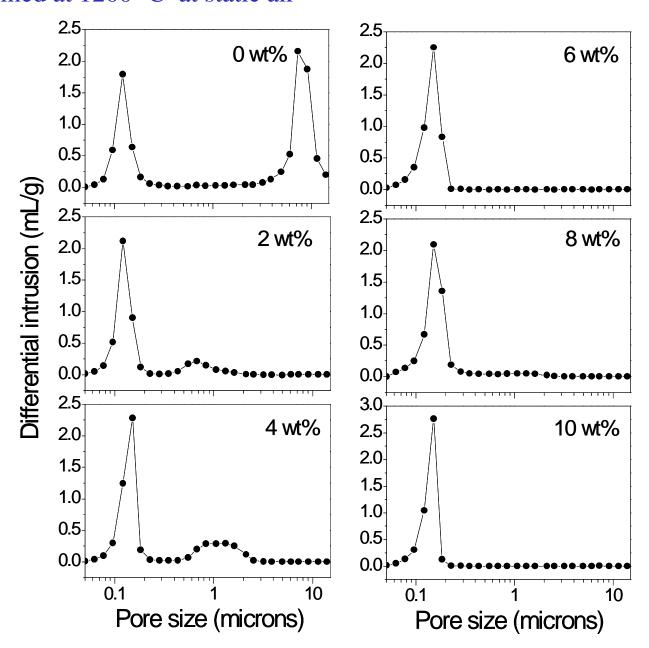




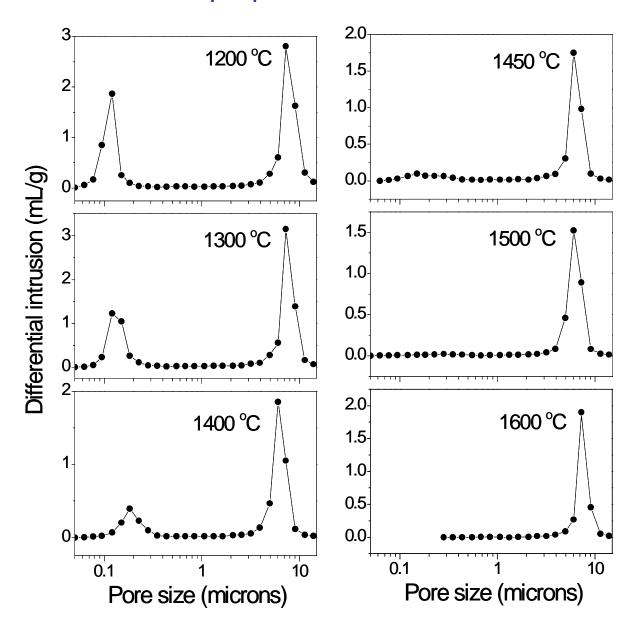
Benjamin KING BURY

B.F.K. Kingsbury, K. Li / Journal of Membrane Science 328 (2009) 134–140

Mercury intrusion data for fibres prepared with 0 wt%-10 wt% water calcined at 1200 °C at static air



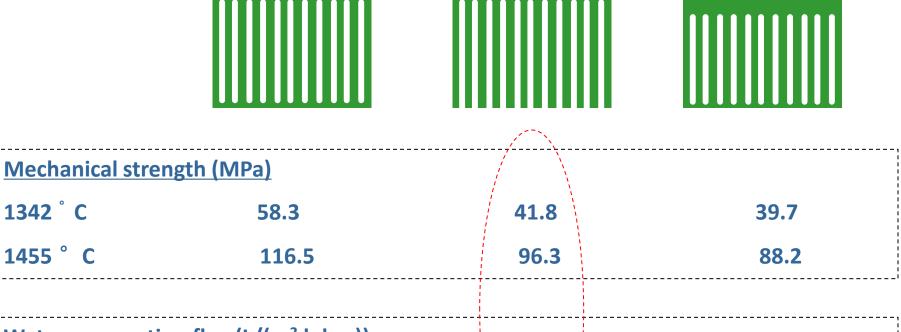
Mercury intrusion data of pore size vs. sintering temperature for an asymmetric hollow fibre prepared with 0% water



Water permeation data

1342 ° C

1455 ° C



Water permeation flux (L/(m² h bar))				
1342	°C	1069	1874	997
1455	°C	753	1088	664

Water flux is comparable with commercial membranes (Metawater 1250@0.1µm, Hyflux 1500@0.2μm and 750@0.1μm), using lower sintering temperatures and less fabricating steps.

Inner surface (5-6 microns) Differential intrusion (mL/g) Sponge-like structure (0.15 microns)

Bi-modal pore size distribution

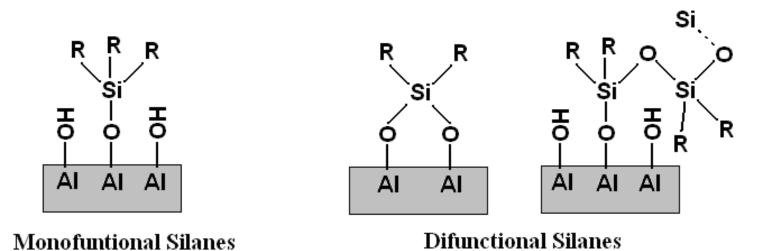
Pore diameter (microns)

100

Possible applications

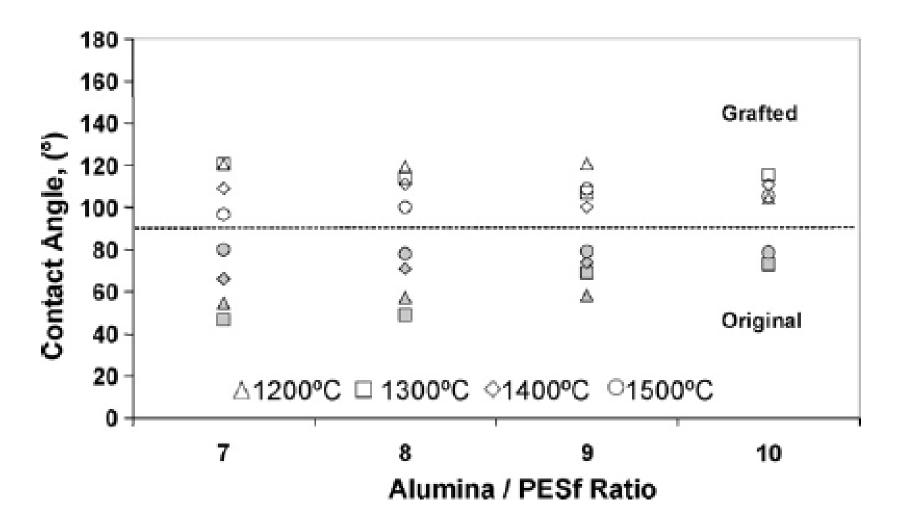
- Warm surfactant/oil /water mixtures usually found in metal cleaning operations,
- Membrane contactor for desalination after surface modification,
- Membrane support for further deposition of nanofiltration or reverse osmosis membranes for desalination,
- The finger-like layer can be used as micro-reactors for enhancement of chemical reactions

Membrane distillation -surface modification by fluoroalkylsilanes (FAS) grafting

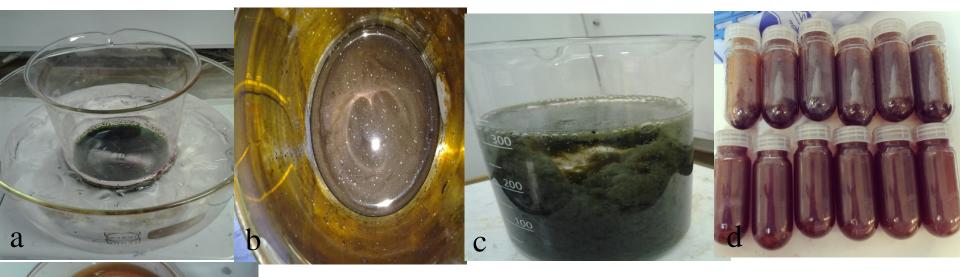


Trifunctional Silanes

The advancing contact angles of the membranes before and after FAS grafting.



3. Graphene based ceramic composite membranes

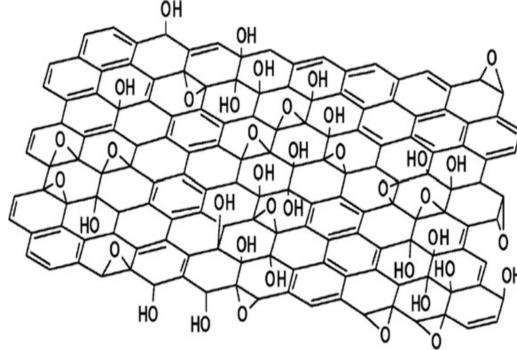




Synthesis of graphene oxide via Modified Hummer's Method

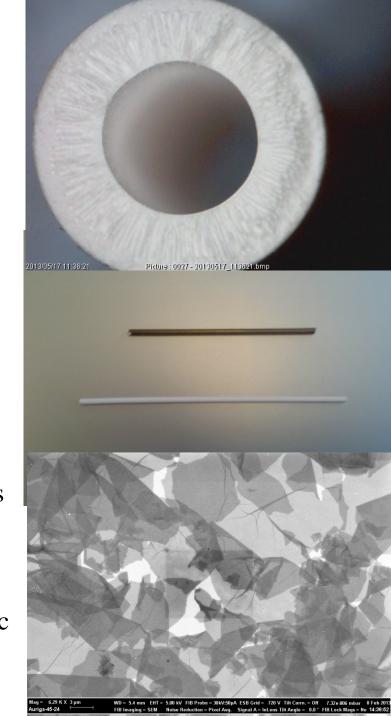
- a) mixture stirred in ice water bath,
- b) mixture stirred after 5 days (turned to dark brown),
- c) addition of water and H_2O_2 into the mixture,
- d) purification of mixture by centrifugation,
- e) drying process of purified graphene oxide

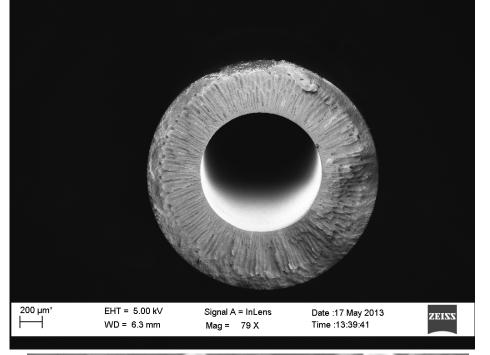
3. Graphene based ceramic composite membranes

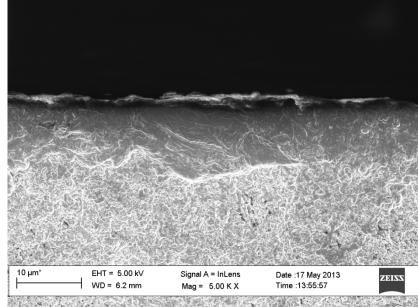


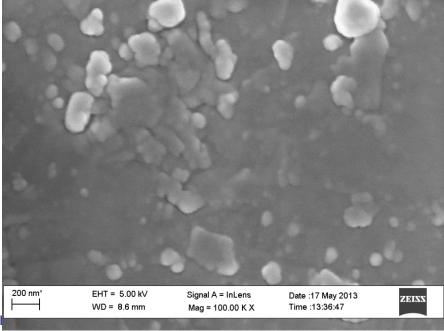
Zeta potential study

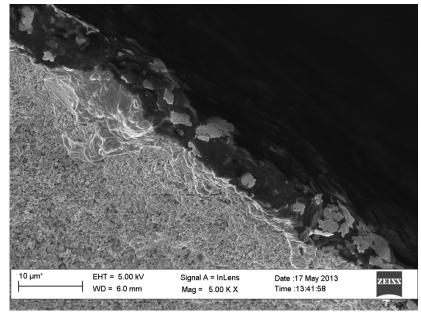
- •Graphene oxide has highly negative surface charges
- •Easily dispersed in water due to ionization of carboxylic acid and the phenolic hydroxyl groups.
- •Its colloidal suspension is stable in water, not only because of the hydrophilicity but also by electrostatic repulsion





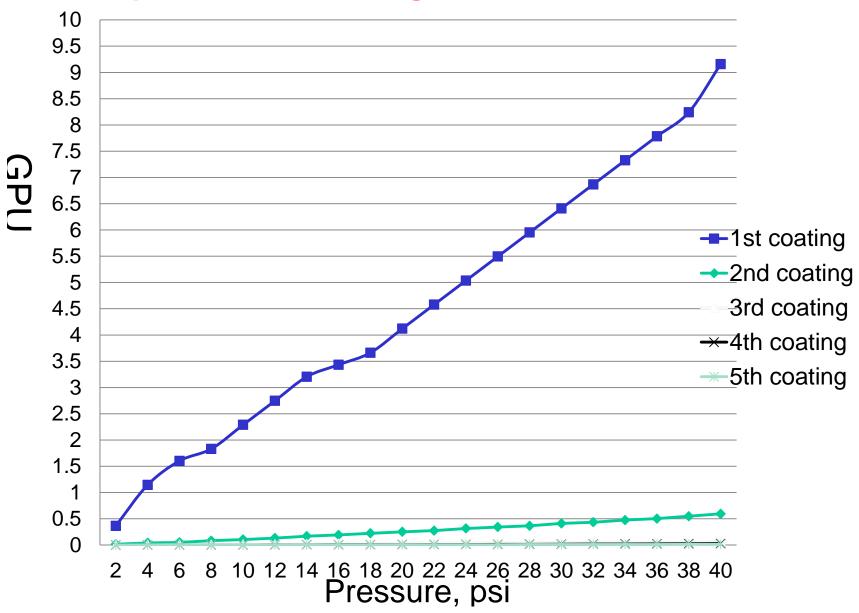




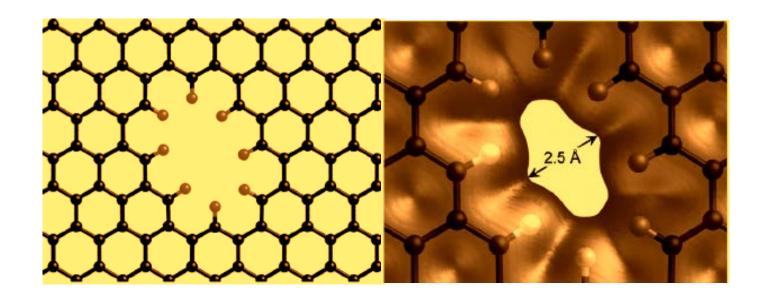


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Graphene Oxide Nitrogen Permeation FLux



Functionalization with hydrogen



- Created by removing 2 neighboring rings from the graphene sheet and then functionalized with hydrogen
- Dimensions are about 2.5 Angstroms by 3.5 Angstroms

Conclusions

- Micro-structured ceramic hollow fibre membranes consisting of two different pore size distributions with mean pore size of ~0.1 mm (sponge-like layer) and ~8 mm (micro-channels) can be prepared in a single step using a viscous fingering induced phase inversion technique,
- The percentage of the sponge layer (or micro-channels) over whole fibre cross section can be varied by controlling the composition of spinning suspension or other spinning parameters such as air-gap and take up velocities, etc.
- The outer sponge layer (~0.1 mm) may serves as a substrate for deposition of nanofiltration or RO membranes for seawater desalination, while micro-channels reduces mass transfer resistances considerably and can also be used as microreactors once an appropriate catalyst is deposited.

Acknowledgements

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Thank Mou

Questions?