Lecture 3: Water Pollution

(Jan 9th 2015) by Dr. Arun Kumar (arunku@civil.iitd.ac.in)

Objective: To introduce water pollutants (chemical, biological) and nutrients which might cause problems

Pathogenic Microorganisms

Typical Pathogens Excreted In Human Feces

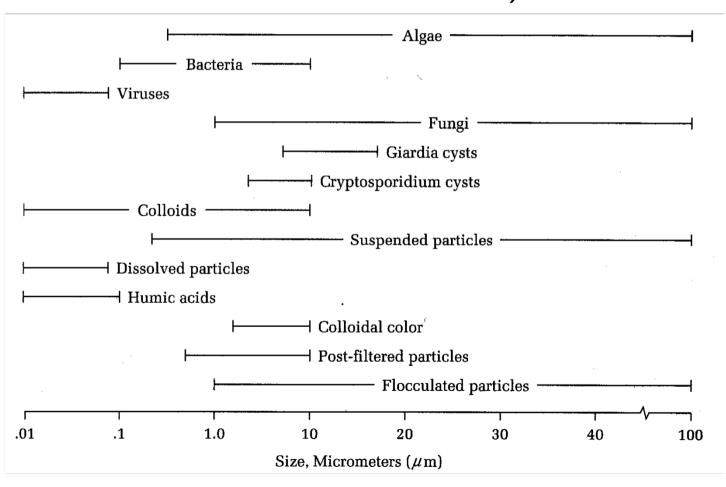
Pathogen Group and Name	Associated Diseases	
Virus		
Adenoviruses	Respiratory, eye infections	
Enteroviruses		
Polioviruses	Aseptic meningitis, poliomyelitis	
Echoviruses	Aseptic meningitis, diarrhea, respiratory infections	
Coxsackie viruses	Aseptic meningitis, herpangina, myocarditis	
Hepatitis A virus	Infectious hepatitis	
Reoviruses	Not well known	
Other viruses	Gastroenteritis, diarrhea	
Bacterium		
Salmonella typhi	Typhoid fever	
Salmonella paratyphi	Paratyphoid fever	
Other salmonellae	Gastroenteritis	
Shigella species	Bacillary dysentery	
Vibrio cholerae	Cholera	
Other vibrios	Diarrhea	
Yersinia enterocolitica	Gastroenteritis	

Pathogenic Microorganisms

Protozoan			
Entamoeba histolytica Giardia lamblia Cryptosporidium species	Amoebic dysentery Diarrhea Diarrhea		
Helminth	Q: Can you list sources		
Ancylostoma duodenale (hookworm) Ascaris lumbricoides (roundworm) Hymenolepis nana (dwarf tapeworm) Necator americanus (hookworm) Strongyloides stercoralis (threadworm) Trichuris trichiura (whipworm)	Hookworm Ascariasis Hymenolepiasis Hookworm Strongyloidiasis Trichuriasis	exposure of pathogenic microorganisms?	

Source: Hammer and Hammer, 1996.

Sizes of Microorganisms and Other Particles (size influences removal through filtration method)



Inorganic Pollutants

- Heavy Metals
- Arsenic
- Nitrates
- chromium
- Asbestos
- Other?

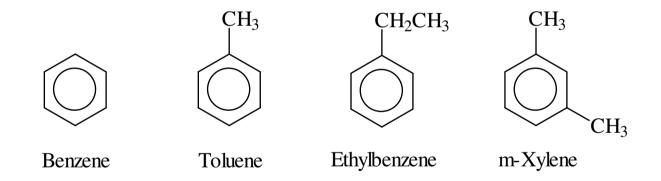
Effect =f(valency type, concentration)

For example: toxic effects of arsenate(AsIV)) and arsenite (As(III)) might be different.

Q: Can you list sources which might give you exposure of these inorganic pollutants?

Volatile Organic Compounds (VOCs)

- Petroleum constituents:
 - benzene and substituted benzenes



- prevalent in gasoline, diesel fuel, heating oil
- most easily transported, slow degradation, toxic

Volatile Organic Compounds (VOCs)

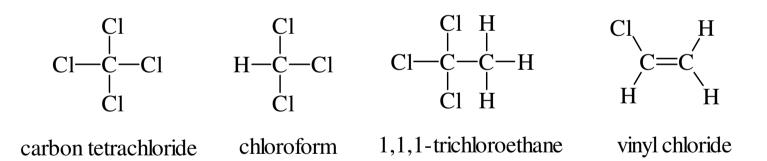
Oxygentated gasoline additives

Methyl tertiary butyl ether (MTBE)

- added to gasoline to improve air quality
- very soluble, resistant to degradation, toxic
- attempt to solve one problem caused another (spills)

Volatile Organic Compounds (VOCs)

- Chlorinated solvents
 - C1 and C2 aliphatics



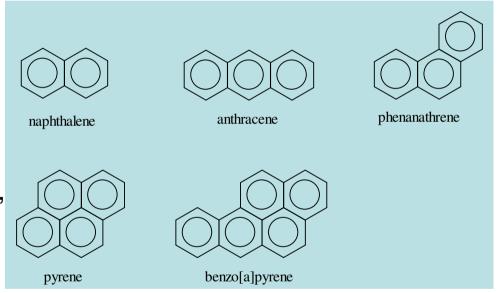
- widely used in degreasing, dry cleaning, extraction
- somewhat soluble, volatile, difficult to degrade

Pesticides (hydrophobic organic compounds)

- Organochlorine Insecticides: were commonly used in the past (e.g. DDT and chlordane).
- Organophosphate Pesticides: most are insecticides, some are very poisonous (they were used in World War II as nerve agents)
- Carbamate Pesticides: affect the nervous system
- Pyrethroid Pesticides: were developed as a synthetic version of the naturally occurring pesticide pyrethrin, which is found in chrysanthemums

Polycyclic Aromatic Hydrocarbons (PAHs)

Polycyclic aromatic hydrocarbons (PAHs) are a group of over 100 different chemicals that are formed during the incomplete burning of coal, oil and gas, garbage, or other organic substances.



They are potent carcinogens, hydrophobic organic compounds and they tend to associate with particles.

Polychlorinated biphenyls (PCBs)

- Polychlorinated biphenyls are mixtures of individual chlorinated compounds (known as congeners).
- PCBs are either oily liquids or solids that are colorless to light yellow.
- PCBs have been used as coolants and lubricants in transformers, capacitors, and other electrical equipment.
- The manufacture of PCBs was stopped in the U.S. in 1977.
- They are hydrophobic and they bio-accumulate in the food chain.

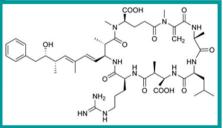
Other Organic Pollutants

- Fertilizers
- Surfactants
- Explosives, propellants
- Chlorination By-Products (e.g. trihalomethanes)
- Antibiotics
- Pharmaceuticals
- Personal care products
- Cyanobacterial toxins
- nanomaterials









Microcystin-LR (hepatotoxin)

Excess Nutrients

- Nitrogen and phosphorus are nutrients required by all living organisms. They are considered pollutants when they are in excess.
- Excessive nutrients often lead to large growths of algae which in turn become oxygen-demanding material whey they die and settle.

Excess Nutrients

106
$$CO_2$$
 + 16 NO_3^- + HPO_4^{2-} + 122 H_2O → $C_{106}H_{263}O_{110}N_{16}P$ + 138 O_2 algae

Algae +
$$O_2 => CO_2 + H_2O + NH_3$$

Nitrogen and phosphorus are typically the limiting factors

Phosphorus

- Phosphorus is typically the limiting nutrient in lakes, and algae growth is linked to phosphorus inputs.
- P Sources
 - fertilizers
 - detergents
 - wastewater
- P can exist in a variety of chemical forms

Nitrogen

- Nitrogen is often the limiting nutrient in ocean waters and some streams
- Nitrogen can exist in numerous forms, but nitrate (NO₃-), nitrite (NO₂-), ammonia (NH₃) are most commonly measured
- Sources are primarily from fertilizers and acid deposition

Factors Controlling Eutrophication

Stoichiometry of photosysnthesis (C,N,P, O & H)

106
$$CO_2 + 16 NO_3^- + HPO_4^{2-} + 122 H_2O \rightarrow C_{106}H_{263}O_{110}N_{16}P + 138 O_2$$

$$\frac{N}{P} = \frac{16 \times 14}{1 \times 31} = 7.2$$

 $\frac{N}{P} = \frac{16 \times 14}{1 \times 31} = 7.2$ It takes ~ 7 times more N than P to produce a given mass of algae It takes ~ 7 times more N than

- Liebig's law of the minimum growth will be limited by the availability of the nutrient that is least available relative to the need
- Most fresh water systems are phosphorus limited

Salts

- Dissolved solids, or salts, may be present as any number of ions
 - cations: Na+, K+, Mg²⁺, Ca²⁺
 - anions: Cl⁻, SO₄²⁻, HCO₃⁻
- Typically measures as total dissolved solids (TDS)
- Water classification
 - freshwater <1500 mg/L TDS
 - brackish water 1500 5000 mg/L
 - saline water >5000 mg/L
 - sea water 30-34 g/L

Salts

Sources

- industrial discharges
- deicing
- evaporative losses
- minerals
- sea water intrusion

Effects

- natural fresh water population threatened
- limits use for drinking
- crop damage/soil poisoning (cannot use for irrigation)

Suspended Solids

- Organic and inorganic particles in water are termed <u>suspended solids</u>
- May be distinguished from <u>colloids</u>, particles that do not settle readily

- Sources
 - storm water
 - wastes
 - erosion

- Problems
 - sedimentation
 - may exert oxygen demand
 - primary transport mechanism for many metals, organics and pathogens
 - aesthetic
 - complicates drinking water treatment

Oxygen-Demanding Wastes

 When <u>organic substances</u> are broken down in water, oxygen is consumed organic C + O₂ → CO₂

For example:

$$CH_3COOH + 2O_2 => 2CO_2 + 2H_2O$$

 $C_6H_{15}O_6N + 6O_2 => 6 CO_2 + 6 H_2O + NH_3$

Oxygen-Demanding Wastes

- High oxygen levels necessary for healthy stream ecology.
- For example:
 - trout require 5-8 mg/L dissolved oxygen (DO)
 - carp require 3 mg/L DO

Oxygen Demanding Wastesmeasurement/estimation

- Estimated stoichiometrically by theoretical oxygen demand (ThOD)
- Measured by oxygen demand potential
 - biochemical oxygen demand (BOD)
 - Nitrogenous oxygen demand (NBOD)
 - chemical oxygen demand (COD)

Next class: tuesday