# Wastewater Instrumentation for Monitoring and Control





# Hach, you know us?

67 years (started in 1946)
An American Company!
70+ Technical Sales Assoc nationwide
80+ Service specialists nationwide



#### PRICE LIST November 1948

DETERMINATION OF TOTAL HARDNESS					b
	5 g.	25 g.	100 g.	500 g.	
pisodium Dihydrogen Versenate, Analytical Reagent Indicator F241 (Ericchromeschwartz T)	52.00	(6.00		010.00	
Calcium Corbonate, Analytical Reagent			.75	2.00	
DETERM MATION OF TOTAL HARDNESS	500	ml. 1	liter	4 liter	
Standard Disodium Dihydrogen Versenate (titer adjusted so that 1.00 ml. is equivalent to 1.00 mg. CaCO3)			£2.00	16.00	
Indicator F241	-		755	20.00	
(2.00 g. per 500 ml. methanol) Huffer Solution	\$3,	.00	4.50		
(a 1:5 mixture of 1 M NH Cl and 1 M NH,	(HO		2.00	4.00	
Standard Calcium Chloride Solution (1.00 ml. equivalent to 1.00 mg. CaCO3)			2.00		
ANALYTICAL REAGENTS	1 g.	10 g.	100 g.	500 g.	
Nioxime (1,2-Cyclohexanedionedioxime) Tetraphenylarsonium Chloride Tetraphenylphosphonium Chloride Alpha Furildioxime Copper Selenite (Kjeldahl Catalyst)	\$1.00 1.00 1.00 1.00	\$5.00 5.00 5.00 5.00	100 E 10 E 100	105.00 105.00 105.00	
PRIMARY STANDARD MATERIALS			100 g.	500 g.	
Constant Boiling Hydrochloric Acid			Too B.	\$ 3.40	
Sulfamic Acid			\$1.40	4.40	1
Potassium Acid Iodate Electrolytic Iron (Ignited in Hydrogen)			9.25	37.50	3
Ferrous Ethylenediamine Sulfate (Cesper's	Salt)		1.80	5.00	

ANALYZEĎ SAMPLES

For analyzed samples for student analysis and standard sample work refer to price list of STANDARD SAMPLE COMPANY, Ames, Iowa. The samples currently available and the number of samples of each kind are:
Barium Chloride (for H<sub>2</sub>O, 3O), Soluble Sulfate (SO<sub>3</sub>, 2O), Soluble Chloride (C1, 25), Iron Ore (Vol. Fe, 2O; Crav. Fe, 16), Sodium Carbonete (Na<sub>2</sub>O, 25), Kjeldahl Nitrogen (N, 25), Limestone (SiO<sub>2</sub>, P<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub>, Al<sub>2</sub>O<sub>3</sub>, P<sub>2</sub>O<sub>5</sub>, CaO, NgO, CO<sub>2</sub>, 65), Brass (Sn, Cu, Fe, R<sub>2</sub>O<sub>3</sub>, Zn, Ni, 35), Copper Ore (Cu iodometric, 27), Lead Ore (Pb, 35).

HACH CHELICAL AND ONYCEN COMPANY

All prices are f.o.b. Ames, Iowa 30 days net . Ames, Iowa

Clifford Hach Harvey Diehl



# What's the Objective Today?

- Think about the appropriate instruments for different applications
- Workshop solutions to common problems
- Products/Ideas/Technologies that might help make your day easier
- Network with others in the room



## **Today's Training Agenda**

- Types of Treatment Systems
- Overview of Process Steps
- Instrumentation and Control Applications
  - Nutrient Removal
  - Aeration
  - Suspended Solids
  - -pH
  - Organics



# **Today's Training Agenda**

- Making Your Data Meaningful
- Troubleshooting Mercoid Switches
- Hands-on instrument review



# Why Do We Treat Wastewater?

To protect public health &

To protect the environment

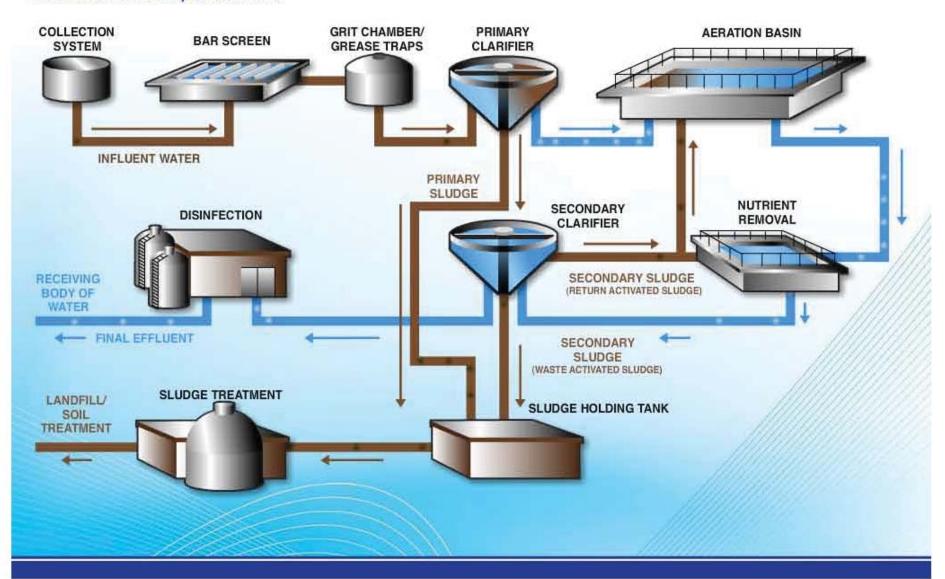


### **How Do We Treat Wastewater?**



# **Process Step Overview**

#### Click on a Process Step to learn more.







Wastewater flows into the collection system through a series of pipes and pumps. Flow measurements and influent sampling are required by NPDES permit and are needed to calculate loading rates on plant processes. Parameters such as pH and conductivity are often measured to determine maintenance needs and to monitor industrial dischargers.

#### **Parameters Tested**

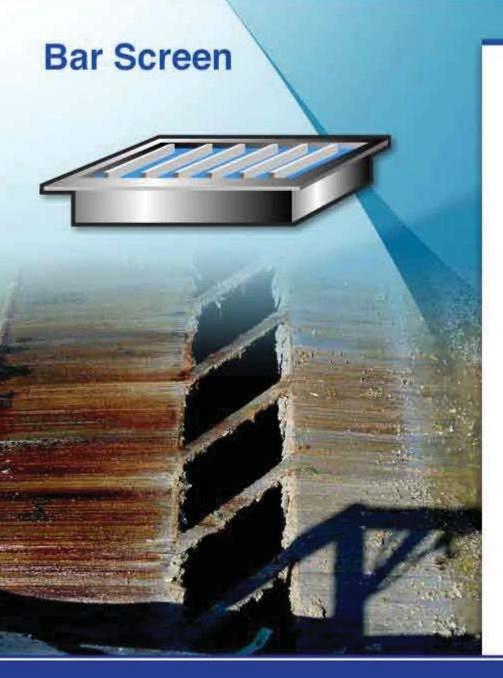
BOD

COD

pH Total Organic Carbon (TOC)

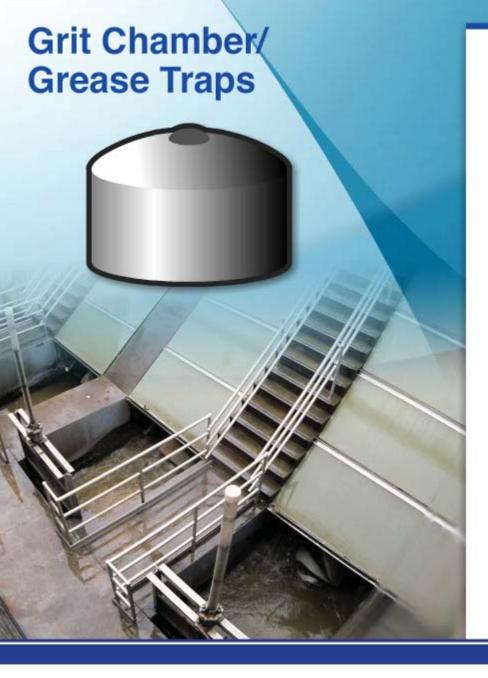
Total Suspended Solids (TSS)





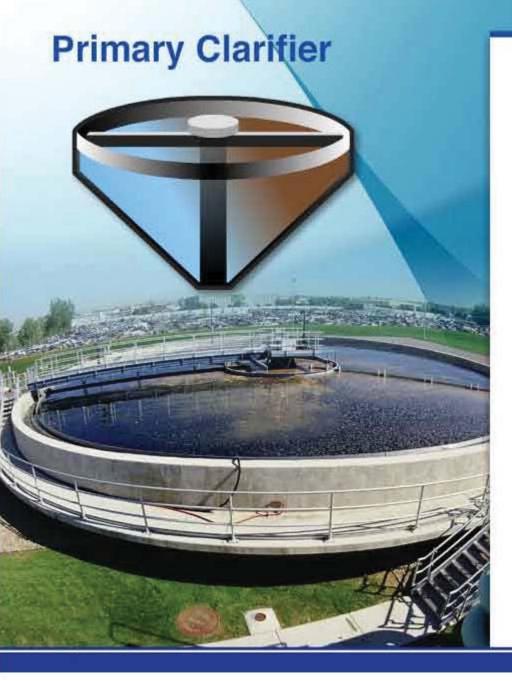
The bar screen is an inclined grid that is designed to remove large debris such as boards, branches, plastic, rags or any large item that may damage the equipment or plug pipes. The coarse solids collected at this step are taken to a landfill for disposal.

Measurements are generally not taken at this step.



The grit chamber removes particles larger than 0.2 millimeters in size through sedimentation. Inorganic solids like sand, cement, cinders, and eggshells as well as some dense organic materials like grains will settle in the grit basin. Air is bubbled through the water to help keep the organic material in suspension. The grit settles to the bottom where it is pumped, dewatered, and hauled to the landfill. Some grit chambers also contain grease traps.

Grease traps remove substances with a lower relative density, such as oils and fats, from kitchens, workshops and gasoline stations. The oils and grease rise to the surface and can be removed with the skimmer. Incineration and recycling are two of the available disposal methods for grease and oil sediment.



Primary clarifiers remove 95% of the settable solids from the wastewater stream. A low flow rate of less than 1 ft/s in the primary clarifier allows heavier suspended solids to settle out and floatable material such as grease to rise to the top of the tank. Flights made of wood or fiberglass scrape the settled sludge on the bottom to a hopper where it is pumped out to the solids handling facility. Then the flights push the grease on the surface to a scum trough for disposal at a sanitary landfill.

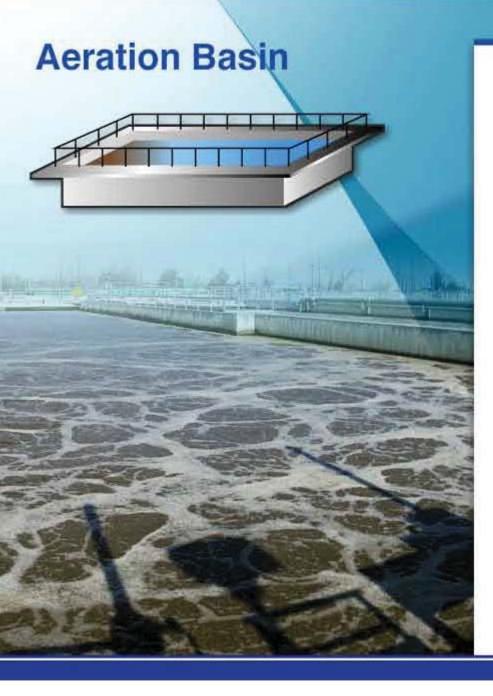
#### **Parameters Tested**

Ammonia

Phosphate

Sludge Level

Total Suspended Solids (TSS)



Ammonia

BOD

COD

Dissolved Oxygen

Nitrate

Nitrite

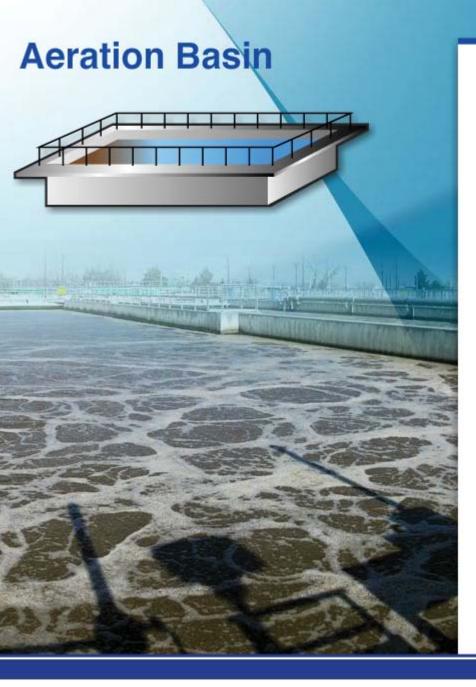
ORP

pH

Phosphate

Total Suspended Solids (TSS)

The Activated Sludge Process is one of the most common secondary treatment processes in use today. This process uses Saprophytic bacteria to breakdown suspended solids and dissolved BOD that flow from the primary clarifier. To maintain an optimal environment for these bacteria, it is important that temperature, pH, oxygen,



As the bacteria or Mixed Liquor Suspended Solids (MLSS) in the aeration basin flows from the aeration basin to the secondary clarifier, the solids settle out. Some of these solids are returned (Return Activated Sludge) to the mix with the primary effluent to continuously treat the wastewater. A percentage of the settled solids are removed from the process by wasting (Waste Activated Sludge) to a handling facility. Solids and Oxygen concentrations are critical to the process.

A number of methods are used to control the solids concentration in an aeration basin.

Sludge age is the average time (in days) that a bacterial cell stays in the biological system. It is calculated by determining the mass of solids in the aeration tank and dividing it by the mass of solids wasted from the system.

Mean Cell Residence Time (MCRT) is very similar to sludge age; however, the solids inventory from the blanket depth in the secondary clarifiers and the solids from the clarifier effluent account for a more accurate and complete solids inventory in the plant.

Using the sludge volume concentration from the aeration basin by performing a 30-minute settling test is another method of controlling the return rate. In addition, the Sludge Volume Index(SVI) concentration can be used to calculate the concentration of solids in the Return Activated Sludge (RAS) line. This value can later be used to calculate the return rate.



#### **Parameters Tested**

Ammonia

BOD

COD

Dissolved Oxygen

Nitrate

Nitrite

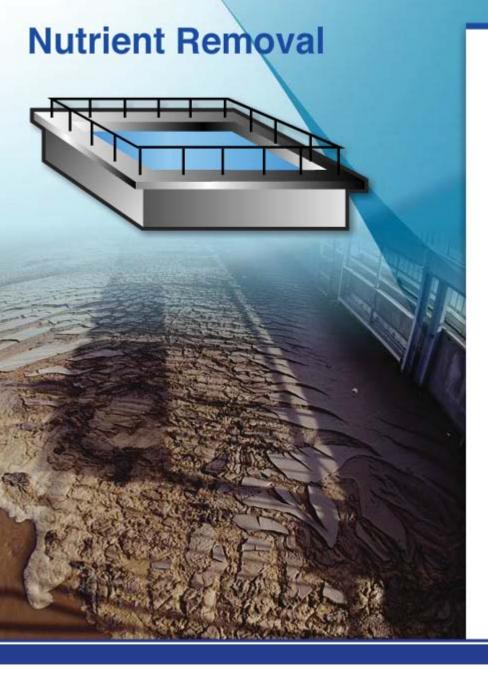
ORP

pH

Phosphorus

Total Suspended Solids (TSS)

Turbidity



Nutrients are essential to the life of the bacteria in the aeration basin. Bacteria break down ammonium, nitrate and organic nitrogen compounds in the course of the metabolic processes involved in cell growth. If supply were to exceed demand, however, incomplete removal of nutrients would result in an effluent that would contribute to the pollution of environmental waters. Ammonium, Nitrate, and Phosphate are the key parameters for advanced nutrient elimination.

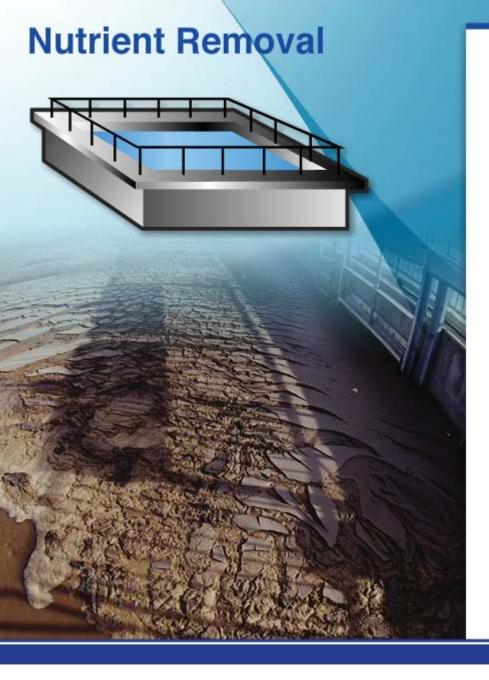
#### Nitrogen Removal

Nitrogen enters wastewater treatment plants as Ammonia and Organic Nitrogen and is typically removed biologically. Utilizing aeration, the nitrification stage utilizes aerobic bacteria to turn Ammonia into Nitrite and then Nitrate. Nitrogen removal happens in the denitrification stage as bacteria consume the Oxygen from Nitrate turning it into Nitrogen gas that escapes into the atmosphere.

#### Phosphorous Removal

Phosphorus does not have a gaseous form, so it must be converted from soluble form to particulate form for removal. Chemical Phosphorus removal uses Iron or Aluminum compounds to combine with and precipitate out Phosphate.

Biological Phosphorus removal utilizes Phosphate accumulating



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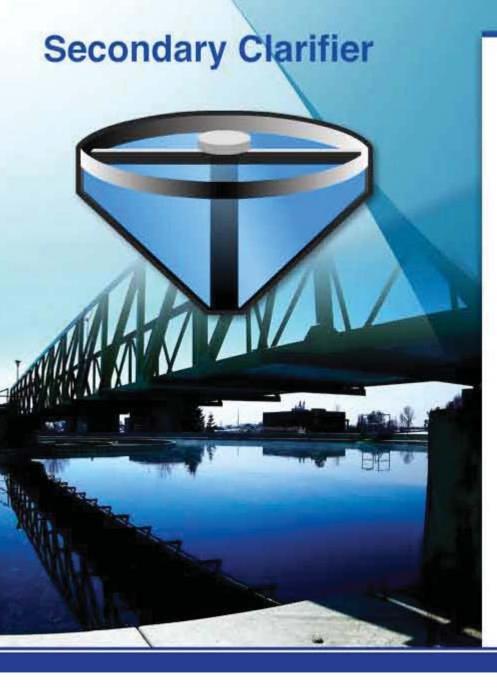
Biological Phosphorus removal utilizes Phosphate accumulating organisms (PAOs). In the anaerobic environment, the PAOs consume volatile fatty acid compounds, while they release Phosphate into the water. In an aerobic environment, the PAOs consume more Phosphate than they can digest, which results in a net removal of Phosphorus from the wastewater. The PAOs settle in the secondary clarifier and are removed when the sludge is wasted.

#### Chemical and Energy Savings

In biological nutrient removal, aeration can be 70% of a plant's budget because most wastewater treatment plants play it safe by running their blowers full speed 24 hours a day 7 days a week.

Monitoring your process in real-time is critical to save you money by knowing when to turn off the blowers.

Typically in chemical Phosphorus removal, the flow weighted average Phosphate is used to set a static dose. This will lead to over and under dosing which could waste money and cause a permit violation! Using real-time data to control dosing can significantly reduce the cost of chemical Phosphorus removal.



The treatment process in the secondary clarifier is similar to that of the primary clarifier. However, a percentage of the sludge that settles in the secondary clarifier is returned (Return Activated Sludge) to the aeration basin to replenish the microbial populations. Excess activated sludge (Waste Activated Sludge) is taken out of circulation by being pumped into the Sludge Treatment Process.

#### **Parameters Tested**

Ammonia

BOD

COD

Dissolved Oxygen

Nitrate

Nitrite

ORP

Phosphorus

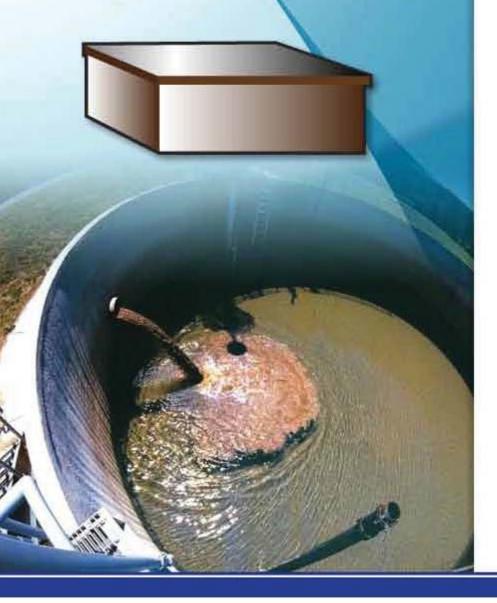
Total Suspended Solids

Sludge Level

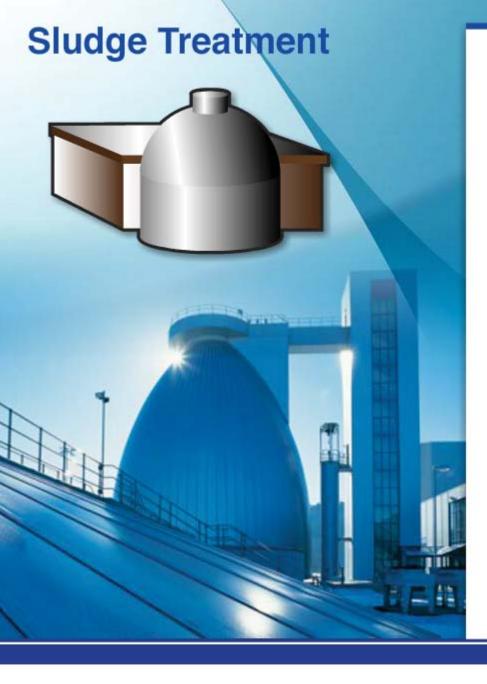
Turbidity

OTOGE





Sludge generated from both the primary and secondary clarifiers is sent to a sludge holding tank where it awaits further processing.



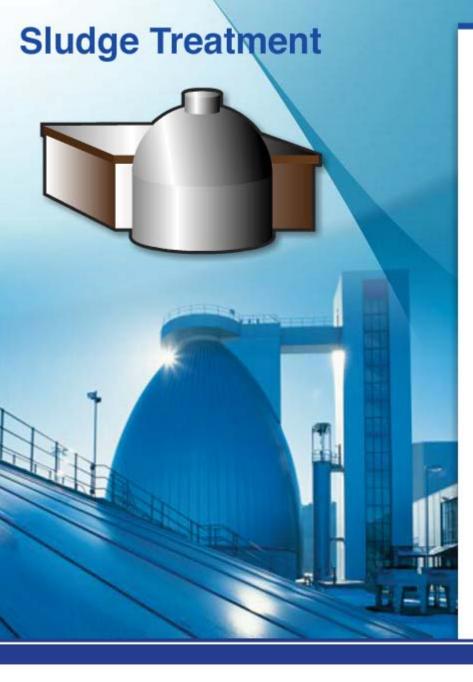
Sludge with a solids content of about 0.1 to 1 percent is formed in the primary and final sedimentation tanks of biological sewage treatment plants. The sludge treatment process and subsequent disposal is one of the highest costs for a wastewater treatment plant often contributing up to one third of the total costs.

#### Sludge Digestion

Aerobic and anaerobic digestion is used to decompose organic matter and to reduce volume. Untreated sludge is approximately 97% water. The dewatering of sludge is a key process step in terms of minimizing the costs of landfill disposal or incineration. Water can be removed from the sludge by using dry beds, vacuum filters, filter presses, and centrifuges resulting in sludge between 50 to 80 percent water. The dehydrated sludge is called sludge cake.

#### **Anaerobic Digestion**

Sludge digestion is a highly sensitive process, requiring 20 to 30 days and at least three types of bacteria, each with its own enzyme system, to carry out completely different tasks. The first bacteria species must break down large molecules into alcohols and fatty acids. The next type of bacteria in the digester immediately converts the alcohols and fatty acids to hydrogen and acetic acid. The third group of bacteria converts acetic acid and hydrogen relatively quickly into methane. The sludge remains in the digester for 20 to 30 days. The physical structures used for this process step are large and dominate the external appearance of many sewage treatment plants.



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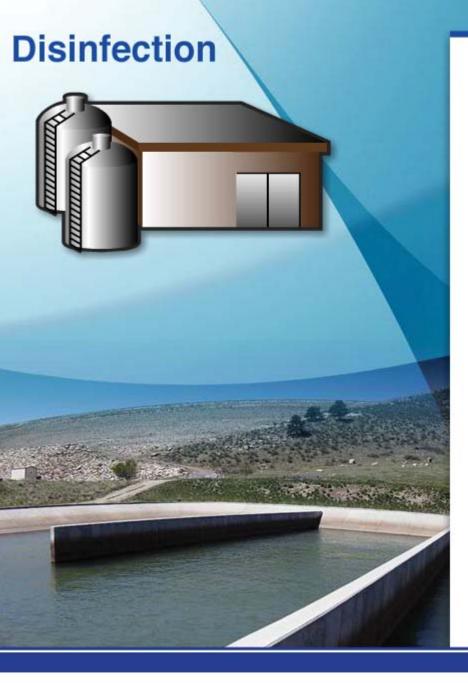
#### Parameters Tested

Dissolved Oxygen (aerobic digester)

ORP (anaerobic digester)

pH

Total Suspended Solids



The biological processes of a wastewater plant are unable to remove all bacteria. Consequently a disinfection step is necessary to ensure that regulated bacterial counts are kept within allowable levels before wastewater is discharged to the environment. While chlorine is most commonly used for disinfection, ozone, chlorine dioxide and ultraviolet light (UV) are becoming more common for wastewater treatment.

High levels of residual chlorine may be harmful to aquatic life in receiving streams. Allowable limits for chlorine release in the effluent are regulated by NPDES permit and may differ based on the receiving body of water. To maintain chlorine discharge within allowable limits, treatment systems often add a chlorine-neutralizing chemical such as sodium bisulfite or sulfur dioxide to the treated water before discharge.

#### Parameters Tested

TOC

Chlorine

UV% Transmittance

# **Process Step Overview**

# NUTRIENTS

Nitrogen and Phosphorus in Wastewater Treatment

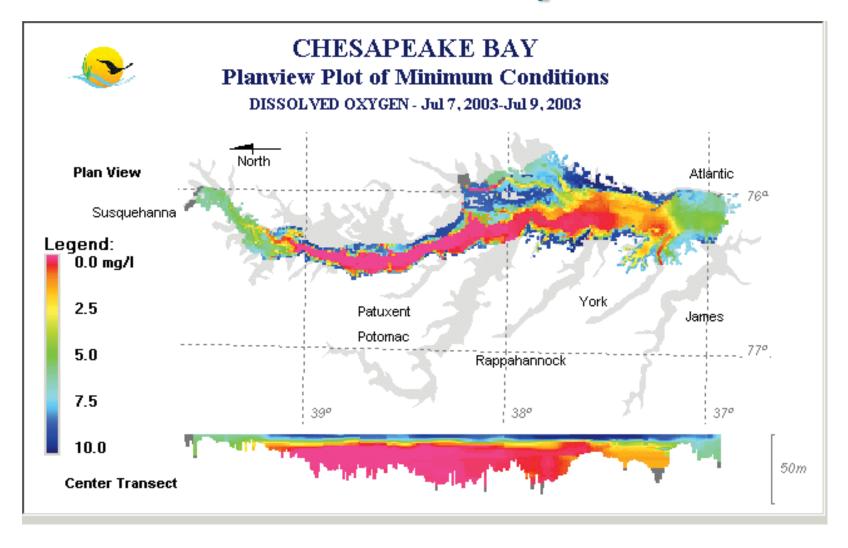


### Introduction

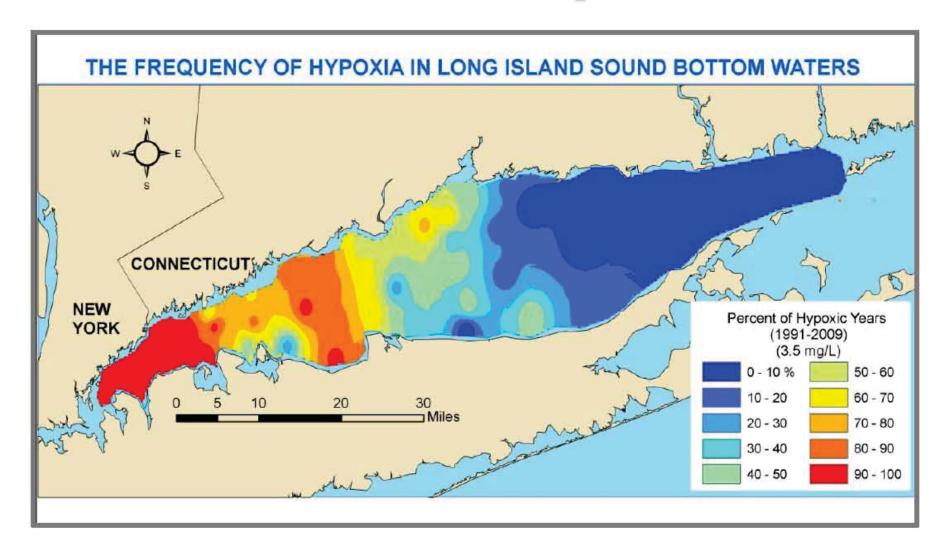
- N & P discharged into Surface Waters
- Algae consume N & P and reproduce
- Bacteria eat algae and "breathe" O<sub>2</sub>
- Low DO kills fish, shellfish, invertebrates



### Effects of Eutrophication



### Effects of Eutrophication



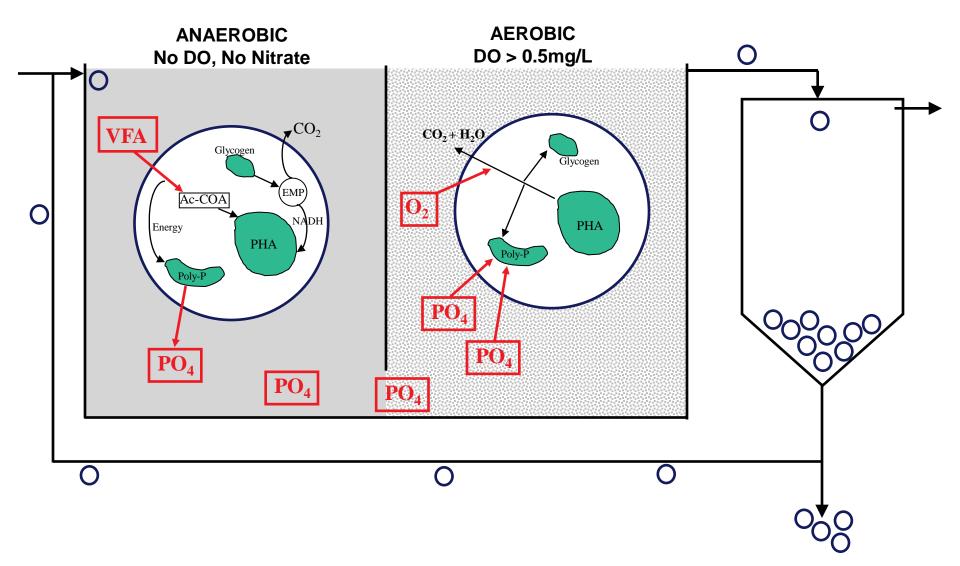
### Impaired Waters

- EPA (2008)
  - 168 Hypoxic Areas in USA
  - 1200 Nitrate Effluent Violations at DW plants
  - Mandate: States to establish nutrient limits in surface waters

- States: Total Maximum Daily Load
  - Set TMDL for N & P in specific watersheds
  - New permits contain N & P limits

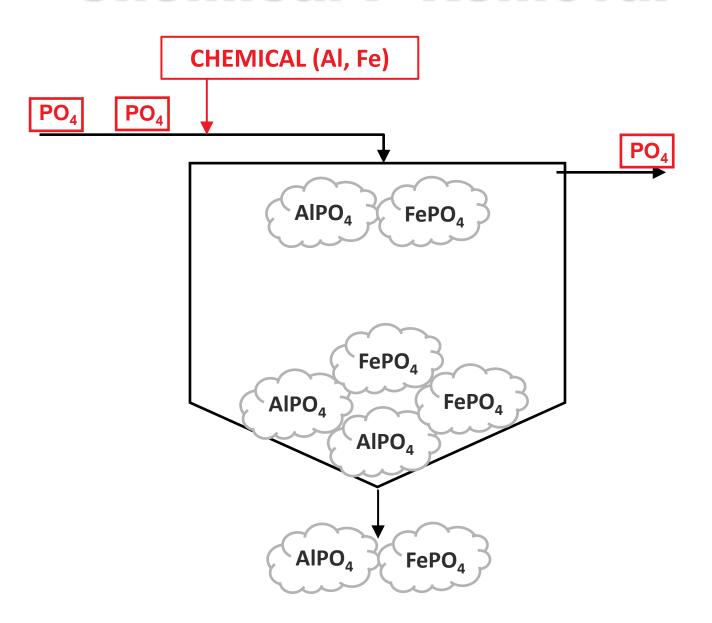
### Biological P Removal

# You can't control bacteria You can only control their *environment*





### Chemical P Removal

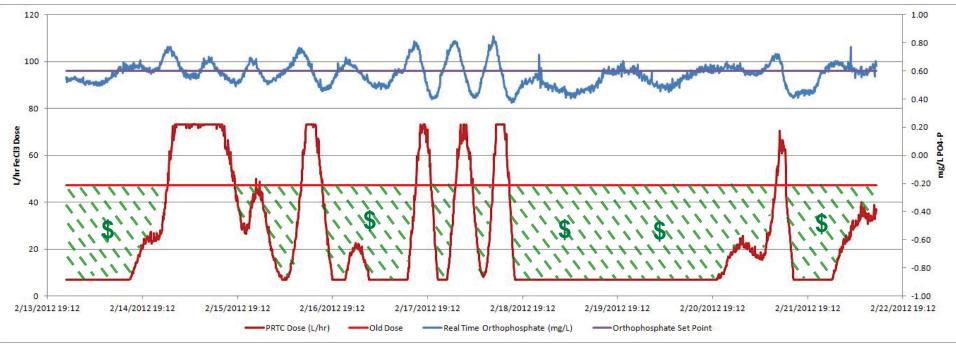






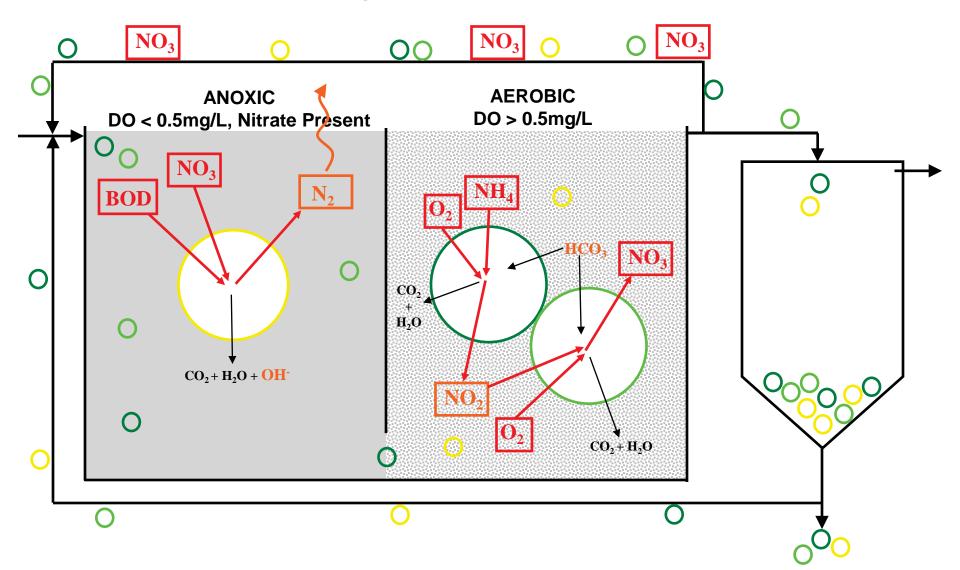
### Phosphorus Real Time Controller

- Measuring Plant Effluent orthophosphate
- Measuring Plant Effluent flow
- Output exact dose to meet 0.60mg/L PO4-P setpoint
- Saving 55% compared to previous static dose



### Biological N Removal

You can't control bacteria
You can only control their *environment* 



#### Nitrification - Ammonia Removal

- Process Characteristics that can Impact Nitrification
  - MCRT (Sludge Age)
    - Generally > 10 day MCRT is needed for stable nitrification in extended air systems
    - Nitrifying bacteria grow very, very slooowly
  - pH
    - Nitrifiers are sensitive to changes in pH
    - Reported "optimum" pHs vary b/w 5.8 8.5...a wide range
  - Alkalinity
    - 7.1 mg of alk is consumed for every mg of Ammonia converted to Nitrate
  - BOD
    - 0.08 lb of inorganic carbon/ lb of Ammonia oxidized
  - DO
    - cBOD oxidation
      - 1.0 1.2 lbs of Oxygen for every lb of BOD
    - Ammonia Oxidation
      - 4.3 lbs Oxygen per lb of NH3-N
      - 4.6 lbs Oxygen per lb of NH4-N



#### Denitrification - Total N Removal

- Other half of biological nitrogen removal
  - Nitrate converted to Nitrogen gas
- Consumes 2.86 grams of BOD per gram of Nitrate reduced
  - Why care: Not enough BOD = denitrification doesn't happen
     Alkalinity not recovered...possible chemical addition
- Forms ~0.5 grams of <u>new cells</u> per gram of Nitrate reduced
  - Why care: WAS = Solids Handling
- Possible problems if Nitrate to Clarifiers >6-8 mg/L
  - Floating Solids, Scum, effluent turbidity issues



# Mallard Creek WRF (NC)



- 12 MGD, NH<sub>4</sub>-N < 1.0 mg/L</li>
- Modified Ludzack-Ettinger process
- Raw wastewater lacks enough alkalinity for complete nitrification
- Dose sodium hydroxide to supplement alkalinity
  - Flow proportional: \$197,283
  - Real time load: \$128,663
  - 34.8% savings

### Ammonia Measurement

- Ammonia Nitrogen
  - Ion Selective Electrode
  - Gas Sensing Electrode

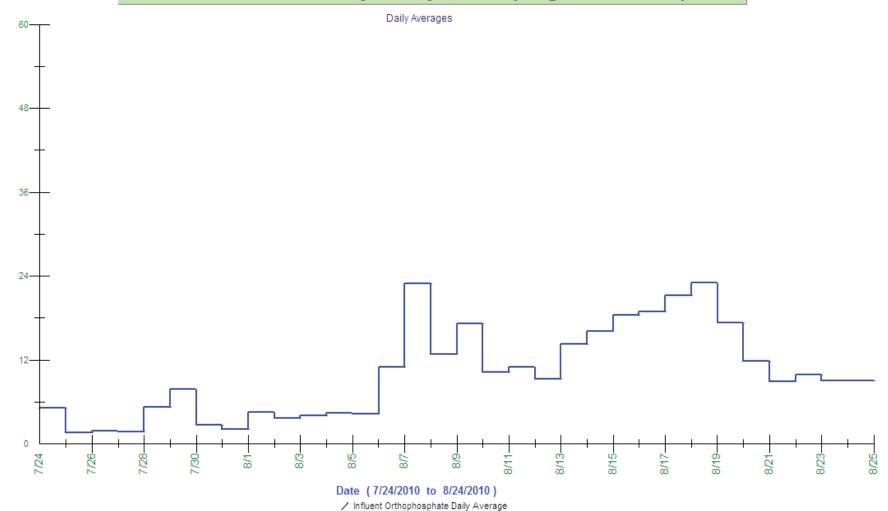




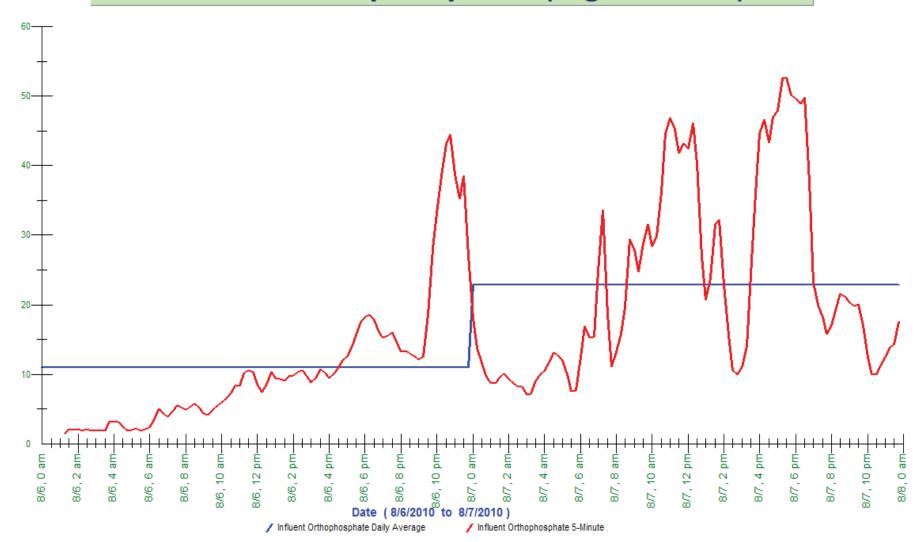
### **Hutchinson WWTP**

- 2.4 MGD, TP < 1.0mg/L monthly average</li>
- 2 Plants
  - Oxidation Ditch & Clarifier
  - Membrane Bioreactor with anoxic selector
- Chemical precipitation with ferric sulfate
- Composite samples & flow proportional dosing were not meeting the effluent limit

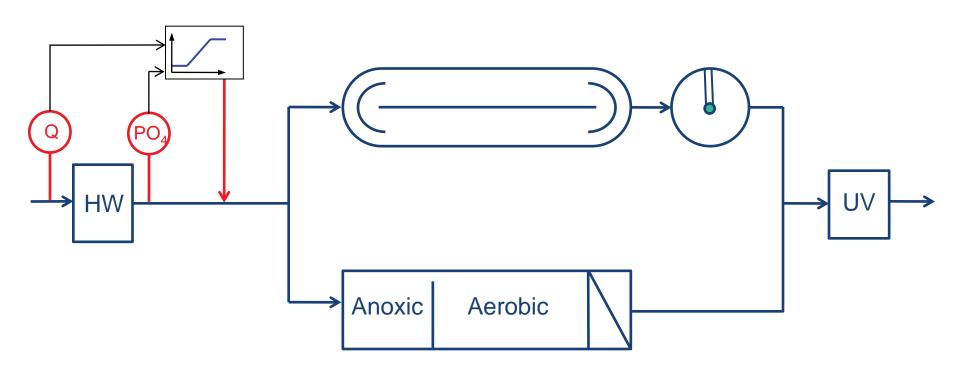
#### Influent Orthophosphate (mg/L PO4-P)



### Influent Orthophosphate (mg/L PO4-P)



### **Hutchinson WWTP**



# Phosphate Analyzer

- Orthophosphate
  - Molybdovanadate
    - "Yellow method"
  - $0.05 15.0 \text{ mg/L PO}_4\text{-P}$
  - +/- 2% accuracy
  - Self Filtering





# Pinery Water & Wastewater Parker, CO



- 2 MGD, TP < 0.05 mg/L</li>
- 5 Stage System with alum addition and mixed media filters
- Instrumentation:
  - 4 ORP Sensors
  - 5 pH Sensors
  - 4 LDO Sensors
  - 2 Phosphate Analyzers
  - 2 Turbidimeters

# Stamford WPCF (CT)



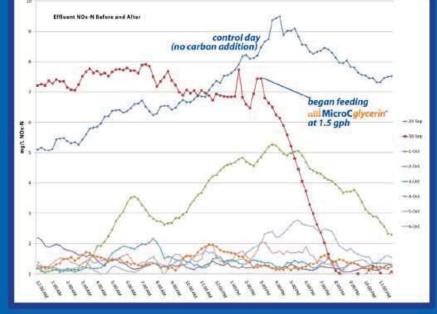
- 24 MGD, TN 1057lbs/day
- 4 Stage System with methanol feed and UV disinfection
- Instrumentation:
  - 4 Ammonia analyzers
  - 1 Ammonia sensor
  - 6 Nitrate/Nitrite sensors
  - 1 MLSS sensor
  - 4 LDO sensor
  - 2 Sludge Blanket sensors
- 2009: Earned \$961,118 with credits

The West Haven Water Pollution Control Facility is a 12 mgd facility located on the Long Island Sound in Connecticut and is required to reduce nitrogen as part of the Long Island Sound TMDL. The facility utilizes an MLE activated sludge process with six separate process trains, each with a pre-anoxic zone. Nitrate is returned in each process train with individual nitrate recycle pumps. The facility is presently upgrading to a 4 stage Bardenpho process utilizing plastic IFAS media in the aerobic zone with carbon addition to the 2nd anoxic zone. The upgraded facilities will not be fully operational for three years.

In 2008 the facility discharged over the nitrogen permit and was required to purchase \$346,000 in nitrogen credits as part of the Connecticut Nitrogen Trading Program. Their 2009 mass based permit is 415 pounds of total nitrogen per day or 7 mg/l at an average flow of 7 mgd.

To reduce costly credit purchases during the construction period, the facility began pilot testing MicroCglycerin™ on September 30, 2009, adding it to the front end of a 1.15 mgd process train, at a flow of 1.5 gph. Within an 5 hour acclimation period effluent nitrate levels were reduced by 4 mg/l, and within 48 hours a stable process was achieved. Nitrates were further reduced by 5.7 mg/l from the control day on September 29th. An online nitrate analyzer provided 24 hour monitoring of anoxic zone nitrate levels.

Acknowledgements: Environmental Operating Solutions, Inc. would like to thank Bill Norton and Abdul Quadir from the City of West Haven, CT, and Kevin Dahl, Facility Manager for OMI, for their participation in this project.









### **Nitrate Sensors**

- Nitrate Nitrogen
  - Reagentless
  - UV absorbance
  - Ion Selective Electrode







# Wyoming Valley San. Auth. (PA)



- 32 MGD, Load Limits
  - TN limit: 584,000 lbs/yr
  - TP limit: 77,000 lbs/yr
- Schreiber™ System
- Pilot Test: control aeration
  - Ammonia Sensors
  - Nitrate Sensors
  - LDO sensors
- Meets new limits, \$20M upgrade unnecessary

# Online Methods for Analysis

### Total Nitrogen

- Digestion with Ozone and Hydroxyl Radical
  - Oxidizes everything to nitrate
  - Measured with UV absorbance
  - 0.3 100,000mg/L N
  - < 8 min. cycle time
  - +/- 3% Accuracy



# Online Methods for Analysis

### Total Phosphorus

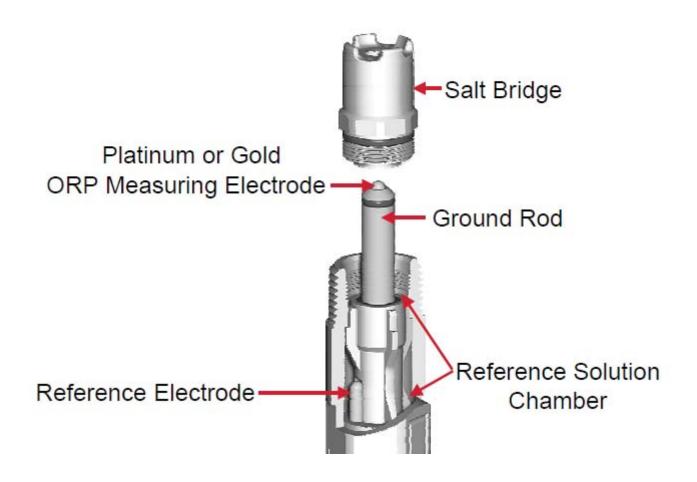
- Digestion with Ozone and Hydroxyl Radical
  - Oxidizes everything to phosphate
  - Measured with molybdovanadate reagent
  - 0.3 100,000mg/L P
  - < 20 min. cycle time
  - +/- 3% Accuracy



#### Other Valuable Tests

- Dissolved Oxygen
- Oxidation Reduction Potential (ORP)
- Total Suspended Solids
- Volatile Fatty Acids
- pH
- Alkalinity
- COD

#### **ORP Sensor**



#### **ORP**

Electron

Acceptors

- Anaerobic:
  - No oxygen, no nitrate

Sulfate present

- Anoxic
  - No oxygen
  - Nitrate Present
- Aerobic
  - Oxygen present

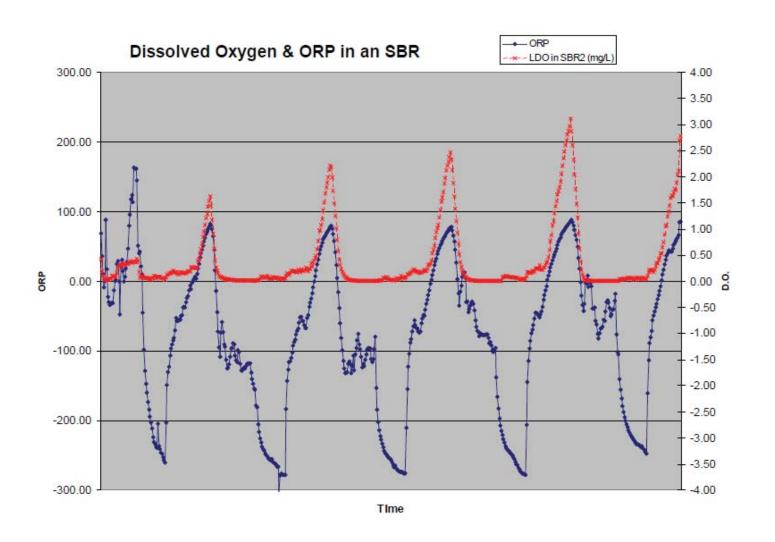
Aerobic zone: DO and ORP (DO: 1.5-3mg/L; ORP: >100mV)

Anoxic zone: ORP between -100mV and +100mV

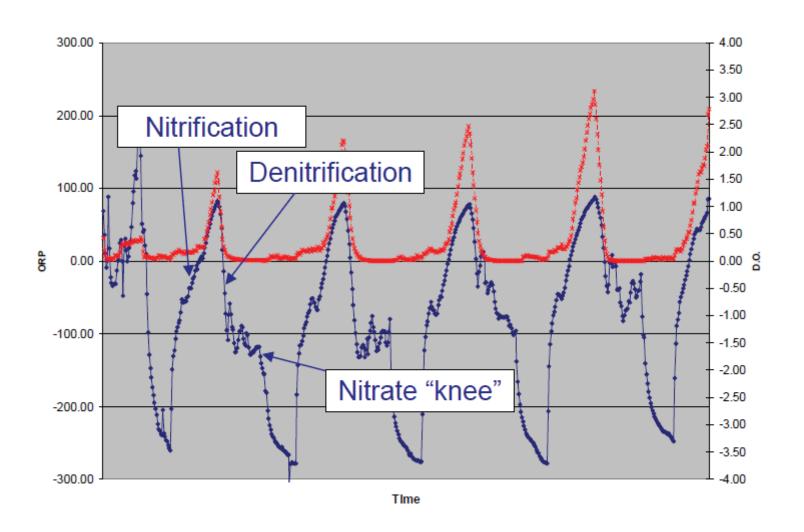
ORP will respond to subtle changes in the anaerobic zone

Should see values < -100mV

#### **ORP in SBR**

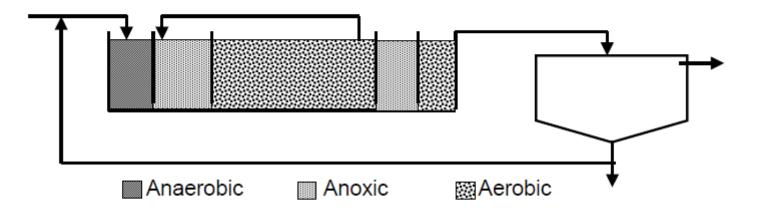


#### **ORP in SBR**



#### ORP in BNR

#### Five Stage Bardenpho



Pinery WWTP (Near Parker, CO: 2mgd, TP<0.05, TN<5.0)

Controlling first aeration at 80mV

Controlling first anoxic at -80mV

Monitoring anaerobic at -250mV

### Take Home Messages

- Nutrient regulations are expanding
- Pioneers have paved the road
- Baseline data is critical for design & operation

# Aeration



### Introduction

- Do we need to have air in the basin?
  - Aerobic Zones- Aeration Present
    - Dissolved Oxygen > 0.5 ppm
  - Anoxic Zones- Oxygen depletion /except chemically bound
    - Dissolved Oxygen < 0.5 ppm</li>
    - Nitrate present –bugs at work!
  - Anaerobic Zones- Atmospheric/ Dissolved oxygen not present
    - Dissolved Oxygen < 0.5 ppm</li>
    - No Nitrate present

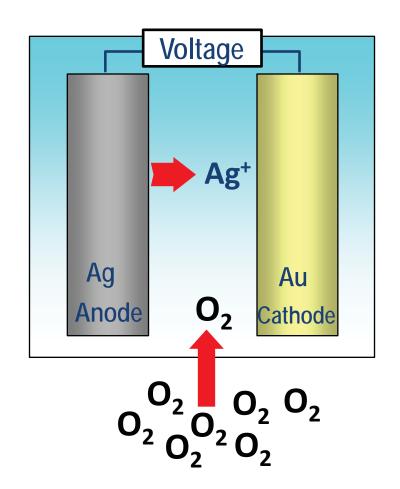






# **Polarographic Method**

- As oxygen permeates the membrane, it is reduced at the cathode
- The current (amperage)
   created is directly
   proportional to the
   dissolved oxygen
   concentration

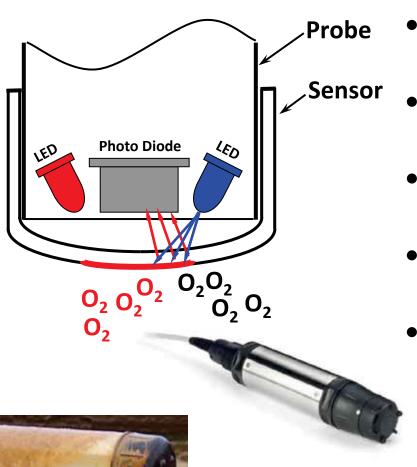


# Steps used to combat the problem

- Began by chlorinating the RAS (common solution to deal with filaments)
  - \$3,000/month
- Manually controlled blowers (increased DO)
  - 14,000 scfm
  - Twice daily DO testing with portable (old) unit.



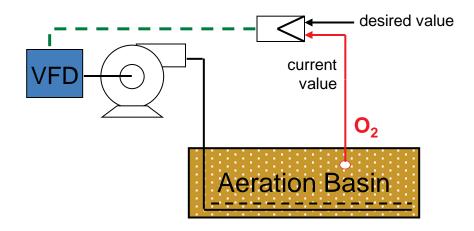
### **Luminescent Dissolved Oxygen**



- The sensor is coated with luminescent material
- Blue light from an LED excites the material
- The excited material emits red light
- Oxygen 'steals' energy from the luminescent material
- The time it takes for the material to stop emitting light directly relates to the amount of dissolved oxygen present in the sample

### **Aeration Control**

- Solution Long Term
  - Automated Blower Control with online luminescent dissolved oxygen sensors (LDO)





# **Accurate Dissolved Oxygen Sensors**

- Stopped chlorinating the RAS
- Eliminated twice per day manual testing
- Reduced blower output from 14000 to 12000 scfm
- Eliminated sensor maintenance 3x/week

= \$70,000 annual savings<sup>1</sup>

<sup>&</sup>lt;sup>1</sup> Kiser, Phil "Advances in Online Dissolved Oxygen Measurement Using Luminescent Technology". Presented at 2006 ISA Automation Week Conference

### Verify online readings with a portable



.39 75.3F

Be Right™

.43 74.6F

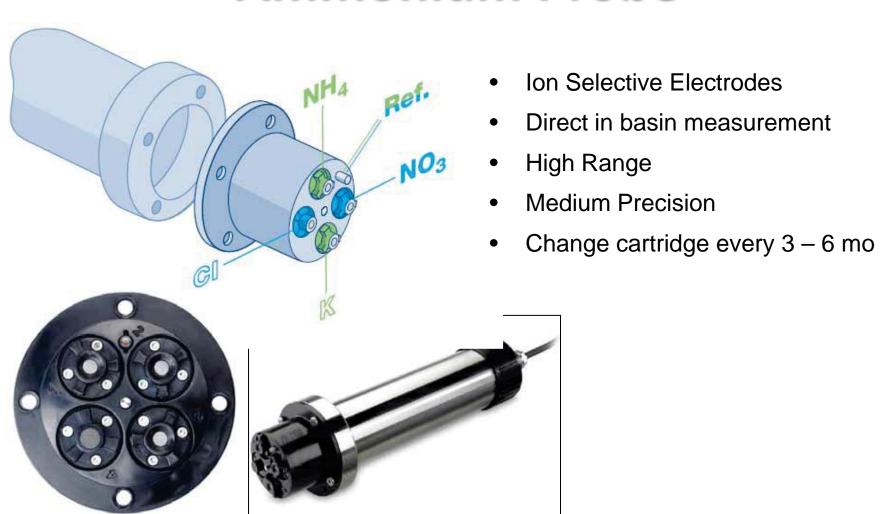


### **Problem**

- Larger WWTP with 10mg/L TN Limit (Total Nitrogen)
- Poor Effluent Quality Forcing Construction
  - Additional Clarifier and Filter
- Have LDO sensors & use for Aeration Control
  - Ammonia sensor was faulty—repairs in Europe
- No budget for additional capital construction
- Operational budget cut by \$50,000

# AMMONIA SENSOR Nitrified recycle DO DO Filter UV

### **Ammonium Probe**





### **Ammonia Analyzer**

- Low Range, High precision applications
- Sample Prep & Delivery:
  - Filter probe for mixed liquor
  - Filtrax for Primary Effluent
- Gasification Method
- Change reagents 3 6 mo

### Results

- Controlling Aeration by Ammonia
- From 9,500 12,000 to 8,000 9,000 kwh/d
- \$90,000/yr Energy Savings
- ROI: 2.8 Months
- Sludge Quality Improved
  - No need for additional Clarifier & Filter
  - Estimated at \$3,200,000

# Aeration



# Suspended Solids

### What are Suspended Solids?

Total Suspended Solids or TSS is defined as - the amount of filterable solids in a water sample





### Why Measure Suspended Solids

- Determine the Mixed Liquor Suspended Solids
- Calculate the Return Activated Sludge (RAS)
- <u>Calculate</u> the Waste Activated Sludge (WAS)
- Calculate the Sludge Retention time (SRT)
- <u>Calculate</u> Mean Cell Residence Time (MCRT or Sludge Age)
- Determine the depth/height of the Sludge Blanket
- Determine the efficiency of the dewatering mechanics
- Solids interfere with disinfection

### Control of Solids Inventory- Can:

- Allow additional capacity to be realized in individual unit processes;
- Potentially allow nitrification to be achieved without the construction of additional biological treatment capacity;
- Reduce the energy use & costs associated with aeration in biological processes
- Reduce biosolids mgnt costs by reducing the quantity of solids requiring processing
- Improve the settling characteristics of the biomass
- Improve the ease and stability of plants operations
- Result in an overall improvement in effluent quality.

# Where do you Measure Suspended Solids?

- Aeration Basin
- Influent
- Effluent
- Centrate/Filtrate
- Clarifier Effluent
- Digester Feed
- 555

### Mixed Liquor Suspended Solids

- How do you currently test for it?
- How long does it take to receive your results?
- What can happen between grab samples?
- How would a real time trend of MLSS help you?
- Do you ever have settleability problems?
- Why do they happen?



### **Food**

- Organics (BOD) in the sewage are the food for the bacteria
  - Typically more than enough food present
  - Typically cannot control quantity of food
- Need to have the right amount of food per bug
  - Food to Microorganism Ratio:

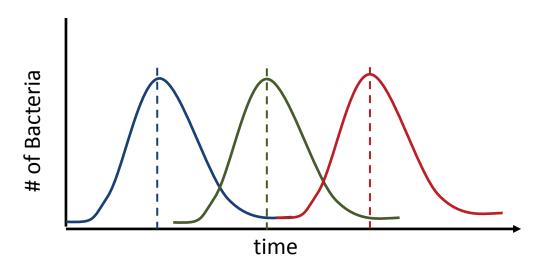
Pounds BOD in Sewage	=	BOD Conc (mg/L) x Flow (MGD) x $8.34$
Pounds of MLSS		MLSS Conc (mg/L) x Volume (M gal) x 8.34

### F:M Ratio

- Typically this is used to determine the Mixed Liquor Suspended Solids concentration (MLSS)
- Then the amount of return sludge is determined
  - Typically as % of influent flow
- The wrong F:M for your plant can lead to settleability issues
- Grab samples are usually sufficient here

# Sludge Age

- The time (days) that an average bacterial cell stays in the biological system
  - "Young": < 4 days</p>
  - "Middle age": 4 and 15 days
  - "Old": > 15 days



# Sludge Age

- The right sludge age for your system means:
  - The bacteria will remove pollutants
  - The sludge will settle

How important do you think sludge age is??!!

# Sludge Age

Calculation:

Pounds of Solids in Aeration Basin

Pounds of Solids Wasted in a Day

Concentration (mg/L) x Volume (M gal) x 8.34

Concentration (mg/L) x Flow (MGD) x 8.34

To calculate this, most wastewater plants take one grab sample from the aeration tank and one grab sample from the WAS once a day. They calculate today's sludge age from yesterday's data.

Is that good enough?

# Mean Cell Residency Time

Calculation:

```
Aeration Solids (lb) + Clarifier Solids (lb)
Wasted Solids (lb/d)+ Effluent Solids (lb/d)
```

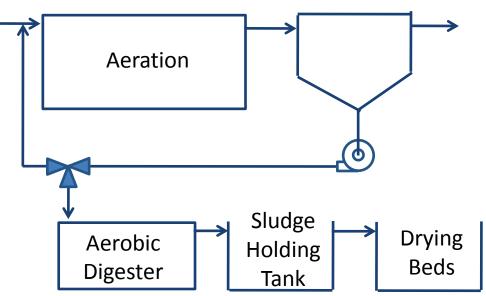
- How accurate do you want this number to be?
- In a perfect world, how soon would you like your Sludge Age/MCRT results?
- Would it save you time/money/sanity if your system did the wasting for you?

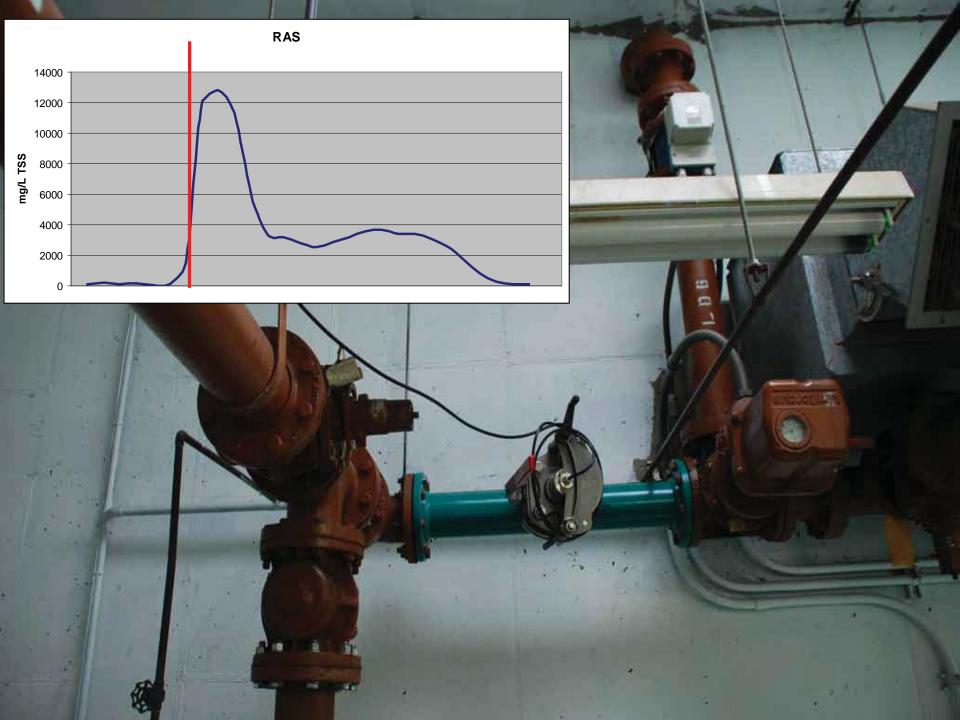
### Why Measure Suspended Solids

- □ Determine the Mixed Liquor Suspended Solids
- □ Calculate The Return Activated Sludge (RAS)
- □ Calculate The Waste Activated Sludge (WAS)
- □ Calculate The Sludge Retention time (SRT) \*Case Study\*
- □Determine the depth/height of the Sludge Blanket
- □ Determine the efficiency of the dewatering mechanics \*Case Study\*

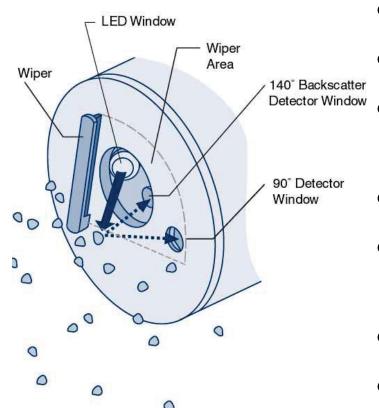
### **Morrison WWTP**

- 100,000 gallon per day
- Extended Aeration Activated Sludge
- One Operator
- Problems with:
  - Oversized RAS pumps
    - "Off" 8 min
    - "On" 2 min
  - Aerobic Digester
    - Clear water in WAS



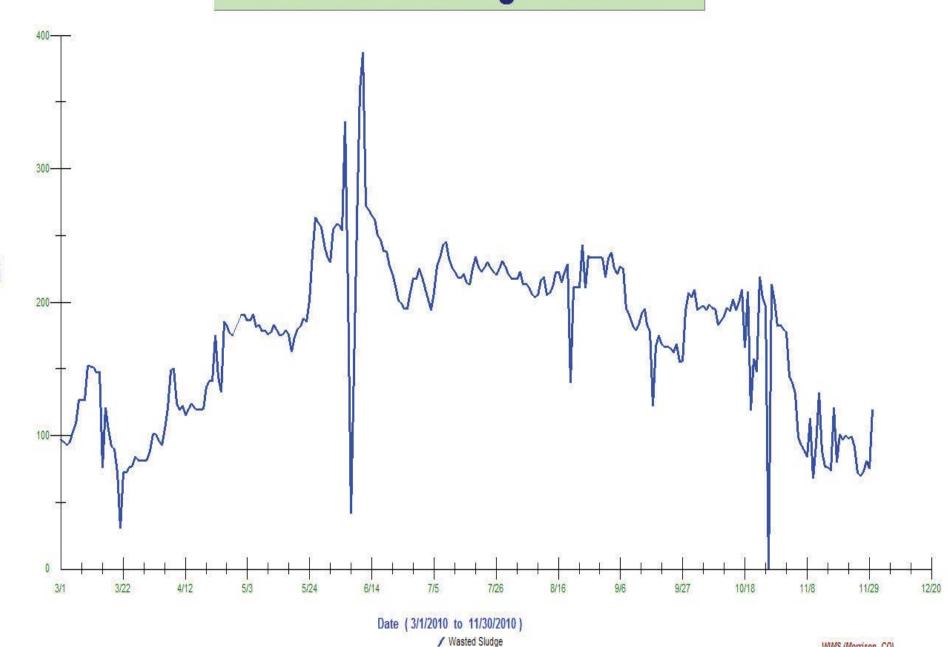


### **TSS Sensors**



- Emit an infrared 860nm light
- 90° detector for turbidity
- 90° and the 140° detectors for Suspended Solids
- Color does not interfere
- Compensates for particle size & shape
- Range: 0.1mg/L 15%+
- +/- 5% Accuracy

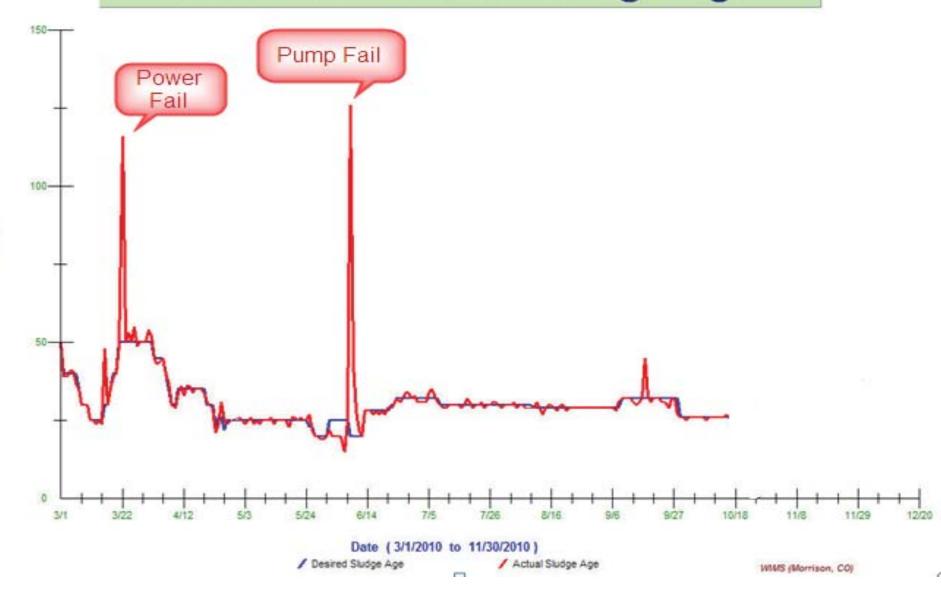
### **Pounds of Sludge Wasted**



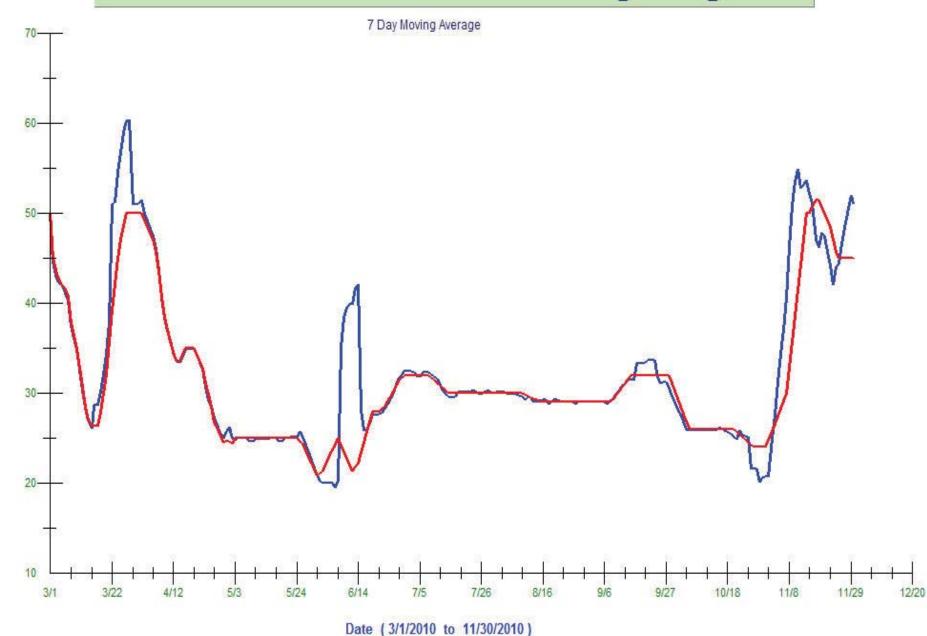
WIMS (Morrison, CO)



### Desired vs Actual Sludge Age



### Desired vs Actual Sludge Age



✓ Actual Sludge Age 7d MA

days

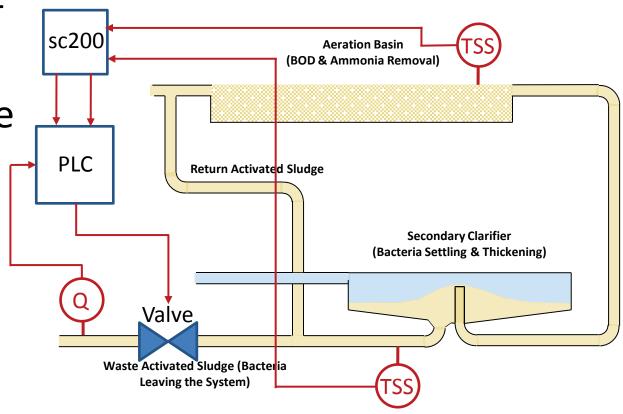
/ Desired Sludge Age 7d MA

# **Automatic Sludge Age Control**

 No guessing if sample was representative

 Use today's data now

Free up operators to other tasks



### **DAFT Polymer Control Issues**

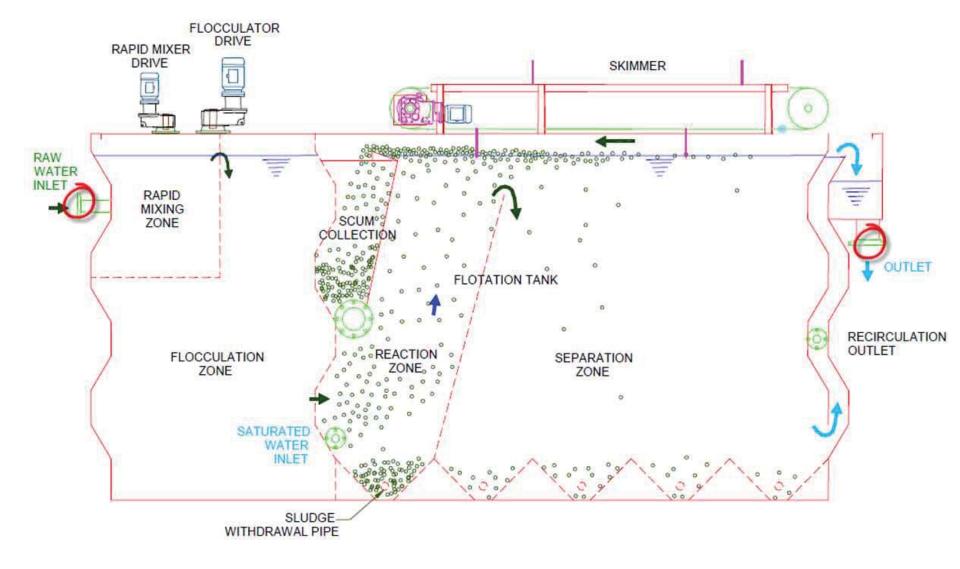
# **King County- Renton Washington**

-Single Loop Cascade control System

- Overfeed of Polymer
- Underfeed of Polymer
- Thickened sludge variability



### DAF- Dissolved Air Flotation



### **DAFT Polymer Control- Solution**

### Control Polymer dose in real time

- Measure flow and solids concentration = solids loading
- Proactive vs. reactive control
- 30% reduction in Polymer
- Maintained 6 6.5% sludge
   Concentration
- Low Maintenance due to self clean wiper
- About \$30,000/year savings

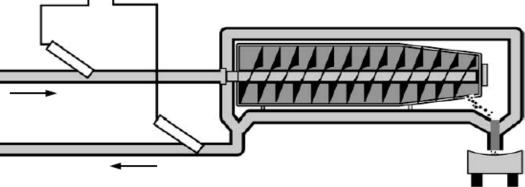


# Removal Efficiency – Centrifuges & Belt Presses

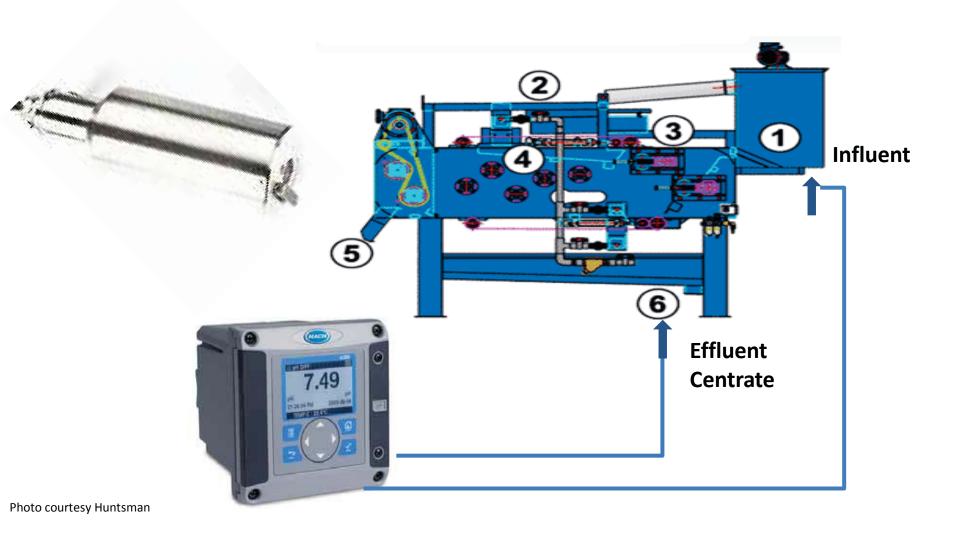
### Control Polymer dose in real time



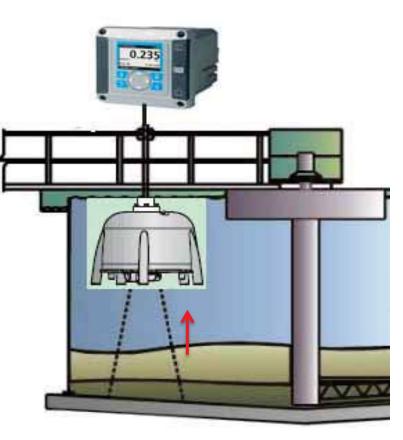
- Measure flow and solids concentration = solids loading
- Measure centrate TSS to trim



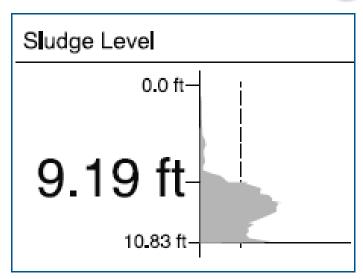
### **Belt Press Application**



### **Sludge Blanket Monitoring**

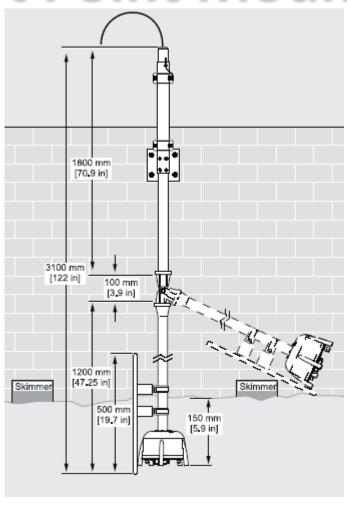


Measuring Principle



- Improved control of sludge pumping
- Early detection of Solids Washout
- Optimize sludge extraction from clarifiers/thickeners
- Reduced labor costs of manual measurement

# **Pivot Point Mounting**



## **Suspended Solids Products**











## **Other Applications**

- Effluent TSS or Turbidity
- Influent TSS
- Clarifier Influent for Polymer Control
- Aerobic Digester TSS
  - Ensure volatile solids reduction

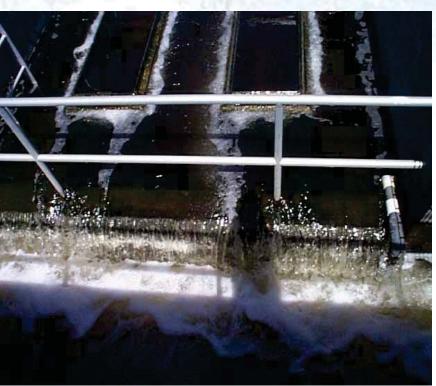
## Suspended Solids

## pH Applications



Wastewater Influent





Wastewater Effluent

**EQ** Basin



## **Typical pH Applications**

- \*Wastewater treatment
- \*Drinking water
- \*Neutralization of effluent
  - Steel
  - Pulp and Paper
  - Food
  - Chemical
  - Pharmaceutical

Chrome destruct

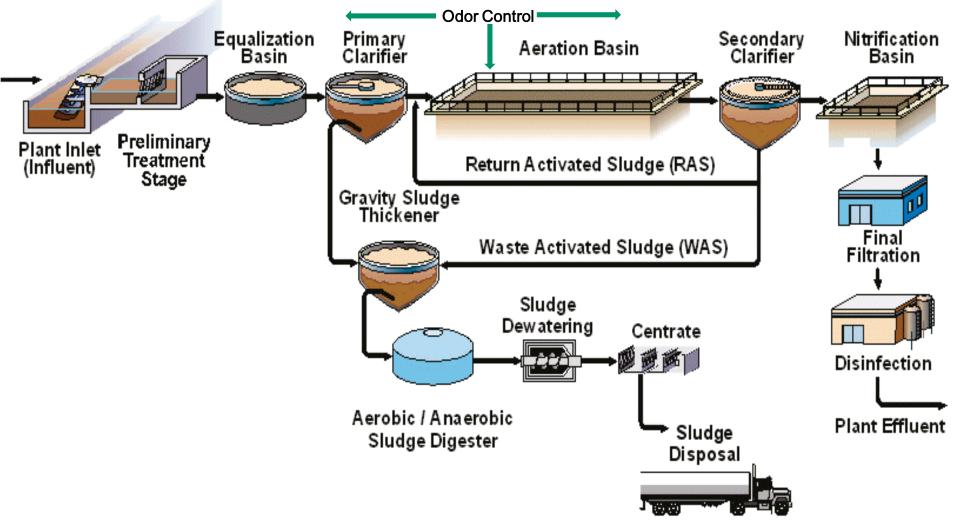
Cyanide destruct

High purity water

- Odor scrubbers
- Pharmaceutical
- Chemical & Petrochemical
- Reverse osmosis
- Cooling tower control
- Food processing
  - Carbon dioxide control
  - Cleaning
  - Canning



## Where do you measure pH?





## Why measure pH in wastewater?

- Inlet
  - protect biological treatment (bacteria)
- Aeration Basins/Denitrification
  - monitor biological activity
- Digester
  - control reactor pH levels (bacteria)
- Outlet:
  - protect environment (fish etc.)
- Neutralization
  - largely industrial requirement



## pH Measurement Problems

- Lab vs. Online
- Fouling
- Reference Electrolyte
- Ground Loops
- Low Conductivity (industrial)
- Lifecycle
- How accurate do you need?

## **Process pH Accuracy**

- Real life reproducibility is +/- 0.1 or 0.2 pH units if:
  - Proper Sensor Chosen
  - Proper Installation
  - Properly Maintained
- "The best results are often obtained by going to three electrodes and leaving them alone."
  - Choose the middle signal
  - Only recalibrate when:
    - Sensor is cleaned & gel layer disturbed
    - Difference between lab & process consistently large

Source: McMillan, G.K. and Cameron, R.A., *Advanced pH Measurement and Control*, 3<sup>rd</sup> Edition, ISA – The Instrumentation, Systems, and Automation Society, 2005.

## Selecting the right Sensor







Industrial Environments

Process Applications

Lab Applications

**High Purity Water** 



**Dirty Samples** 



Complete Solutions



Technical Support



**Quick Sampling** 

## **Choose the Right Sensor**



#### **Industrial Environments**

- Replenish "dirty" probes with rebuildable references
- Rugged, non-glass probes keep process running safely



- Electrode design optimized for fast response time
- High flowing reference junctions ensure quick reading and calibration



Process Applications Lab Applications









- Rugged portable meters for rapid analysis throughout plant
- Measure multiple parameters right inside the meter – no probe needed



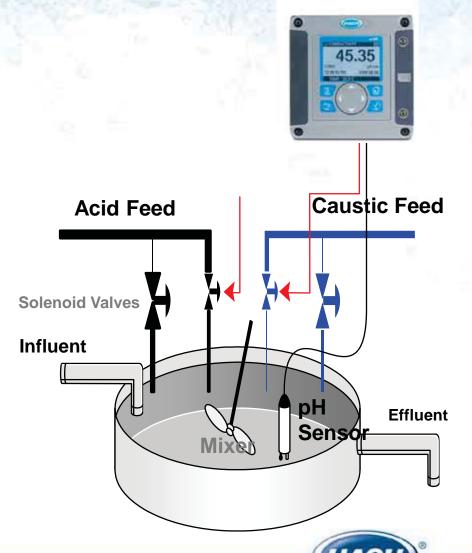
- Probe life extended with references protected from sulfides & proteins
- Clogging reduced with junctions designed for particulates and suspensions





## **Industrial Pre-treatment**

- Industrial dischargers may send their waste directly to municipal wastewater plants.
- Acids
- Heavy metals
- Solvents
- -These substances must be monitored to ensure the health of the bio-system and ensure effluent quality is not adversely affected.

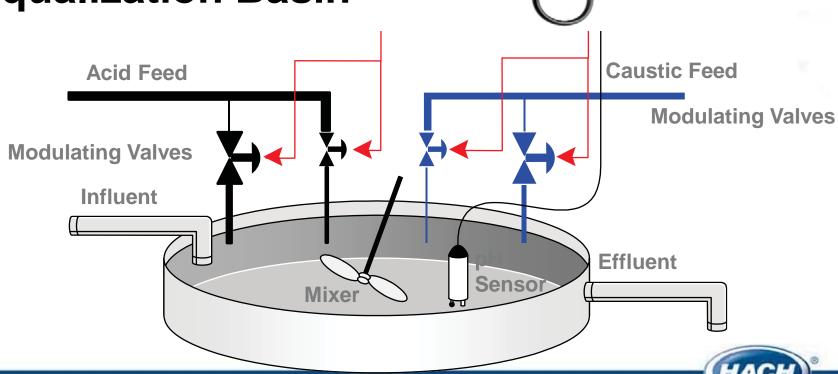


Be Right"

## pH Control

**PID Control** 

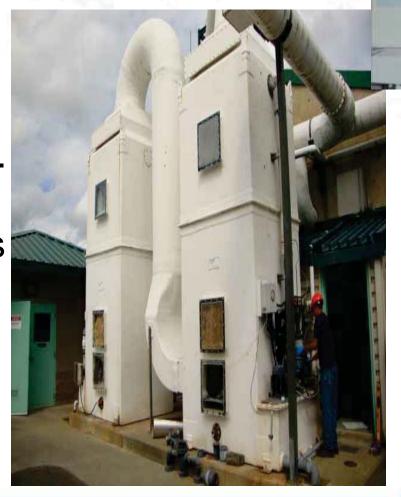
**Equalization Basin** 



Be Right™

### **Odor Control**

Odors are released as they enter the primary clarifier and sludge drying areas. There are mechanical systems that use pH measurement to make chemical feed adjustment to control odor.



## Specific pH Applications



## pH Applications



Wastewater Influent





Wastewater Effluent

**EQ** Basin



Managing Organics

potassium hydrogen phthalate (KHP)

urea

CHO

нс-он

HO-CH

HOOC

соон

phenylalanine

ОΗ

glucose

glutamic acid

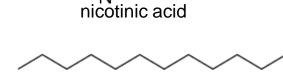
humic acid

ОH

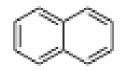
cellulose

carboxylic acids

estradiol



dodecane (and other alkanes/paraffins)



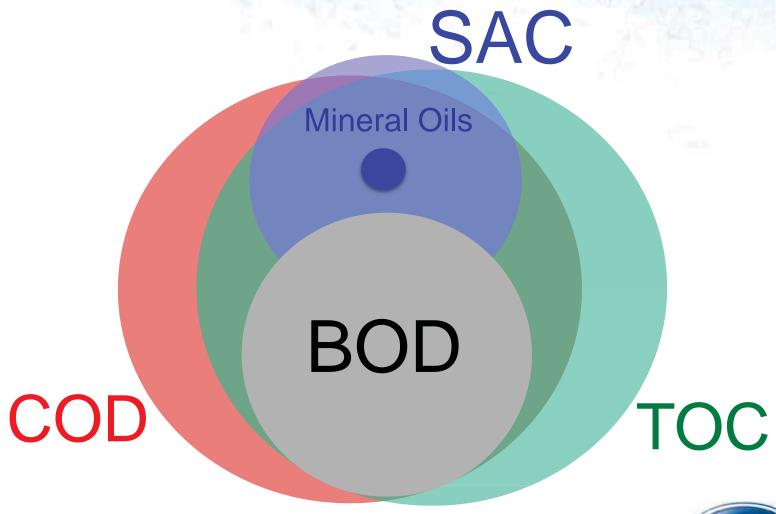
naphthalene



#### Biggest Issues with Organics

- BOD is there anything that is good about it?
- Comparing BOD, TOC, UV, COD can this be done?
- Can it be used to tune the food to microorganism ratio (F/M)? What about DO?
- Waiting 2 hours or 5 days doesn't allow me to see what is coming into my plant right now, how do I know if I have a large organic slug?

#### How do we Measure Organics?



#### What we are seeing...

Туре	Definition	Accuracy*
BOD	Bio-available organics and other compounds that can be oxidized through consumption of dissolved oxygen	15%
CBOD	Bio-available organics with an inhibiting agent added to prevent nitrogen oxidation through consumption of dissolved oxygen	15%
COD (Cr)	Chemical oxygen demand, measuring the reduction of chromium: $\text{Cr}_2\text{O}_7^{2-}$ (orange-colored) $\rightarrow$ $\text{Cr}^{3+}$ (green-colored)	<3%
COD (Mn)	Chemical oxygen demand, measuring the reduction of manganese: Manganese Mn(III) purple-colored → Mn(II)	<3%
TOC	Total organic carbon, measuring the amount of CO <sub>2</sub> generated after oxidation of carbon	<3%
SAC	Spectral absorbance coefficient, measuring the absorbance of 254nm by certain organic molecules with double & triple bonds	Trending Tool

<sup>\*</sup> As determined by recovery of a known standard



Organics Discussion

potassium hydrogen phthalate (KHP)

$$O$$
 $H_2N$ 
 $C$ 
 $NH_2$ 

urea

humic acid

phenylalanine

ОΗ

glucose

glutamic acid

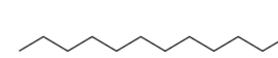
ОH

nicotinic acid

cellulose

carboxylic acids

estradiol



dodecane (and other alkanes/paraffins)



naphthalene



# Key Applications and Issues with Each Measurement



#### BOD & CBOD

Average number of hours/week a typical WW facility spends on BOD?

15 hours

Average number of bottles set per week for BOD? 21 bottles

Average number of times per week BOD is run? 2 sets

Percent of BOD tests that FAIL to meet Standard Methods Guidelines?

5-10%



#### Issues and Root Causes

Failure Mode	Key Root Cause(s)	
Blank depletion	Organic contamination in dilution water or bottles Drift in DO sensor	
Effluent fails to achieve 2 ppm depletion	Insufficient nutrient or seed Toxicity	
Fail G/GA reference	Inaccurate G/GA Toxicity Insufficient seed	

www.boddoctor.com



#### Products for BOD













#### Chemical Oxygen Demand

 COD is a measure of the oxygen demand of a sample that is susceptible to oxidation by a strong chemical oxidant

 Not all organics are susceptible to oxidation by the COD method

#### Common Uses of COD

Please share how you use COD...

- Food estimate?
- Tracking industrial dischargers?
- Anyone charging industrial discharges based on COD instead of BOD?

#### COD for Rapid Organic Estimates

## Hach COD— Treating Brewery Process Water On Site

#### Process Water Conservation, Treatment & Recycling Minimizes Water Consumption

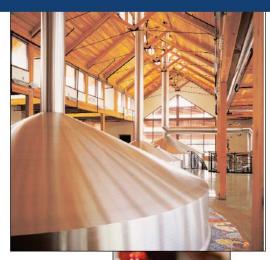
by Mandy Miller

New Belgium Brewing Company today reuses much of its process water for evaporative cooling, cleaning, and landscape purposes. On-site digestion by aerobic and anaerobic bacteria to reduce the organic components in process effluent, coupled with a continuous testing program to accurately monitor the most important sources of high organic loadings within the brewery, have been important keys to the brewery's successful process water treatment and re-use program.

For every barrel of beer a typical brewery produces, an average of eight to 10 barrels of water are consumed. New Belgium Brewing Company in Fort Collins, Colorado, produces one barrel of beer using about four barrels of water, thanks in part to a strong company-wide commitment to resource recovery that includes on-site process water treatment and an ambitious water re-use program.

New Belgium Brewing Company is the fifth largest craft brewery in the country; it produced more than 320,000 barrels of beer in 2004, including the breweries flagship brew, Fat Tire Amber Ale. The company strives to be on the cutting edge of technology to best meet its high objectives for environmental has a rated capacity of 80,000 gallons/day. Average daily flows for 2004 was 58,274 gallons, and the plantís total outlet flow for the year was about 20 million gallons.

The automated process water treatment plant, supplied by German systems provider Von Nordensköld, includes a three-stage anaerobic digester, an aerobic lagoon,





#### The COD Problem...Waste...

- EZ COD recycling
  - Fill bucket with used COD Vials
  - Heritage Environmental will pick up and recycle



- Mercury Free COD
  - TNTplus packaging
  - Not EPA approved
  - Only for low-chloride waters



#### Products for COD











## Rapid Organics Assessment

Total Organic Carbon Fluorescence UV254



## Time...the Achilles Heel of Organics Measurements

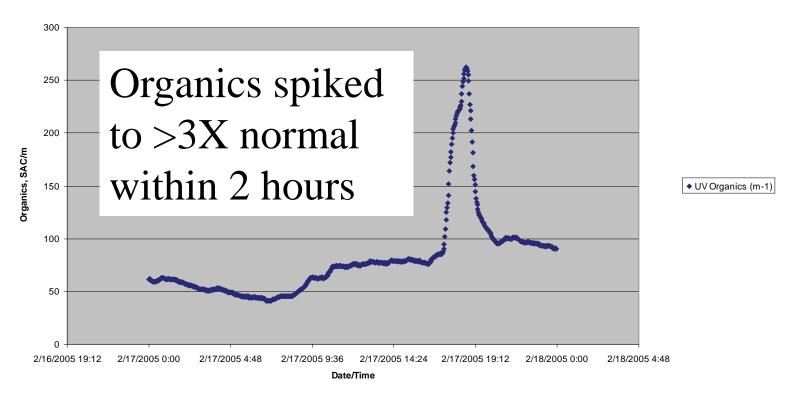
- BOD5 5 Day Procedure
- COD 2-3 Hour Procedure

 Organics can be monitored using other technologies for more rapid assessment

#### What if this was your facility?

- Organic spike from 75 sac/m to >250 sac/m
- What do you do?

**UV Organics Concentration** 



#### Total Organic Carbon

Anyone tried TOC on Wastewater???

Most facilities that try on-line TOC pull them off line

 Why??? They are complex analyzers with tiny tubes that require frequent calibration, maintenance, and rebuild

## TOC Organics Discussion

potassium hydrogen phthalate (KHP)

urea

phenylalanine

"OH <sup>™</sup>ОН HO\* OH

ОΗ

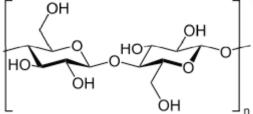
glucose

glutamic acid

humic acid

ОH

nicotinic acid

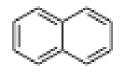


cellulose

carboxylic acids

estradiol

dodecane (and other alkanes/paraffins)



naphthalene

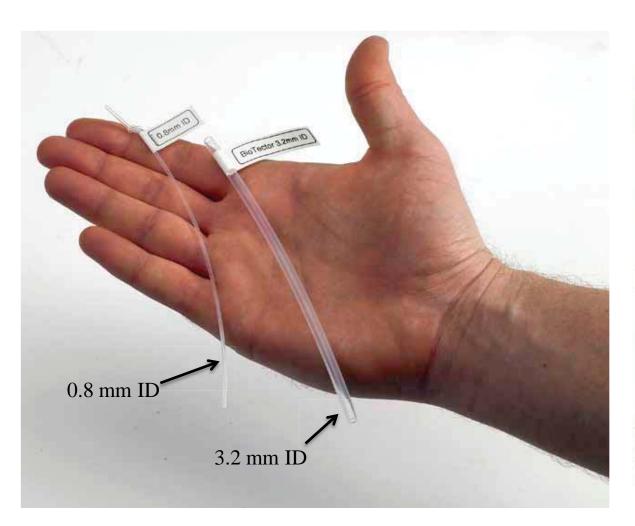


#### Online Total Organic Carbon

- Digestion with Ozone and Hydroxyl Radicals
  - Converts all organic carbon to CO<sub>2</sub>
  - Direct measurement of CO<sub>2</sub>
- Range: 0.3 100,000 mg/L C
- Accuracy: +/- 3%
- Avg. < 8 min. cycle time
- Automatic cleaning
- Scheduled (routine) maintenance every 6 months, period!



#### Online Total Organic Carbon







UV254 Organics Discussion

potassium hydrogen phthalate (KHP)

ωOH

<sup>™</sup>ОН

glucose

phenylalanine

glutamic acid

ОH

cellulose

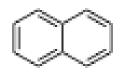
humic acid

carboxylic acids

Ĥ HO estradiol

nicotinic acid

dodecane (and other alkanes/paraffins)



HOOC

COOH

naphthalene

#### Spectral Absorbance Coefficient

- Absorbance of 254nm light (UV254)
  - Double and triple bonded organics
  - Single bonded organics do not absorb 254nm
- No chemicals or reagents necessary
- Lab method with Spectrophotometer



#### Spectral Absorbance Coefficient

- Online Organics Probe
  - Displays as BOD, COD, TOC, SAC, %T
  - Reagent-free measurement
  - Wiper automatically cleans optics



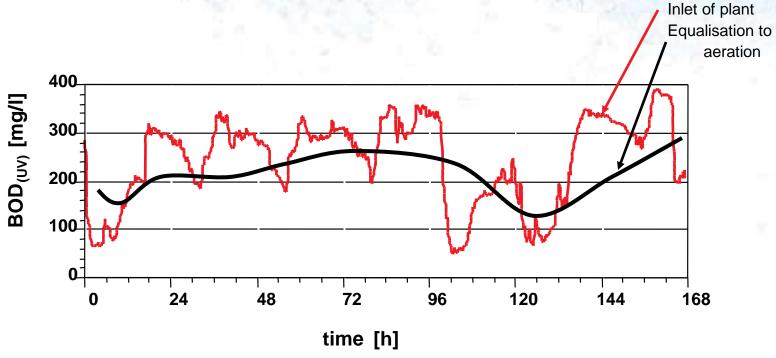


#### Influent Monitoring

- Allows operator to react
  - Divert & blend (dilute)
  - Step feed
  - Increase air
  - Increase MLSS



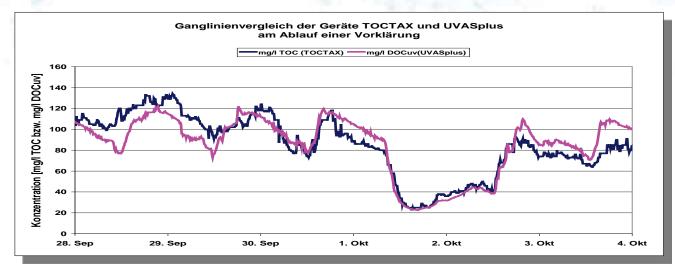
# Organics Probe: Load Equalization



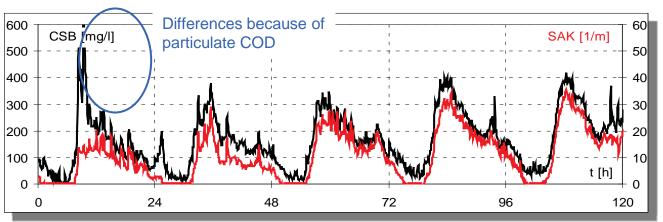
- <u>Target:</u> steady loading to the aeration basin
   ->optimal denitrification and biological phosphorus removal
- Solution: bypass the primary clarifier during low loading
  - supplement external carbon sources



# Comparison between SAC/TOC and SAC/COD



SAC / TOC Primary Effluent



SAC / COD
Raw Influent



# Dissolved Organics













# Wastewater Instrumentation for Monitoring and Control



# Transforming Data into Information to Ensure Wastewater Quality

Making Data Meaningful



#### Some Words of Wisdom

"The goal is to transform data into information and information into insight"

- Carly Fiorina, Former CEO, Hewlett-Packard



#### Some Words of Wisdom

Data is dumb.

More data is not better ...

unless you can do something with it to make it meaningful.



#### DATA... what is it?

- Individual facts, terms, numbers
- It is meaningless without CONTEXT
- Requires CONTEXT for conversion

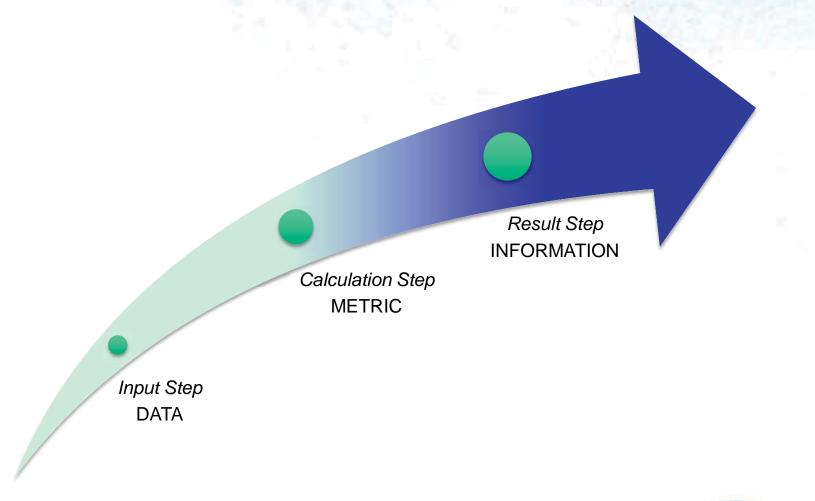


#### CONTEXT... what is it?

- Drawing data from various sources provides context
- Often referred to as "data about data" (metadata)
- Context provides meaning to data
- Appropriate metrics for converting data into information
  - Calculations
  - Charts
  - Statistical Analysis
  - Summary Reports

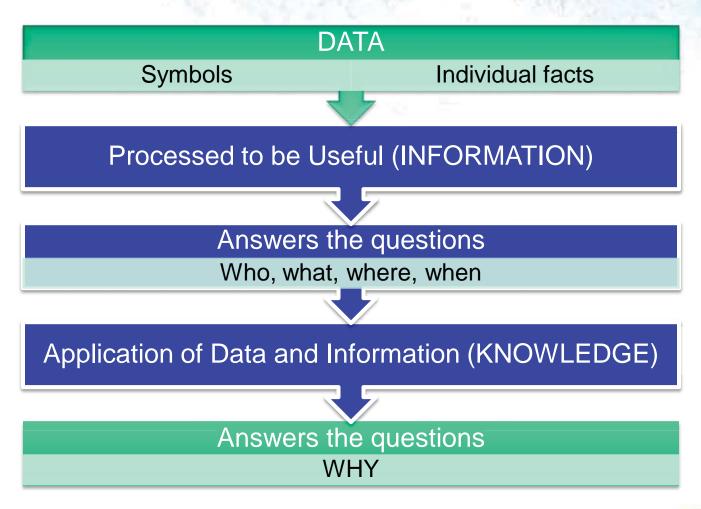


#### How do we transform data into information?





# Another way to think about it





#### Acronyms ... what do they mean?

#### SCADA

- <u>Supervisory</u> <u>Control</u> <u>and</u> <u>Data</u> <u>Acquisition</u> (Single Data Source)
- Not SCADAR
  - Reports were never part of the plan
  - DMR, MOR, regulatory reports never considered
    - Data set is incomplete
- Not SCAIR either
  - Acquires data... not information
  - Data for process control

#### LIMS

- <u>L</u>aboratory <u>Information <u>M</u>anagement <u>S</u>ystem (Single Data Source)
  </u>
  - Collects data from instruments, technicians
  - Tracks samples, assigns unique identifiers, creates worksheets
  - Applies metrics and workflows
  - Reports lab information



### Acronyms ...what do they mean?

- Central Database Solutions (Data Repository)
  - A central repository of DATA
  - Provides CONTEXT to DATA
    - Combine data with data from many sources
      - » SCADA
      - » LIMS
      - » Contract Labs
      - » Bench Sheets
      - » Manual data entry
  - DATA is processed and transformed into INFORMATION
    - For compliance reporting
    - Optimize plant operations
    - Troubleshooting
  - KNOWLEDGE is the application of data and information
    - Allows for understanding of why
      - » Risk management
      - » Corrective actions

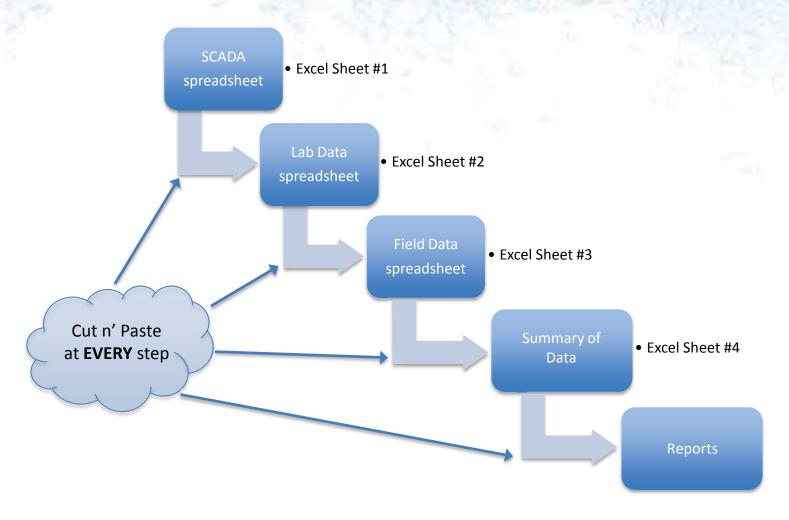


#### Excel... for information management?





# Excel for Reporting





#### Do you know this person?

#### MCHUMOR.com by T. McCracken



"I've been here for 30 years. I've forgotten what my exact role is, but I do finally know how to fill out all the forms."

@T McCracken mchumor com

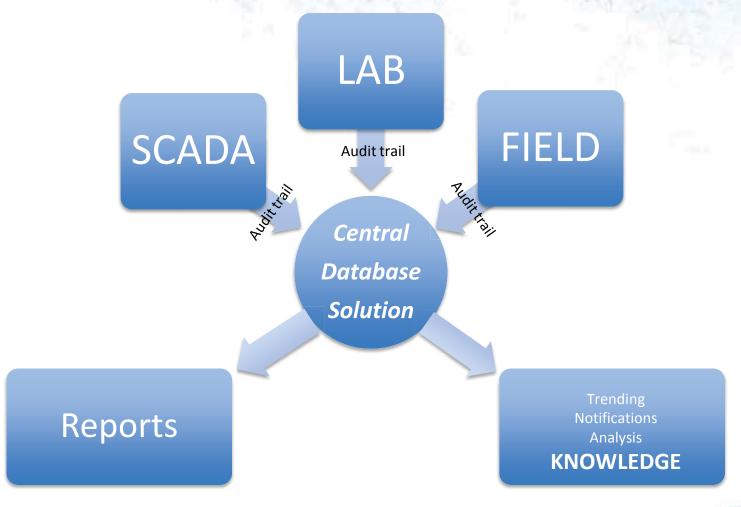


### Information Management Benefits

- Collect all data from different sources
- Use validated metrics to convert DATA into INFORMATION
- Eliminates the use of cut/paste strategies that:
  - Waste time
  - Waste human resources better spent on more important tasks
  - Increase error rate, re-entry of data, dyslexia
  - Not supported by a commercial vendor
    - "Joe wrote a great macro 10 years ago, but he retired"
- Honestly, how much time does it REALLY take?
  - One customer shared that it took 14, 12-hour days to create his reports
    - No time left for data review
    - No time to identify and correct any deviations



## **KNOWLEDGE** Management



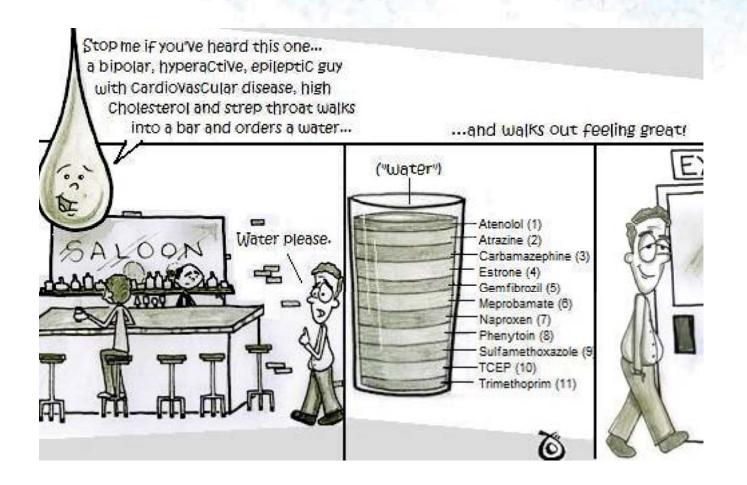


# Big plants, Regulatory Agencies Small plants, Regulatory Agencies





### Ever-changing environmental concerns





#### More data, less time Limited human resources

- More stringent reporting requirements
  - e-Reporting a reality
- More contaminants
  - New instrumentation required
  - More sources of data
- Less time to:
  - Navigate your data terrain
  - Manually put data into context ... and into Excel
  - Convert data into information... and manually generate reports
  - Combine data and information to gain knowledge to
    - Optimize plant processes
    - Enable "Predictive Troubleshooting"
    - Take action before it creates a problem AND
    - Manage risk, avoid violations, consent decrees



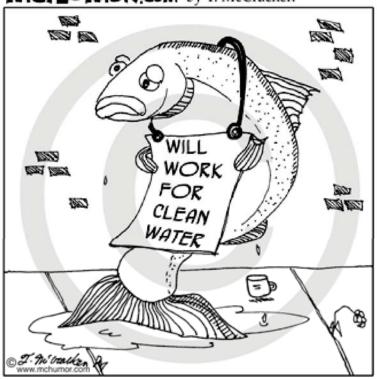
### **Available Options**

- DO NOTHING
  - Wait for things to happen, hope for the best
- DO THE SAME THING
  - Try to react to the ever-changing environment we live in
- EMPLOY A INFORMATION MANAGEMENT STRATEGY
  - Maintain all of your data into a single database
  - Put into into context
  - Transform it into information
  - Create knowledge by bringing your data and information together
  - Use that knowledge to make the best decisions in the shortest time
  - Manage your risk, before it manages you.



### One common goal



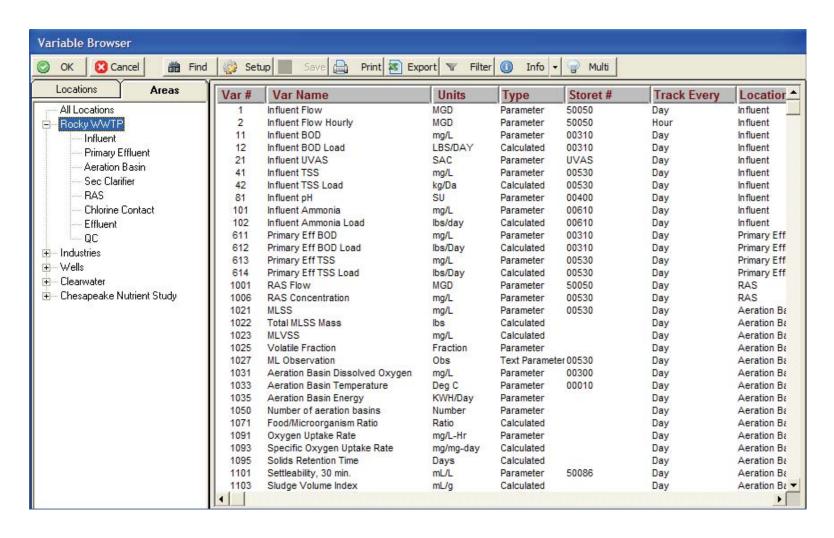


©T. McCracken mchumor.com



Monitor & Improve Operations

#### Central location for all of your data





#### **User Defined Dashboards**

# Lancaster Area Sewer Authority Susquehanna Water Pollution Control Facility



Wednesday, November 03, 2010

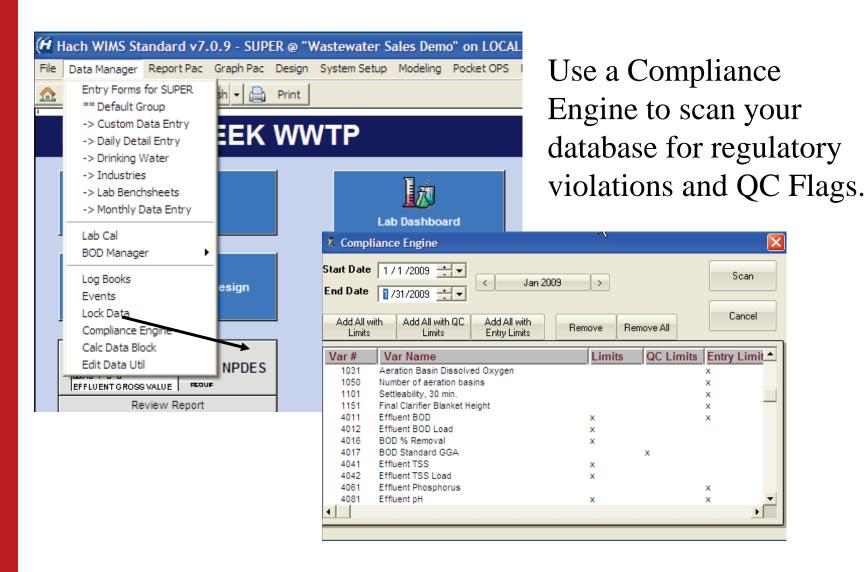




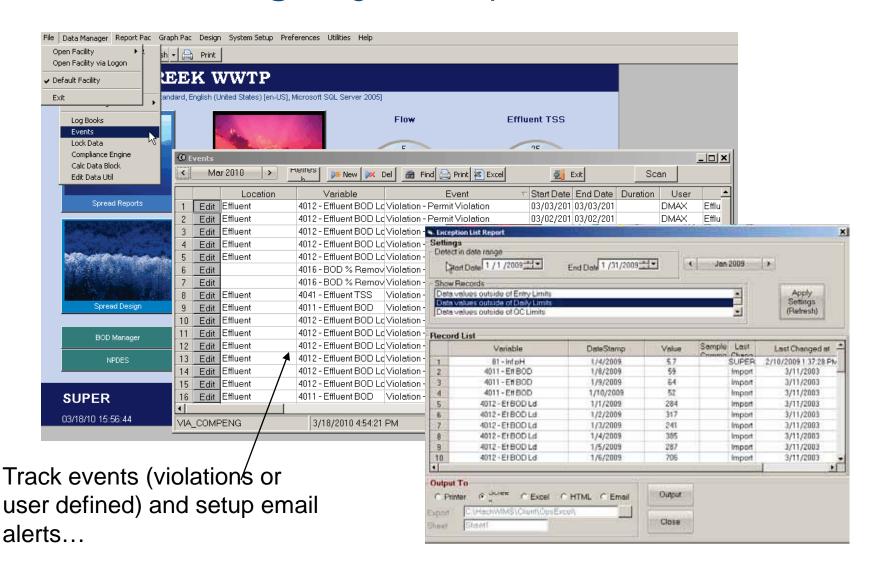
#### **User Defined Dashboards**

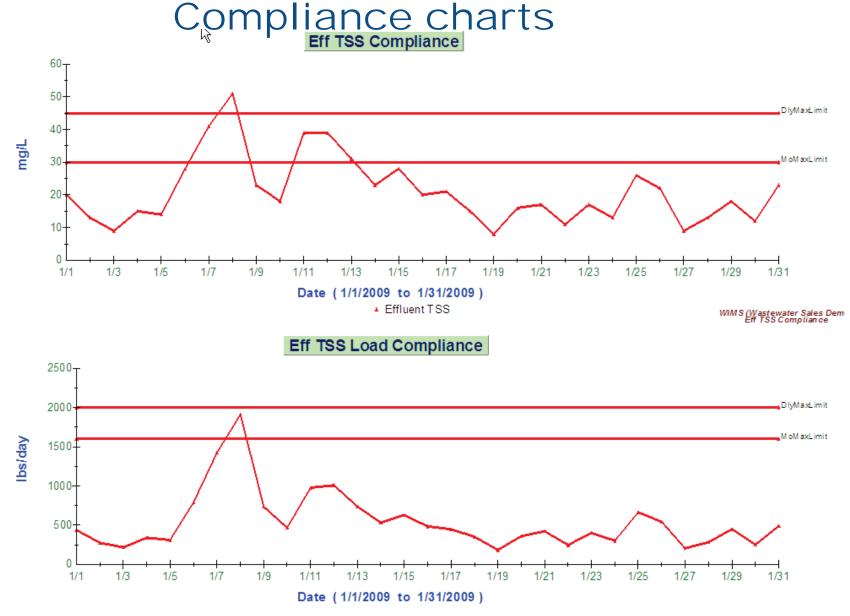


#### Manage by exception...



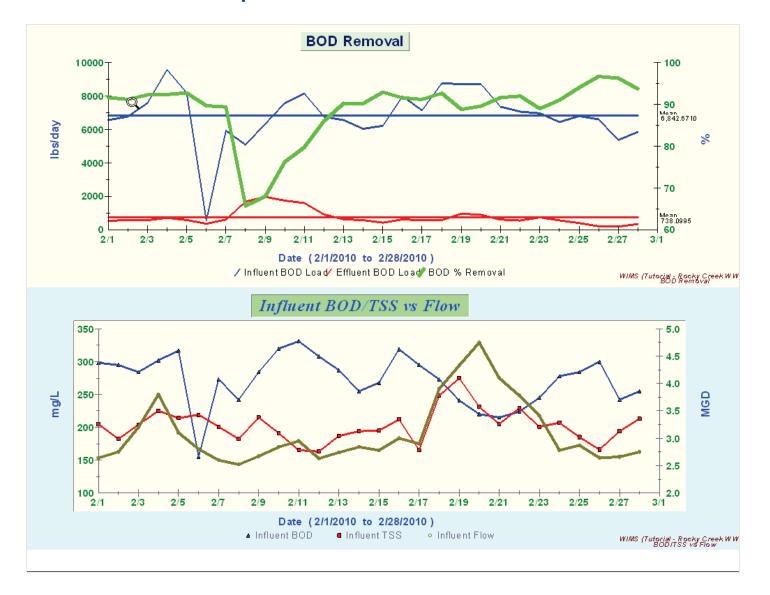
#### Manage by exception...





Automatically chart data for key plant parameters.

#### Compliance charts

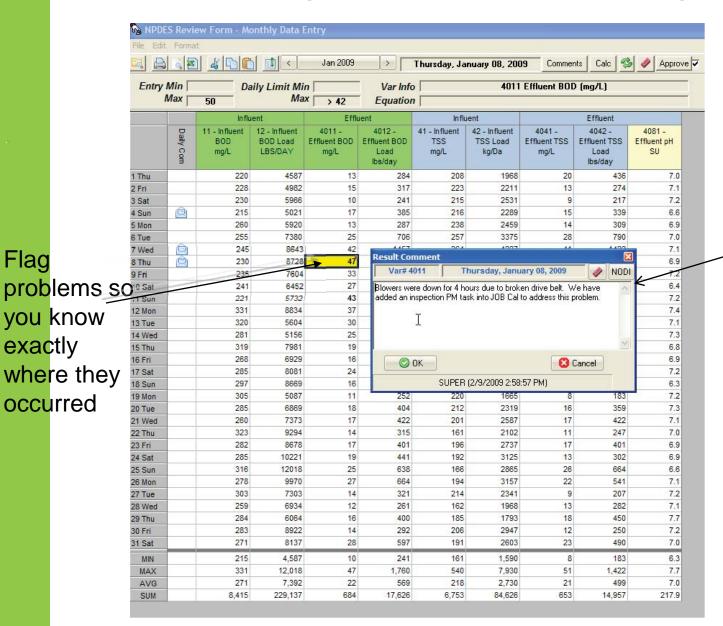


### Troubleshoot Issues

#### Let Software do the Detective Work

- Gather information and find answers to:
  - System upsets
  - Cost overruns
  - Compliance issues
  - Customer complaints
- Use predictive modeling tools to prevent future issues from occurring
  - Develop "what if" scenarios
- Perform simple or complex search queries
  - Find the exact information you need

#### Automatically Compare & Verify All Info



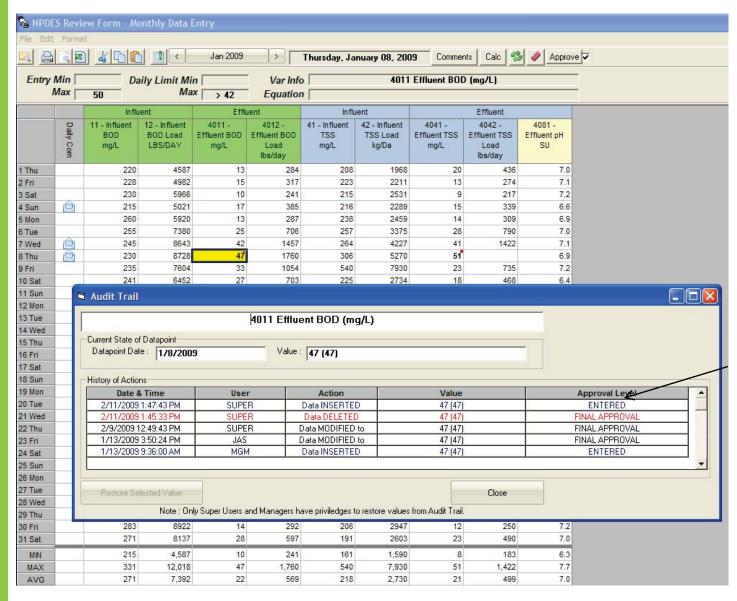
Flag

exactly

occurred

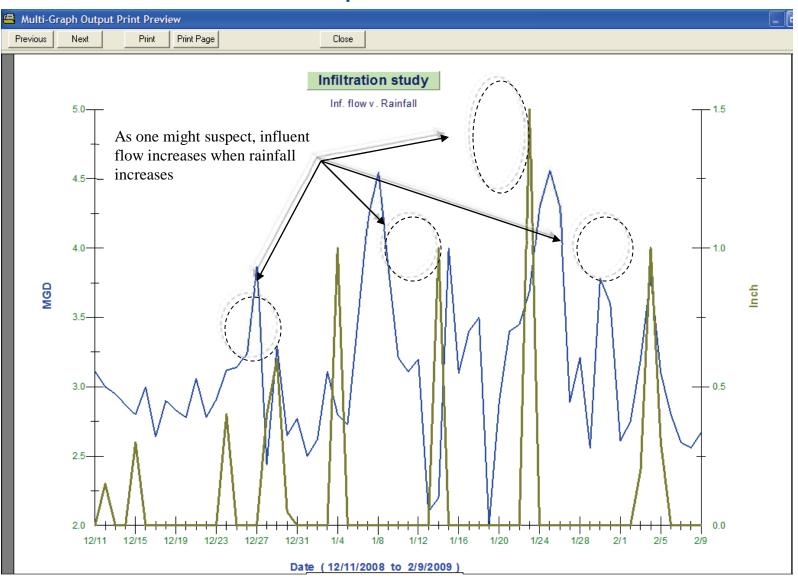
**Immediately** identify change in the system

#### Maintain Accurate Records with Audit Trails

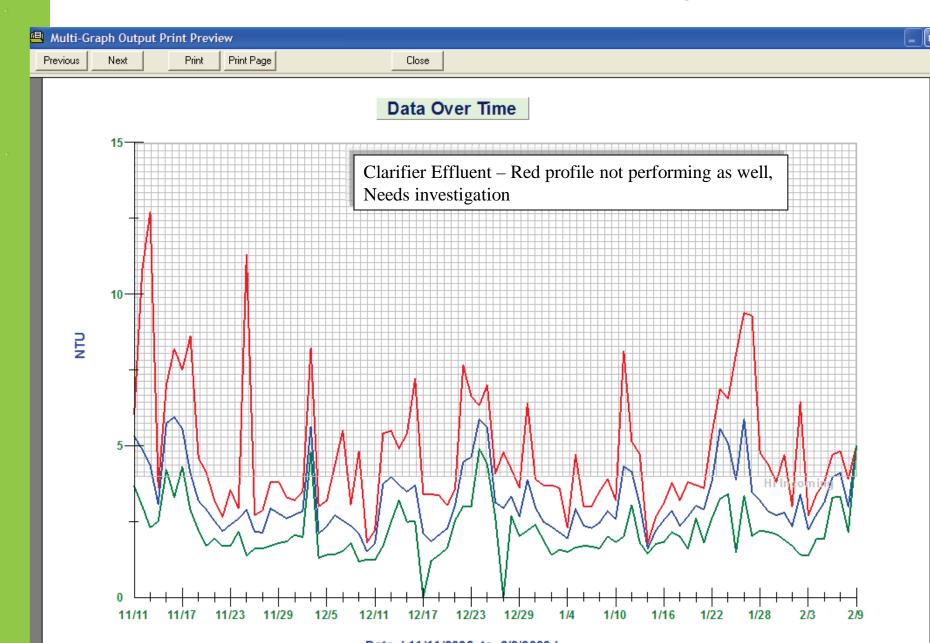


Audit trails show who touched the data

# Waste water example-Inf flow v. rainfall



### Waste water example—turbidity profile

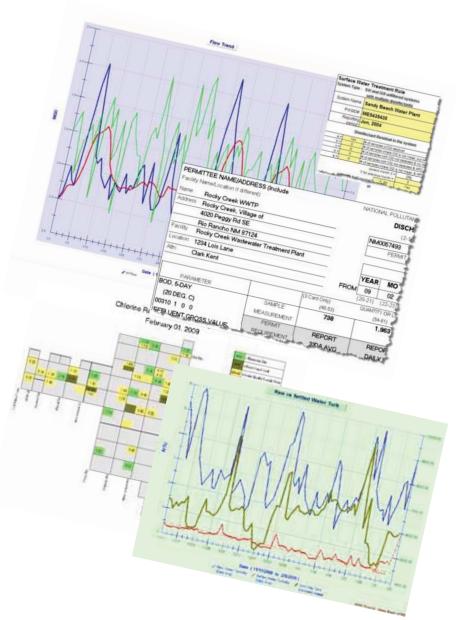


#### Prepare Regulatory & Internal Reports

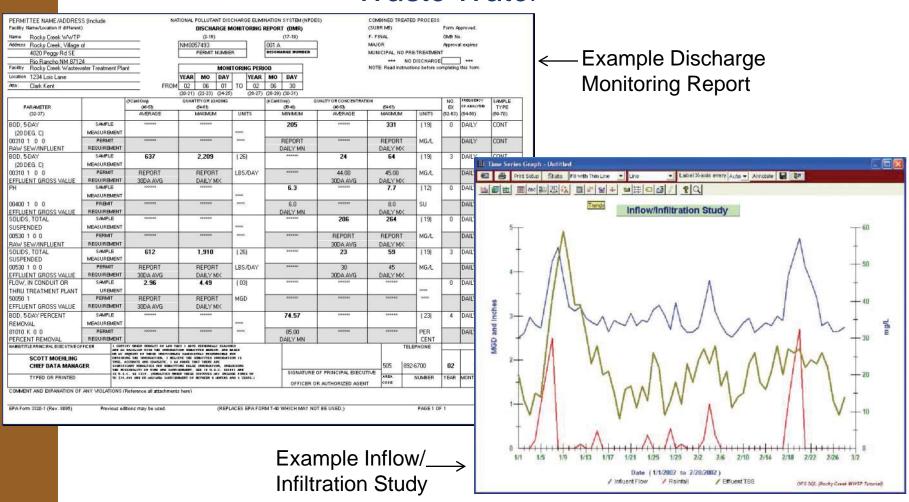
You already have the raw data, use it to your advantage

#### Turn Raw Data into Actionable Information

- Create business and regulatory reports instantly
- Schedule automatic report output to the screen, print or email
- Quickly configure standard reports using templates and wizards
- EPA and state report templates (SWTR, DBR, NPDES, DMR, eDMR, MOR, SDWA, CCR, industrial pretreatment compliance, and more)



# Powerful Information at your Fingertips Waste Water

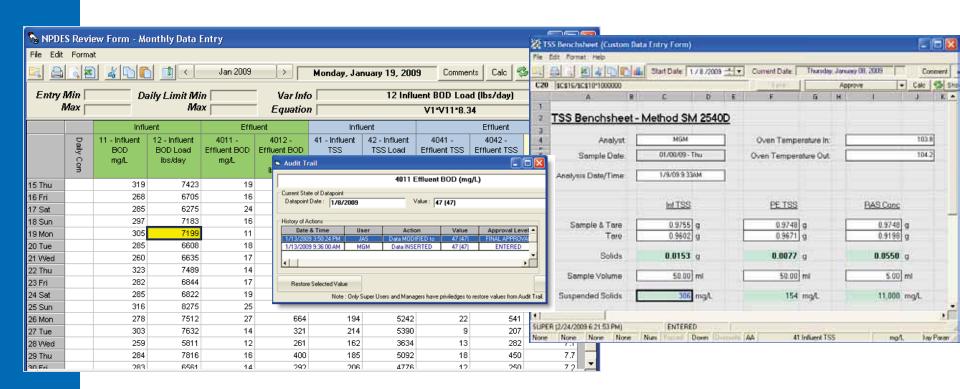


#### Compile Data Easily

More data is not better unless you can do something with it

# ee Your Whole Operation: Combine Data From Field, Lab & Operations

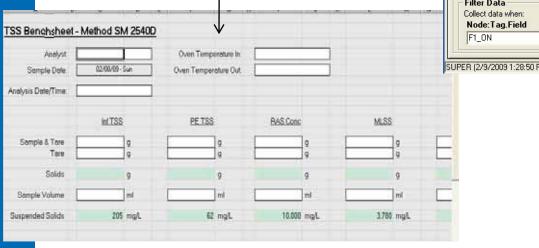
- Access to data, reports, entry forms, audit information
- Track dosages, flows, concentration, summarized data like daily average flow, 15 minute turbidity maxi, hourly DO min

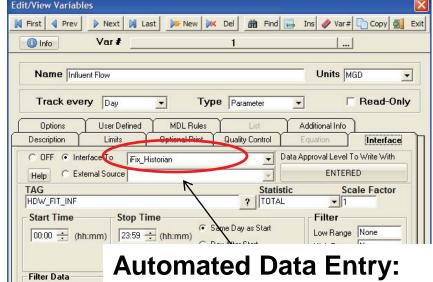


Transform Data Dumps Into Actionable Insight

#### **Manual Data Entry:**

- Built-in templates for data entry
- Screens that match spreadsheet, daily log and bench sheet formats





- SCADA
- **Dataloggers**
- LIMS

Node: Tag. Field

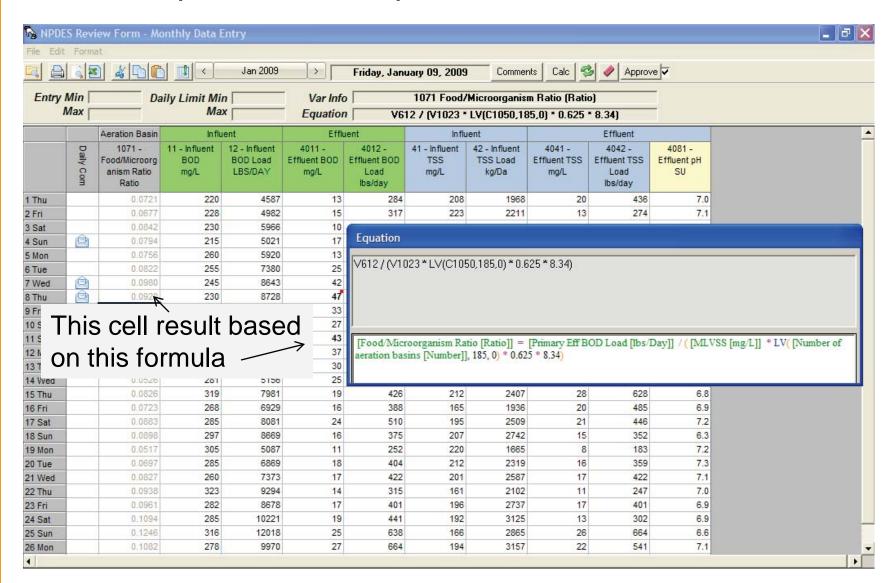
F1 ON

- Commercial Laboratory Reports
- Download data from portable field devices

#### Manage Complex Calculations

Stop worrying about making mistakes in complex calculations

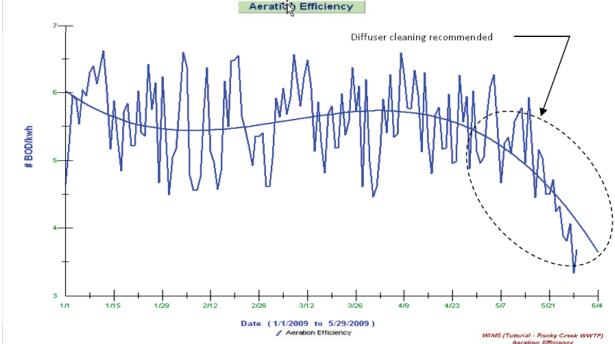
#### Example of Complex F-to-M Ratio



# Case Studies Making Data Meaningful

# Case 1: Improving Aeration Efficiency

Oxygen transfer efficiency is a function of bubble size. The smaller the bubble, the higher the efficiency. Ceramic or membrane diffusers will foul in time, causing the bubble size to increase. To minimize energy used and reduce downtime, it is important to determine when diffusers need to be cleaned.

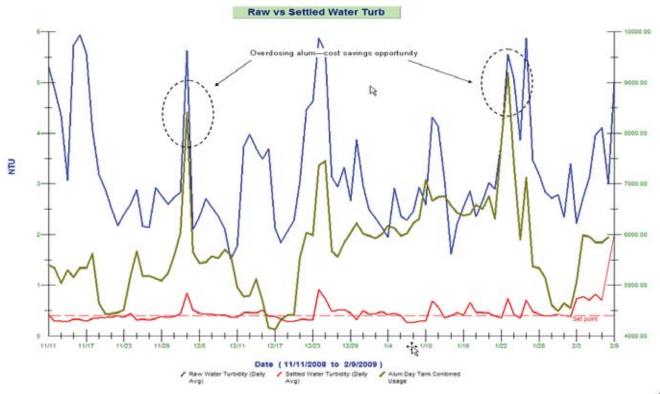


Determine cleaning cycle by benchmarking pounds of BOD removed per KW of electricity used.



# Case 2: Lowering Chemical Costs

In the chart of raw and settled water turbidity against alum dosage, increased alum dosing (circled) did not improve clarified turbidity. This led to an opportunity to save up to 10 percent in alum dosage.



#### Case 3: Real-time estimation of BOD

The time-lag in obtaining BOD results, a 5-day lab test, make it challenging for plant operations to adjust treatment processes to adverse levels of BOD. A site-specific correlation between on-line UV absorbance against lab BOD measurements was obtained. With this BOD can be estimated in real-time, allowing for the optimization for treatment processes and avoidance of potential violations.

