# Disinfection By-Products

Adapted by Peter Nathanson, NMRWA from the NMED-Drinking Water Bureau

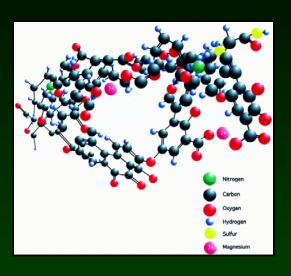




## DBP Chemistry

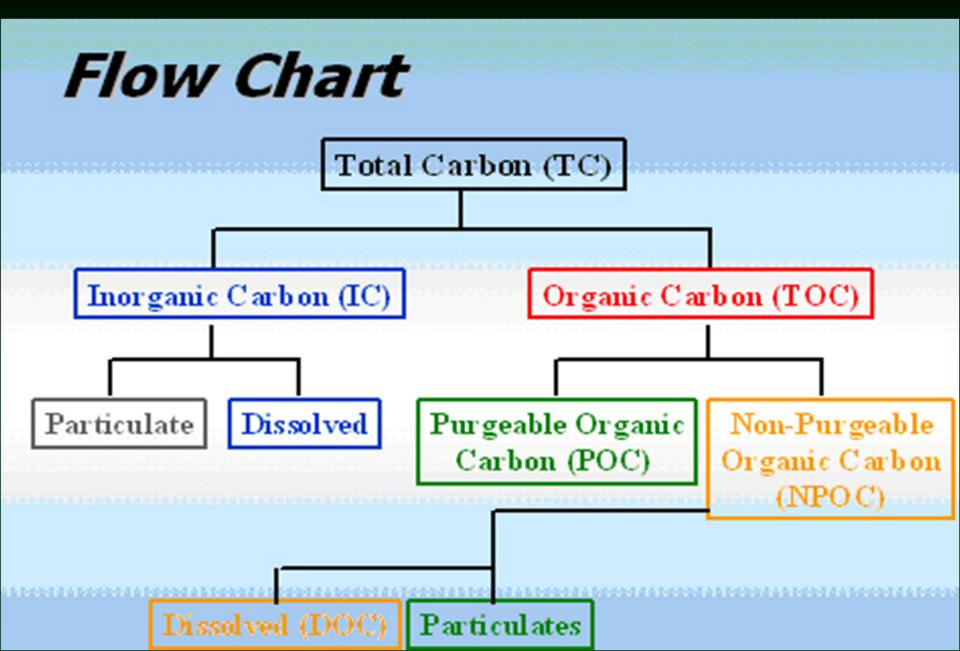
### **Organic Matter**

 Natural Organic Matter (NOM) is found in all surface water and comes from decaying plant and animal matter.



- Total Organic Carbon (TOC) is a measure of all organic carbon in water and includes NOM, petroleum products, pesticides, etc.
- Dissolved Organic Carbon (DOC) is that portion of TOC that is dissolved. It is usually the largest fraction of TOC and of the most concern.

### Flow Chart Illustrating Total Carbon Constituents



### **Total Carbon Constituents**

- ➤ Total Carbon (TC): all the carbon in the sample, including both inorganic and organic carbon
- ➤ Total Inorganic Carbon (TIC): often referred to as inorganic carbon (IC), carbonate, bicarbonate, and dissolved carbon dioxide (CO2)
- ➤ Total Organic Carbon (TOC): material derived from decaying vegetation, bacterial growth, and metabolic activities of living organisms or chemicals
- ➤ Non-Purgeable Organic Carbon (NPOC): commonly referred to as TOC; organic carbon remaining in an acidified sample after purging the sample with gas
- ➤ Purgeable (volatile) Organic Carbon (POC): organic carbon that has been removed from a neutral, or acidified sample by purging with an inert gas. These are the same compounds referred to as Volatile Organic Compounds (VOC) and usually determined by Purge and Trap Gas Chromatography
- ➤ Dissolved Organic Carbon (DOC): organic carbon remaining in a sample after filtering the sample, typically using a 0.45 micrometer filter
- Suspended Organic Carbon (SOC): also called particulate organic carbon (PtOC); the carbon in particulate form that is too large to pass through a filter

## **DBP Chemistry**

 Disinfection Byproducts (DBPs) are caused by the reaction of a disinfectant with DBP precursors.
The most significant precursor is NOM:

Disinfectant + NOM → DBPs

- There are hundreds of different DBP compounds, many of which can have serious human health impacts (e.g. birth defects and carcinogenicity).
- Only a few DBPs are regulated:
  - Trihalomethanes (THM)
  - Haloacetic Acids (HAA)
  - Chlorite
  - Bromate

## **DBP Chemistry**

#### **Trihalomethanes**

- The MCL is for the combined concentration of 4 trihalomethanes, Total Trihalomethanes (TTHM)
- Results from use of Chlorine as disinfectant

### Haloacetic Acids

- The MCL is for the combined concentration of 5 Haloacedic Acids (HAA5)
- Results from use of Chlorine as disinfectant

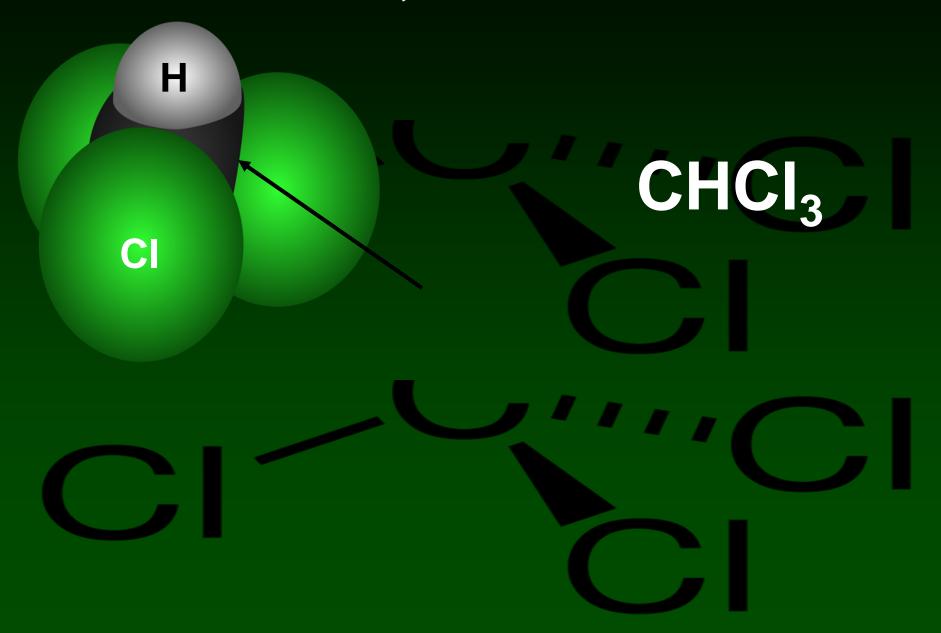
### **Chlorite**

Results from dissociation of Chlorine dioxide as disinfectant (doesn't require NOM)

### **Bromate**

Results from use of Ozone and requires Bromide as a precursor

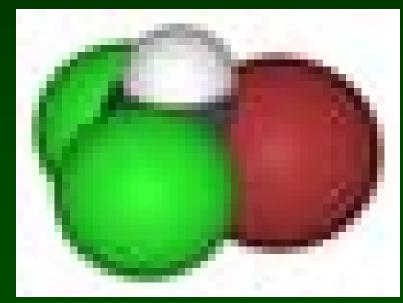
## TTHMs – Ex., Trichloromethane



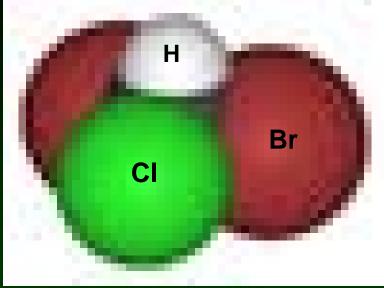
## All Four (4) TTHMs



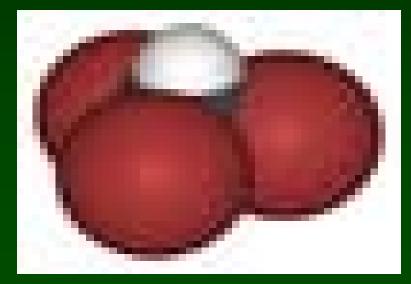
**Chloroform/Trichloromethane - CHCl<sub>3</sub>** 



Bromodichloromethane - CHBrCl<sub>2</sub>

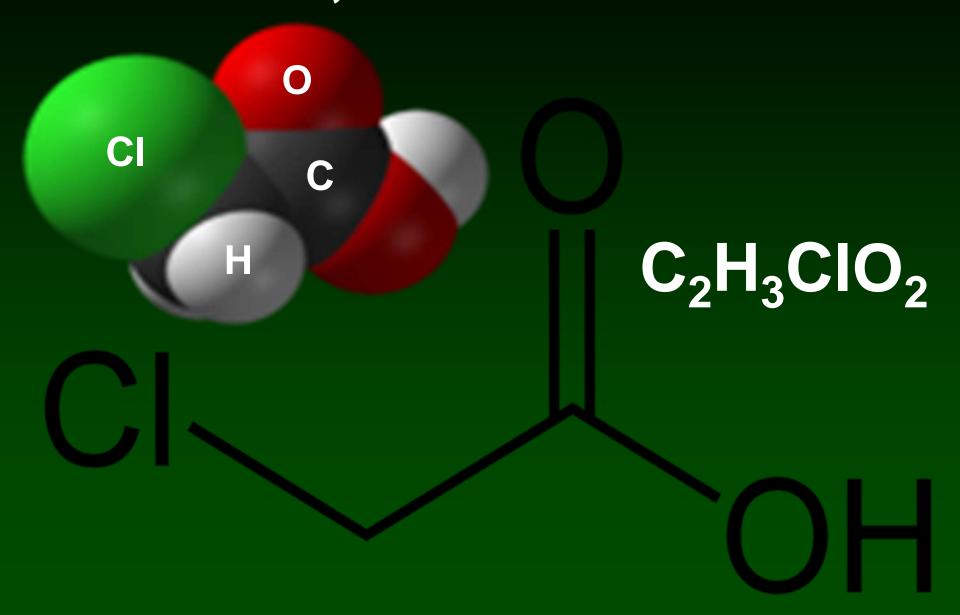


Chlorodibromomethane - CHBr<sub>2</sub>Cl

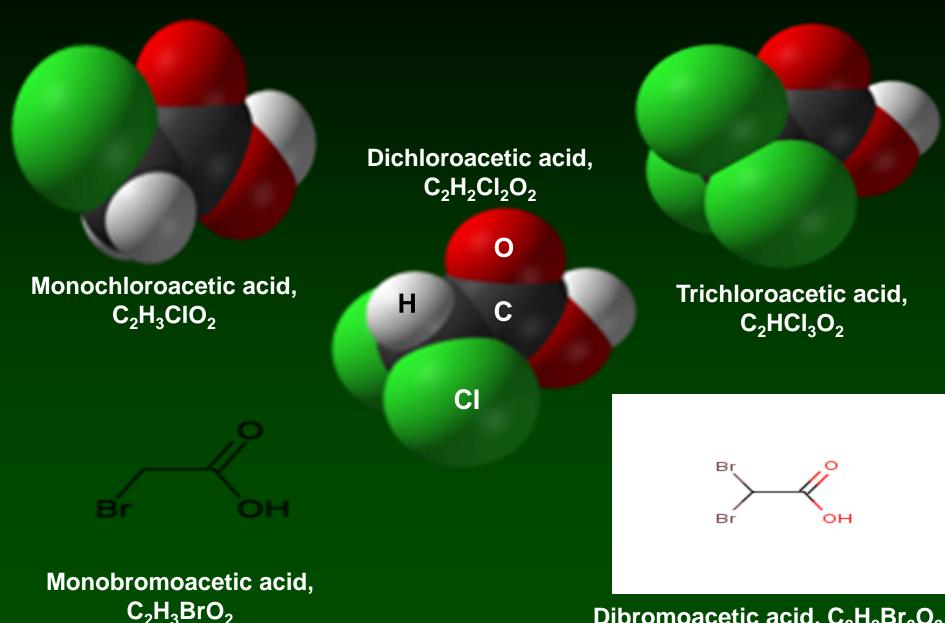


Bromoform/Tribromomethane - CHBr<sub>3</sub>

## HAA5s - Ex., Monochloroacetic Acid



### All Five (5) HAA5s



Dibromoacetic acid, C<sub>2</sub>H<sub>2</sub>Br<sub>2</sub>O<sub>2</sub>

## **DBP Chemistry**

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Results from use of Ozone and requires Bromide as a precursor

## **DBP Chemistry**

### **DBP Formation Factors**

- Temperature Increasing temperature results in increased DBP formation rate.
- pH THMs increase somewhat with pH, HAAs increase with decreasing pH.
- Time Reaction is rapid for the first few hours and then decreases. Reaction will continue as long as there is disinfectant and precursors.
- Disinfectant Dose Increasing dose results in increasing DBP formation
- Type and concentration of NOM

- Published in 1998.
- Stage 1 DBPR regulates the allowable levels of disinfectants, DBPs, and DBP precursors in drinking water.
- Applies to CWS and NTNC systems that add a disinfectant, and transient systems using chlorine dioxide.

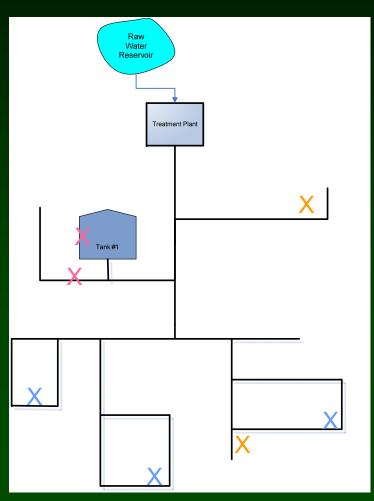
### Monitoring

- Systems required to monitor must have and follow a monitoring plan, to include specific locations and schedules.
- Monitoring must include:
  - DBPs
  - Disinfectant residuals
  - TOC and alkalinity (conventional treatment only)
- DBP sampling locations must be representative of Maximum Residence Time (MRT)

Find active locations of MRT:

 Locations most distant from point of chlorination

- Areas of low usage
- Water within or downstream of tanks



### Monitoring Schedule

Analyte	Applicability	Frequency	
	Surface water and GWUDI serving ≥ 10,000	4/plant/quarter	
	Surface water and GWUDI serving 500 - 9,999	1/plant/quarter	
TTHM/HAA <sub>5</sub>	Surface water and GWUDI serving < 500	1/plant/year in month of warmest water temperature	
	Ground water serving ≥ 10,000	1/plant/quarter	
	Ground water serving < 10,000	1/plant/year in month of warmest water temperature	
Bromate	Ozone plants	Monthly	
Chlorite	Chlorine dioxide plants	Daily at entrance to distribution system; monthly in distribution system	
Chlorine Dioxide	Chlorine dioxide plants	Daily at entrance to distribution system	
Chlorine/Chloramines	All systems	Same location and frequency as TCR sampling	
DBP Precursors	Conventional filtration plants	Monthly for total organic carbon and alkalinity	

### Compliance Requirements:

- Maximum Contaminant Levels are:
  - 0.08 mg/l for TTHM
  - 0.06 mg/l for HAA5
  - 0.01 mg/l for Bromate
  - 1.0 mg/l for Chlorite
- Systems collecting <u>quarterly</u> samples base MCL on a running annual average (RAA) of 4 consecutive quarters.
- Systems collecting more than 1 sample per quarter average all samples over past year for RAA.

RAAs are calculated each quarter using the most recent 4 quarters of data for all samples. Is the following system in compliance?

Quarter	Q1	Q2	Q3	Q4	Q5	Q6
TTHM	47 µg/l	108 µg/l	94 µg/l	52 µg/l	67 µg/l	95 µg/l
RAA	n/a	n/a	n/a	75.2 µg/l	80.25 µg/l	77.0 µg/l

- Maximum Residual Disinfectant Level (MRDL) for chlorine is 4 mg/l.
- Compliance with MRDL is based on RAA of one consecutive year's worth of monthly averages of all samples.
- Total chlorine residual cannot be undetectable in distribution in more than 5% of samples/month.

- TOC must be monitored monthly at conventional treatment systems.
- A source water sample must be tested for TOC and alkalinity and a CFE sample must be tested for TOC at the same time.
- Rules of thumb:
  - The lower the TOC, the lower the percentage removal of TOC
  - The higher the alkalinity, the lower the percentage removal of TOC

 The table gives the percent TOC removal required for conventional treatment systems:

Source Water TOC, mg/l	Source Water Alkalinity, mg/l as CaCO3			
	0 - 60	> 60 - 120	> 120	
> 2.0 – 4.0	35%	25%	15%	
> 4.0 – 8.0	45%	35%	25%	
> 8.0	50%	40%	30%	

- Removal is based on the RAA of monthly samples.
- Failure to meet removal will require:
  - Step 1: Adjustments to treatment
  - Step 2: Determination of alternative removal requirement

- Published in 2006.
- Stage 2 builds on Stage 1 and accomplishes the following:
  - Broadens applicability
  - Requires Initial Distribution System Evaluation (IDSE) to identify locations of greatest DBP concentrations
  - Requires routine monitoring plan be adjusted to reflect results of IDSE
  - Stricter compliance requirements

- Stage 2 applies to all CWS and NTNC that deliver water that has been treated with a disinfectant (other than UV).
- Therefore, consecutive systems (i.e. purchase systems) are subject to Stage 2.
- For combined systems, the wholesale and consecutive systems are on the same schedule.

- IDSE has 4 options for determining Stage 2 sampling locations:
  - Standard Monitoring, which is one year of DBP samples. The number of locations and the sampling frequency varies with the size of the system.
  - System Specific Study using either sufficient existing data or modeling.
  - 40/30 certification means there were no sampling violations under Stage 1 and all samples analyzed were less than 40 μg/l TTHM and 30 μg/l HAA5. No additional sampling required.
  - Very Small System Waiver for systems serving < 500 and have collected all Stage 1 samples. No additional sampling required.

 Results of IDSE are used to determine routine monitoring locations. Monitoring frequency is:

Source Water Type	Population Size	Monitoring Frequency	Distribution System Monitoring Locations
Subpart H	< 500	Per year*	2
	500 – 3,300		2
	3,301 – 9,999		2
	10,000 – 49,999		4
	50,000 – 249,999	Per quarter	8
	250,000 – 999,999		12
	1,000,000 – 4,999,999		16
	≥ 5,000,000		20
Groundwater	< 500	D*	2
	500 – 9,999	Per year*	2
	10,000 – 99,999		4
	100,000 – 499,999	Per quarter	6
	≥ 500,000		8

<sup>\*</sup> During warmest time of year

- Compliance based on a Locational Running Annual Average (LRAA) which is an RAA, but computed for each monitoring location. If any of the LRAAs are above MCL, system is out of compliance.
- If a system monitors once per year, compliance is determined by the one sample.

- Operational Evaluation Levels were introduced to try and reduce peaks in DBPs.
- Operational Evaluation Levels (QEL) must be computed quarterly for each sample location for TTHM and HAA5:

$$QEL = (Q_1 + Q_2 + 2 \cdot Q_3)/4$$

where  $Q_3$  is the current quarterly result, and  $Q_1$  and  $Q_2$  are the results from the two previous quarters.

- If the QELs exceed the MCLs, the system must perform an operational evaluation to include consideration of treatment and distribution operational practices.
- An operational evaluation report must be submitted to the state within 90 days of receiving the analytical results.
  Failure to submit report is a M&R violation

- System review for addressing high disinfection byproducts should include the following:
  - Treatment modifications
  - Disinfectant usage
  - Storage design, operation and maintenance
  - Distribution management

### **Treatment:**

- Goal is to reduce DBP precursors prior to chlorination.
- Modifications to process control should be reviewed:
  - Is the operator properly adding coagulant?
  - Is there a better coagulant option?
  - Should a coagulant aid be used?
- Does the operator have sufficient time at the plant?

### Disinfectant:

- Is the system pre-chlorinating? If so, can this be eliminated?
- Can the chlorine dose be decreased?
  - Make sure CT is being met
  - Would this require a new disinfection profile?
  - Can booster chlorination be considered?
  - Do consecutive systems need to be considered?

### Storage:

- Goal is to reduce water age.
- Are tanks poorly mixed? Possible solutions include
  - Decreasing inflow/outflow pipe diameter
  - Baffling tank
- Can tank operation be modified to increase turnover rate?
- Is there excessive storage capacity?

### Distribution:

- Goal is to minimize residence time in distribution.
- Ensure an adequate flushing program.
- Looping dead ends when appropriate.
- Ensure proper valving.

### Questions

- What are the minimum and maximum chlorine residuals allowed in distribution?
  At the system entry point?
- If the RAA of a system collecting annual samples for TTHM is 80.51 mg/l, are they in compliance?

### **Quick Reference Guides**

#### Stage 1 D/DBP Rule

http://www.epa.gov/safewater/mdbp/qrg\_st1.pdf

#### Stage 1 D/DBP Rule for Laboratories

http://www.epa.gov/safewater/mdbp/pdfs/st1labqrg-final02rev.pdf

Stage 2 D/DBP Rule for Schedule 1-4 systems (population based) <a href="http://www.epa.gov/safewater/disinfection/stage2/pdfs/qrg\_stage\_2\_dbpr\_qrg\_sc">http://www.epa.gov/safewater/disinfection/stage2/pdfs/qrg\_stage\_2\_dbpr\_qrg\_sc</a>

h1\_final.pdf (>100,00)

http://www.epa.gov/safewater/disinfection/stage2/pdfs/qrg\_stage\_2\_dbpr\_qrg\_sc h2\_final.pdf (50,000-99,999)

http://www.epa.gov/safewater/disinfection/stage2/pdfs/qrg\_stage\_2\_dbpr\_qrg\_sc h3\_final.pdf (10,000-49,999)

http://www.epa.gov/safewater/disinfection/stage2/pdfs/qrg\_stage\_2\_dbpr\_qrg\_sc h3\_final.pdf (<10,000)

#### **Distribution Systems: A Best Practices Guide**

http://www.epa.gov/safewater/smallsystems/pdfs/guide\_smallsystems\_dist\_system 08-25-06.pdf