

Lectures on sterilization and disinfection

- Principle of sterilization and disinfection
- Individual sterilization and disinfection processes
- Media-specific disinfection (water and wastewater)
- Media-specific disinfection (air and surfaces)
- Media-specific disinfection (infectious solids)

Common disinfectants in water/wastewater treatment processes

- Free chlorine
- Combined chlorine
- Chlorine dioxide
- Ozone
- UV

Key points

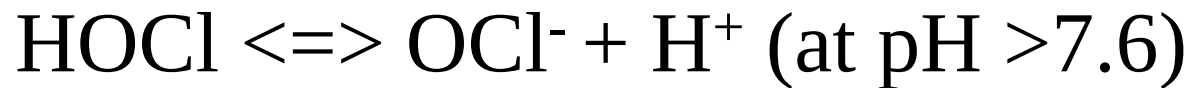
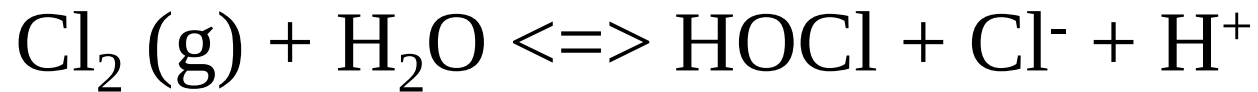
- Basic chemistry and principle
- Method of application
- Effectiveness on microbes
- Advantages/disadvantages

Chemical disinfectants

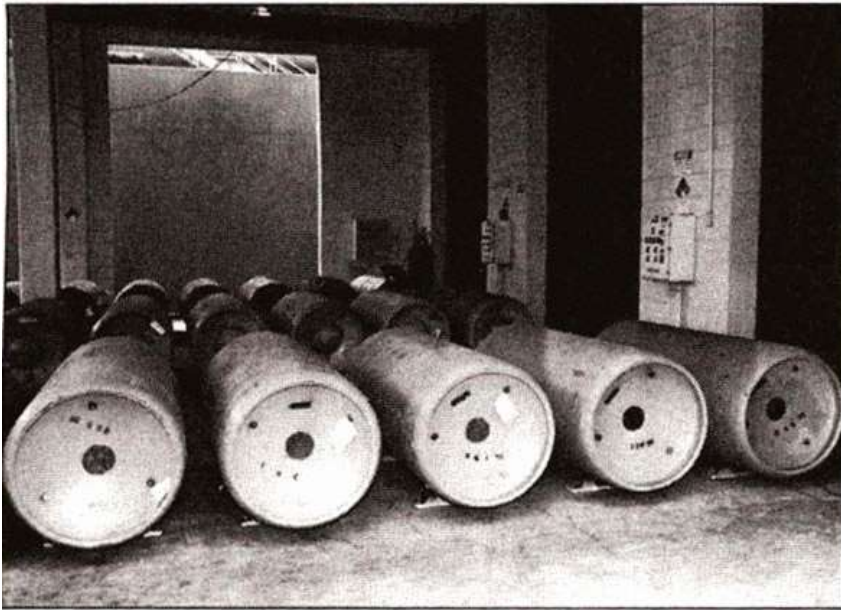
Free chlorine: Chemistry

- Three different methods of application
 - Cl_2 (gas)
 - NaOCl (liquid)
 - Ca(OCl)_2 (solid)

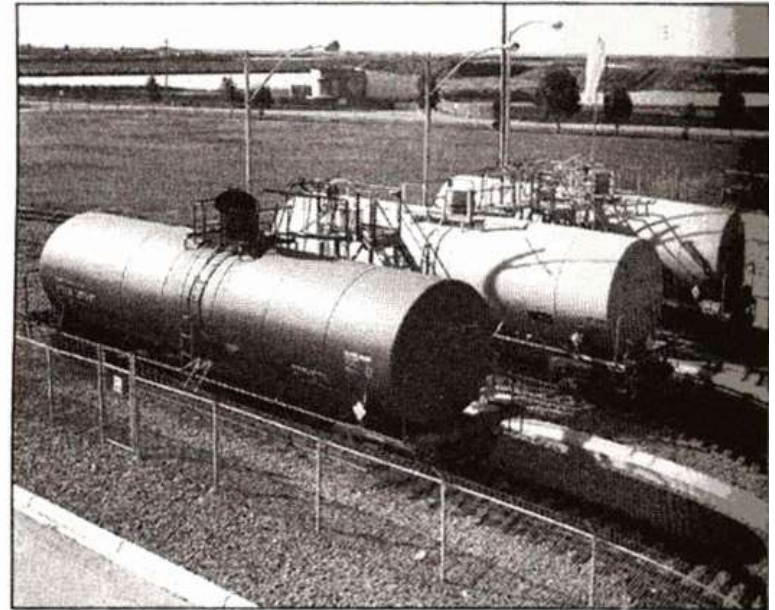
- Reactions for free chlorine formation:



Chlorine application (I): containers

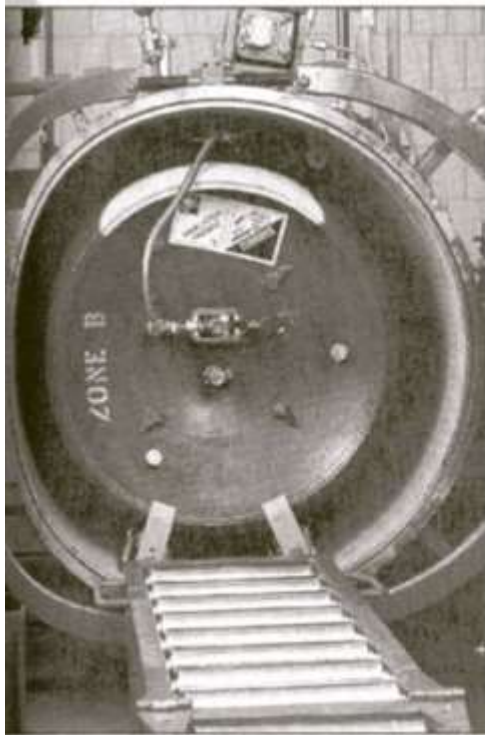


(a)

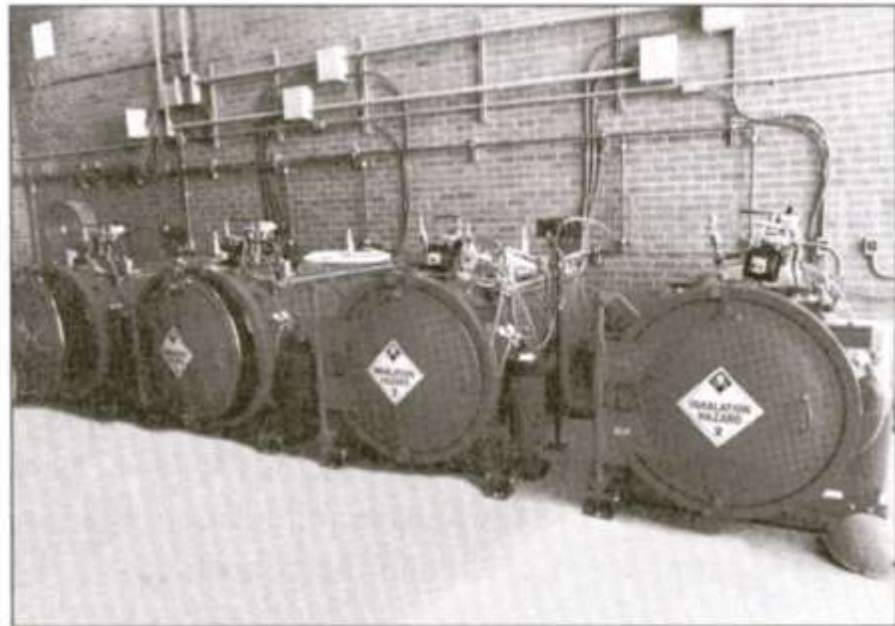


(b)

Chlorine application (II): containment vessels

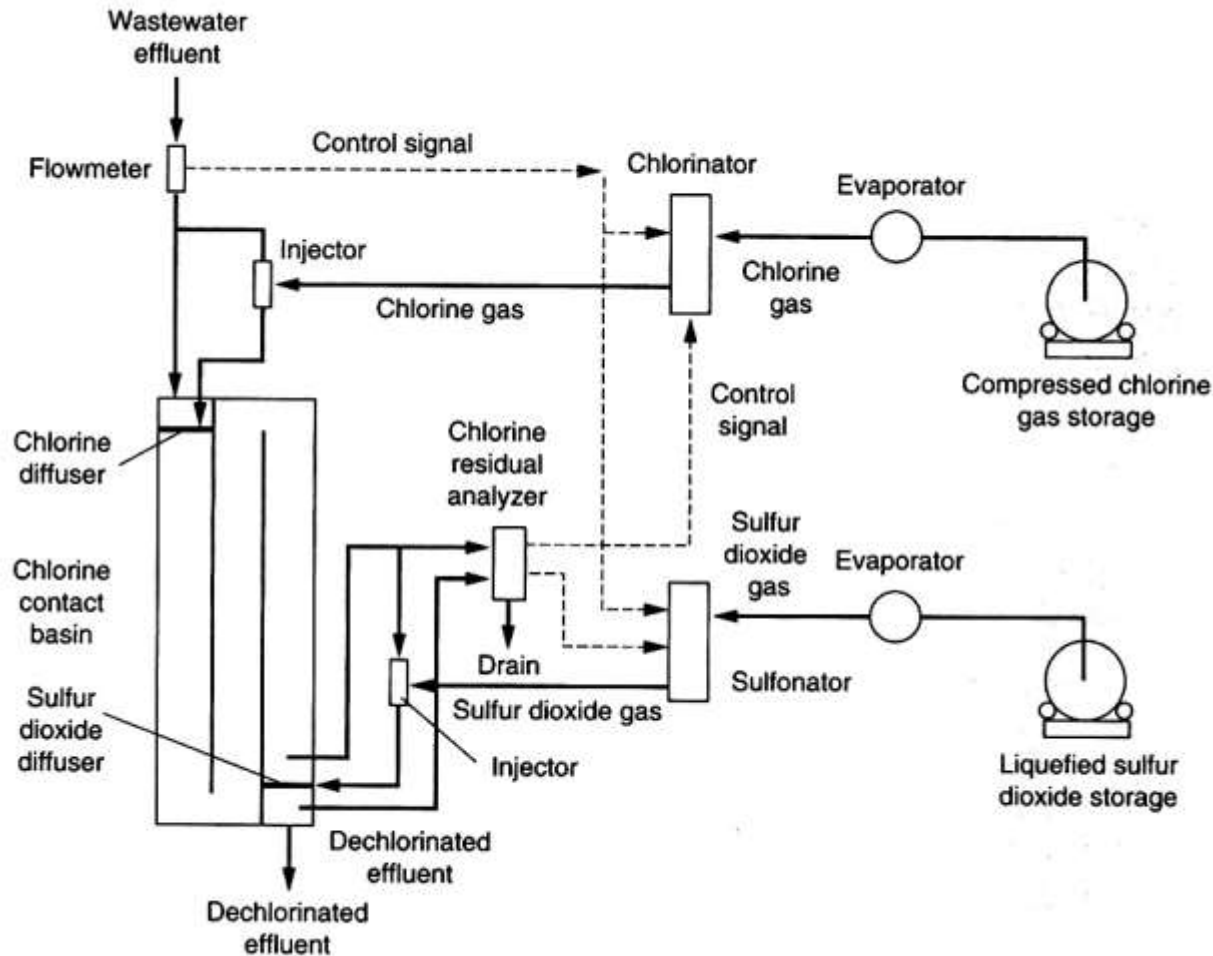


(a)



(b)

Chlorine application (III): flow diagram



(a)

Chlorine application (IV): Injectors

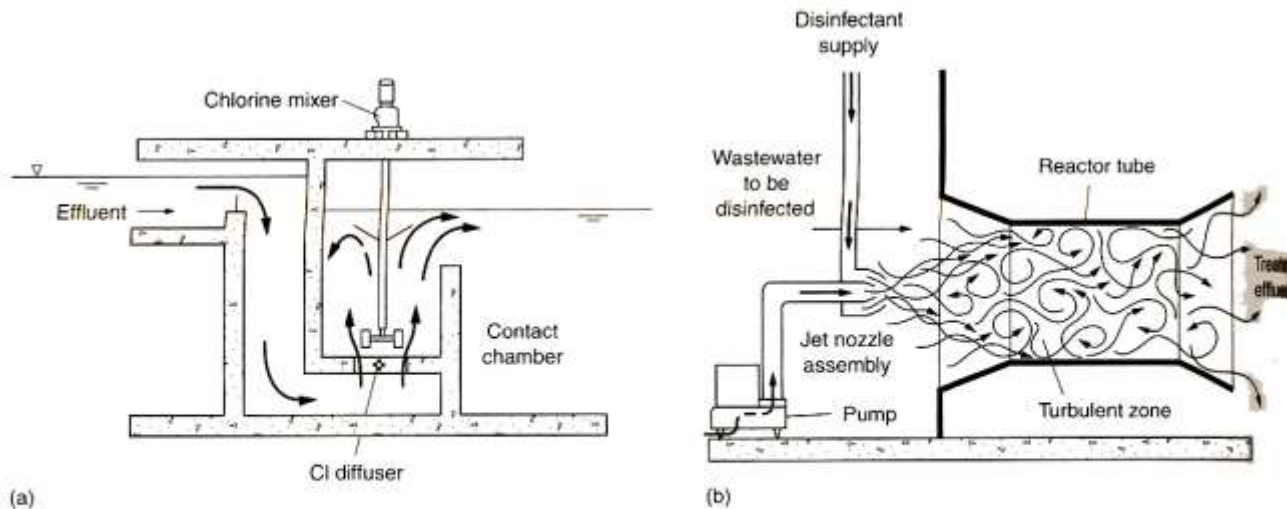


Figure 12-22

Typical mixers for the addition of chlorine: (a) in-line turbine mixer and (b) injector pump type. (From Pentech-Houdaille.) For additional types of chlorine mixers see Fig. 5-14 in Chap. 5.

Chlorine application (V): Contact chambers



(a)



(b)

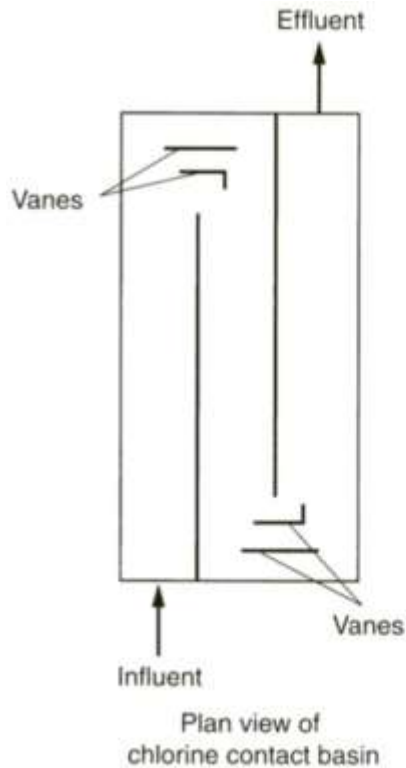


(c)

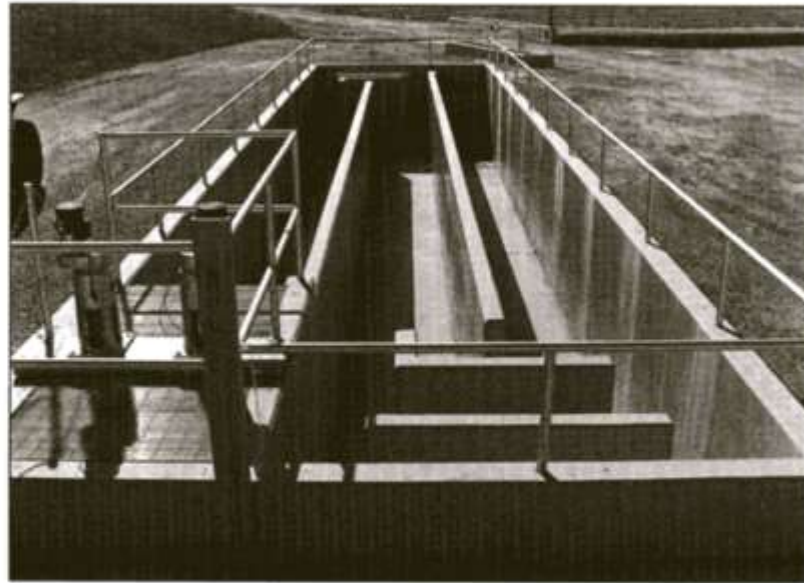


(d)

Chlorine application (VI): Contact chambers



(a)



(b)

Free chlorine: effectiveness (I)

TABLE 2 Inactivation of Health-related Bacteria in Water by Free Chlorine

Bacteria	Water	Cl ₂ Residual (mg/l)	Temp. (°C)	pH	Time (min.)	Reduction (%)	Est. C.t.*	Reference
<u>E. coli</u>	BDF	0.2	25	7.0	15	99.998-	ND	Ram and Malley, 1984
						99.996		
<u>E. coli</u>	BDF + N-organics	0.2	25	7.0	15	99.999-	ND	
						99.992		
<u>E. coli</u>	BDF	0.1	23	10	3.5	90	0.6	Haas et al., 1986
<u>E. coli</u>	" 0.1M KNO ₃	0.1	23	10	0.8	90	0.15	
<u>E. coli</u>	BDF, <7 µm*	0.5	5	7.0	30	ND	0.9	Berman et al., 1988
<u>E. coli</u>	BDF, >7 µm*	0.5	5	7.0	30	ND	2.7	
<u>E. coli</u>	CDF	1.5	4	?	60	ca. 99.9	ca. 2.5	LeChevalier
<u>E. coli</u> + GAC*	CDF	1.5	4	?	60	<<10	>>60	et al., 1984b
HPC	CDF	1.7	4	?	60	>99.995	2.5	
HPC + GAC	CDF	1.7	4	?	60	<10	>>60	
HPC	treated*	0.25-1.3	24	8.0	15	99.82	ND	Wolfe et al., 1984
						(42->99.99)		
HPC	treated	0.25-1.3	24	8.0	30	99.94	ND	
						(46->99.99)		
<u>C. jejuni</u>	BDF	0.1	4	8.0	5	99.98-	ND	Blaser et al., 1986
						>99.998		
<u>L. pneumophila</u> (water grown)	tap	0.25	20	7.7-7.8	58	99	15	Kuchta et al., 1985
<u>L. pneumophila</u> (media grown)	tap	0.25	20	"	4	99	ca. 1.1	
<u>L. pneumophila</u>	tap	4-6	25	?	52	99.9	ND	Muraca et al., 1987
<u>M. chelonae</u>	BDF	0.3	25	7.0	60	40	>>60	Carson et al., 1978
<u>M. chelonae</u>		0.7	25	7.0	60	99.95	46	
<u>M. chelonae</u>	BDF	1.0	?	7.0	60	96	ca. 80	Pelletier and Du Moulin 1987
<u>M. fortuitum</u>	BDF	0.15	?	7.0	60	0	>720	
<u>M. fortuitum</u>	BDF	1.0	?	7.0	30	99.4	ca. 28	
<u>M. intra-cellulare</u>	BDF	0.15	?	7.0	60	70	>>480	"

* C.t = product of disinfectant concentration (C) in mg/l and contact time (t) in minutes for 99% inactivation; ND = not done or no data; BDF = buffered demand-free; CDF = chlorine demand free; <7 µm = with <7 µm particles; >7 µm = with >7 µm particles; GAC = granular activated carbon; treated = conventionally treated surface water.

Free chlorine: effectiveness (II)

TABLE 3 Inactivation of Health-related Viruses and Protozoan Cysts in Water by Free Chlorine

Microbe	Water	Cl ₂ Residual (mg/l)	Temp. (°C)	pH	Time (min.)	Reduction (%)	Est. C:t	Reference
VIRUSES:								
Parvo- H-1	PBS*	0.2	20	7.0	6	99.9	0.53	Churn et al., 1984
Parvo- H-1	PBS*	0.2	10	7.0	11	99.9	0.85	Berman and Hoff, 1984
SAll, disp. *	BDF	ca. 0.5	5	10	1.1-1.65	99	0.63	Raphael et al., 1987
" , cell-ass. *	BDF	ca. 0.5	5	10	2.4-4.4	99	1.8	Harekash & Butler, 1984
Rota, Wa	treated	0.75	22	8.3-8.6	60	94.3	ND	Vaughn et al., 1986
Human rota-	effluent	1.1	15	7.2	15	40	>>15	Grabow et al., 1983
Human rota-	effluent	2.2	15	7.2	10	60	>>15	Sobsey et al., 1988
Rota, SAll	BDF	0.1	4	8.0	0.5	99.9	0.03	"
Rota, Wa	BDF	0.1	4	8.0	0.65	99.9	0.03	"
Rota, SAll	BDF	0.4-0.28	25	10.0	1.1	99.99	ca. 4.0	"
Hepatitis A	BDF	0.42-0.06	25	6	0.7	99.99	ca. 3.0	"
Hepatitis A	BDF	0.4-0.28	25	10	2.5	99.99	ca. 5.5	"
Hepatitis A	BDF	0.5	5	6.0	6.5	99.99	ca. 1.8	"
Hepatitis A	BDF	0.5	5	10.0	49.6	99.99	ca. 12.3	"
Coliphage MS2	BDF	0.5	5	6.0	1.2	99.99	ca. 0.25	"
Coliphage MS2	BDF	0.5	5	10.0	26.5	99.99	ca. 6.9	"
PROTOZOAN CYSTS:								
<u>G. lamblia</u>	BDF	1-4	5	7.0	3-32	90	90-170	Jarroll et al., 1981
<u>G. lamblia</u>	BDF	1.5	25	6.0-8.0	47	99	<15	Rice et al., 1982
<u>G. lamblia</u>	BDF	2.5	5	6.0-8.0	19-26	90	ca. 120	Hibler et al., 1987
<u>G. lamblia</u>	BDF	0.2-3.0	5.0	6.0	---	99	54-87	"
<u>G. lamblia</u>	BDF	0.2-3.0	5.0	7.0	---	99	83-133	"
<u>G. lamblia</u>	BDF	0.2-3.0	5.0	8.0	---	99	119-192	"
<u>G. muris</u>	BDF	2.5	5	6.0	30	90	>150	Rice et al., 1982
<u>G. muris</u>	BDF	2.5	5	8.0	48	90	>150	"
<u>G. muris</u>	BDF	23.8-78.5	5	7.0	5.7-42.6	99	449-1012	Leahy et al., 1987
<u>G. muris</u>	BDF	2.8-7.1	25	7.0	3.6-16	99	25.5-44.8	"
<u>G. muris</u>	BDF	11.1	25	9.0	15.6	99	177	"
<u>G. muris</u>	BDF	4.4	25	5.0	16.3	99	71	"
<u>Naegleria</u> (2 species)		0.5-1.0	25	7.3-7.4	1-3 hr?	99.99	12-18	de Jonckheere and van de Voorde, 1976
<u>Acanthamoeba</u> (2 species)		4-8	25	7.3-7.4	24 hr.	99.99	960-7200	"
<u>N. gruberi</u>	BDF	2.64	25	5.0	2.8	99	7.3	Rubin et al., 1983
<u>N. gruberi</u>	BDF	2.2	25	7.0	5.2	99	11.4	"
<u>N. gruberi</u>	BDF	15.4	25	9.0	15.4	99	177	"
Crypto-sporidium	gut homog. in PBS	30,000	4	ca. 7.0	18 hr.	<95	>>18 hrs.	Campbell et al., 1982

* PBS = phosphate buffered saline; for other abbreviations see footnote to Table 2.

Free chlorine: advantages and disadvantages

- Advantages
 - Effective against (almost) all types of microbes
 - Relatively simple maintenance and operation
 - Inexpensive
- Disadvantages
 - Corrosive
 - High toxicity
 - High chemical hazard
 - Highly sensitive to inorganic and organic loads
 - Formation of harmful disinfection by-products (DBP's)

Free chlorine: other applications

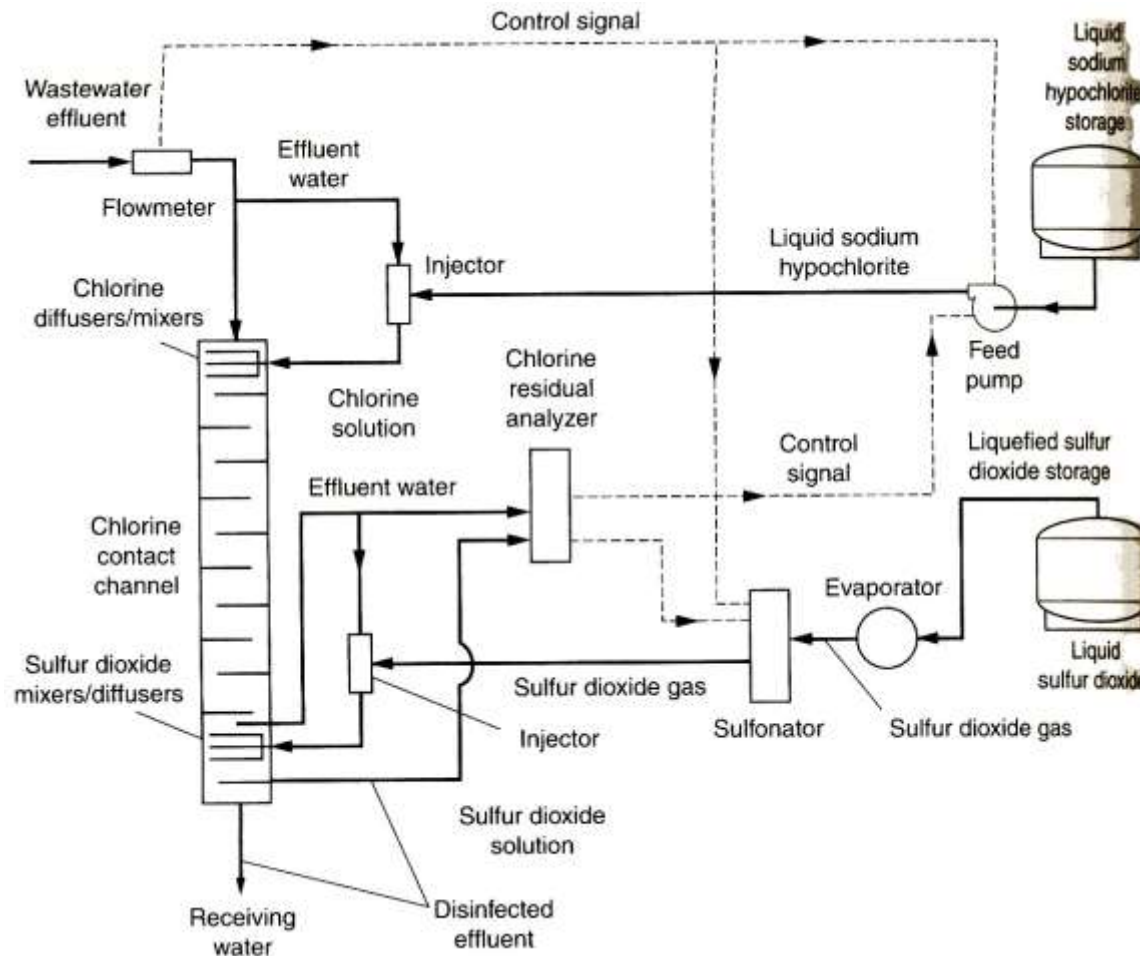
- Swimming pool/spa/hot tube water disinfection
- Industrial water disinfection (canning, freezing, poultry dressing, and fish processing)
- (Liquid and solid chlorine)
 - General surface disinfectant
 - Medical/household/food production

Questions?

Chloramines: Chemistry

- Two different methods of application (generation)
 - chloramination with pre-formed chloramines
 - mix hypochlorite and ammonium chloride (NH_4Cl) solution at $\text{Cl}_2 : \text{N}$ ratio at 4:1 by weight, 10:1 on a molar ratio at pH 7-9
 - dynamic chloramination
 - Reaction of free chlorine and ammonia *in situ*
- Chloramine formation
 - $\text{HOCl} + \text{NH}_3 \rightleftharpoons \text{NH}_2\text{Cl}$ (monochloramine) + H_2O
 - $\text{NH}_2\text{Cl} + \text{HOCl} \rightleftharpoons \text{NHCl}_2$ (dichloramine) + H_2O
 - $\text{NHCl}_2 + \text{HOCl} \rightleftharpoons \text{NCl}_3$ (trichloramine) + H_2O
 - $\frac{1}{2} \text{NHCl}_2 + \frac{1}{2} \text{H}_2\text{O} \rightleftharpoons \frac{1}{2} \text{NOH} + \text{H}^+ + \text{Cl}^-$
 - $\frac{1}{2} \text{NHCl}_2 + \frac{1}{2} \text{NOH} \rightleftharpoons \frac{1}{2} \text{N}_2 + \frac{1}{2} \text{HOCl} + \frac{1}{2} \text{H}^+ + \frac{1}{2} \text{Cl}^-$

Application of chloramines (preformed monochloramines): flow diagram



Chloramines: effectiveness

TABLE 4 Inactivation of Health-related Microbes in Water by Chloramines

Microbe	Water	Residual (mg/l)	Temp. (°C)	pH	Time (min.)	Reduction (%)	Est. C-t	Reference
BACTERIA:								
<u>E. coli</u>	BDF *	1.9-2.2	5	9.0	51-59	99	113	Scarpino, 1984
<u>E. coli</u>	BDF	1.0	22	6.0	5-8	99	ND *	Ward et al., 1984
<u>E. coli</u>	BDF	1.0	22	8.0	29-46	99	ND	"
<u>K. pneumoniae</u>	BDF	1.0	22	8.0	8.5-22	99	ND	"
<u>S. typhimurium</u>	BDF	1.0	22	8.0	12-14	99	ND	"
HPC	treated *	1.6	24	8.0	15	95 (35-99.73)	ND	Wolfe et al., 1985
HPC	treated	1.6	24	8.0	30	99.68 (88.4-→99.99)	ND	
Coliforms, <u>S. typhimurium</u> & <u>S. sonnei</u>	tap + 1% sewage	0.4-1.5	20	6.0	ND	90	8.5	Snead et al., 1980
"	"	"	"	8.0	ND	90	40	Engelbrecht et al., 1977
<u>M. fortuitum</u>	BDF	3.25	20	7.0	50	90	2667	Pelletier and Du Moulin, 1988
<u>M. fortuitum</u>	BDF	3.0	17	7.0	ca. 116	90	ND *	
<u>M. avium</u>	BDF	3.0	17	7.0	1320	90	ND	
<u>M. Intra-cellulare</u>	BDF	3.0	17	7.0	660	90	ND	
VIRUSES:								
Polio 1	BDF	5-22	5	9.0	170	99	1420	Scarpino, 1984
Polio 1	1 ⁰ effl.	1-10	25	7.5	60-308	99	ca. 345	Fujioka et al., 1983
Polio 1	tap	0.5	24	7.0-8.2	15	50	ND	Keswick et al., 1985
Polio 1	+ 10% effl.	0.5	24	"	15	0	ND	
Polio 1	+ 10% effl.	2.0	24	"	15	99.7	ND	
Hepatitis A	BDF	10	5	8.0	117	99.99	ca. 592	Sobsey et al., 1988
Coliphage MS2	BDF	10	5	8.0	>>60		ca. 2100	
Rotavirus SA11 dispersed	BDF	10	5	8.0	366-402	99	4034	Berman and Hoff, 1984
cell-assoc.	BDF	10	5	8.0	570-636	99	6124	
PROTOZOAN CYSTS:								
<u>G. muris</u>	BDF	1.5-2.6	3	6.5-7.5	188-296	99	430-580	Meyer, 1982
<u>G. muris</u>	BDF	6.35	5	7.0	220	99	ca. 1400	Rubin, 1988
<u>G. muris</u>	BDF	1.5-30	15	9.0	---	99	ca. 600	Rubin, 1988
<u>G. muris</u>	BDF	"	"	6.0-7.0	---	99	ca. 1000	Rubin, 1988

Chloramines: advantages and disadvantages

- Advantages
 - Less corrosive
 - Low toxicity and chemical hazards
 - Relatively tolerable to inorganic and organic loads
 - No known formation of DBP
 - Relatively long-lasting residuals
- Disadvantages
 - Not so effective against viruses, protozoan cysts, and bacterial spores

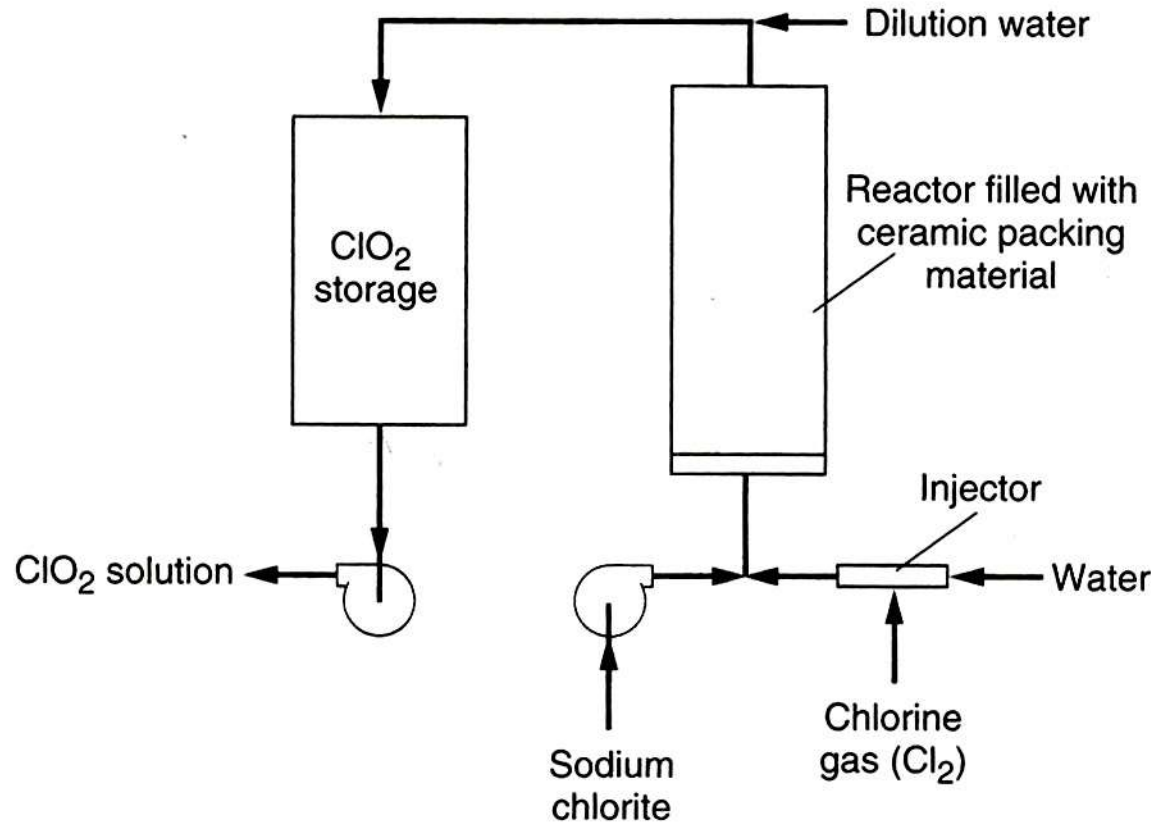
Chloramines: other applications (organic chloramines)

- Antiseptics
- Surface disinfectants
 - Hospital/household/food preparation
- Laundry and machine dishwashing liquids

Chlorine dioxide: Chemistry

- The method of generation
 - On-site generation by reaction of chlorine (either gas or liquid) with sodium chlorite
- Formation of chlorine dioxide
 - $2 \text{NaClO}_2 + \text{Cl}_2 \rightarrow 2 \text{ClO}_2 + 2 \text{NaCl}$
 - Highly soluble in water
 - Strong oxidant: high oxidative potentials
 - 2.63 times greater than free chlorine, but only 20 % available at neutral pH
 - » $\text{ClO}_2 + 5\text{e}^- + 4\text{H}^+ = \text{Cl}^- + 2\text{H}_2\text{O}$ (5 electron process)
 - » $2\text{ClO}_2 + 2\text{OH}^- = \text{H}_2\text{O} + \text{ClO}_3^- + \text{ClO}_2^-$ (1 electron process)

Generation of chlorine dioxide



Application of chlorine dioxide: flow diagram

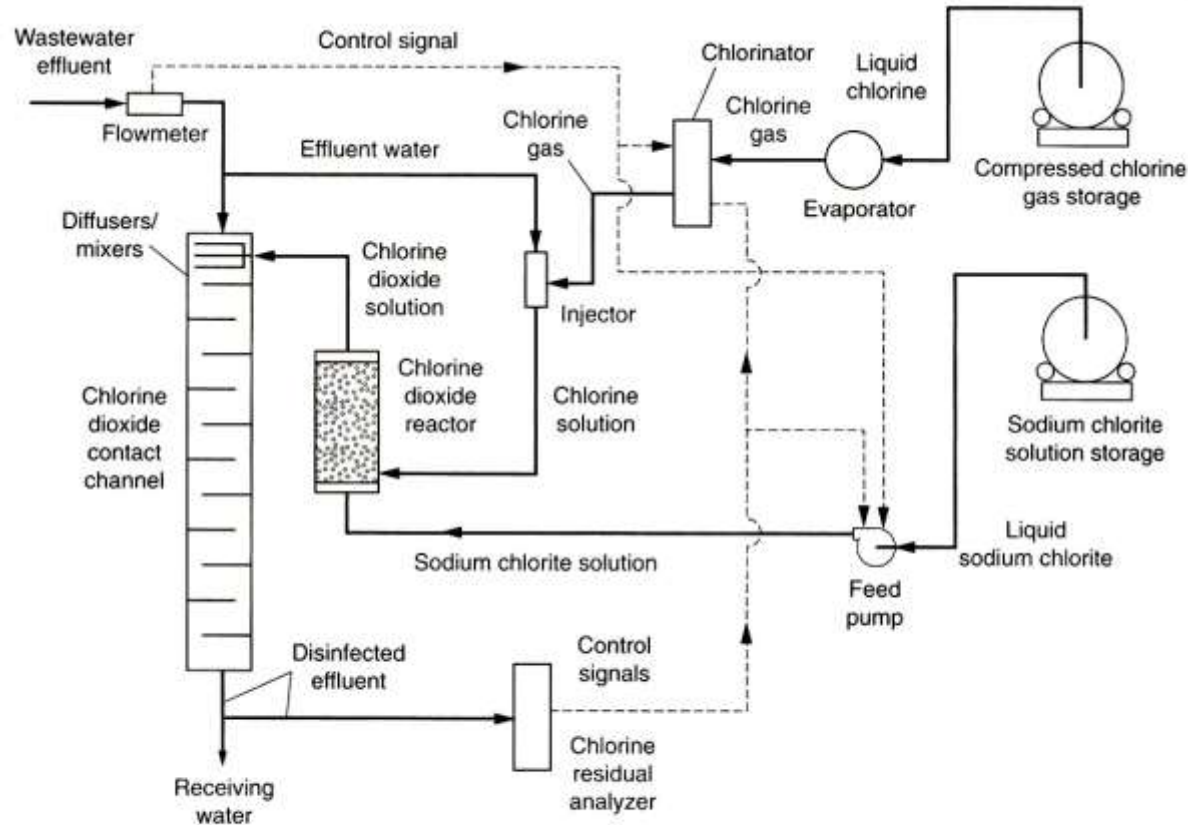


Figure 12-20

Typical flow diagram for the addition of chlorine dioxide.

Chlorine dioxide: effectiveness

TABLE 6 Inactivation of Health-Related Microbes in Water by Chlorine Dioxide

Microbe	Water	ClO ₂ Residual (mg/l)	Temp. (°C)	pH	Time (min.)	Reduction (%)	C·t	Reference
BACTERIA:								
<i>E. coli</i>	BDF*	0.3-0.8	5	7.0	0.6-1.8	99	0.48	Cronier, 1977
<i>E. coli</i>	PBS	0.4-0.8	?	7.0	5	>99.999	ND	Fujioka et al., 1986
<i>Fec. colif.</i>	effl.	1.9	?	?	10	99.94	ND	"
<i>Fecal strep.</i>	effl.	1.9	?	?	10	99.5	ND	"
<i>C. perfringens</i>	effl.	1.9	?	?	10	0	ND	"
<i>E. coli</i>	BDF	0.3-0.5	5	?	15	99.9-99.999	ND	Berg et al., 1982; 1985
<i>L. pneumophila</i>	BDF	0.5-0.35	23	?	15	99.9-99.99	ND	Harakeh et al., 1985
<i>K. pneumonia</i>	BDF	ca. 0.12	23	?	15	99.3-99.7	ND	
VIRUSES:								
Coliphage f2	BDF	1.5	5	7.2	2	99.994	ND	Olivieri et al., 1985b
Polio 1	BDF	2.0	5	7.2	5-10	99-99.96	ND	Scarpino et al., 1979
Polio 1	BDF	0.4-14.3	5	7.0	0.2-11.2	99	0.2-6.7	Fujioka et al., 1986
Polio 1	PBS	0.4-0.8	?	7.0	5	99.9-99.97	ND	
Polio 1	effl.	1.9	?	?	10	99.4	ND	
Rota SAll:								
dispersed	BDF	0.5-1.0	5	6.0	0.2-0.6	99	0.2-0.3	Berman and Hoff, 1984
cell-assoc.	BDF	0.45-1.0	5	6.0	1.2-4.8	99	1.0-2.1	"
cell-assoc.	BDF	0.46-0.52	5	10.0	0.3-0.4	99	0.16-0.2	Battigelli & Sobsey, 1987
Hepatitis A	BDF	0.14-0.23	5	6.0	8.4	99	1.7	
Hepatitis A	BDF	0.2	5	9.0	<0.33	>99.9	<0.04	
Coliphage MS2	BDF	0.15	5	6.0	34	99	5.1	in preparation
Coliphage MS2	BDF	0.15	5		<0.33	>99.95	<0.03	
PROTOZOAN CYSTS:								
<i>N. gruberi</i>	BDF	0.8-1.95	5	7.0	7.8-19.9	99	15.5	Chen et al., 1985
<i>N. gruberi</i>	BDF	0.46-1.0	25	5.0	5.4-13.2	99	6.35	"
<i>N. gruberi</i>	BDF	0.35-1.26	25	7.0	4.0-14.2	99	5.51	"
<i>N. gruberi</i>	BDF	0.42-1.1	25	9.0	2.5-6.7	99	2.91	"
<i>G. muris</i>	BDF	0.1-5.55	5	7.0	1.3-168	99	10.7	Leahy, 1985;
<i>G. muris</i>	BDF	0.26-1.2	25	5.0	4.0-24	99	5.8	1987;
<i>G. muris</i>	BDF	0.21-1.12	25	7.0	3.3-28.8	99	5.1	Rubin, 1988
<i>G. muris</i>	BDF	0.15-0.81	25	9.0	2.1-19.2	99	2.7	"

* BDF = buffered demand free; PBS = phosphate buffered saline

Chlorine dioxide: advantages and disadvantages

- Advantages
 - Very effective against all type of microbes
- Disadvantages
 - Unstable (must be produced on-site)
 - High toxicity
 - $2\text{ClO}_2 + 2\text{OH}^- = \text{H}_2\text{O} + \text{ClO}_3^-$ (Chlorate) + ClO_2^- (Chlorite): in alkaline pH
 - High chemical hazards
 - Highly sensitive to inorganic and organic loads
 - Formation of harmful disinfection by-products (DBP's)
 - Expensive

Chlorine dioxide: other applications

- Hospital/household surface disinfectant
 - ‘stabilized’ chlorine dioxide and ‘activator’
- Industrial application
 - bleaching agent: pulp and paper industry, and food industry (flour, fats and fatty oils)
 - deordoring agent: mildew, carpets, spoiled food, animal and human excretion
- Gaseous sterilization

Ozone: Chemistry

- The method of generation
 - generated on-site
 - generated by passing dry air (or oxygen) through high voltage electrodes (ozone generator)
 - bubbled into the water to be treated.
- Ozone
 - colorless gas
 - relatively unstable
 - highly reactive
 - reacts with itself and with OH^- in water

Generation of ozone

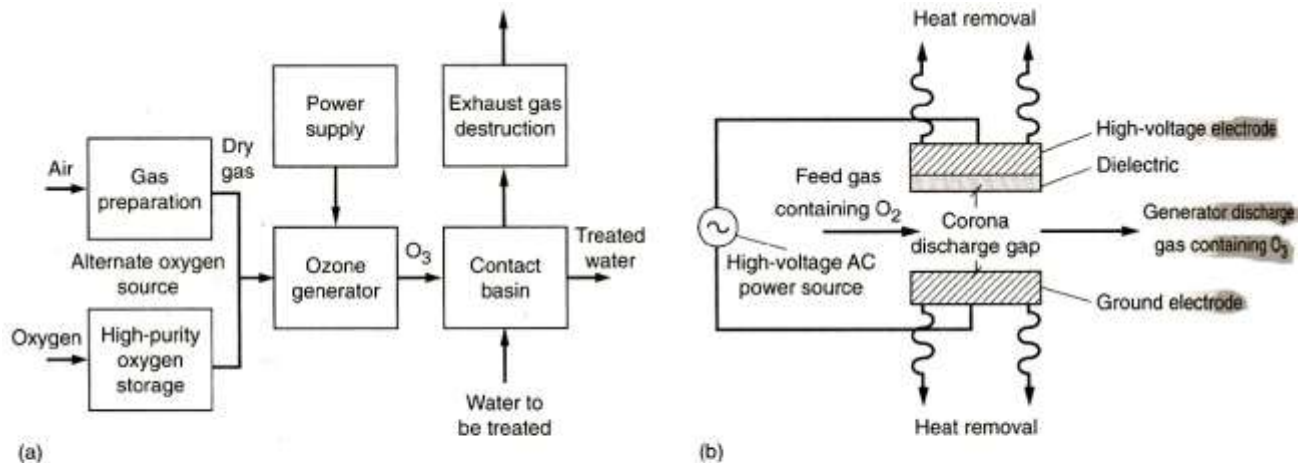


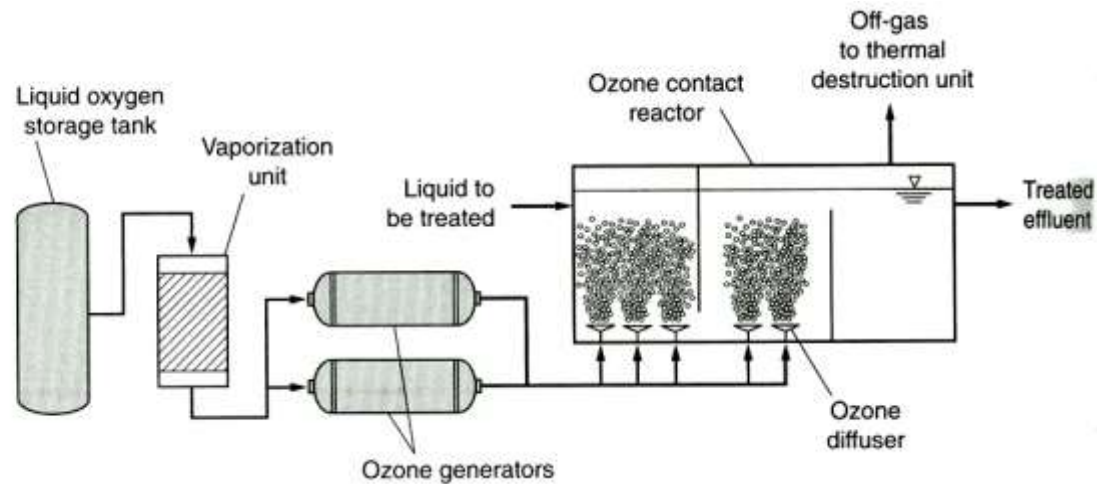
Figure 12-29

Elements of ozone disinfection system: (a) schematic flow diagram for complete ozone disinfection system and (b) schematic detail of the generation of ozone. [Adapted from U.S. EPA, 1986.]

Application of ozone: flow diagram

Figure 12-30

Typical flow diagram for the application of ozone for disinfection.



Ozone: reactivity

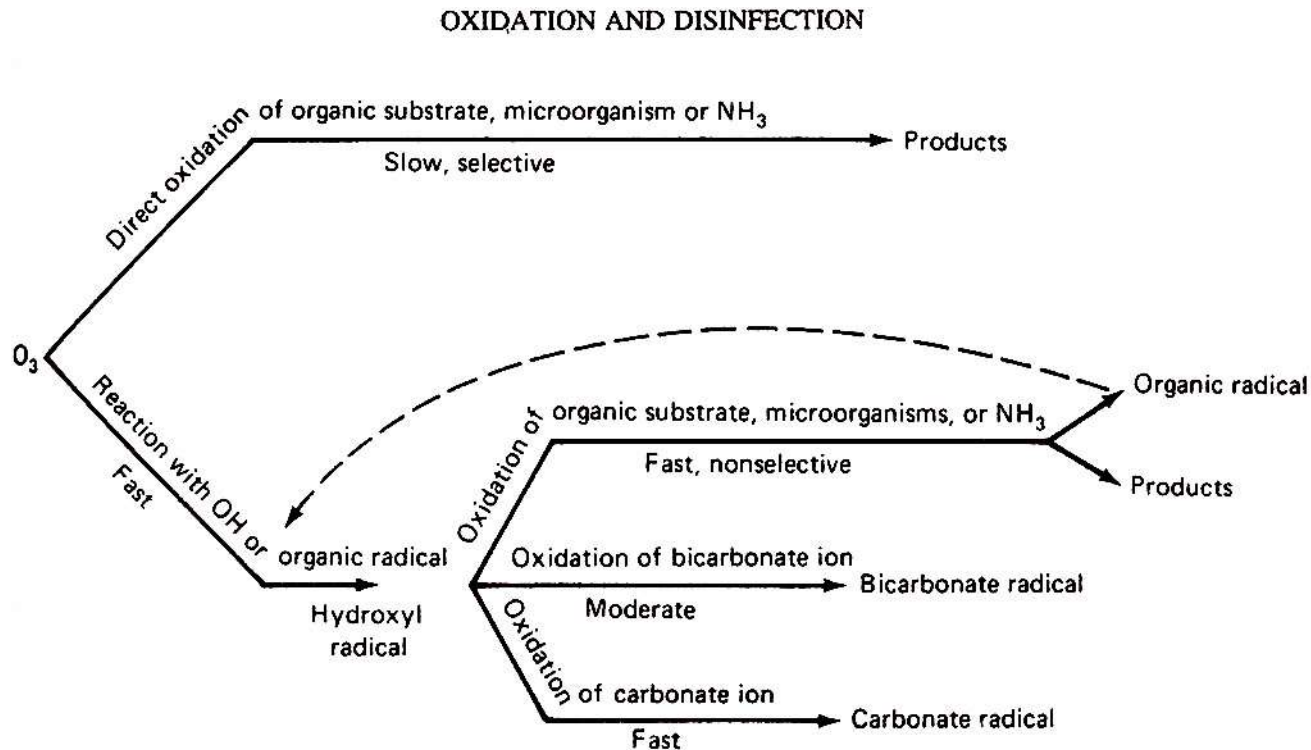


FIGURE 10.19 Reaction pathways of ozone in water. (Source: J. Hoigne and H. Bader, "Role of Hydroxyl Radical Reactions in Ozonation Processes in Aqueous Solutions," *Water Resources Bulletin*, vol. 10, 1976, p. 377.)

Ozone: effectiveness

TABLE 5 Inactivation of Health-related Microbes in Water by Ozone

Microbe	Water	O ₃ Residual mg/l	Temp. (°C)	pH	Time (min.)	Reduction (%)	Est. C·t	Reference
<u>E. coli</u>	?	0.04-0.07	1	7.2	0.08-0.5	99	0.006-0.02	Katzenelson et al., 1974
Tot. colif.	effl.	5.0	12.6	7.1	18	99.89	ND	Netzer et
Fecal strep.	effl.	5.0	12.6	7.1	18	99	ND	al., 1979
<u>E. coli</u>	effl.	0.29-0.36	24	7.4	0.6	99.94	ND	Farooq and
<u>S. typhi-</u> <u>murium</u>	effl.	"	"	"	"	99.9998	ND	Akhlaque, 1983
<u>M. fortuitum</u>	effl.	0.29-0.36	24	7.4	0.6	89	ND	"
<u>M. fortuitum</u>	BDF	0.8-1.08	24	7.0	0.58	99	0.53	Farooq et al., 1977
<u>VIRUSES:</u>								
Polio 1	BDF	0.15-0.2	5	7.2	0.4-1.5	99	0.2	Roy et al.,
Polio 2	BDF	0.15	25	7.2	4.83	99	0.72	1982
Polio 1	effl.	0.29-0.36	24	7.4	0.6	99.5	ND	Farooq and
								Akhlaque, 1983
Polio 1 effl. + HCO ₃		0.2	20	7.0	10-15	97		Harakeh and
Polio 1 effl.		0.2	20	7.0	10-15	80		Butler, 1985
Rota SA11	BDF	0.1-0.3	4	6.0-8.0	0.12-0.19	99	0.019-0.064	Vaughn et
Human rota	BDF	0.05-0.3	4	6.0-9.0	0.0.12-0.19		0.006-0.036	al., 1987
Rota SA11	effl.	0.26	15	7.2	15	96	ND	Harakeh and
Human rota	effl.	0.26	15	7.2	15	40	ND	Butler,
Human rota	effl.	0.4	15	7.2	15	99.99	ND	1985
<u>PROTOZOAN CYSTS:</u>								
<u>G. muris</u>	BDF	0.15-0.7	5	7.0	2.8-12.9	99	1.94	Wickramanayake
<u>G. lamblia</u> *	BDF	0.11-0.48	5	7.0	0.94-5.0	99	0.53	et al., 1984;
<u>N. gruberi</u>	BDF	0.55-2.0	5	7.0	2.1-7.8	99	4.23	1985
<u>N. (6 sp.)</u> *	DW	0.4	25	7.0	4	>98.9	ND	Langlais and
<u>A. polyphaga</u> *	DW	0.4	25	7.0	4	95	ND	Perrine,
<u>A. (3 sp.)</u>	DW	0.4	25	7.0	4	>98.9	ND	1986

* BDW = buffered demand free water; DW = distilled water; G. = Giardia; N. = Naegleria;
A. = Acanthamoeba.

Ozone: advantages and disadvantages

- Advantages
 - Highly effective against all type of microbes
- Disadvantages
 - Unstable (must be produced on-site)
 - High toxicity
 - High chemical hazards
 - Highly sensitive to inorganic and organic loads
 - Formation of harmful disinfection by-products (DBP's)
 - Highly complicated maintenance and operation
 - Very expensive

Ozone: other applications

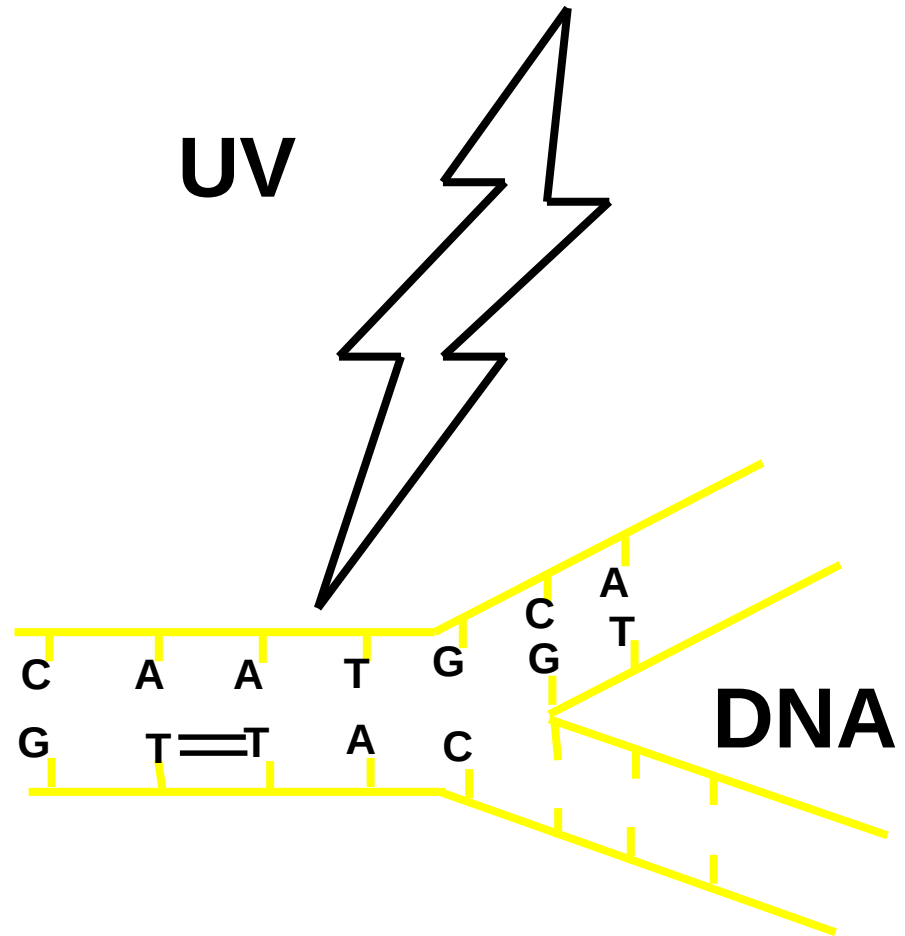
- Industrial applications
 - aquaria, fish disease labs, and aquaculture
 - cooling towers
 - pharmaceuticals and integrated circuit processing (ultra-pure water)
 - pulp and paper industry
- Gaseous sterilization
 - cleaning and disinfection of healthcare textiles

Questions?

Physical disinfectants

Ultraviolet irradiation: mechanism

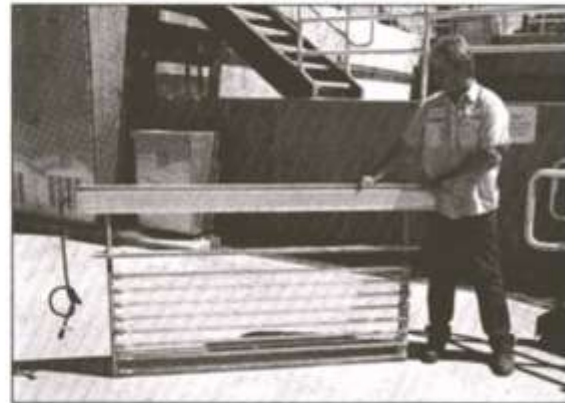
- Physical process
- Energy absorbed by DNA
 - pyrimidine dimers, strand breaks, other damages
 - inhibit replication



Low-pressure (LP) UV: wastewater



(a)

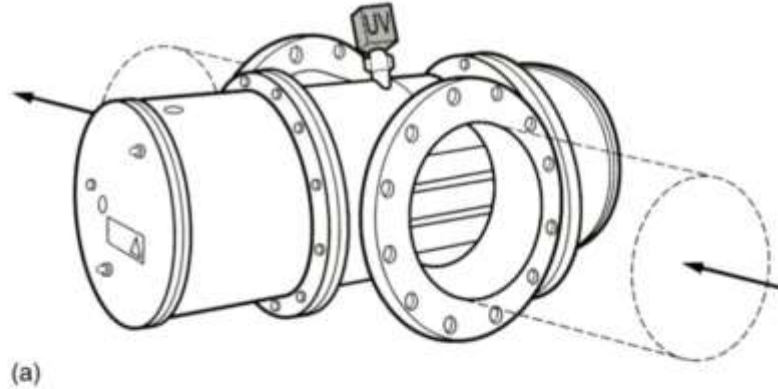


(b)

Medium-pressure (MP) UV: drinking water

Figure 12-36

Typical closed-channel UV disinfection system with lamps placed perpendicular to the flow:
(a) Isometric cutaway view of disinfection reactor and (b) view of typical installation.
(Courtesy Aquionics Inc.)



UV disinfection: effectiveness

TABLE 7 Inactivation of Health-Related Microbes in Water by Ultraviolet Radiation

Microbe	Water	UV Dose ₂ (mW-s/cm ²)	Temp. (°C)	pH	Reduction (%)	Reference
BACTERIA:						
<u>E. coli</u>	BDW*	6.5	room	?	99.9	Chang,
Total coliforms	effl.	8.2	room	?	99.9	et al.,
HPC	effl.	14	room	?	99.9	1985
<u>B. subtilis</u> spores	BDW	60	room	?	99.9	"
Total coliforms	effl.	6.5	?	?	99	Qualls et
Total coliforms	effl.	9.6-52	?	?	99.97	al., 1985
Total coliforms	filt. effl.	5.7	?	?	99	"
Total coliforms	tea	7.2-28	?	?	>99.95	Tobin et
HPC	tea	7.2-28	?	?	88-96	al., 1983
Fecal coliforms	sewage	ca. 275	?	?	99.9	Zukovs
Enterococci	sewage	ca. 275	?	?	99.9	et al.,
<u>P. aeruginosa</u>	sewage	ca. 280	?	?	99.9	1986
<u>E. coli</u> recA-, uvrA-	on medium	10.3	?	?	99.9	Knudson,
<u>E. coli</u> , B/r	"	0.1	?	?	99.9	1985
<u>Legionella</u> (6 sp.)	"	0.8-5.5	?	?	99.9	"
<u>L. pneumophila</u>	tap	30	25	?	99.9	Muraca et
						al., 1987
<u>L. pneumophila</u>	dist. water	1.8	room	?	99.9	Butler et
<u>Y. enterocolitica</u>	dist. water	2.7	room	?	99.9	al., 1987
<u>C. jejuni</u>	dist. water	5.0	room	?	99.9	"
VIRUSES:						
Polio 1	PBS*	21	room	7.5	99.9	Chang et
Rota SA11	PBS	25	room	7.5	99.9	al., 1985
Polio 1	BDW	29	?	7.2	99.9	Harris et
Reo 1	BDW	45	?	7.2	99.9	al., 1987
PROTOZOAN CYSTS:						
<u>Giardia lamblia</u>	BDW	63	?	?	70	Rice and
						Hoff, 1981
<u>A. castellanii</u>	BDW	100	room	?	99.9	Chang et al.,
						1985

* BDW = buffered distilled water; PBS = phosphate buffered saline

UV disinfection: advantages and disadvantages

- Advantages
 - Very effective against bacteria, fungi, protozoa
 - Independent on pH, temperature, and other materials in water
 - No known formation of DBP
- Disadvantages
 - Not so effective against viruses
 - No lasting residuals
 - Expensive

UV disinfection: other applications

- Disinfection of air
- Surface disinfectant
 - Hospital/food production
- Industrial application
 - Cooling tower (Legionella control)
 - Pharmaceuticals (disinfection of blood components and derivatives)

Disinfection Kinetics

Disinfection Kinetics

- Chick-Watson Law:

$$\ln N_t/N_0 = -kC^n t$$

where:

N_0 = initial number of organisms

N_t = number of organisms remaining at time = t

k = rate constant of inactivation

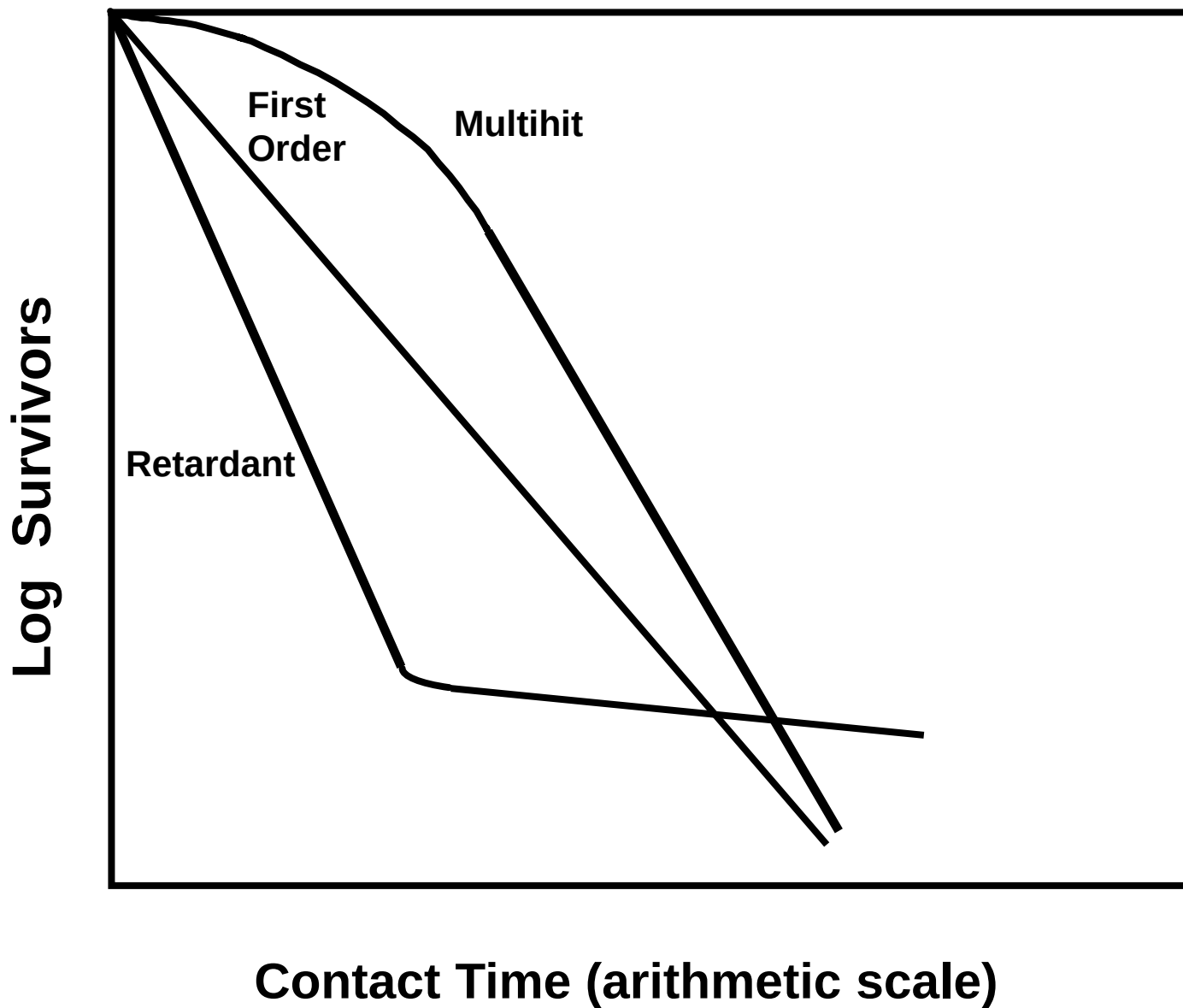
C = disinfectant concentration

n = coefficient of dilution

t = (exposure) time

- Assumptions
 - Constant disinfectant concentration
 - Homogenous microbe population: all microbes are identical
 - “Single-hit” inactivation: one hit is enough for inactivation
- When k , C , n are constant: first-order kinetics
- Decreased disinfectant concentration over time or heterogeneous population
 - “tailing-off” or concave down kinetics: initial fast rate that decreases over time
- Multi-hit inactivation
 - “shoulder” or concave up kinetics: initial slow rate that increase over time

Chick-Watson Law and deviations



CT Concept

- Based on Chick-Watson Law
- “Disinfection activity can be expressed as the product of disinfection concentration (C) and contact time (T)”
- The same CT values will achieve the same amount of inactivation

Disinfection Activity and the CT Concept

- Example: If $CT = 100$ mg/l-minutes, then
 - If $C = 1$ mg/l, then T must = 100 min. to get $CT = 100$ mg/l-min.
 - If $C = 10$ mg/l, T must = 10 min. in order to get $CT = 100$ mg/l-min.
 - If $C = 100$ mg/l, then T must = 1 min. to get $CT = 100$ mg/l-min.

C*t₉₉ Values for Some Health-related Microorganisms (5°C, pH 6-7)

Organism	Disinfectant			
	Free chlorine	Chloramines	Chlorine dioxide	Ozone
<i>E. coli</i>	0.03 – 0.05	95 - 180	0.4 – 0.75	0.03
Poliovirus	1.1 – 2.5	768 - 3740	0.2 – 6.7	0.1 – 0.2
Rotavirus	0.01 – 0.05	3806 - 6476	0.2 – 2.1	0.06-0.006
<i>G. lamblia</i>	47 - 150	2200	26	0.5 – 0.6
<i>C. parvum</i>	7200	7200	78	5 - 10

$I^*t_{99.99}$ Values for Some Health-Related Microorganisms

Organism	UV dose (mJ/cm²)	Reference
<i>E.coli</i>	8	Sommer et al, 1998
<i>V. cholera</i>	3	Wilson et al, 1992
Poliovirus	21	Meng and Gerba, 1996
Rotavirus-Wa	50	Snicer et al, 1998
Adenovirus 40	121	Meng and Gerba, 1996
<i>C. parvum</i>	< 3	Clancy et al, 1998
<i>G. lamblia</i>	< 1	Shin et al, 2001