

FUNDAMENTALS OF ENVIRONMENTAL ENGINEERING

CE 356

Water Treatment



Adopted from Dr.
Fernando Cadena

©1997



Learning Objectives

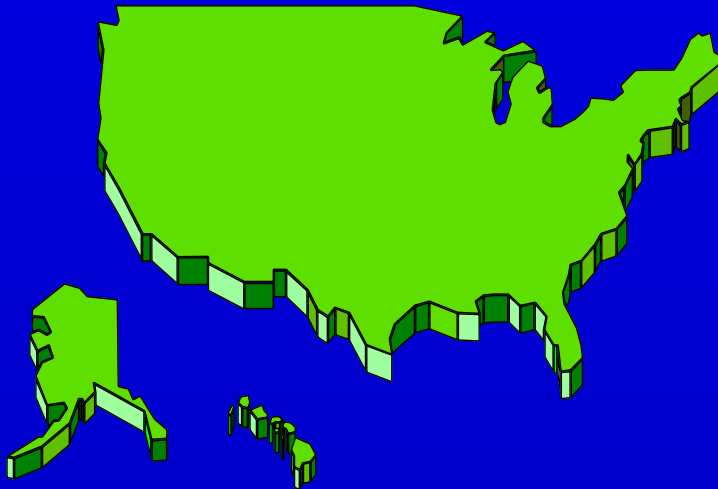
- 1 Differentiate between surface and ground water treatment requirements:
 - Ground water vs surface water quality
 - Treatment technologies
- 2 List in appropriate order and describe the four basic unit operations used for treating surface water:
 - Rapid mix/coagulation
 - Flocculation
 - Sedimentation
 - Rapid sand filtration



Sources of Water

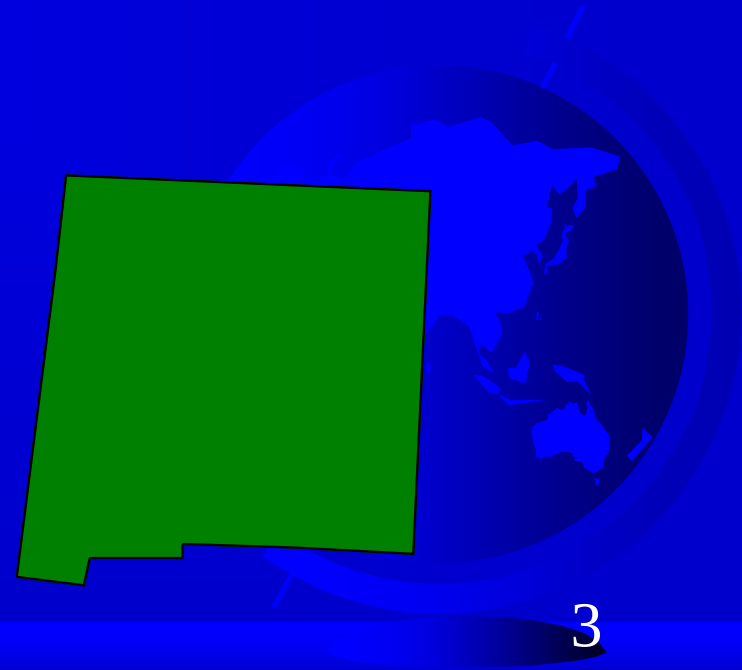
☞ National

- 50% surface (rivers or lakes)
- 50% ground water



☞ New Mexico

- 95% ground water



Groundwater Quality



- Dissolved minerals & chemicals
- No turbidity, few microorganisms: filtered out by soil
- Iron & manganese
- Other specific SDWA contaminants

Groundwater Treatment



- ➡ Pump
- ➡ Disinfection
- ➡ Fluoridation (if less than 1 ppm)
- ➡ Storage and Distribution



Surface Water Quality

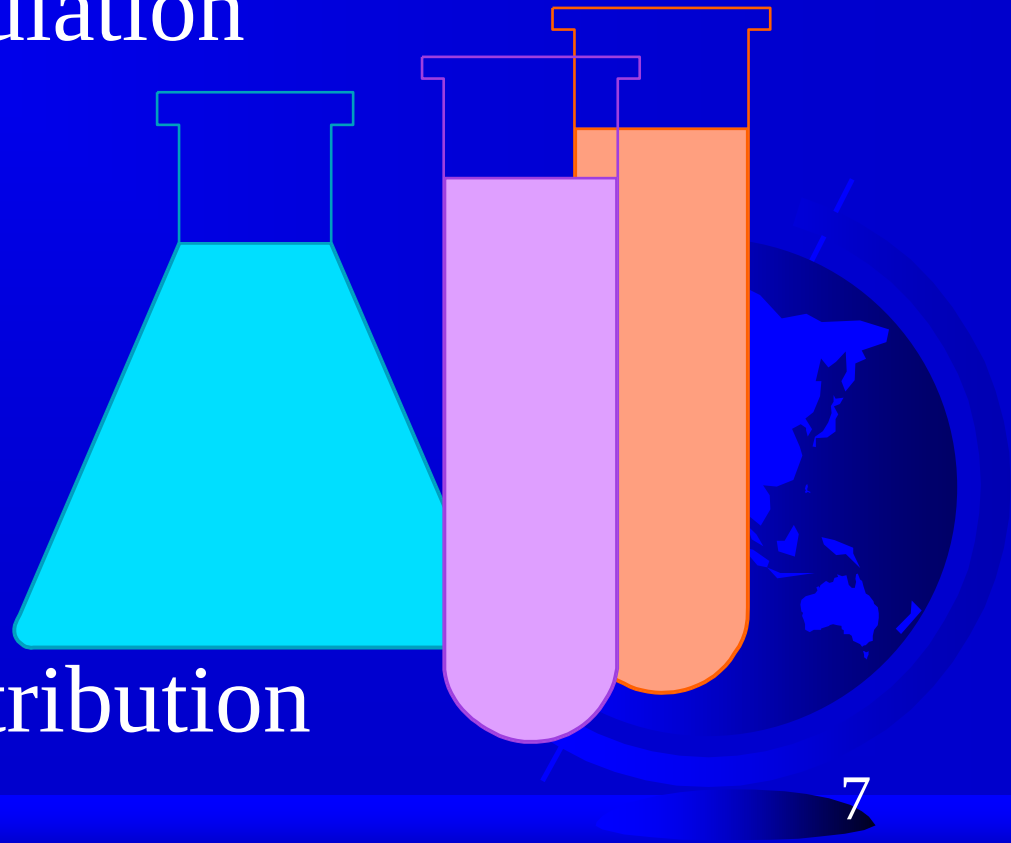
☞ Contaminants:

- Suspended solids, soil (turbidity)
- Pathogens (coliform indicator)
- Color (decaying vegetation, algae)
- Taste & odor
- Other SDWA contaminants

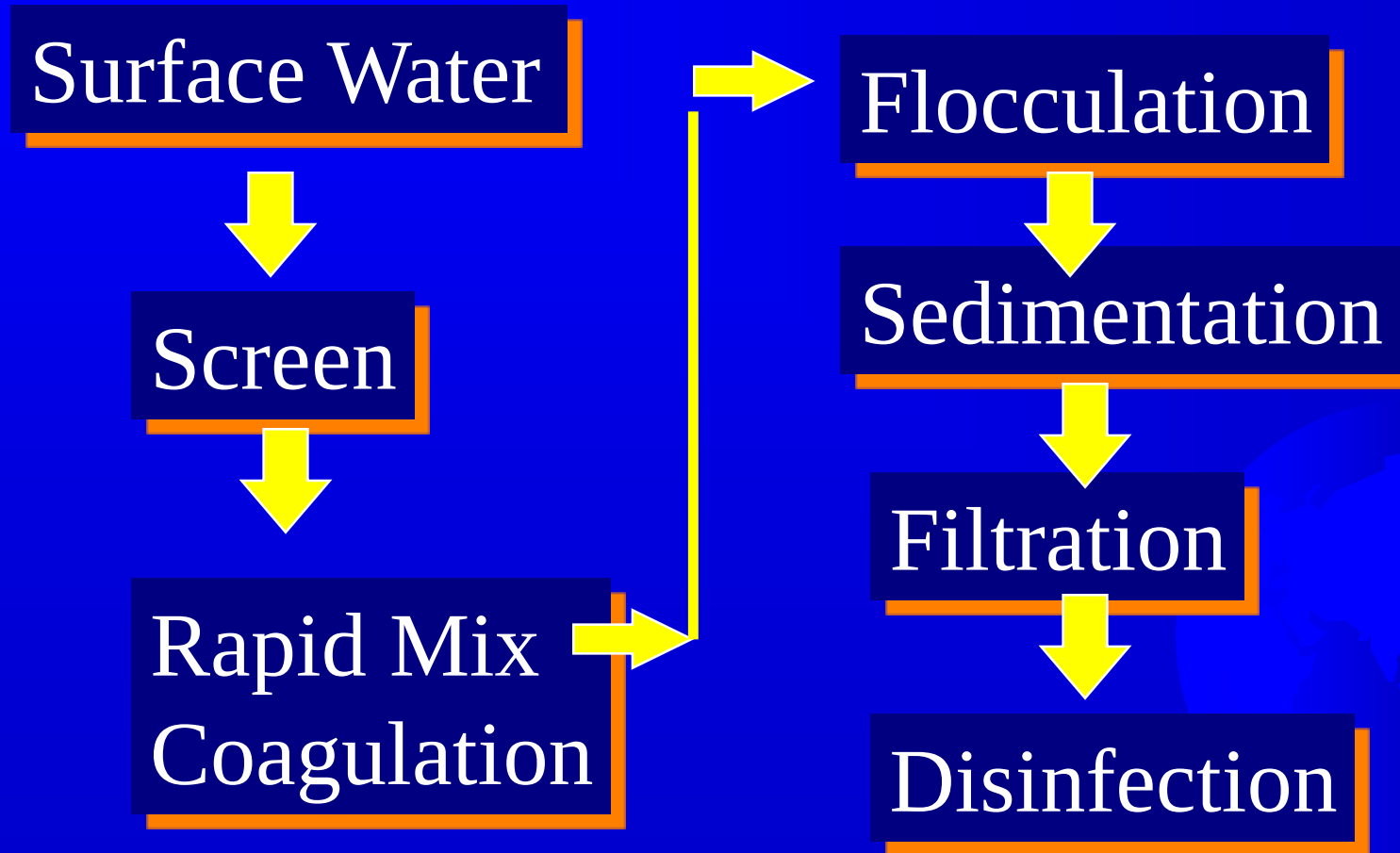


Surface Water Treatment

- ☞ Screening
- ☞ Rapid mix/coagulation
- ☞ Flocculation
- ☞ Sedimentation
- ☞ Sand filtration
- ☞ Disinfection
- ☞ Storage and Distribution

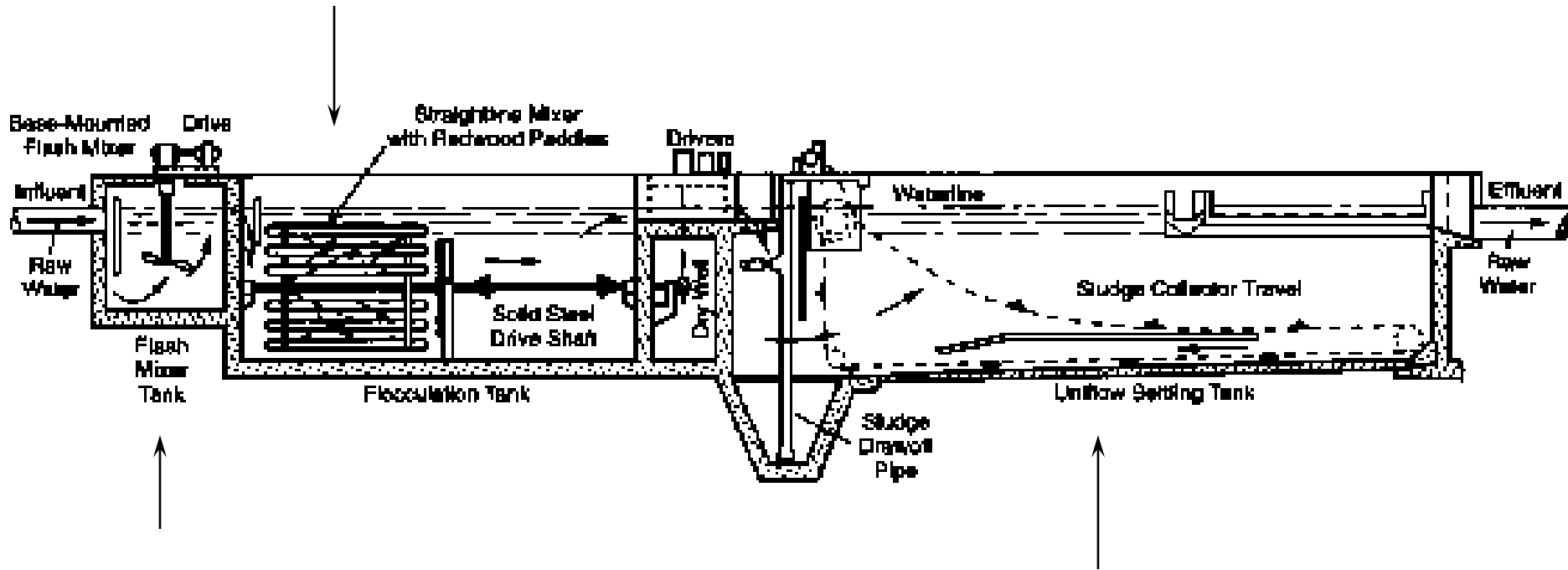


Conventional Treatment



Treatment Plant Layout

Flocculation

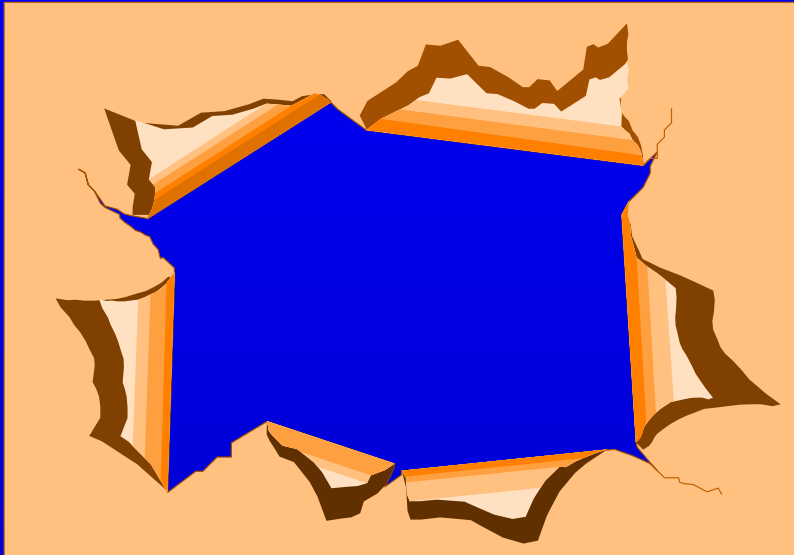


Rapid Mix
& Coagulation

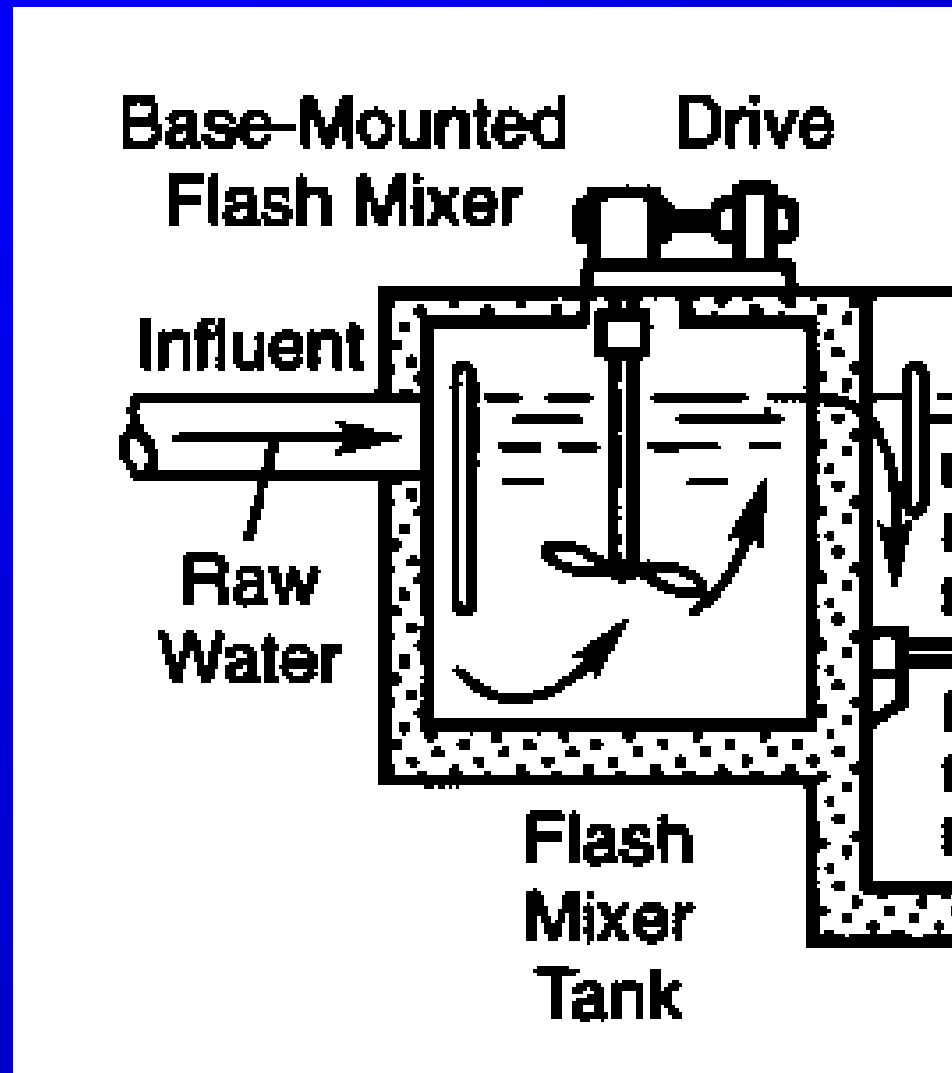
Sedimentation

Screening

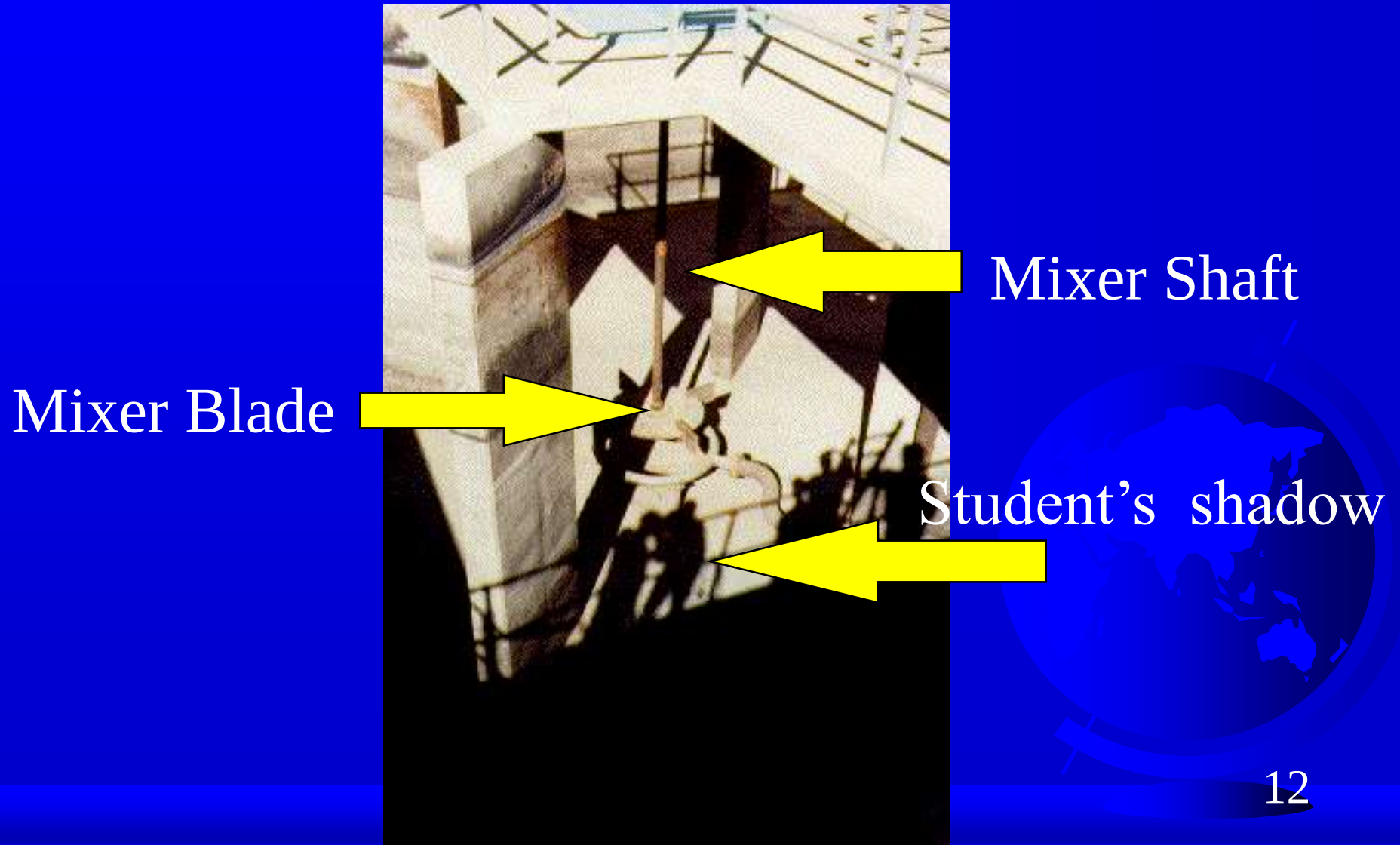
- ☞ Removes trash, leaves, fish, etc. at the water source



Rapid Mix



Rapid Mixer



Coagulation

- ☞ Chemicals added to remove small particles
- ☞ Particles repelled from each other by negative electric charge

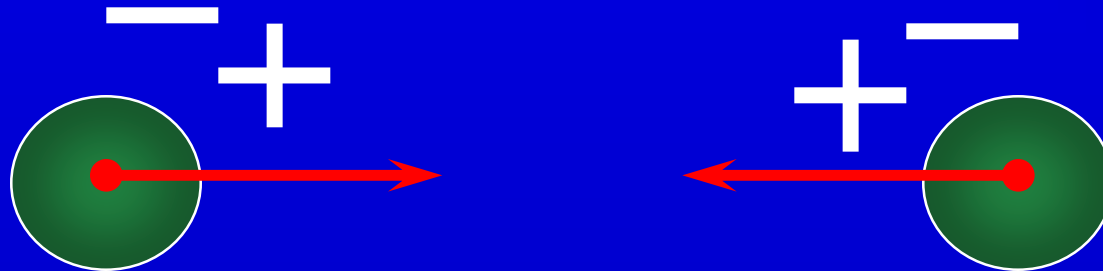


Coagulation

☞ Apply positive charges (ions) to colloids.

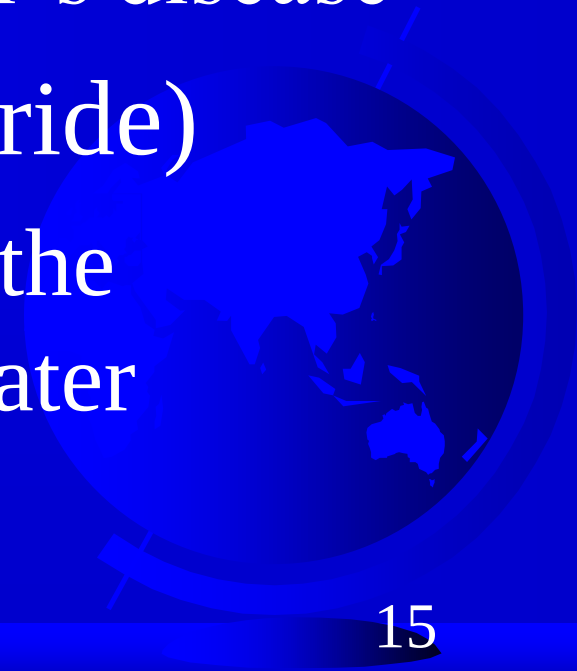
☞ Attractive gravitational forces

☞ Positive ions (cations): Fe^{3+} , Al^{3+}



Coagulants

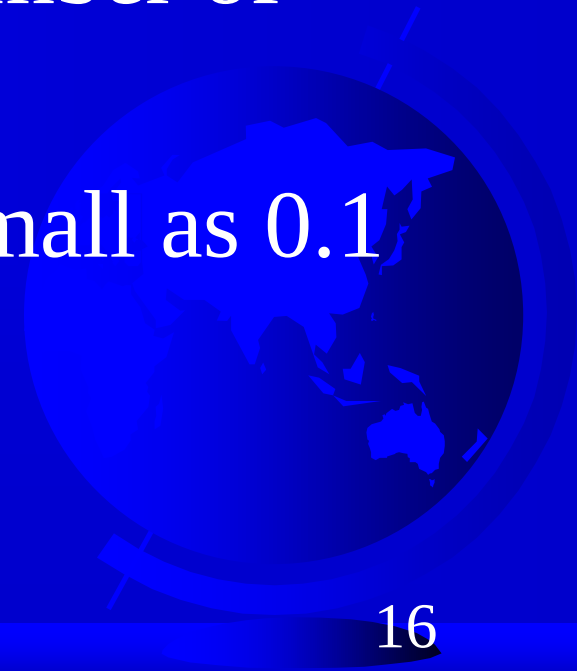
- ☞ Aluminum sulfate (alum)
 - Most commonly used
 - Concerns about Alzheimer's disease
- ☞ Ferric chloride (iron chloride)
 - Effectiveness depends on the chemistry of the source water



Coagulant Aids

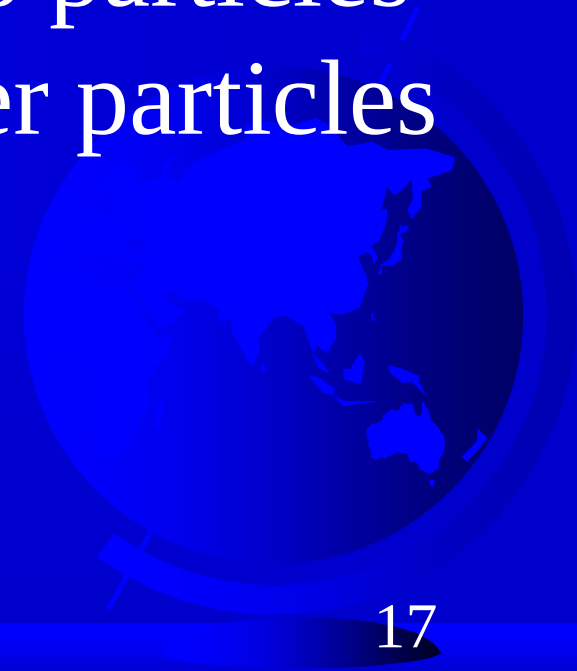
☞ Polymers

- “Liquid plastic”
- Highly branched, high number of “reactive” sites
- Very effective, doses as small as 0.1 mg/l

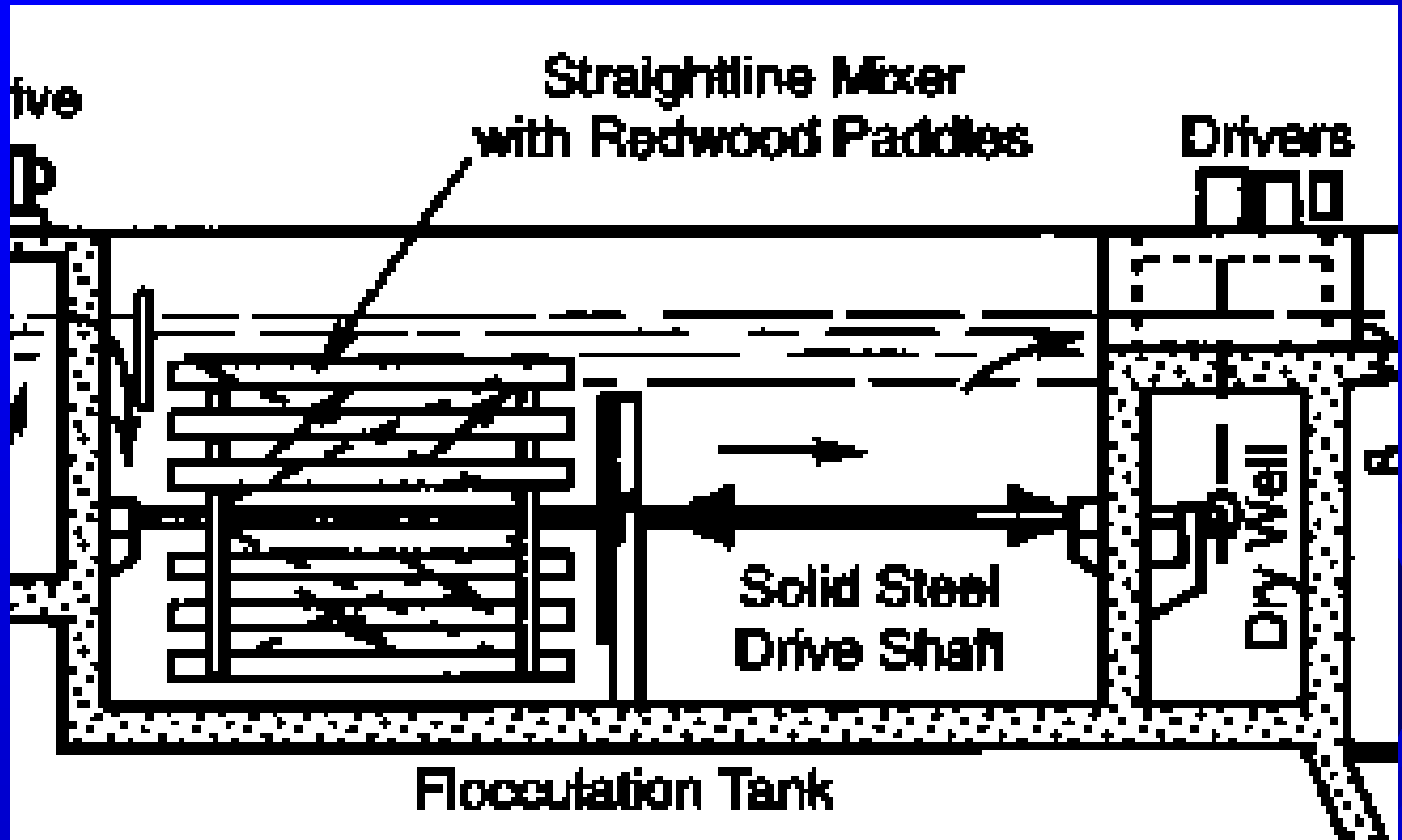


Flocculation

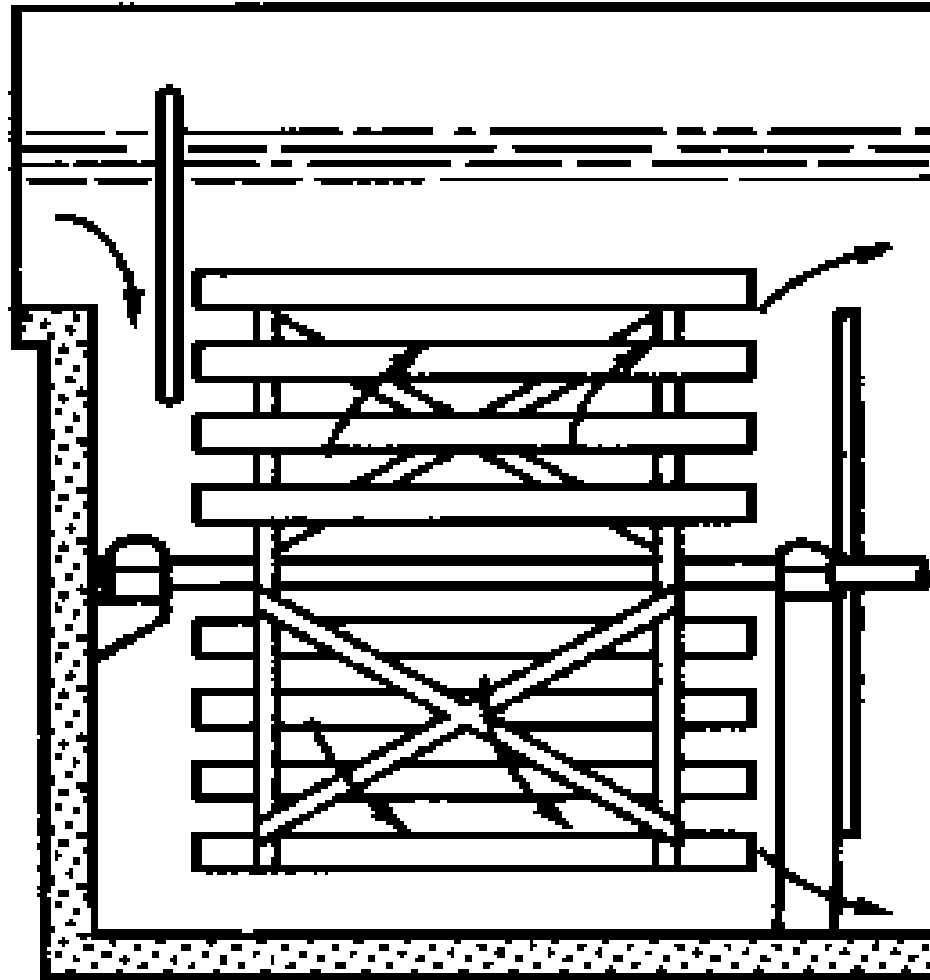
- ☞ Slow mixing encourages collision of particles
- ☞ Coagulant chemicals help particles stick together, make larger particles (flocs)



Flocculator



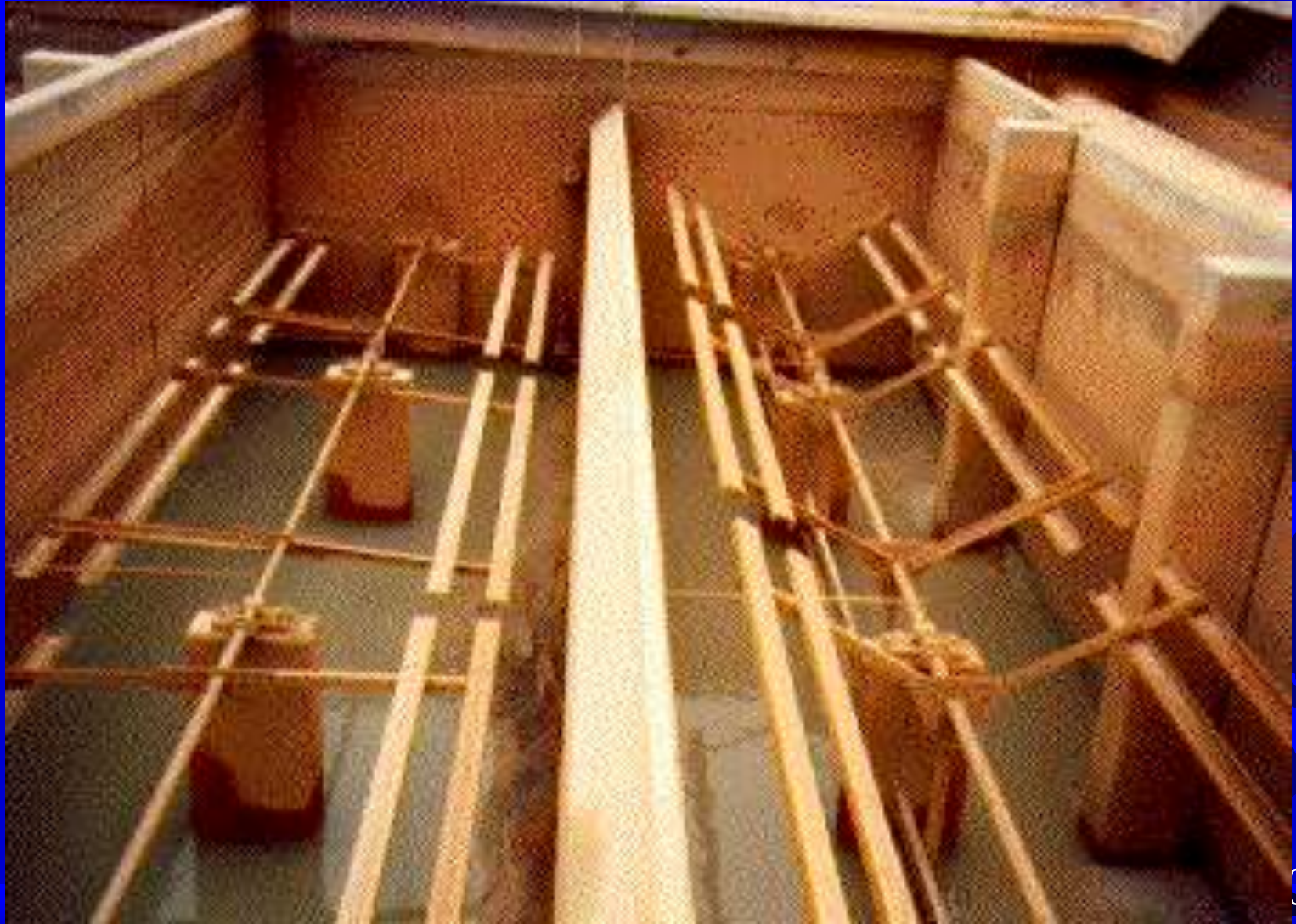
Flocculator



Enlarged View of Paddle Flocculator



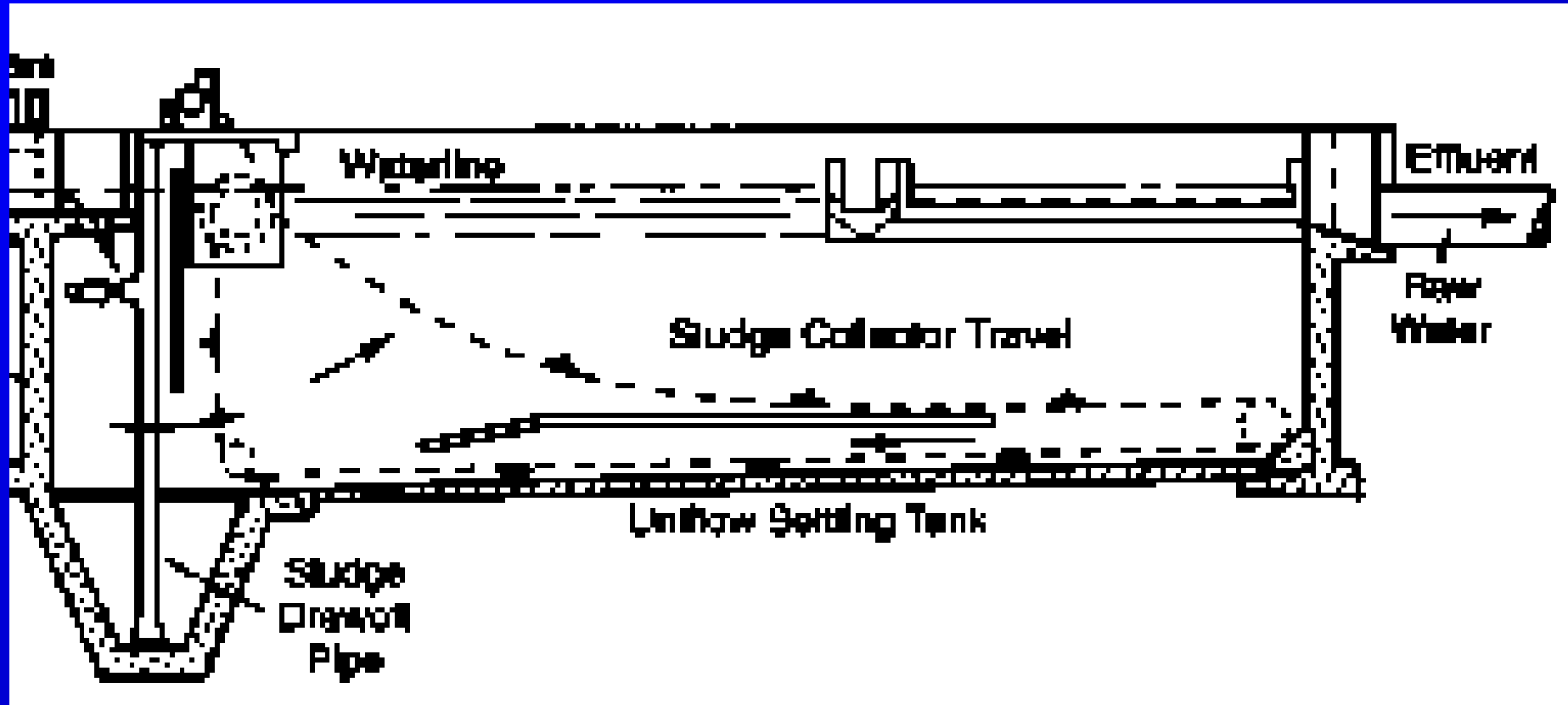
Flocculator Paddles



Sedimentation

- ☞ Floc particles allowed to settle in a large basin
- ☞ Force of gravity causes settling
 - Large, dense particles settle the fastest (dependent on flocculation)
- ☞ Solid material removed as a sludge

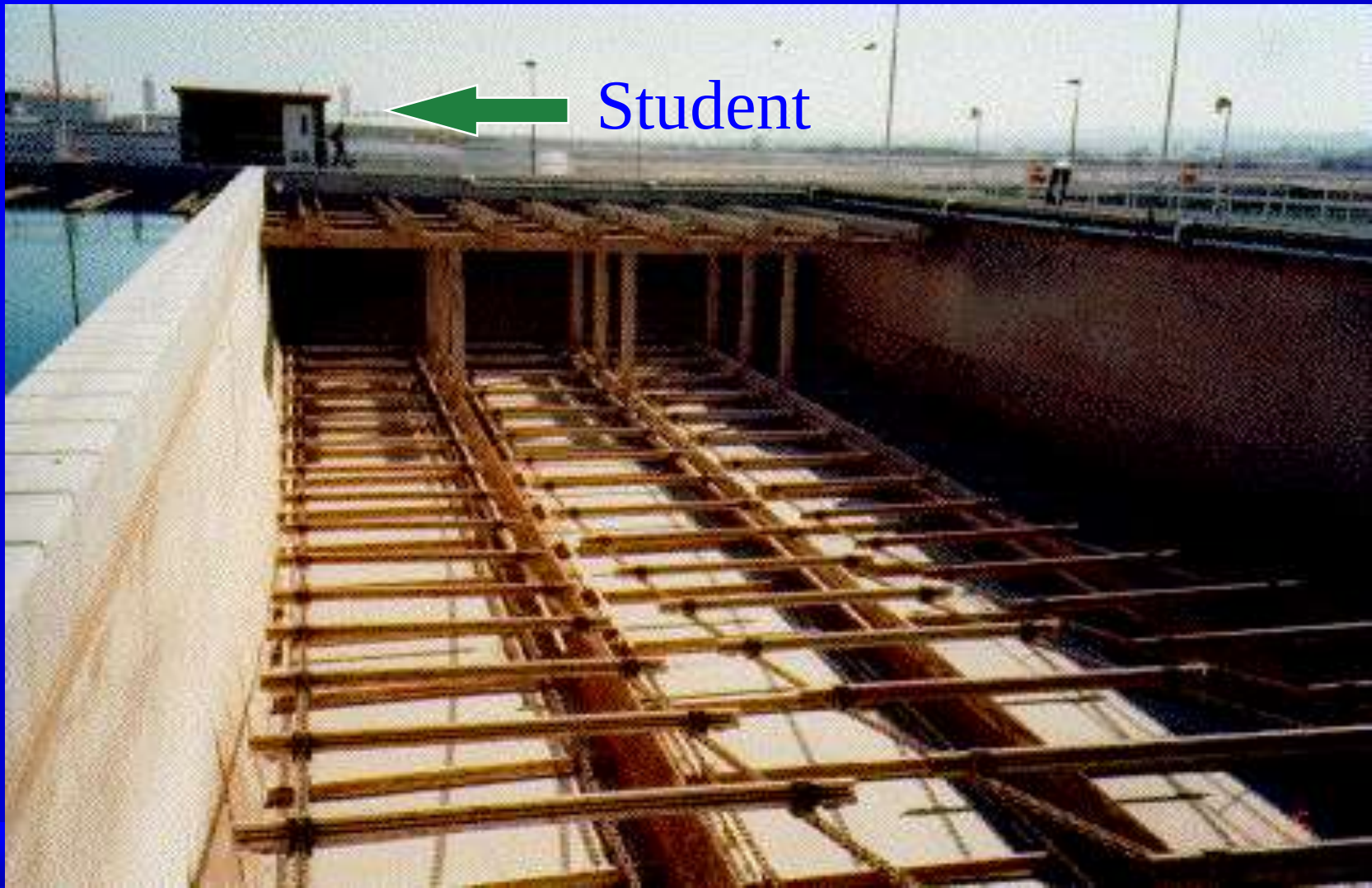
Settling Tank



Rectangular Clarifier (Inlet End)



Rectangular Clarifier (Outlet End)



Circular Clarifier



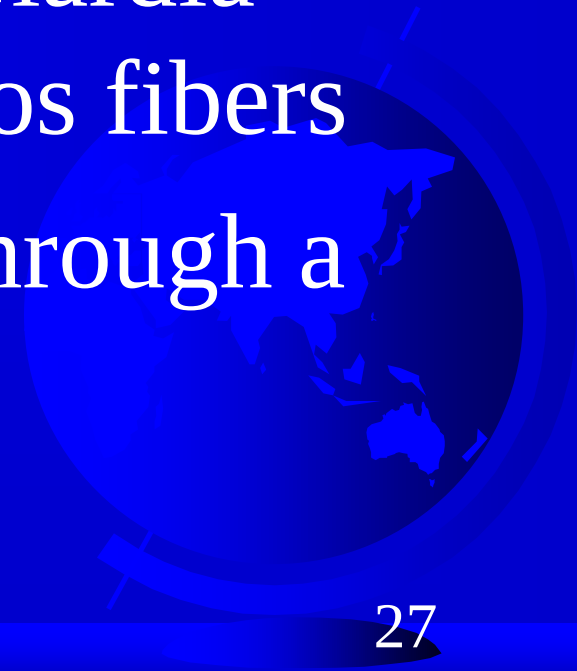
Sludge Treatment

- ☞ Sludge is pumped into a settling pond
- ☞ Water from settling pond is either
 - Recycled back into plant, or
 - Disposed
- ☞ Solids are either left in the pond, or removed and landfilled



Filtration

- ☞ Needed for fine particles not removed by sedimentation
- ☞ Filters can also capture Giardia cysts, viruses, and asbestos fibers
- ☞ Water flows downward through a bed of sand and gravel

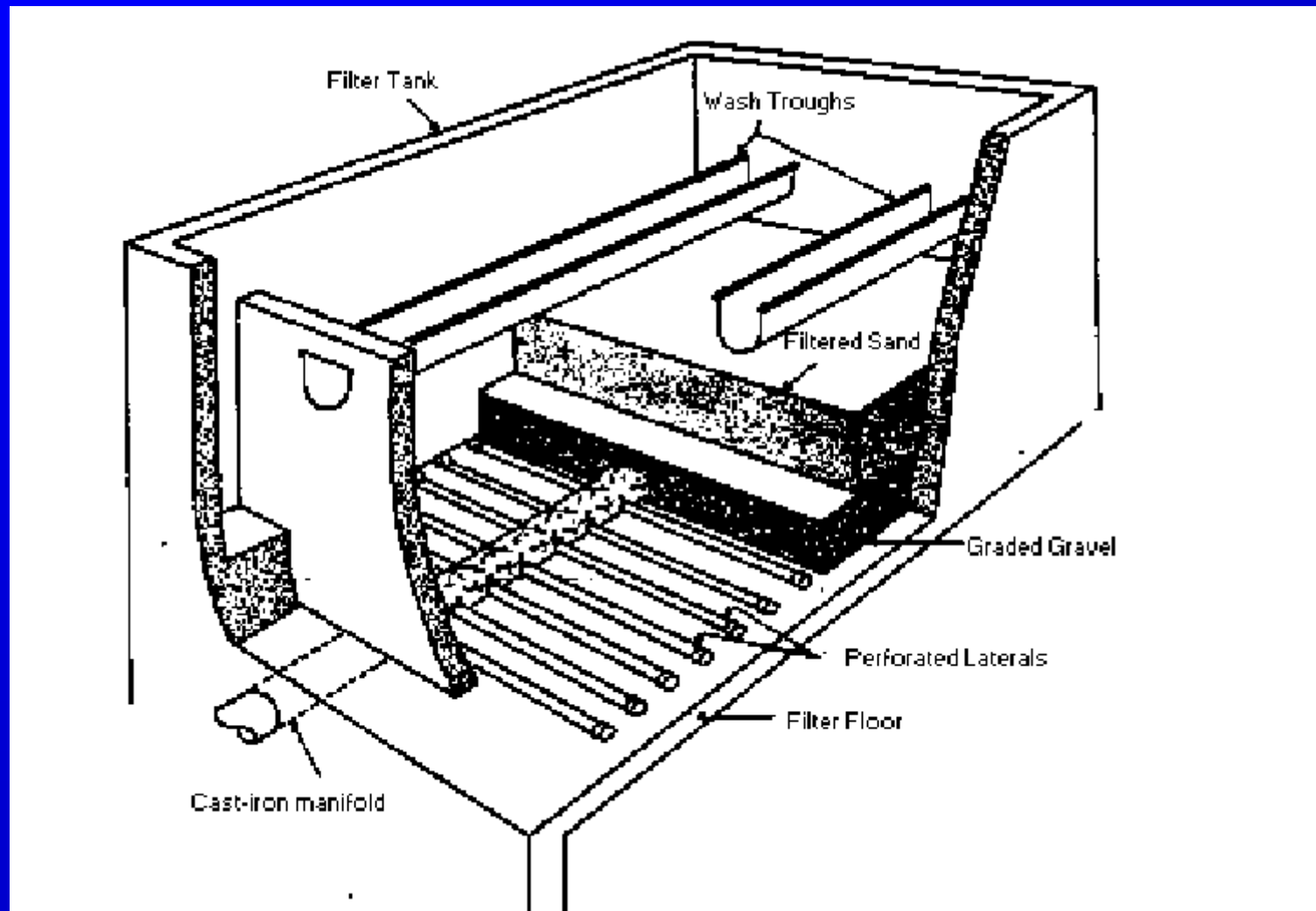


Filtration

- ☞ Particles are captured on and between sand grains
- ☞ Filtered water is collected in the underdrain, sent to disinfection
- ☞ Some filters also contain activated carbon, to remove dissolved organics



Sand Filter



Sand Filter



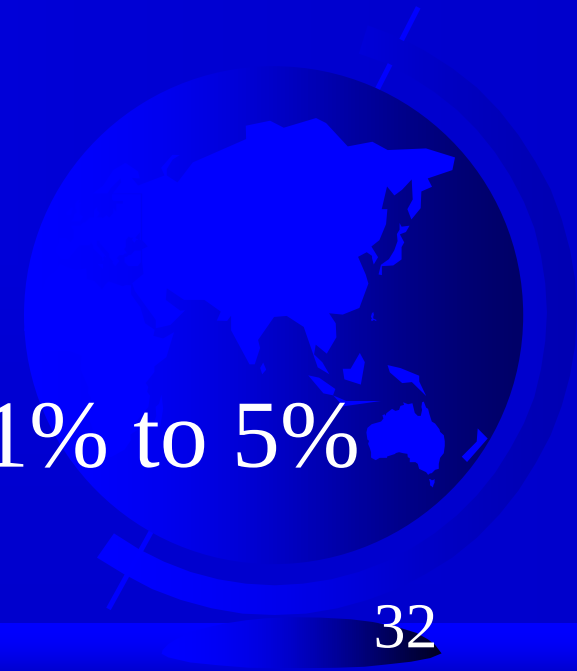
Filter Backwash

- ☞ Sand is backwashed when
 - It becomes clogged, or
 - Turbidity of filtered water gets too high
- ☞ During backwash, water is pumped upwards through the sand bed



Filter Backwash

- ☞ Sand becomes “fluidized”, and particles are flushed from the sand
- ☞ Dirty backwash water is pumped into a settling pond, and either
 - Recycled back into plant, or
 - Disposed
- ☞ Backwashing can consume 1% to 5% of a plant’s production



Disinfection

- ☞ Disinfectants used in water treatment have included:
 - Chlorine, ozone, Ultra-violet light, bromine, iodine, and heat



Chlorination

- ☞ Chlorine is most common disinfectant in water treatment
- ☞ Added at the end of treatment process. Slight excess is used, to prevent bacterial growth in distribution system.
- ☞ Typical chlorine residual is 0.5 to 1 mg/L.



Chlorination

☞ Disadvantages:

- Chlorine is a dangerous chemical
- Can form Trihalomethanes, THMs, which are possible carcinogens.



Other Disinfectants

➡ Other disinfectant chemicals:

- Ozone gas (O_3)
- Ultraviolet light (UV)

➡ Not as desirable because:

- More expensive than chlorine
- Can't maintain a residual concentration

