

Fuel Gas Storage – The Challenge of Methane

Adam Pugsley, Nuno Bimbo, Andrew Physick, Antonio Noguera-Díaz, Jessica Sharpe, Valeska P Ting and Timothy Mays.

URL: <http://people.bath.ac.uk/cestjm>; : <http://people.bath.ac.uk/vt233>

Department of Chemical Engineering, University of Bath, UK

Centre for Doctoral Training in Sustainable Chemical Technologies, University of Bath, UK

Methane

- **Methane combustion emits less carbon dioxide (high H to C ratio) than other fossil fuels and less SO_x and NO_x**
- **Can be used as a transition fuel for the use of even cleaner alternatives (e.g. hydrogen energy)**
- **Has a higher heating value of 55.50 MJ kg⁻¹ (compared with hydrogen's 141.80 MJ kg⁻¹ and gasoline's 47.30 MJ kg⁻¹)**

Methane storage

- As hydrogen, it has a very poor volumetric density (also a gas at normal pressure and temperature)
- To be used in vehicles, it has to improve on its volumetric density (amount per volume) using gas compression, liquefaction or by adsorption
- The goal is to test new porous materials for methane storage and investigate how adsorptive storage compares with other methods

Equipment



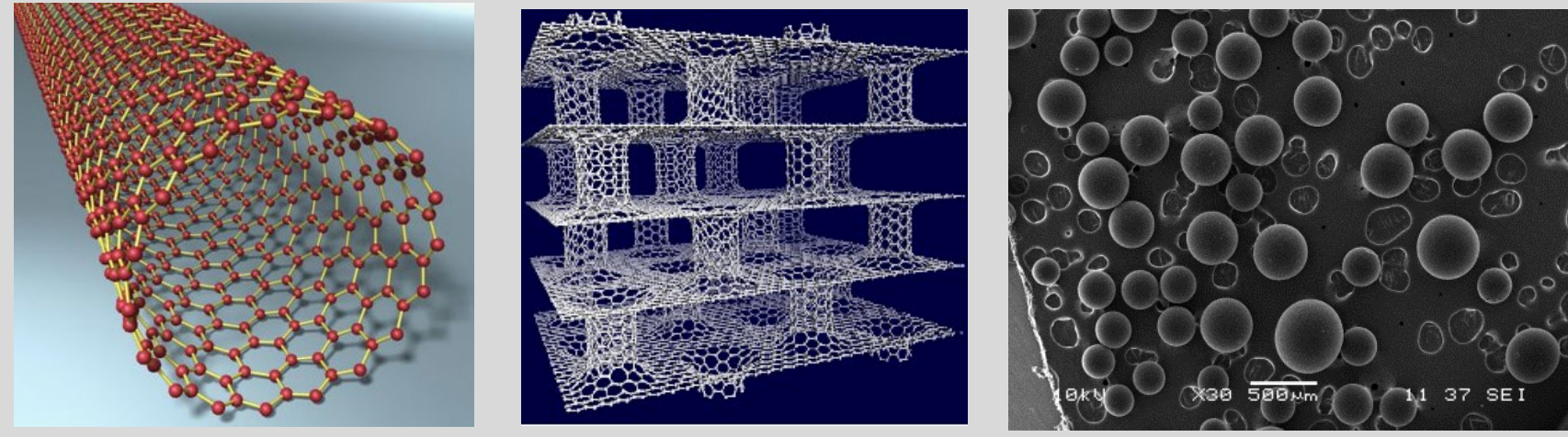
Clockwise from top left: X-ray diffractometer; IsoEx apparatus, Thermal Gravimetric analyser, HTP-1 volumetric sorption analyser, ASAP 2020 sorption analyser (centre), Helium pycnometer and IGA gravimetric sorption analyser

Materials

Carbons

Advantages:

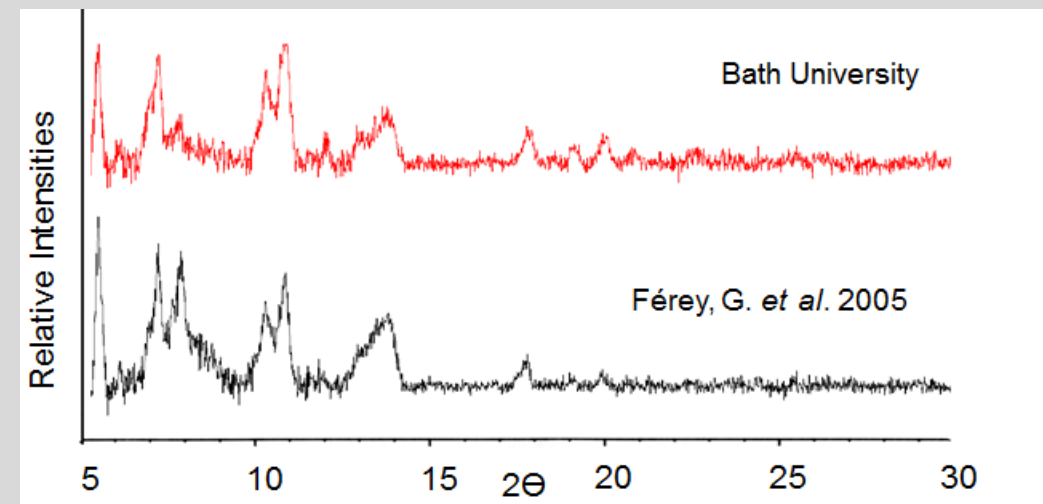
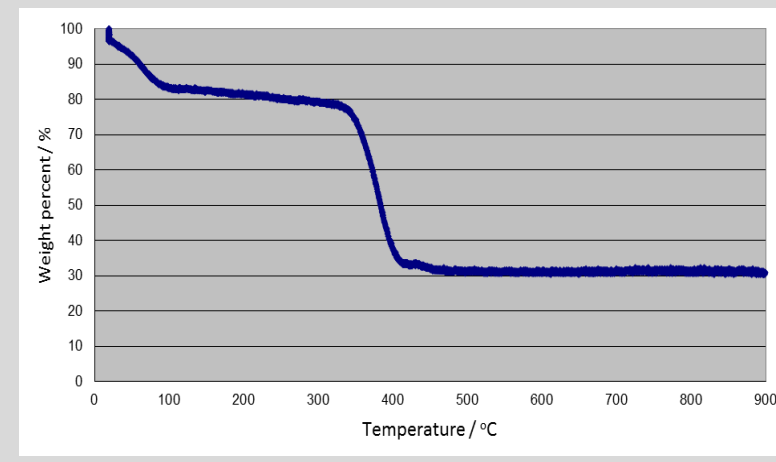
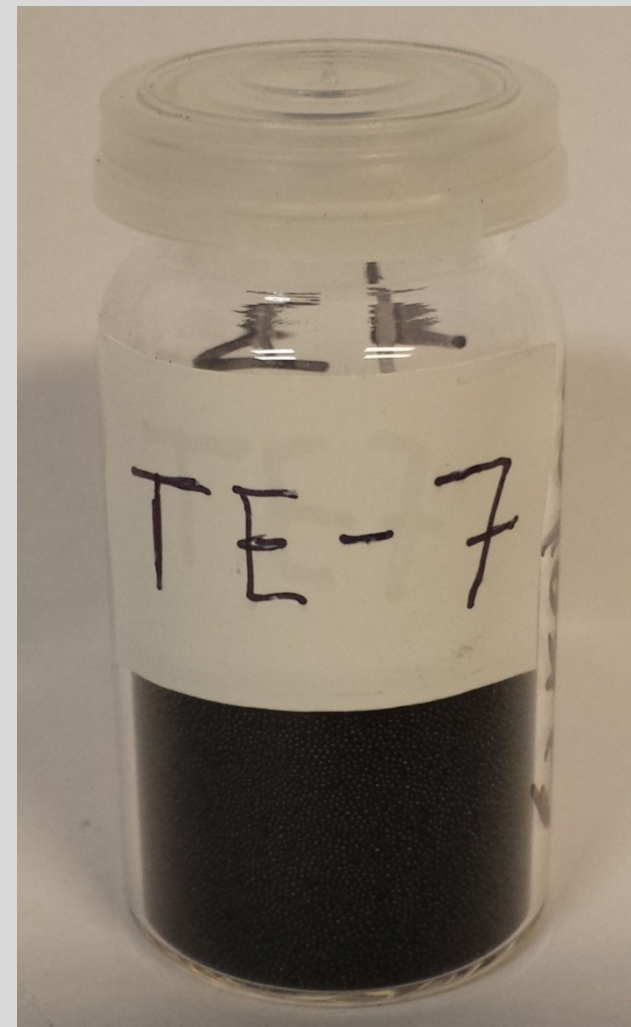
- **Reversible, lightweight and cheap**
- **Wide variety of structural forms**
- **Good thermal stability**
- **Ability to modify the structure**



Nanotube, Pillared Graphene, carbon beads

Porous Materials

MAST TE7 Carbon Beads



MIL-101(Cr) XRD and TGA

Metal-organic frameworks

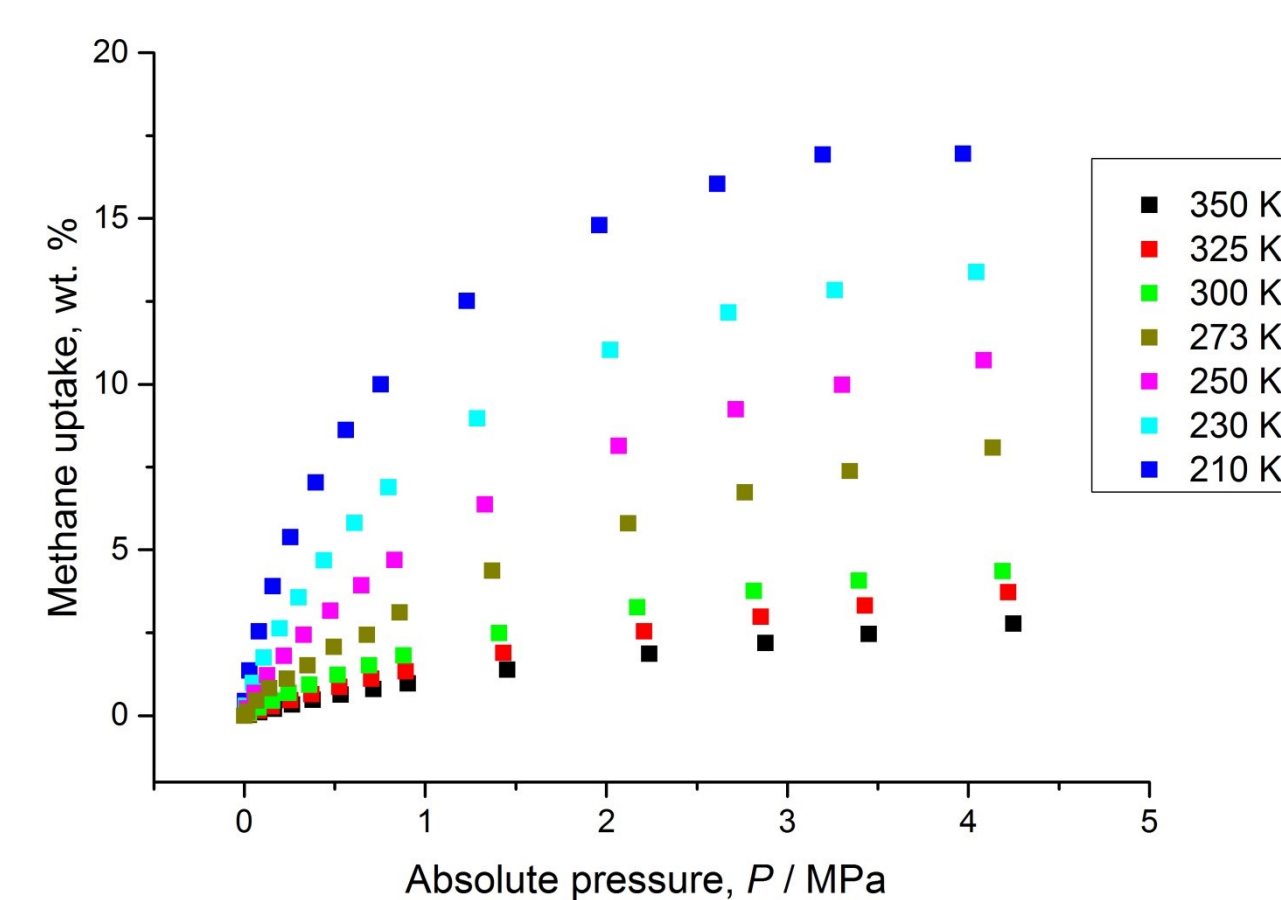
- Metal centres strongly bonded to organic linkers
- High surface area
- Highly tuneable



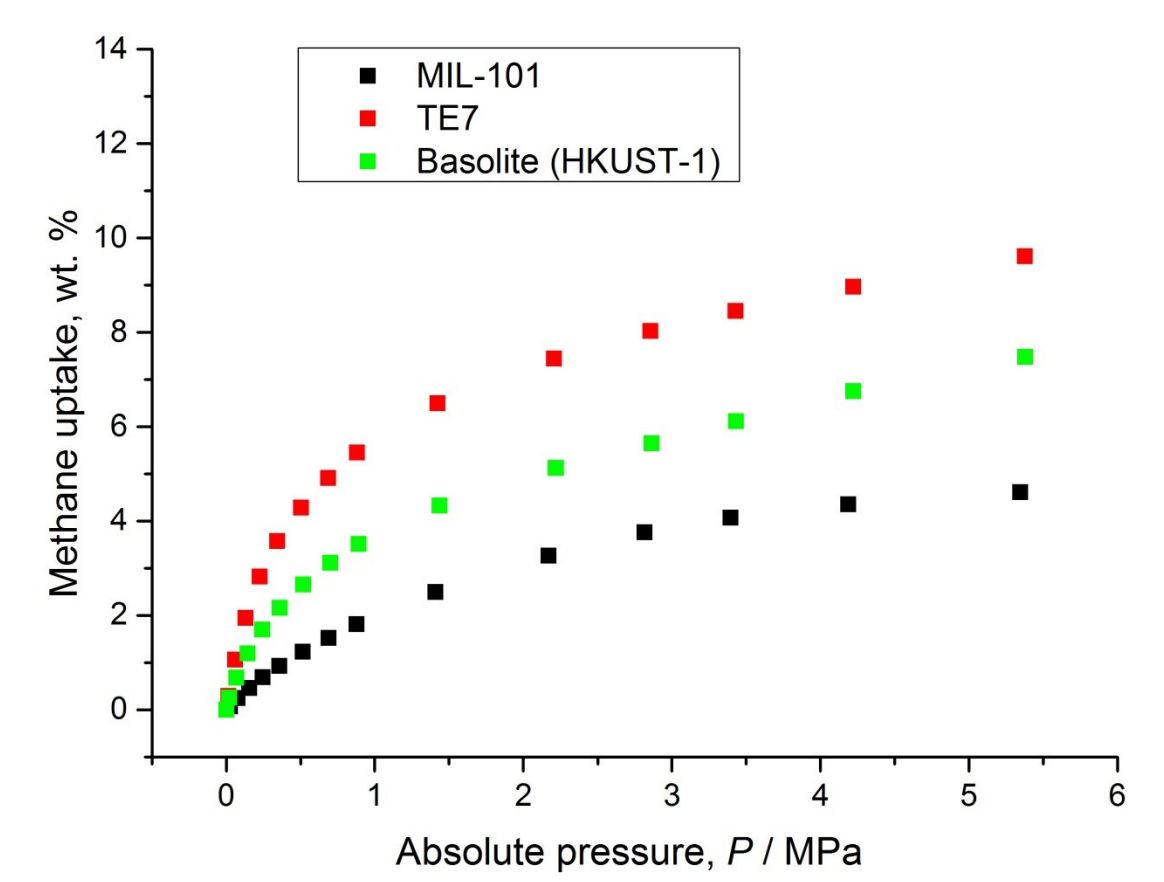
MIL-101 (Cr) and Basolite samples (HKUST-1)

Results

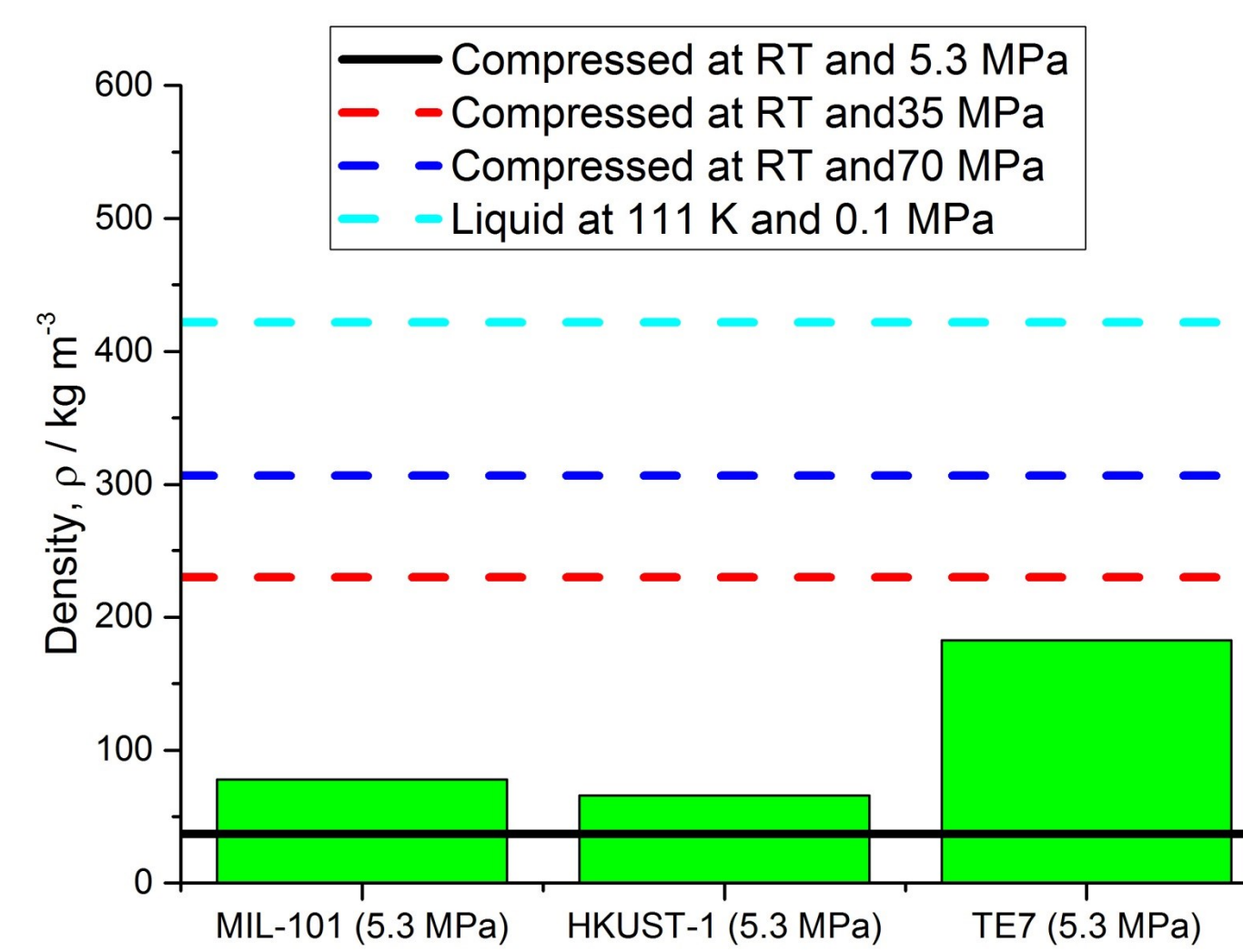
High – pressure methane isotherms



Experimental high-pressure methane excess for MIL-101



Experimental high-pressure methane excess at 300 K for MIL-101, TE7 and HKUST-1



Comparative density of adsorbed methane at 300 K

TE7	MIL-101	HKUST-1
1.90	1.69	0.88

**Density of materials
(in g cm⁻³)**

Group



References

- Peng et al., J. Am. Chem. Soc. 2013, 135, 11887–11894
- Mason et al., Chem. Sci. 2014, 5, 32-51