

#### Effect of Desalination Discharges on Coastal Environments

Dr. Ibrahim Al-TisanDr. Mohamed Saeed



## Outline

Theme

- Overview
- About SWCC
- Some impacts of the environment on desalination plants
  Effect of discharges on coastal environments

#### Theme

- The brine discharged from SWCC's plants has a benign or (at worst) minimal impact on the marine coastal environment.
- However, the plants themselves are subject to severe impacts from adjacent water and surrounding air environments.

## Overview

- Much publicity is devoted to the negative effect of desalination plants on the environment
- This claim is rarely supported by experimental evidence
- On the contrary, operation of plants is often jeopardized by problems arising from surrounding environments.

## The Saline Water Conversion Corporation (SWCC)

- SWCC is a Government Agency of the Kingdom of Saudi Arabia, responsible for the production of desalinated water.
- ≈60% of freshwater requirement of the Kingdom is met from seawater desalination plants on the shores of the Red Sea and Arabian Gulf.

## **Production Capacity**

- Present capacity of SWCC is ≈3 millions M<sup>3</sup> per day of desalinated water produced from 26 plants at 15 sites:12 sites on the Red Sea and 3 sites on the Arabian Gulf.
- Additional 1.25 millions M<sup>3</sup> are being added from new plants
- Additionally, 74000 MWH of power are also generated by the SWCC plants, with 2400 to be added from new plants.

SWCC needs a clean source water to feed its plants. Therefore, SWCC has great interest in keeping clean environment and has dealt with this issue seriously. As a consequence:

- The Research Institute is tasked with the responsibility of environmental assessment and corrective remedies.
- SWCC also established environment committees in the East and West coasts with environmental personnel in each plant for on-site monitoring.

## Size of Feed and Discharged Water

The product water constitutes only a small fraction of the feed water withdrawn from the sea for desalination.

The larger fraction, a huge quantity of water, is returned to sea in the form of brine reject.

- For example: the Jubail plants withdraw 400,000 m<sup>3</sup> of seawater/hour.
- Of this quantity, 340,000 m<sup>3</sup>/hour is discharged back into the sea.

## IMPACT

- SWCC plants could affect and in turn be affected by the environment.
- The impact involves air and coastal water environments.

# Some impacts of environment on Desalination plants include:

- Marine shells clogging intake structures and impeding water flow and heat transfer.
- Seasonal water currents and tide bringing suspended matter and creating filtration problems.
- Dust storms fertilizing coastal water creating algal blooms and associated die offs and filtration problems.
- > The nagging problems of membrane fouling.

## Effect of Discharges

#### on Coastal Environments

## Intake and Discharge System of Jubail plants



 $S_1$  sampling station from open sea,  $S_2$  from Intake bay,  $S_3$  from discharge site

## **Physico-chemical parameters**

Distribution of major seawater quality parameters during different seasons in the near-shore waters of Jubail Desalination and Power Plants

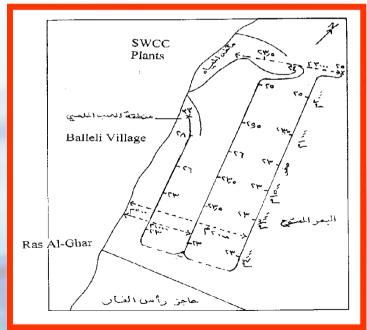
	Sampling sites			
Parameters/Seasons	Intake Bay	Open Sea	Outfall Mixing Bay	Recovery Zone (1 Km)
Sea surface temperature ( OC) Winter Spring Summer Fall	$17.90 \pm 0.85 \\ 24.42 \pm 5.10 \\ 30.25 \pm 0.35 \\ 27.00 \pm 1.41$	$17.80 \pm 1.0624.00 \pm 4.0930.75 \pm 1.0627.00 \pm 1.41$	$27.30 \pm 2.47 \\33.08 \pm 4.06 \\37.25 \pm 0.35 \\34.50 \pm 0.71$	$20.50 \pm 4.95 25.91 \pm 5.59 34.38 \pm 3.71 30.00 \pm 2.82$
Conductivity (Milli seimens/cm) Winter Spring Summer Fall	$57.28 \pm 4.70 \\ 58.83 \pm 1.33 \\ 63.85 \pm 1.77 \\ 61.15 \pm 1.49$	$57.58 \pm 5.90 \\ 59.56 \pm 3.11 \\ 63.73 \pm 1.66 \\ 61.60 \pm 0.00$	$67.33 \pm 1.23  65.55 \pm 2.65  69.53 \pm 2.65  67.40 \pm 4.53$	$60.15 \pm 6.15 \\ 61.21 \pm 1.62 \\ 68.58 \pm 3.57 \\ 64.43 \pm 1.66$
<i>pH</i> Winter Spring Summer Fall	$8.36 \pm 0.00 \\ 8.29 \pm 0.06 \\ 8.34 \pm 0.06 \\ 8.60 \pm 0.22$	$8.38 \pm 0.02 \\ 8.31 \pm 0.06 \\ 8.35 \pm 0.06 \\ 8.61 \pm 0.21$	$8.39 \pm 0.02 8.32 \pm 0.05 8.34 \pm 0.04 8.63 \pm 0.24$	$\begin{array}{c} 8.39 \pm 0.00 \\ 8.31 \pm 0.05 \\ 8.34 \pm 0.06 \\ 8.67 \pm 0.17 \end{array}$
Dissolved Oxygen (mg/L) Winter Spring Summer Fall	$6.88 \pm 0.566.66 \pm 0.525.24 \pm 0.685.22 \pm 1.09$	$6.98 \pm 0.41  6.85 \pm 0.79  5.46 \pm 0.27  4.89 \pm 0.69$	$6.36 \pm 0.616.18 \pm 0.495.34 \pm 0.444.86 \pm 0.45$	$6.65 \pm 0.00 \\ 6.27 \pm 0.54 \\ 5.17 \pm 0.29 \\ 5.17 \pm 0.00$

## **Temperature profile**

Temperature profile at Jubail plants at 500,1000 and 2000m from intake and discharge sites.

 Temperature stabilizes at 500-1000m beyond the discharge point





## Effect of Discharges on Primary Productivity in Terms of Chlorophyll

	Chlorophyll Concentration (mg/m <sup>3</sup> )					
Location	Discharge site	500m from discharge	1000m from discharge	Open sea		
Jubail (depth ~4m)	0.50	2.50	2.60	2.60		
Jeddah (depth 30m)	*0.51	0.32	0.22	0.22		

**Note:** At Jubail normal primary production regained at 500m from discharge site.

At Jeddah discharge site is more productive than open sea.

## Plankton groups and numbers in Jubail and Jeddah Coast

Group		Location : Juba	Jeddah		
	Open sea <sup>1</sup>	Intake bay <sup>2</sup>	Discharge site <sup>3</sup>	Intake zone	Discharge site
A. Phytoplankton (cell/m <sup>3</sup> )				A.Phytoplankton <sup>4</sup>	
Diatoms	3.42 x 10 <sup>5</sup>	2.14 x 10 <sup>5</sup>	1.48 x 10 <sup>5</sup>	2.98 x 10 <sup>4</sup>	5.79 x 10 <sup>4</sup>
Dinoflagellates	4.29 x 10 <sup>4</sup>	2.71 x 10 <sup>4</sup>	7.67 x 10 <sup>3</sup>	-	-
Blue-green bacteria	6.84 x 10 <sup>5</sup>	1.39 x 10 <sup>5</sup>	1.23 x 10 <sup>6</sup>	-	-
B. Zooplankton (No./m <sup>3</sup> )				B. Zooplankton <sup>5</sup>	7.10 x 10 <sup>4</sup>
Protozoa	4.93 x 10 <sup>3</sup>	5.93 x 10 <sup>3</sup>	8.09 x 10 <sup>3</sup>		
Coelenterates	1.59 x 10 <sup>3</sup>	1.85 x 10 <sup>3</sup>	9.92 x 10 <sup>3</sup>		
Nematodes	2.19 x 10 <sup>3</sup>	1.06 x 10 <sup>3</sup>	1.92 x 10 <sup>3</sup>		
Annelida	5	4	10		
Mollusca	2.75 x 10 <sup>3</sup>	5.5 x 10 <sup>2</sup>	1.92 x 10 <sup>3</sup>		
Crustaceans	4.41 x 10 <sup>4</sup>	7.52 x 10 <sup>4</sup>	3.56 x 10 <sup>4</sup>		
Echinoderms	1.25 x 10 <sup>3</sup>	12	50		
Chordata	4.09 x 10 <sup>4</sup>	8.70 x 10 <sup>4</sup>	3.17 x 10 <sup>4</sup>		
(fish eggs, fish larvae and					
tunicates)					

## Notes on Plankton

- The distribution of the major plankton groups are similar in feed and discharge zones
- Only Echinoderms seem to be impacted by the brine discharge, and reasons other than brine may contribute to their scarce presence in the brine discharge area e.g. impingement and entrainment
- The major groups of phyto- and zooplankton could form a healthy base of food chains in feed and discharge water zones.

#### **Concentration of nutrients at Jubail and Jeddah**

	Jubail			Jeddah	
Nutrient (µg/l)	Open sea	Intake bay	Discharge	Open sea	Discharge site
			site		
A. A.Inorganic nutrients (µg/l)					
1. Ammonia -N	1.7 – 8.9	0.8 – 8.5	0.5 – 0.7	1.4 – 4.6	1.7 – 3.5
2. Nitrite -N	0.5 – 1.4	0.2 – 3.0	0.05 – 0.1	0.02 – 0.2	0.03 – 0.2
3. Nitrate-N	1.4 - 6.0	1.6 – 5.3	0.5 – 2.0	0.5 – 0.9	0.2 – 1.2
4. Phosphate-P	1.7 – 6.5	1.9 – 4.6	0.2 – 0.3	0.03 – 0.2	0.1 – 0.3
5. Silicate –S	BDL – 0.1	BDL – 0.1	BDL	BDL	BDL
B. Organic nutrients (mg/l)					
1. Dissolved sugars	0.5 – 0.8	0.9 – 1.5	3.9 – 5.0	0.34 – 0.42	0.17 – 0.25
2. Dissolved nitrogen	0.2 – 0.3	1.5 – 2.1	4.9 – 6.1	0.02 – 0.05	0.02 – 0.03
3. TOC	2.1 – 2.6	1.6 – 2.4	2.9 – 3.5	1.9 – 2.9	1.9 – 2.9

**BDL = Below detection limit** 

Increased organic matter in discharge due to organic decomposition by chlorine

#### Metals in the Coastal Waters of Jubail and Jeddah

	Concentration (µg/l)				
Trace Metal <sup>1</sup>		Jubail	Jeddah		
	Open sea	Intake bay	Discharge site	Open sea	Discharge site
1. Iron	0.34 – 5.86	0. 45 – 6.20	0.6 – 7.0	2.9 – 3.9	3.9 – 4.2
2. Nickel	0.16 - 1.40	0.25 - 1.40	0.25 – 1.50	0.90 – 1.0	0.17 – 1.2
3. Copper	0.35 – 2.60	0.65 – 3.70	0.6 - 4.0	1.0 - 1.0	0.95 – 1.1
4. Chromium	0.60 – 0.20	0.00 - 0.17	0.00 – 0.05	ND	ND

<sup>1</sup>These metals are normally associated with corrosion - No difference between source and discharge waters ND = Not detected

## Toxic Effects of Brine Discharge

- Cumulative findings of our research show that the brine discharged from SWCC's plants has a benign or (at worst) minimal impact on marine coastal environment.
- Still there are claims that effluents are potentially harmful to coastal environment.
- A direct way to address these concerns involves exposing selected marine organisms to brine discharge, measuring their biological response and assessing any deviation from norm.

## **Toxic Effects of Brine Discharge**

• We assessed the comparative *in vitro* toxicity of water from the discharge site of Jubail compared to feed water using *artemia* cysts and a bioluminescent bacterium

#### **RESULTS:**

- No difference in the hatching and survival of larval artemia.
- No difference in emission of light by the bioluminescent bacterium.

#### **Brine Shrimp Hatching and Larval Mortality Rates**

Source Hatching		Larval Mortality Rates (%)			
water	Rate (%)	24h	48h	72h	96h
Discharge	≥98	None	None	$3.3\pm7.5^{a}$	$\textbf{26.7} \pm \textbf{14.9^{b}}$
Feed	≥ <b>98</b>	None	None	$\textbf{5.7} \pm \textbf{9.3}^{a}$	21.2±17.0 <sup>b</sup>

<sup>a,b</sup> Means with same letter superscript are not different (n = 50, ANOVA and t-test, P=0.05)
 No difference in hatching rate or larvae survival between discharge and feed waters

## **Stages in Brine Shrimp Life Cycle**



**Hatching Larvae** 

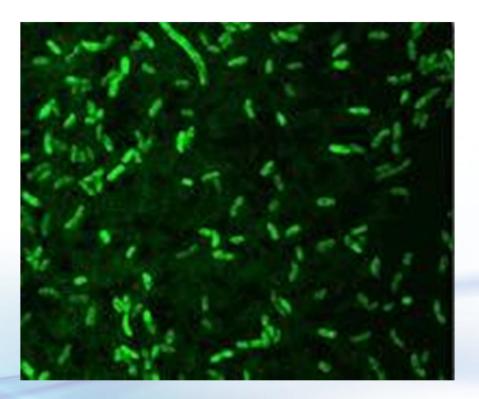
Adult

#### **Inhibition of Bacteria Bioluminescence**

Source	% Inhibition	Difference*
Open seawater	$\textbf{8.2} \pm \textbf{5.3}$	Control
Intake bay	$\textbf{9.9} \pm \textbf{5.8}$	Not Significant
Discharge site	9.7 ± 4.9	Not Significant
Feed water + Antifoam	$\textbf{9.0} \pm \textbf{5.1}$	Not significant
Feed water + Antiscalant	$\textbf{10.4} \pm \textbf{5.9}$	Not significant

\*(n=10, ANOVA and t-test, P=0.05)





#### **Luminescent** bacteria



## Reasons for Benign Effects of Discharge

## 1. Inherent mitigating design

- The discharge channel of Jubail plants is designed as such that it dissipates temperature.
- The channel is ~1.5 Km long and is cascading to the discharge point with strong mixing and air contact that reduce temperature and replenish Oxygen.
- The added volume of cooling water in the discharge also helps in diluting chemical additives and salinity.



## 2. Dilution Effect

# Significant and immediate dilution by cooling water (heat rejection)



## **Dilution Effect of Cooling Water**



Total intake (ex: Jubail)	12 (≈ 12 millions m³/day)
Cooling (heat rejection)	9
Product	1
Reject	2
Brine discharge	11
Assume salinity	40‰
2/3 of make-up is rejected	
1/3 product	3
Salinity of rejected portion	of make-up is $\frac{3}{2} \ge 40 = 60\%$
Salinity of 9 parts cooling is	s 40‰ <sup>2</sup>
Final Salinity $\frac{2}{11} \times 60 + \frac{9}{11} \times 60$	40 = 10.9 + 32.7 = 43.6

.:. Salinity increases by only 3.6% or 9%

## CONCLUSIONS

- An environmental database has been established for the Arabian Gulf and Red Sea coastal and open sea waters opposite the SWCC Jubail and Jeddah desalination and power plants.
- The data clearly show that the brine discharged from SWCC's plants has a benign or (at worst) minimal impact on the marine coastal environment.
- Desalination plants should not be falsely implicated in any negative impact upon coastal water environments. Any report about coastal water pollution from the desalination plants should be interpreted with caution.

## Thank You

