



# Successful Starts in LC-MS/MS

## Critical considerations when choosing a syringe filter for HPLC, UHPLC, and LC-MS/MS analysis

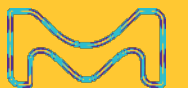
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MilliporeSigma is the U.S. and  
Canada Life Science business of  
Merck KGaA, Darmstadt, Germany.



# LC-MS/MS Analytical Methods

## Three Components for Success

HPLC Columns



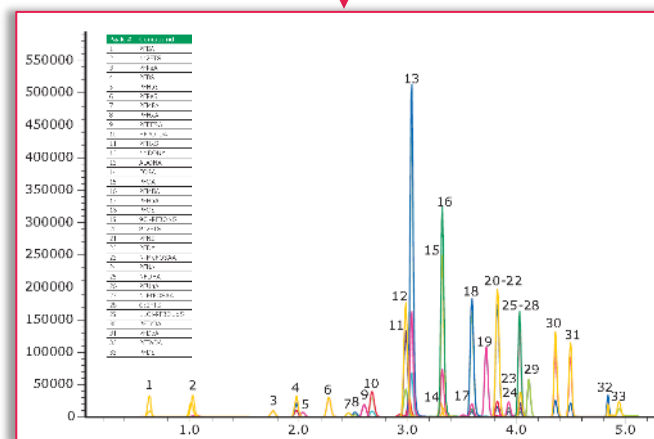
Mobile Phase Components



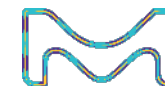
Sample preparation removes particles, contaminants and interferences from the sample, which brings it to a form amenable to downstream analysis.



- Analyte properties
- Impurities present
- Matrix components
- Choice of water



- For high-particulate matrices, SPE and filtration are common

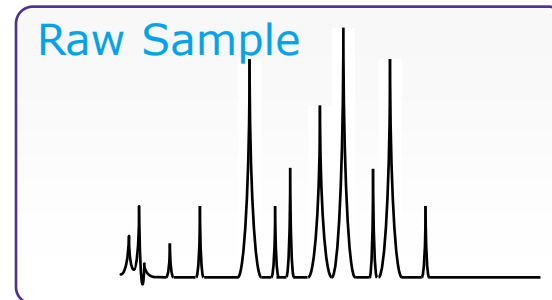
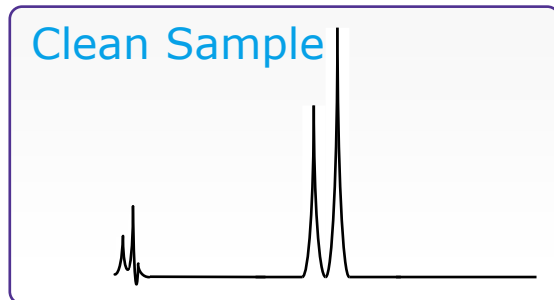


# Why Sample Preparation

## Importance of sample preparation

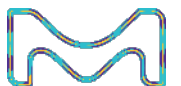
Without sample preparation, instrument performance and data quality will not be optimal:

- Undissolved particles in a sample / mobile phase clog and reduce the life of the chromatography column.
- Sample matrix containing lots of impurities makes chromatograms challenging to interpret.
- Particles held up on the column can clog the column and/or leach contaminants into the eluent (and affect current and subsequent samples), thereby increasing background noise.



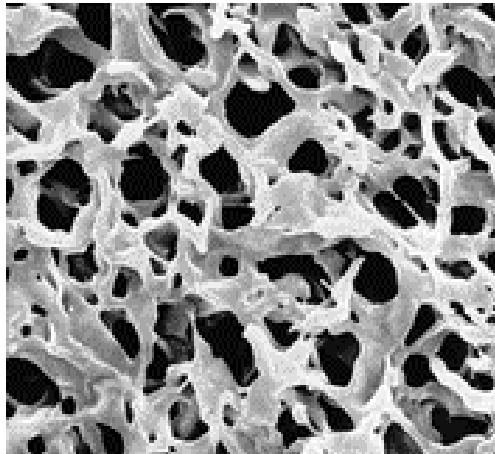
- Sample matrix impurities
- Undissolved particles
- Particle leaching

80%

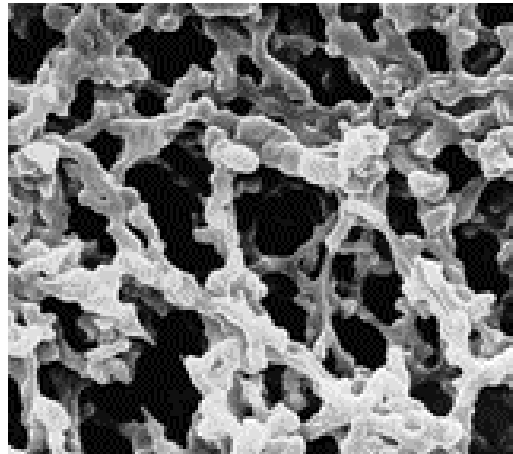


# Which Membrane, and Why?

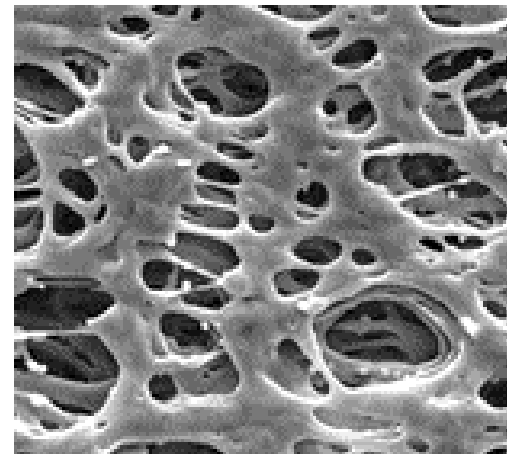
## Membrane Morphology & Applications



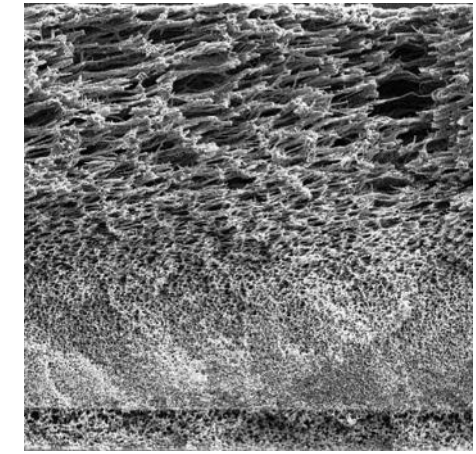
**Polyvinylidene Fluoride (PVDF):** *Low binding and fast flow for protein sample prep*



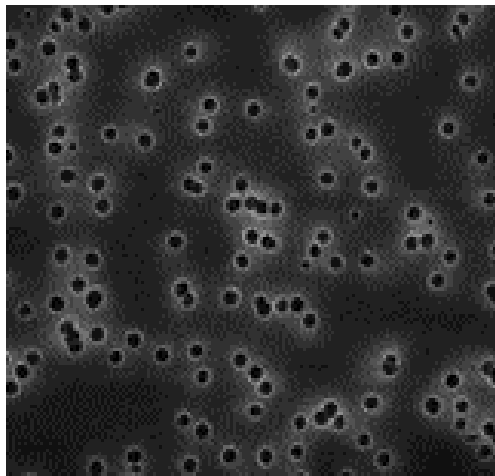
**Mixed Cellulose Esters (MCE):** *Biologically inert, versatile, smooth and uniform*



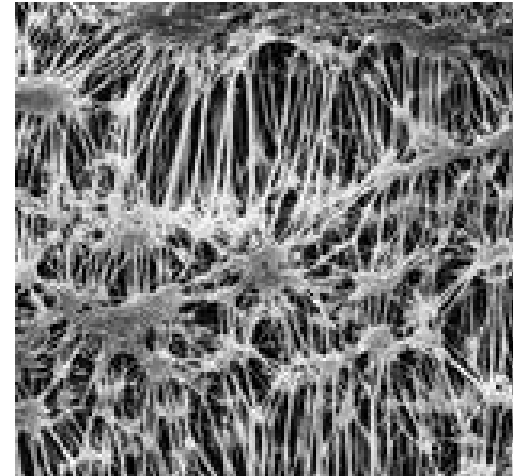
**Polyethersulfone (PES):** *Quick flow and high capacity, asymmetric for high-particulate water samples*



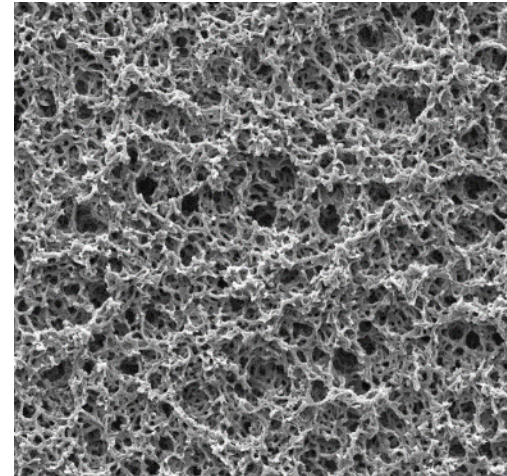
**PES Cross-section**



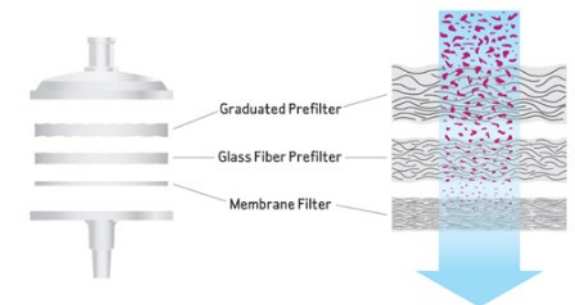
**Polycarbonate (PC):** *preferred for microscopy and cell-based applications*



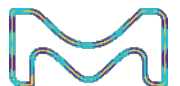
**Polytetrafluoroethylene (PTFE):** *Low extractables and high chemical compatibility*



**Nylon:** *Broad compatibility and commonly used for HPLC*



**Nylon with a Glass Fiber Prefilter (HPF):** *One-step cleanup of large and small particulates without clogging*



# Which Membrane, and Why?

## Chemical Compatibility, Extractables and Retention

	Housing Type		Membrane Type			
	HDPE	PP	Nylon	PES	GFF	PTFE*
AA	E	ND	P	E	E	E
ACN	E	E	E	G	E	E
MeOH	E	E	G	G	E	E
EtOH	E	E	E	E	E	E
3M NaOH	E	E	P	P	ND	E
NH <sub>4</sub> OH	E	E	P	P	E	E
Na <sub>2</sub> CO <sub>3</sub> Solution	G	E	P	P	ND	E
1N HCl	E	E	G	E	E	E
Brine	E	E	E	E	ND	E
SDS	G	G	G	G	ND	E

**E** Excellent      **G** Good      **P** Poor

**Abbreviations:** HDPE = high-density polyethylene; PP = polypropylene; PES = polyethersulfone; GFF = glass fiber filter; MCE = mixed cellulose ester; AA = acetic acid; ACN = acetonitrile; MeOH = methanol; EtOH = ethanol; SDS = sodium dodecyl sulfate; ND=not determined/testing recommended

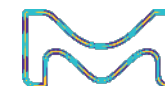
\*PTFE/PVDF not recommended for PFAS testing applications

**Pore size** is important to consider with respect to your column. For U/HPLC columns with <3µm packing, use a 0.2µm, for sizes above this, a 0.45µm is sufficient.

**Retention doesn't always depend on pore size**, but also membrane manufacturing method, symmetry and chemical structure, and even lot-by-lot

Material	0.2/0.22µm Syringe Filters	0.45µm Syringe Filters
	% Retention of 0.24µm PS Beads	% Retention of 0.5µm PS Beads
<b>Nylon</b>	100.0±0.1	100.0±0.05
<b>Nylon-HPF</b>	54.2±27.3	-- <sup>a</sup>
<b>PES</b>	69.4±28.1	99.5±0.79
PTFE (Vendor 1)	49.8±31.8	99.98±0.10
PP (Vendor 1)	25.3±0.9	100.0±0.06
RC (Vendor 2)	15.8±2.2	48.16±4.27

<sup>a</sup>Pore size not available.





## Case study: PFAS analysis

# Analytical Methods for PFAS Developed at a Rapid Pace

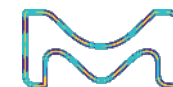
## Regulatory Landscape

Method(s)	Update/Revision	Matrix/Matrices	Sample Prep	Analytical Method
EPA 537.1	Jun. 2020	Drinking water	SPE	LC-MS/MS
EPA 533	Dec. 2019	Drinking water	SPE	LC-MS/MS
SW-846 Method 8327*	Jul. 2021	Non-potable groundwater, surface water, wastewater	SPE, <b>filtration</b>	LC-MS/MS
ASTM D7968-17a	Sep. 2017	Environmental solids	Solvent extraction, <b>filtration</b>	LC-MS/MS
ASTM D7979-19	Sep. 2021	Water matrix (no drinking water)	Solvent extraction, <b>filtration</b>	LC-MS/MS
ISO 21675	Oct. 2019	Drinking, natural and wastewater	SPE, <b>filtration</b> as needed	LC-MS/MS
FDA C-010.02	Dec. 2021	Foods	QuEChERS, SPE, <b>filtration</b>	LC-MS/MS
OTM-45	Jan. 2021	Air Emissions (stationary sources)	Sampling train: <b>filtration</b> , impingers	LC-MS/MS
EPA Draft 1633	Feb. 2022	Aqueous, soil, biosolids, sediment, tissue	SPE, <b>filtration</b>	LC-MS/MS
EPA Draft 1621*	Apr. 2022	Aqueous matrices	TSS, GAC column cleanup	CIC

- Almost exclusively LC-MS/MS based methods
- SPE & filtration are common sample preparation
- Increased focus on high-particulate matrices
- Additional watchouts for PFAS sample preparation

*\*screening method only*

**Abbreviations:** SPE = solid phase extraction; TSS = total suspended solids determination; GAC = granular activated carbon; CIC = combustion ion chromatography  
Selected methods; does not include all drinking water and international methods

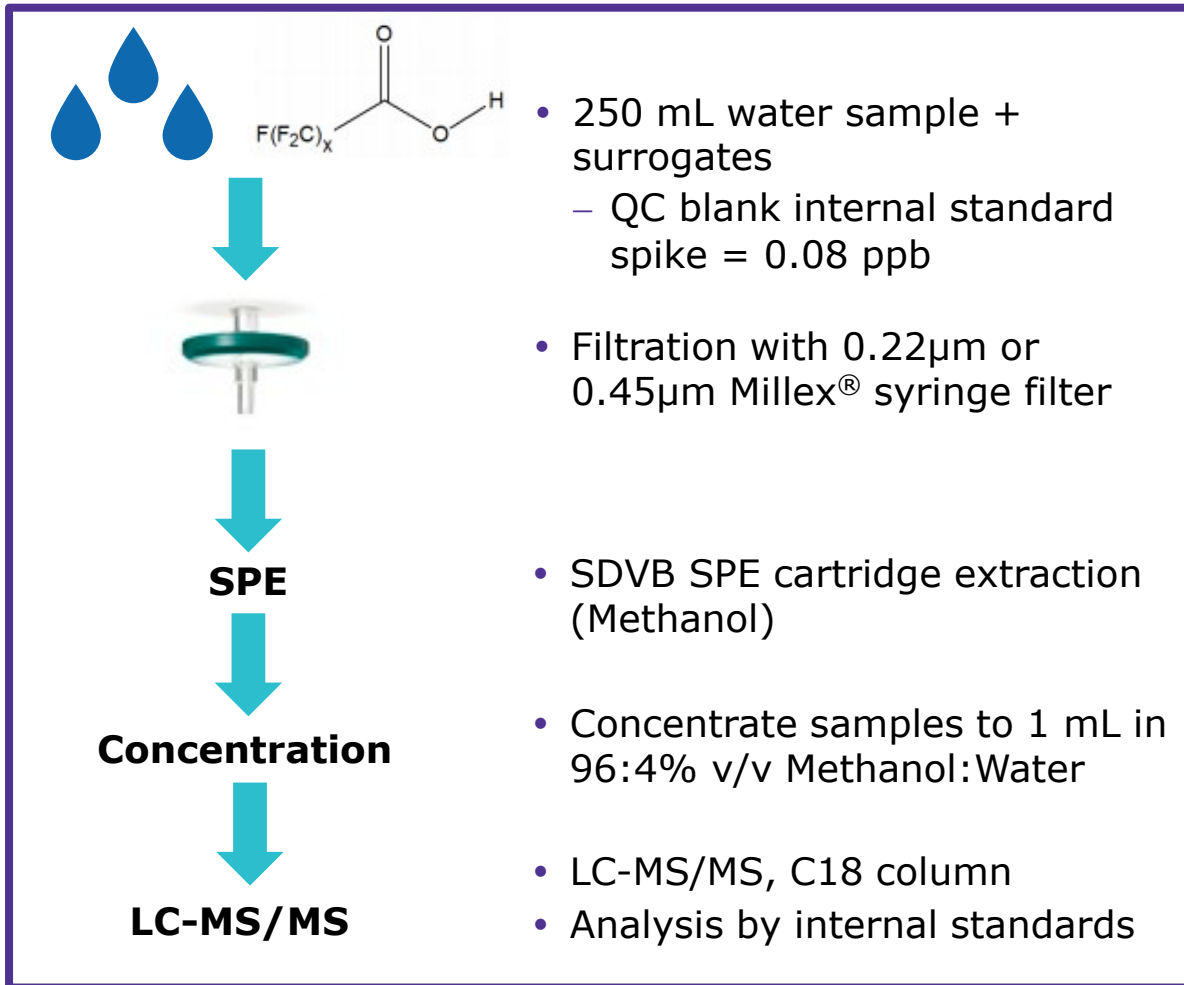




# Do Nylon or PES Syringe Filters Contaminate Samples with PFAS?

## Experimental

### Overview of Modified EPA 537.1

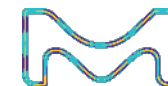


### LC-MS/MS Conditions

<b>Column:</b>	C18, 100 x 2.1mm ID, 2.7 $\mu$ m superficially porous particles			
<b>Mobile phase:</b>	[A] DI Water, 0.1% (v/v) acetic acid; [B] Methanol (MeOH), 0.1% (v/v) acetic acid			
<b>Gradient:</b>	<b>Time (min)</b>	<b>A %</b>	<b>B %</b>	<b>Flow (mL/min)</b>
	0-0.0	65%	35%	0.4
	0-7.0	0%	100%	0.4
	7.0-10.0	0%	100%	0.7
	10.0-11.0	0%	100%	0.7
	11.0-15.0	65%	35%	0.4
<b>Flow rate:</b>	See gradient table			
<b>Detection:</b>	MS/MS, ESI(-), details of MS/MS conditions can be requested from the author			
<b>Column temp:</b>	50.0 $^{\circ}$ C			
<b>Injection volume:</b>	3-5 $\mu$ L autosampler injection			
<b>Sample :</b>	SPE eluate concentrated to 1 mL methanol: water, 96:4% (v/v)			

Data collected in collaboration with:

**SGS**  
(Orlando, FL)



# Results

Category	Compound	Abbreviation	RL (ppb)	MDL (ppb)	SLGP033 0.2µm PES			SLHP033 0.45µm PES			SLGN033 0.2µm Nylon			SLGNM25 0.2µm Nylon-HPF	
					Lot1	Lot2	Lot3	Lot1	Lot2	Lot3	Lot1	Lot2	Lot3	Lot1	Lot2
<b>[1] Perfluoroalkylcarboxylic acids</b>	Perfluorobutanoic acid	PFBA	0.0040	<b>0.0020</b>	<b>ND –</b> Not detected in filtrate						<b>ND –</b> Not detected in filtrate				
	Perfluoropentanoic acid	PFPeA	0.0020	<b>0.0010</b>											
	Perfluorohexanoic acid	<b>PFHxA</b>	0.0020	<b>0.0010</b>											
	Perfluoroheptanoic acid	<b>PFHpA</b>	0.0020	<b>0.0010</b>											
	Perfluorooctanoic acid	<b>PFOA</b>	0.0020	<b>0.0010</b>											
	Perfluorononanoic acid	<b>PFNA</b>	0.0020	<b>0.0010</b>											
	Perfluorodecanoic acid	<b>PFDA</b>	0.0020	<b>0.0010</b>											
	Perfluoroundecanoic acid	<b>PFUnDA</b>	0.0020	<b>0.0010</b>											
	Perfluorododecanoic acid	<b>PFDODA</b>	0.0020	<b>0.0010</b>											
	Perfluorotridecanoic acid	<b>PFTTrDA</b>	0.0020	<b>0.0010</b>											
Perfluorotetradecanoic acid	<b>PFTeDA</b>	0.0020	<b>0.0010</b>												
<b>[2] Perfluoroalkylsulfonic acids, Perfluorooctane sulfonamides, and Perfluorooctanesulfonamidoacetic acids</b>	Perfluorobutanesulfonic acid	<b>PFBS</b>	0.0020	<b>0.0010</b>	<b>ND –</b> Not detected in filtrate						<b>ND –</b> Not detected in filtrate				
	Perfluoropentanesulfonic acid	PFPeS	0.0020	<b>0.0010</b>											
	Perfluorohexanesulfonic acid	<b>PFHxS</b>	0.0020	<b>0.0010</b>											
	Perfluoroheptanesulfonic acid	PFHpS	0.0020	<b>0.0010</b>											
	Perfluorooctanesulfonic acid	<b>PFOS</b>	0.0020	<b>0.0010</b>											
	Perfluorononanesulfonic acid	PFNS	0.0020	<b>0.0010</b>											
	Perfluorodecanesulfonic acid	PFDS	0.0020	<b>0.0010</b>											
	PFOSA	PFOSA	0.0040	<b>0.0020</b>											
	N-MeFOSAA	<b>MeFOSAA</b>	0.0040	<b>0.0020</b>											
N-EtFOSAA	<b>EtFOSAA</b>	0.0040	<b>0.0020</b>												
<b>[3] Fluorotelomer sulfonates and Next Generation PFAS Analytes</b>	4:2 Fluorotelomer sulfonate	8:2 FTS	0.0080	<b>0.0020</b>	<b>ND –</b> Not detected in filtrate						<b>ND –</b> Not detected in filtrate				
	6:2 Fluorotelomer sulfonate	6:2 FTS	0.0080	<b>0.0020</b>											
	8:2 Fluorotelomer sulfonate	8:2 FTS	0.0080	<b>0.0020</b>											
	HFPO-DA	<b>GenX</b>	0.0040	<b>0.0020</b>											
	ADONA	<b>ADONA</b>	0.0080	<b>0.0020</b>											
	9C1-PF3ONS (F-53B Major)	--	0.0080	<b>0.0020</b>											
	11C1-PF3OUdS (F-53B Minor)	--	0.0080	<b>0.0020</b>											

# PFAS Analysis

## The Known Challenges

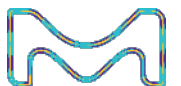
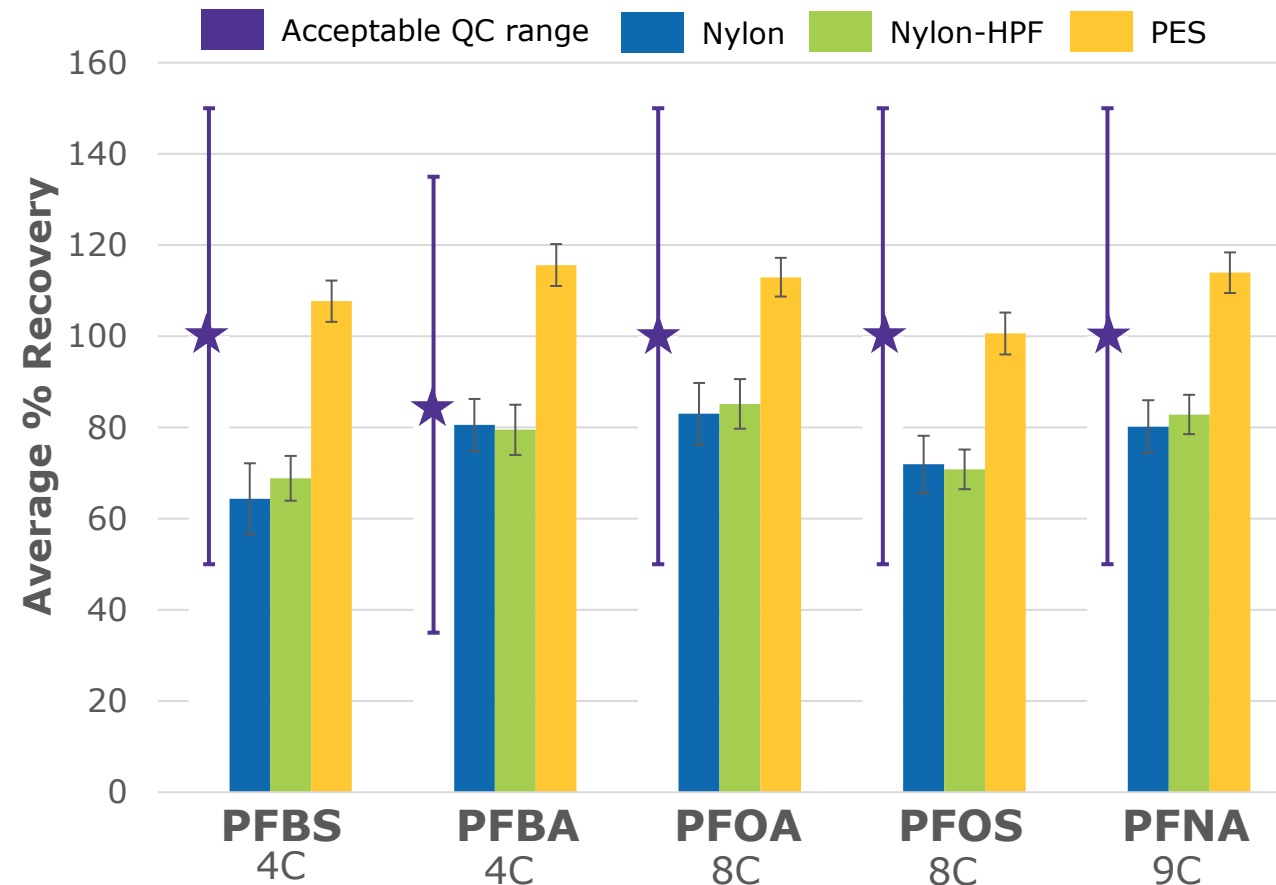
### Known Challenges

- Contamination
- Volatility
- Longer chain compounds
- Recovery and absorption
- Consumable vs. process

### Solutions

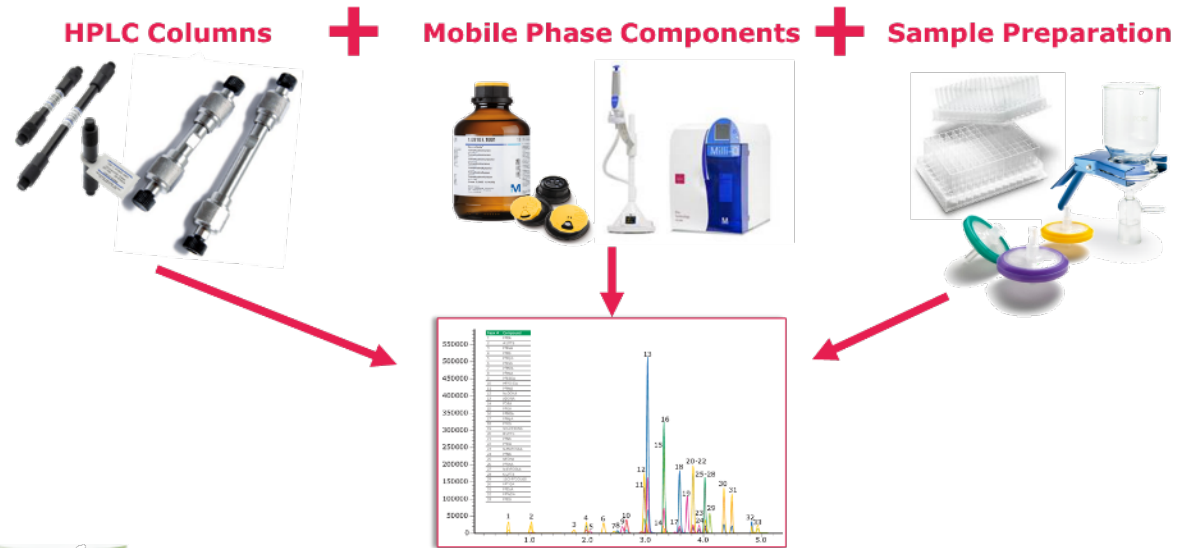
- Know your membrane & vendor
- For nylon, a simple wash can reduce absorption from detectable to undetectable levels [1,2]
- Choose the proper membrane for targeted compounds

### Average Recovery of C13 Labeled Standards



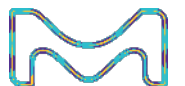
# Successful Starts in LC-MS/MS Summary

- Sample preparation is one of the critical components of successful LC-MS/MS analysis
- With higher particulate matrices, sample preparation needs increase (ex., filtration, SPE)
- **Choosing the right filter for sample preparation depends on many properties:**
  - Material, chemical compatibility, pore size & retention, prefiltration, sample volume
- For PFAS analysis specifically, always consider the known challenges when choosing a consumable for sample preparation:
  - **Risk of contamination**
  - **Risk of analyte loss**



We tested PES (0.2 $\mu$ m and 0.45 $\mu$ m), Nylon (0.2 $\mu$ m) and Nylon with a glass fiber prefilter (0.2 $\mu$ m) for PFAS extractables using a modified EPA 537.1 LC-MS/MS method

**None of the filters demonstrated PFAS extractables**, and are thus appropriate for prepping high particulate samples for PFAS analysis



## Acknowledgements

- Amy Laws
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# PREPARED FOR PFAS TESTING

- <https://sigmaaldrich.com/pfas-testing>
- <http://sigmaaldrich.com/pfassamplefiltration>

