

Fouling Control for RO/NF Systems

April 2009

×15k

0004

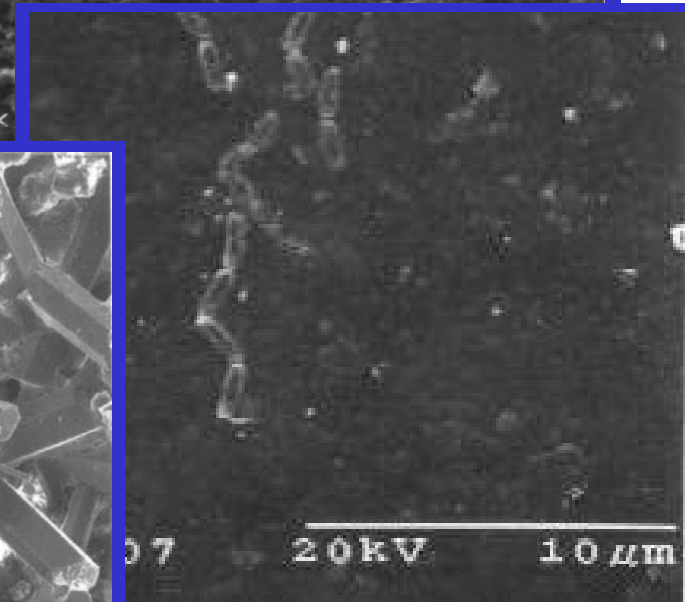
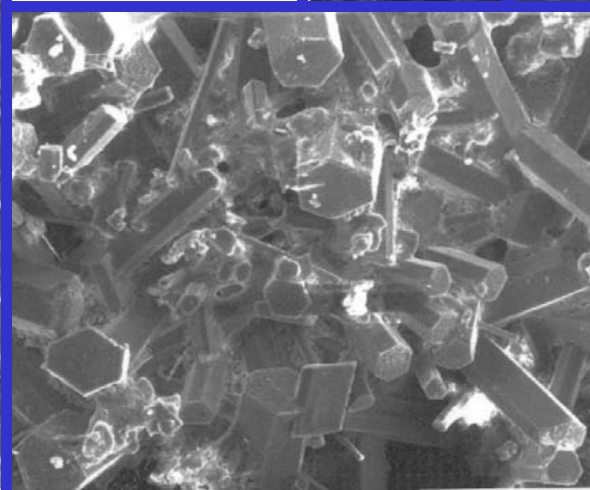
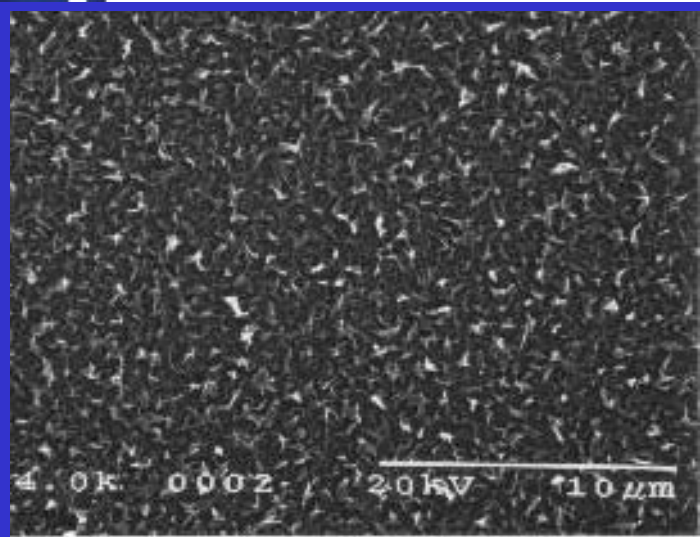
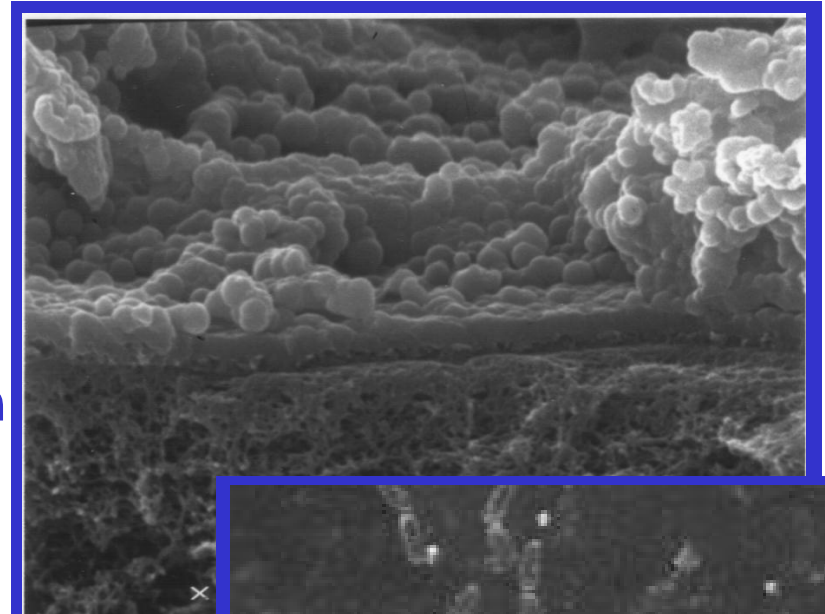
20kV

2 μm

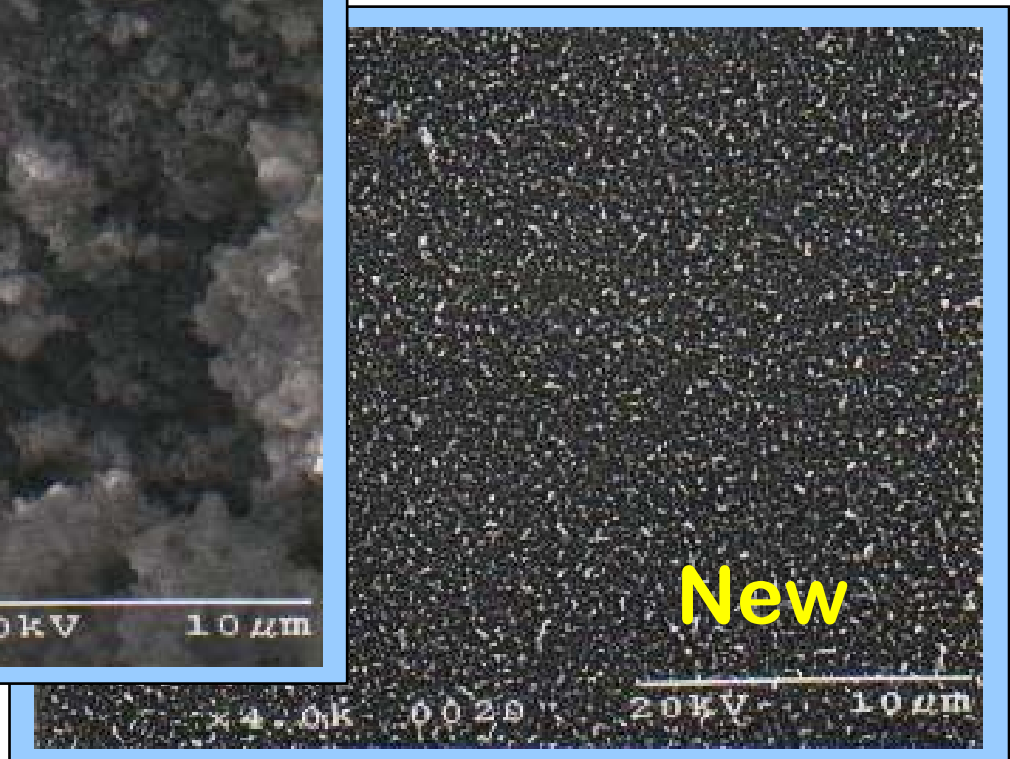
Keys to Treatment with Membrane Technology

Fouling Control

- Colloidal Material
- Biogrowth
- Scaling
- Organic Adsorption
- Coagulant post-precipitation

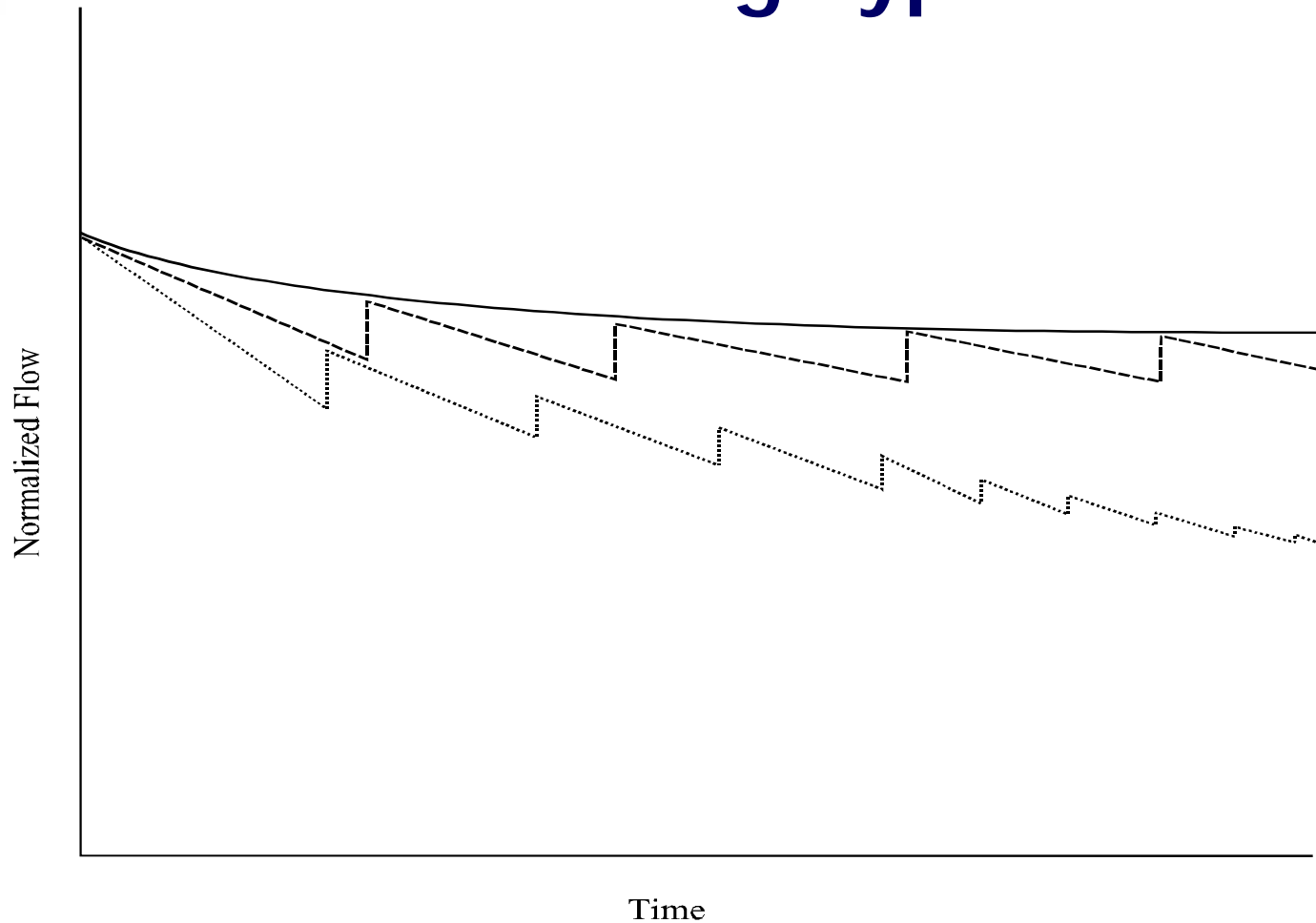


Membrane Fouling



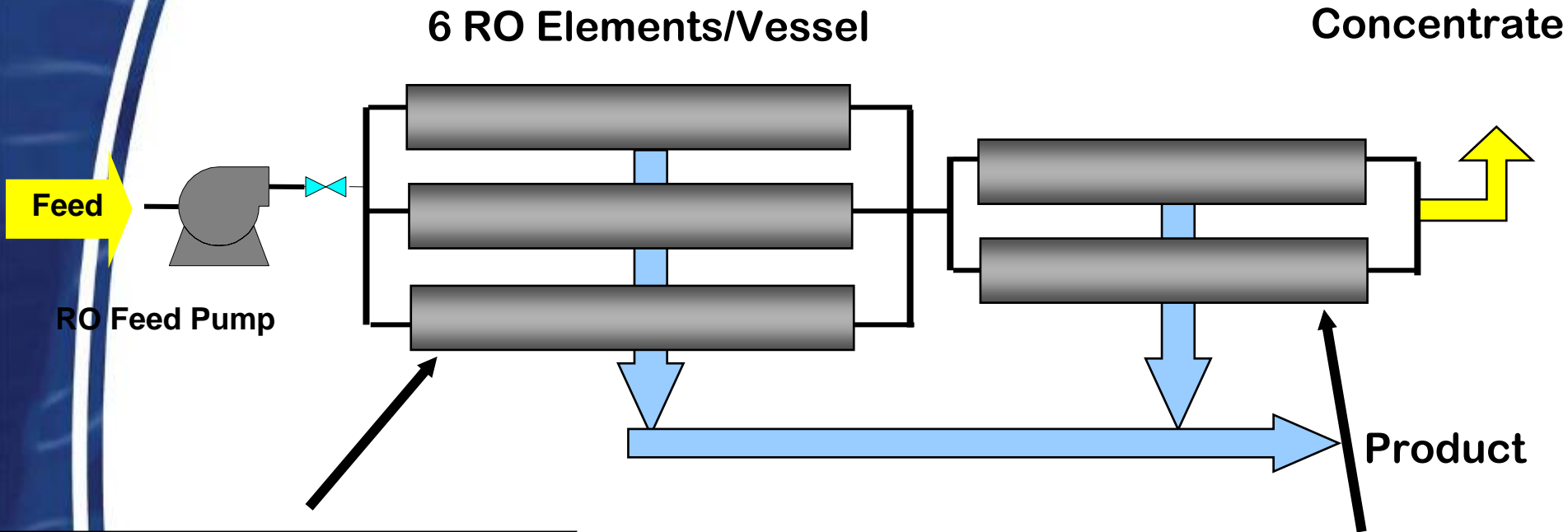
Coating of Membrane Surface
Flux Reduction / High Operating Press
Frequent Chemical Cleaning
Shorter Membrane Life

Fouling Types



- Proper Pretreatment
- - - Marginal Pretreatment
- Inadequate Pretreatment with Frequent Cleaning

Types of Fouling



Lead Element Fouling

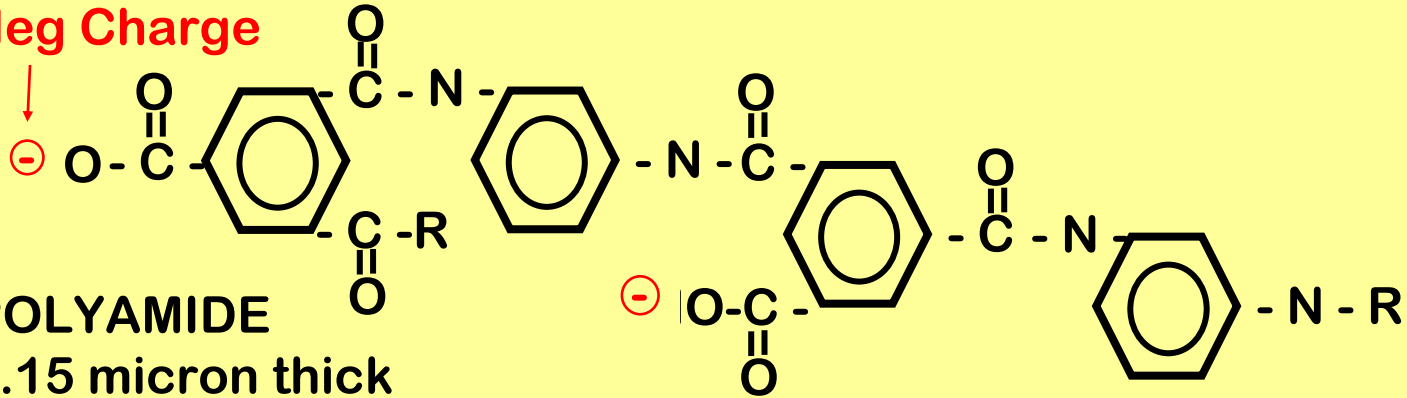
- Colloidal
- Biological
- Dissolved Organics

Tail Element Fouling

- Scaling by Silica, Sulfates, and Carbonates
- Dissolved Organics

Chemical Composition of Typical RO Membrane

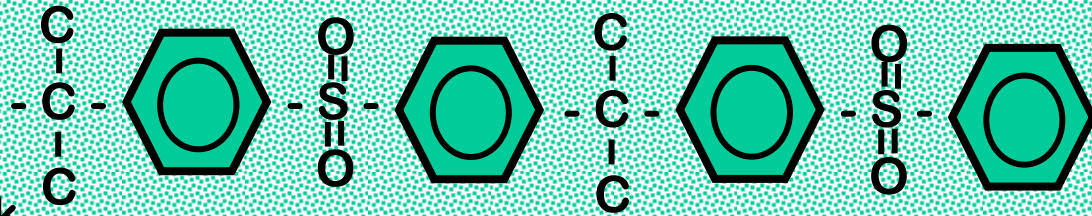
Neg Charge



POLYAMIDE

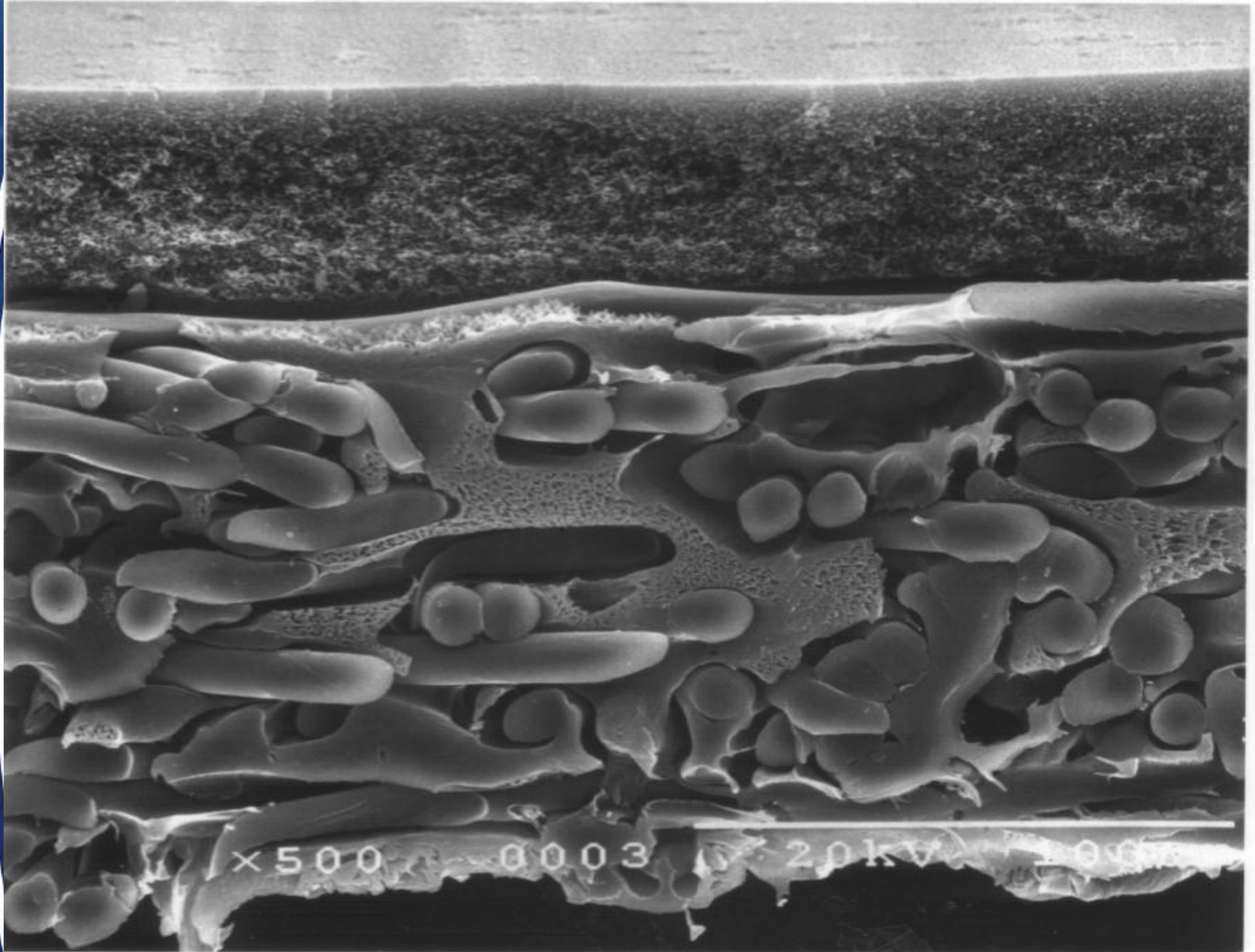
0.15 micron thick

POLYSULFONE
50 microns thick

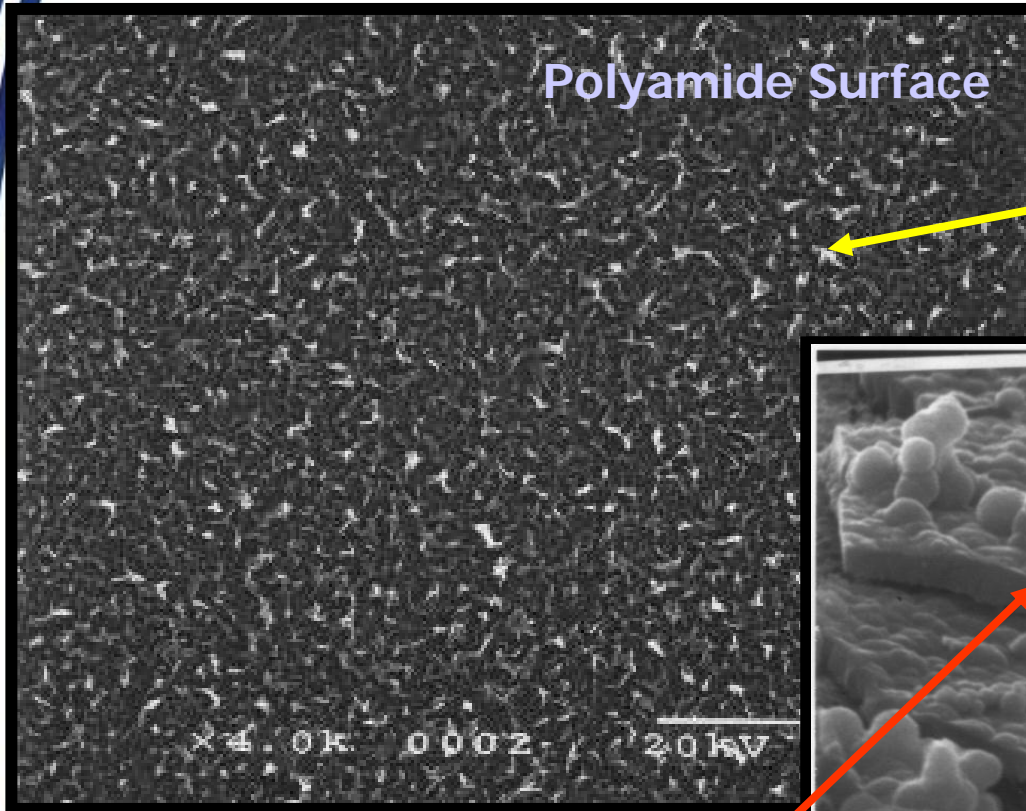


POLYESTER Fabric
150 microns thick

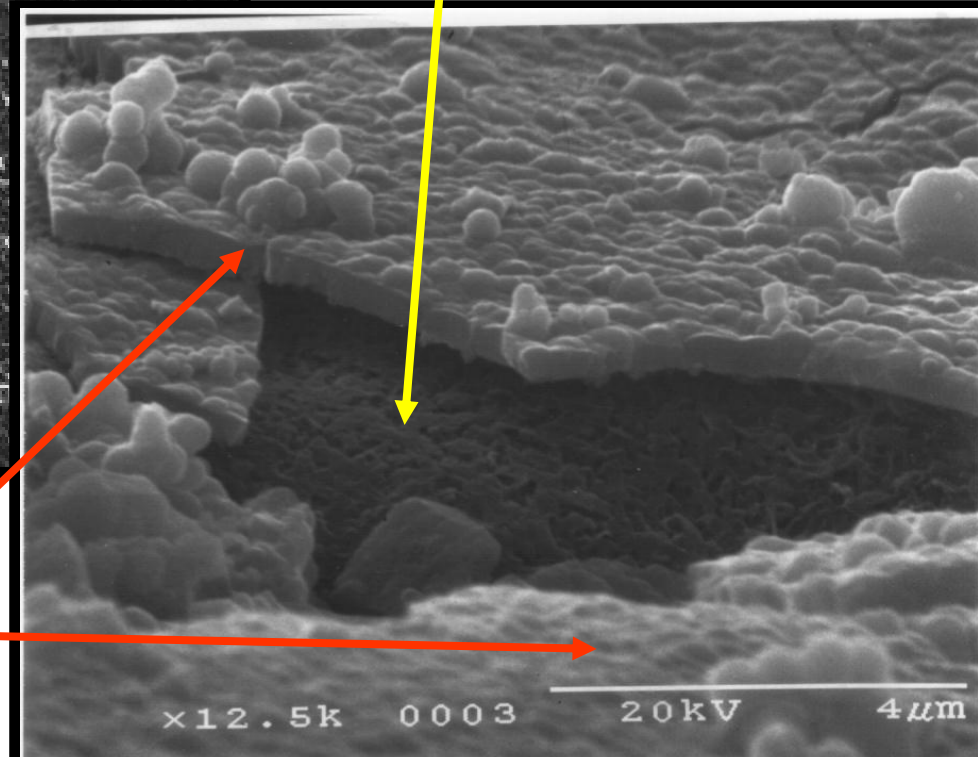
Composite PA Membrane Cross-Section



Keeping RO Membranes Clean

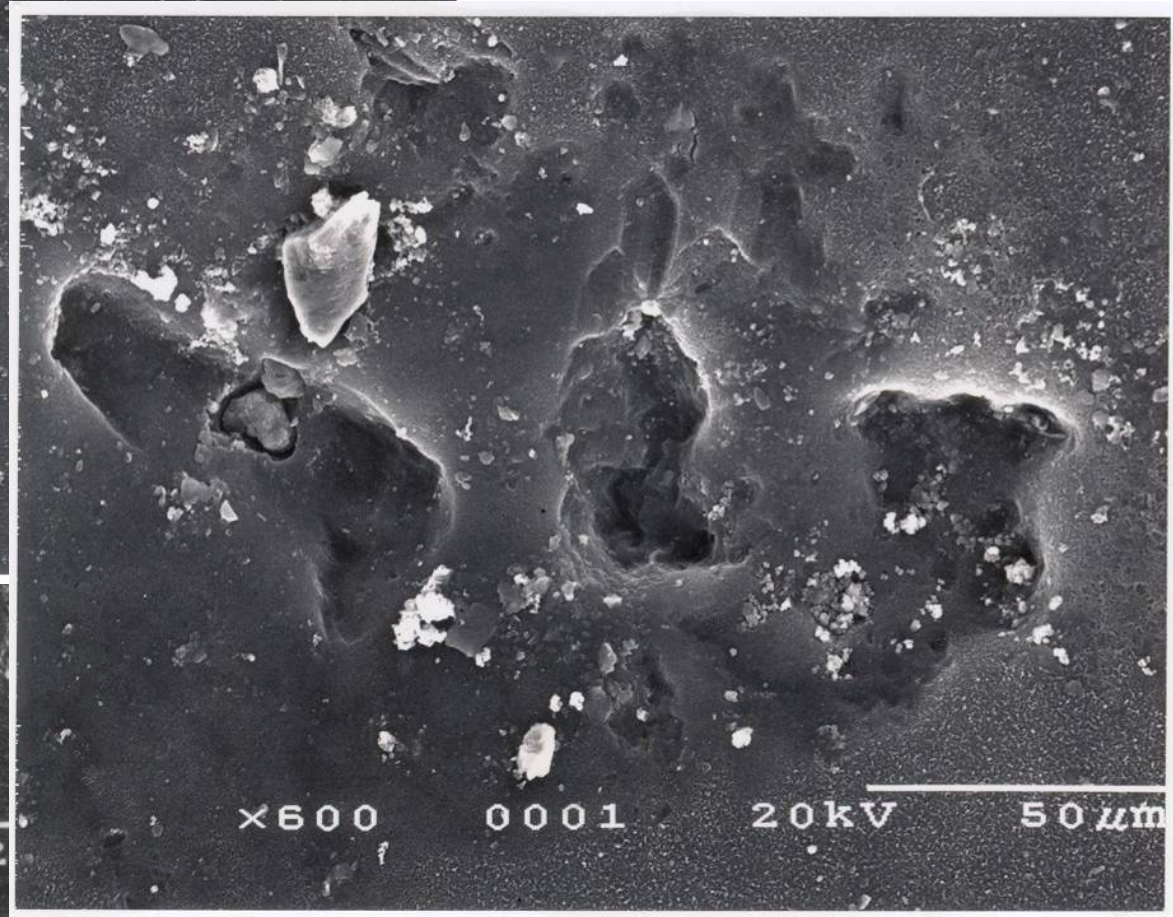
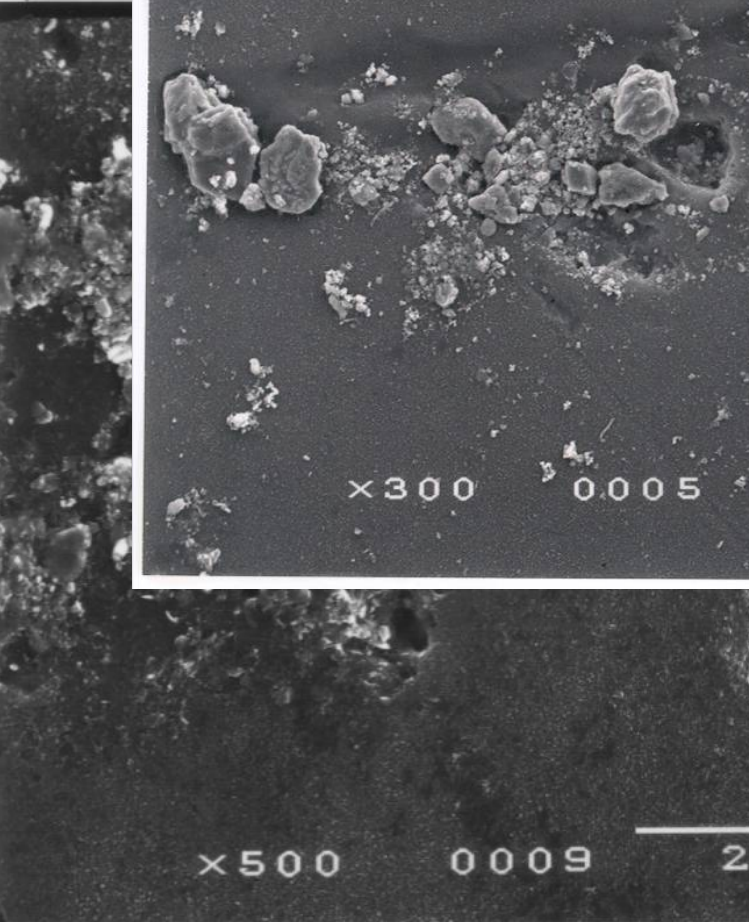
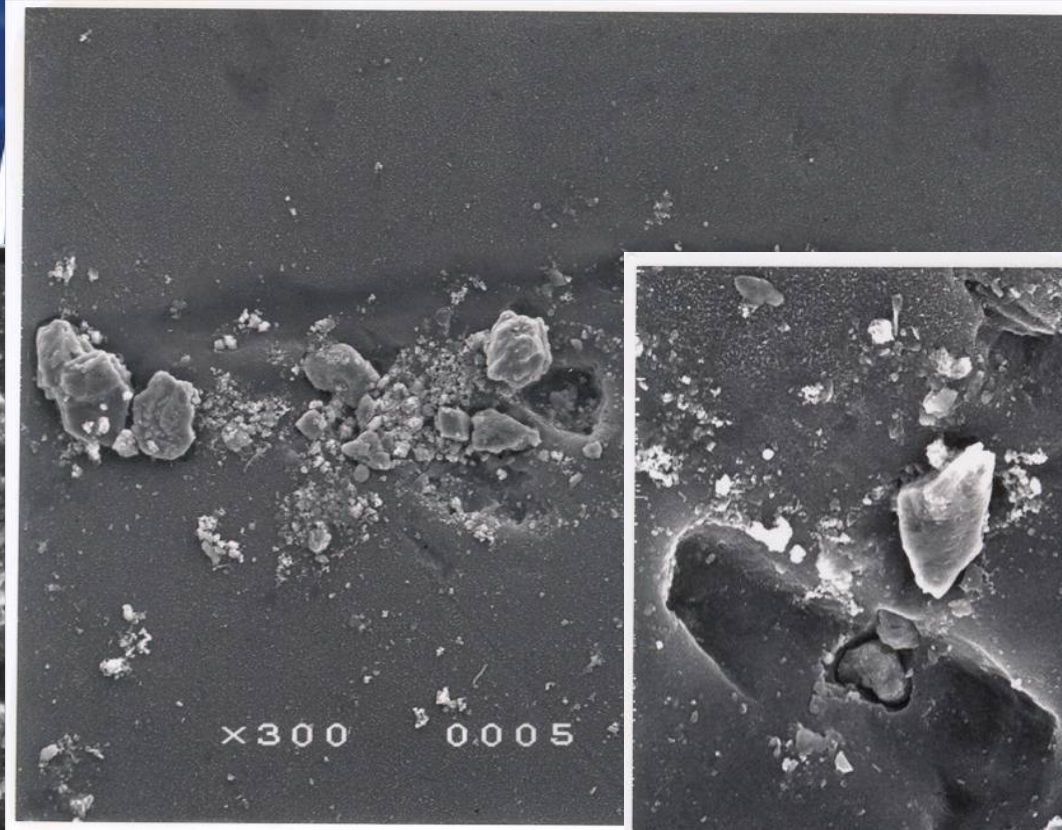


Membrane Surface



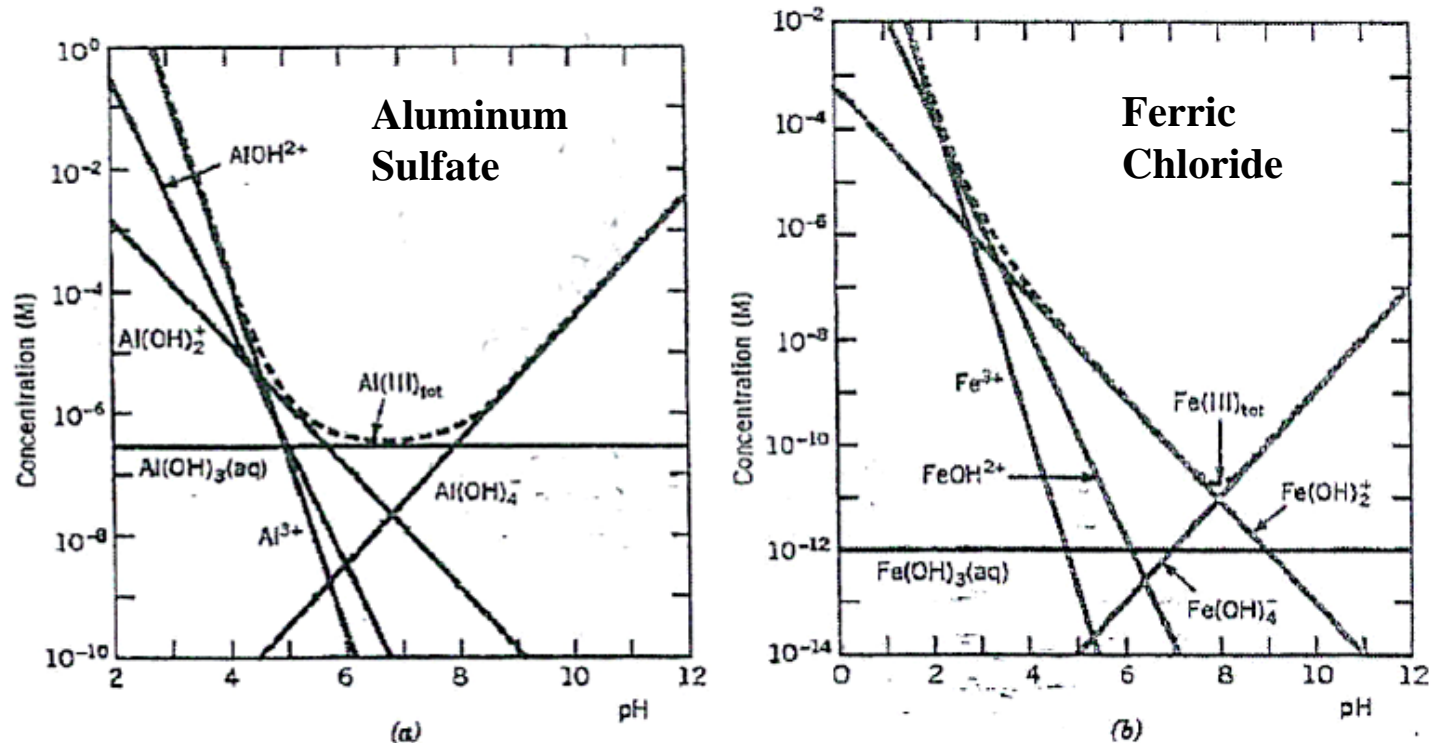
Colloidal Fouling

Surface Damage Caused by Particulate Matter on RO Surface



Coagulant Formation of Colloidal Foulant

C.J. Gabelich et al. / Desalination 150 (2002) 15–30



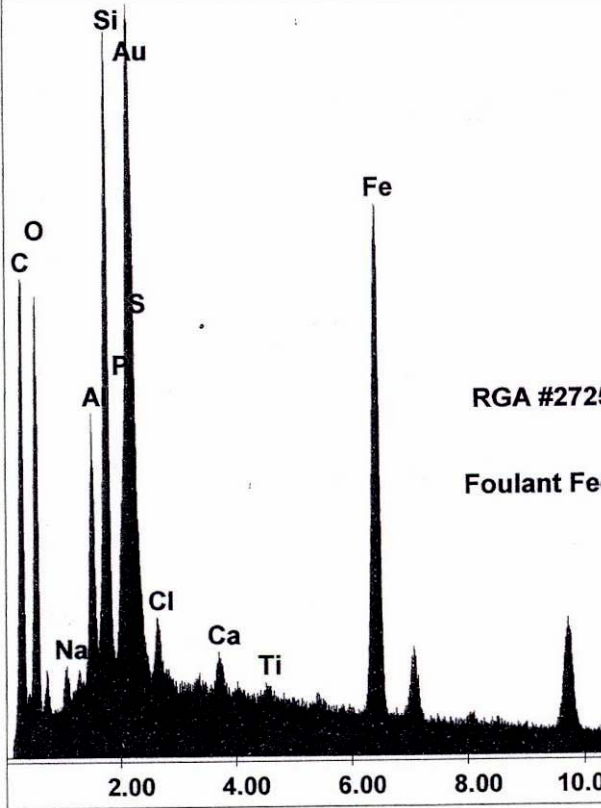
Solubility of amorphous (a) $Al(OH)_3$ and (b) $Fe(OH)_3$. Polynuclear complexes not included [25].



At pH 8.0, the solubility of iron is about 1 ug/l as $Fe(OH)_2^+$ and is much lower than for aluminum, which is 0.5 mg/l as $Al(OH)_4^-$

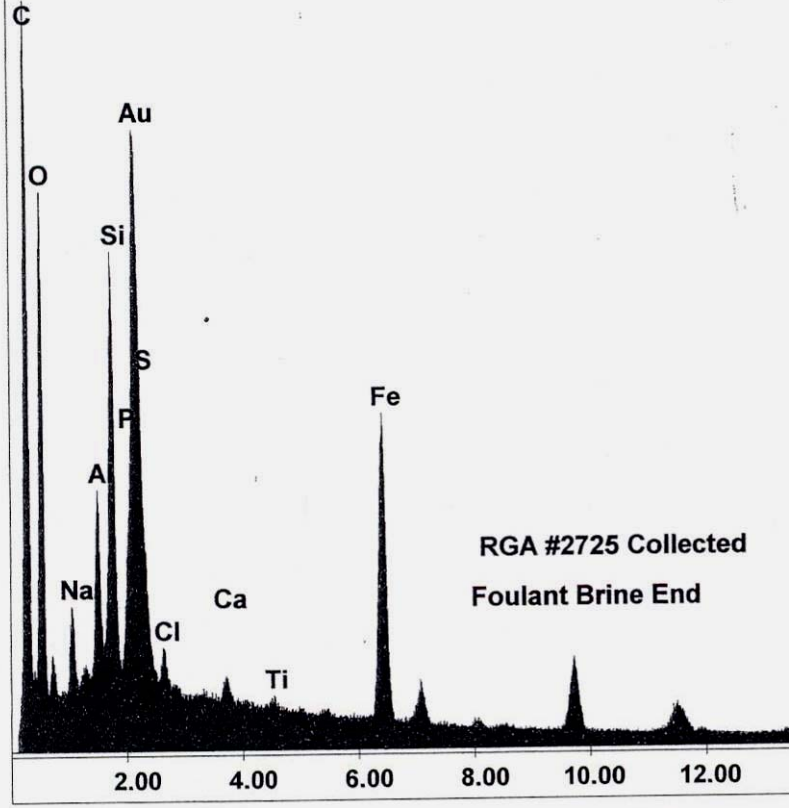


Membrane
Fouling
End



EDAX ZAF Quantification (Standardless)
Element Normalized

Element	Wt %	At %	K-Ratio	Z	A
C K	24.48	49.29	0.0467	1.10	0.2085
O K	17.64	26.66	0.0357	1.08	0.1851
NaK	0.82	0.86	0.0021	1.01	0.2803
AlK	3.05	2.73	0.0149	1.01	0.5220
SiK	6.87	5.92	0.0418	1.03	0.6298
P K	1.86	1.45	0.0120	1.01	0.6953
AuM	24.33	2.99	0.1958	0.74	1.1777
S K	2.03	1.53	0.0111	1.04	0.5874
ClK	1.03	0.70	0.0064	0.99	0.6788
CaK	0.63	0.38	0.0050	1.01	0.8286
TiK	0.32	0.16	0.0027	0.92	0.9068
FeK	16.94	7.33	0.1538	0.93	0.9792
Total	100.00	100.00			



EDAX ZAF Quantification (Standardless)
Element Normalized

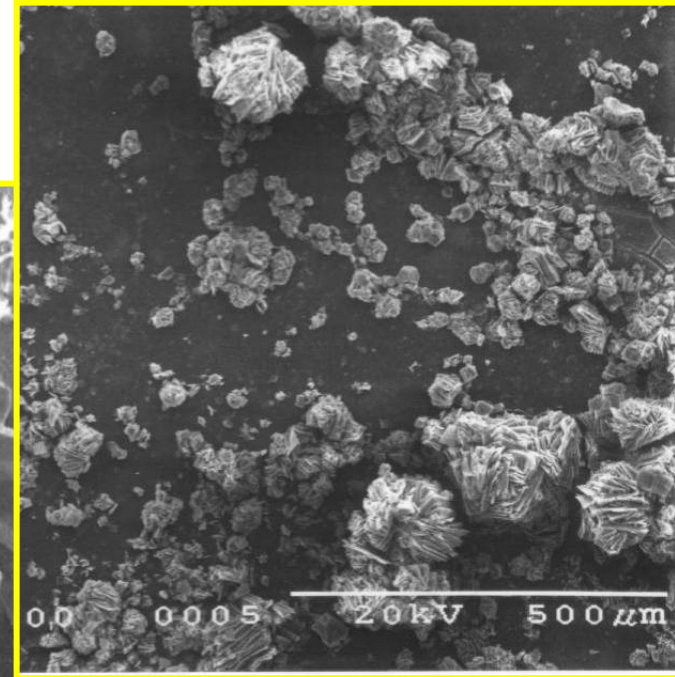
Element	Wt %	At %	K-Ratio	Z	A
C K	33.22	55.73	0.0746	1.0776	0.2085
O K	23.24	29.27	0.0456	1.0594	0.1851
NaK	1.71	1.50	0.0048	0.9912	0.2803
AlK	2.22	1.66	0.0114	0.9860	0.5220
SiK	4.54	3.26	0.0291	1.0146	0.6298
P K	1.36	0.89	0.0094	0.9851	0.6953
AuM	19.98	2.04	0.1695	0.7199	1.1777
S K	1.92	1.21	0.0114	1.0112	0.5874
ClK	0.57	0.33	0.0038	0.9653	0.6788
CaK	0.39	0.20	0.0032	0.9892	0.8286
TiK	0.23	0.10	0.0019	0.9039	0.9068
FeK	10.60	3.82	0.0948	0.9042	0.9792
Total	100.00	100.00			

Particulate on Spiral Wound Element

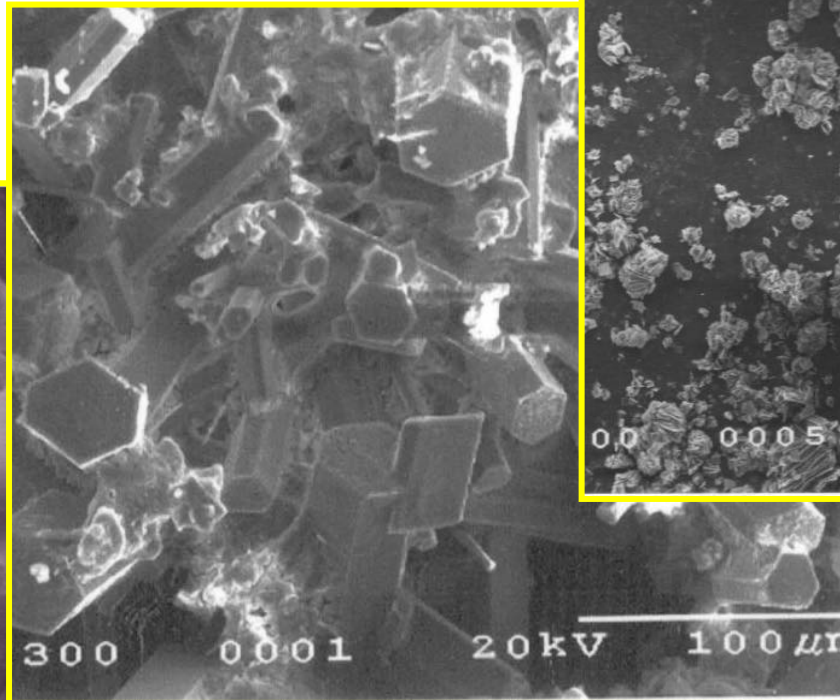


Scaling – Precipitation of Sparingly Soluble Salts

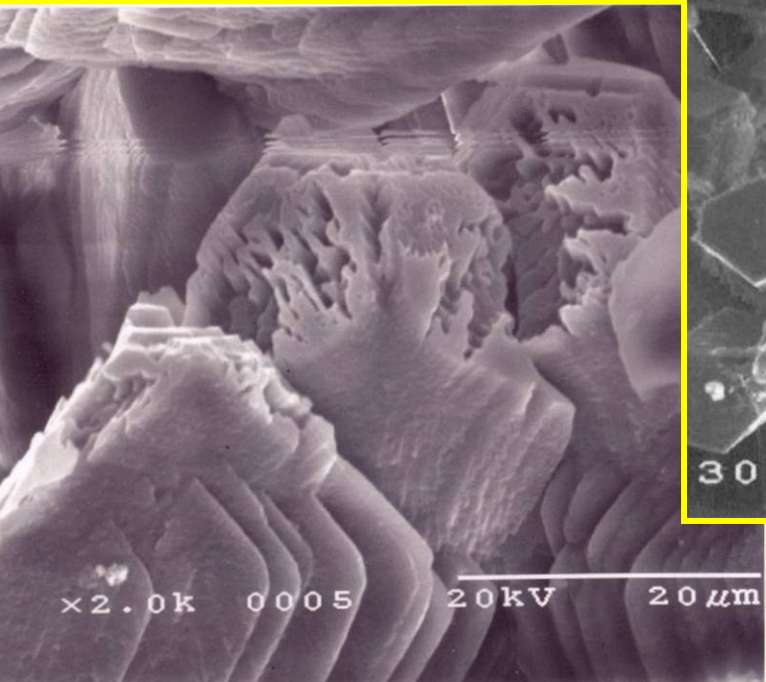
Calcium Phosphate



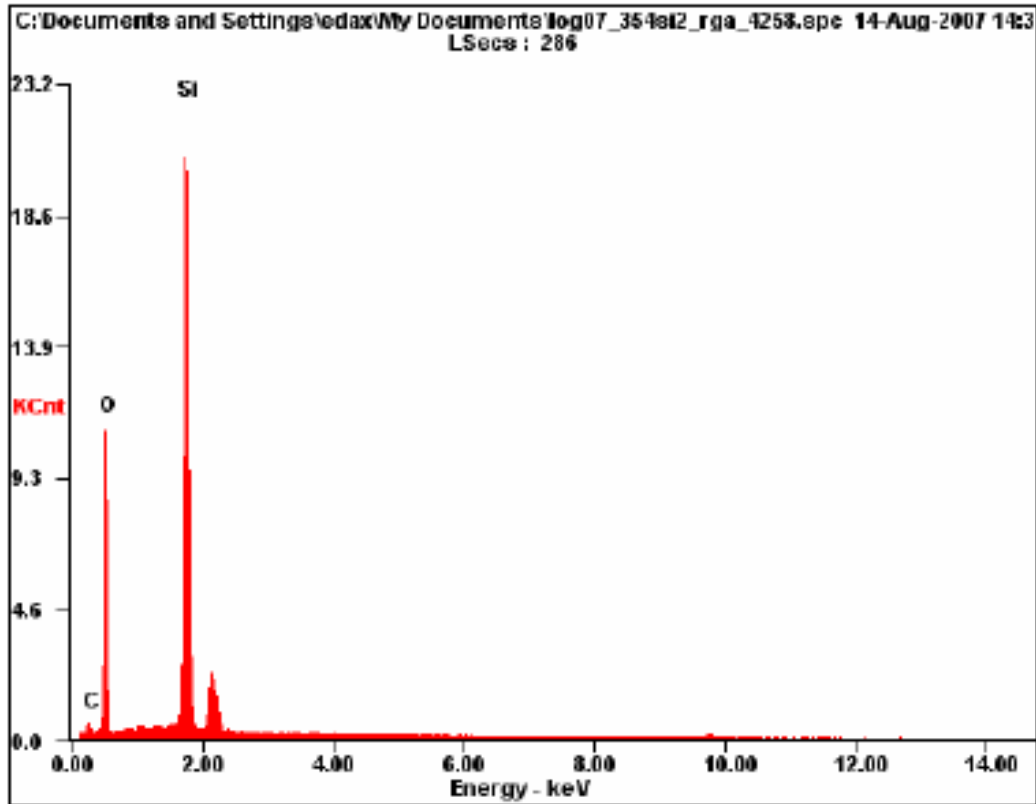
Calcium Sulfate



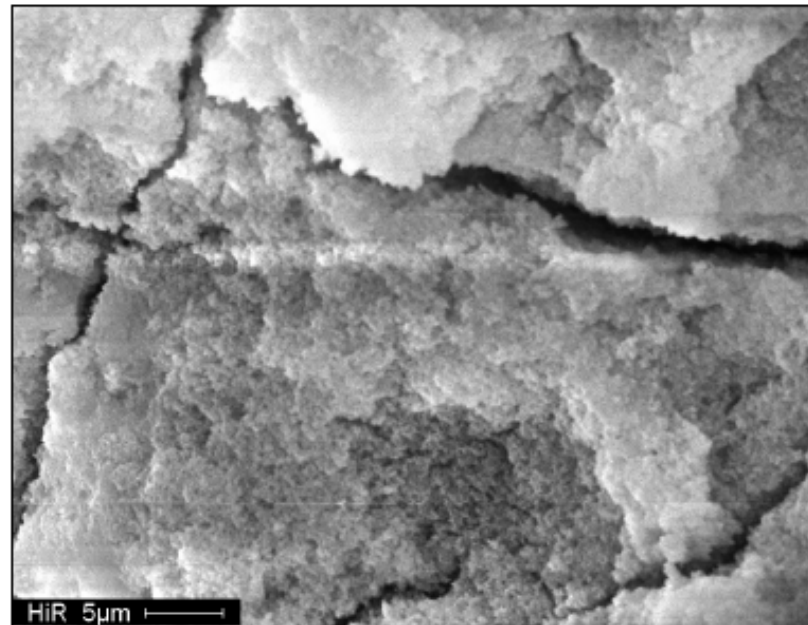
Calcium Carbonate



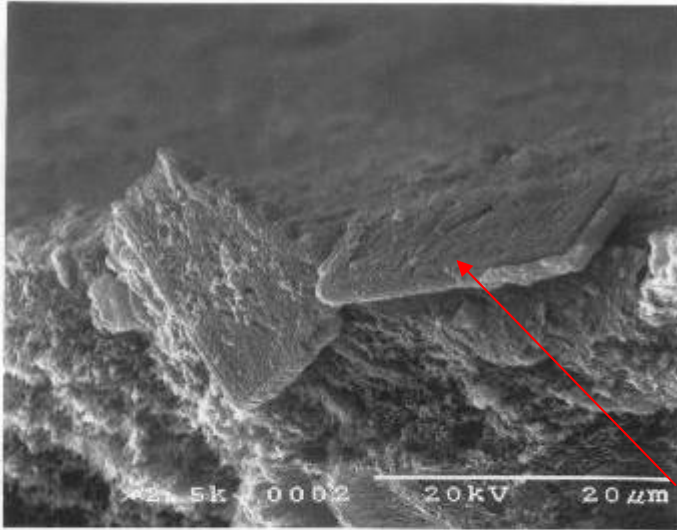
Silica Scale



<i>Element</i>	<i>Wt%</i>	<i>At%</i>
<i>CK</i>	10.22	16.23
<i>OK</i>	44.49	53.02
<i>SiK</i>	45.29	30.75
<i>Matrix</i>	Correction	ZAF



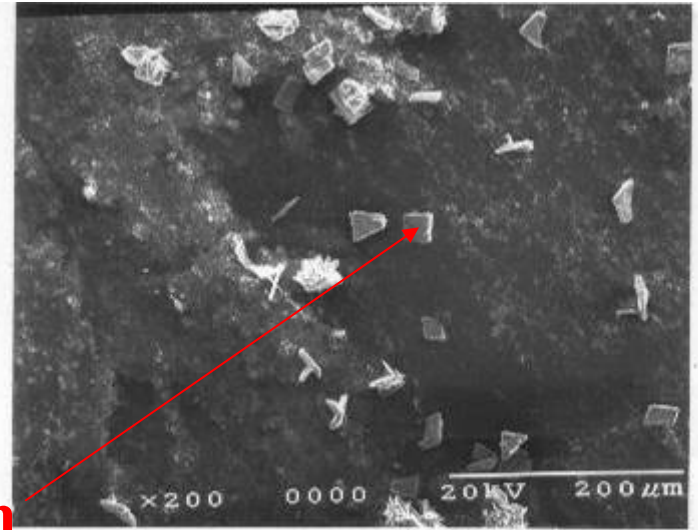
Damage of Membrane Surface by Calcium Phosphate Scale



265-86-2

RGK# 2289

Photo# 2

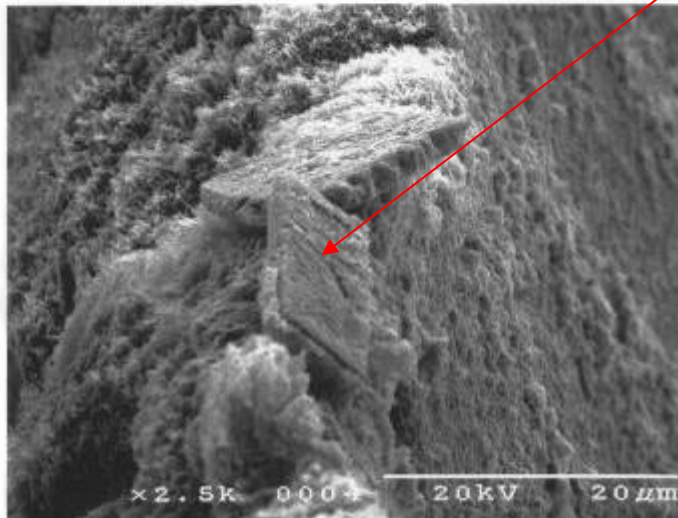


265-81-1

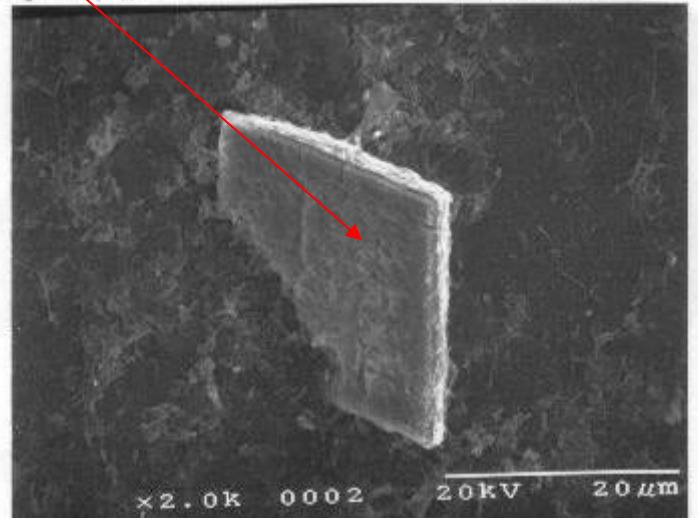
RGK# 2289

Photo# 2

**Calcium
Phosphate**



$\times 2.5k$ 0004 20kV 20 μm

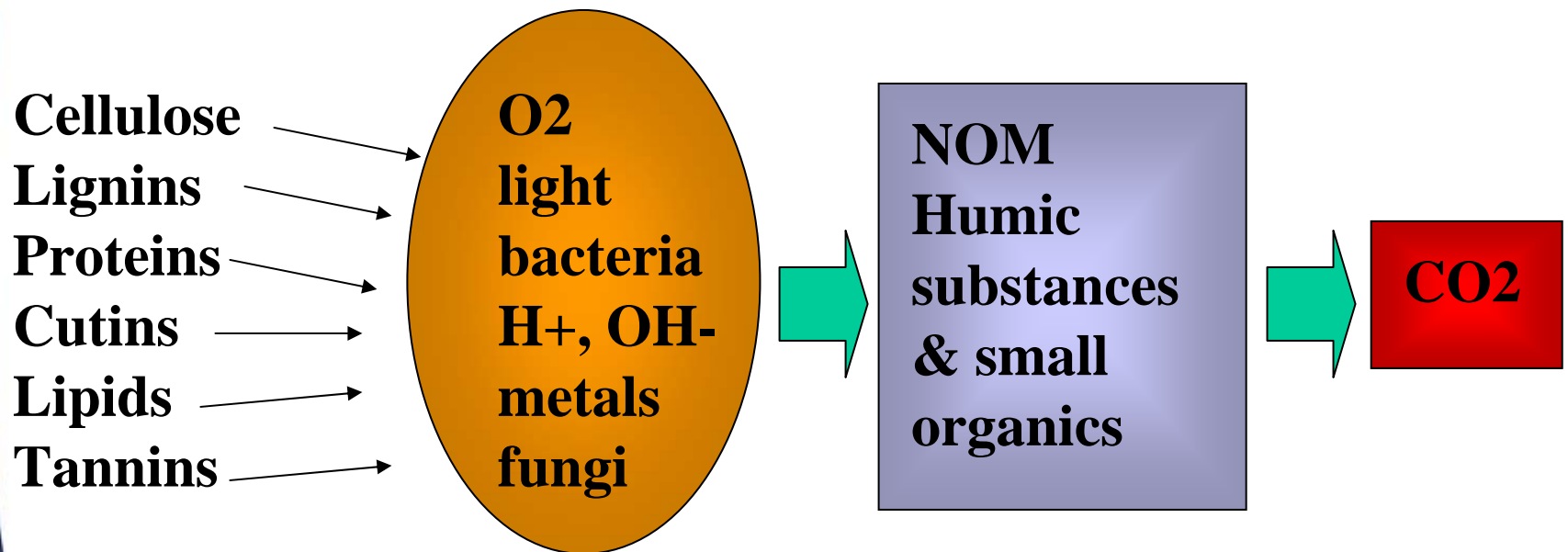


$\times 2.0k$ 0002 20kV 20 μm

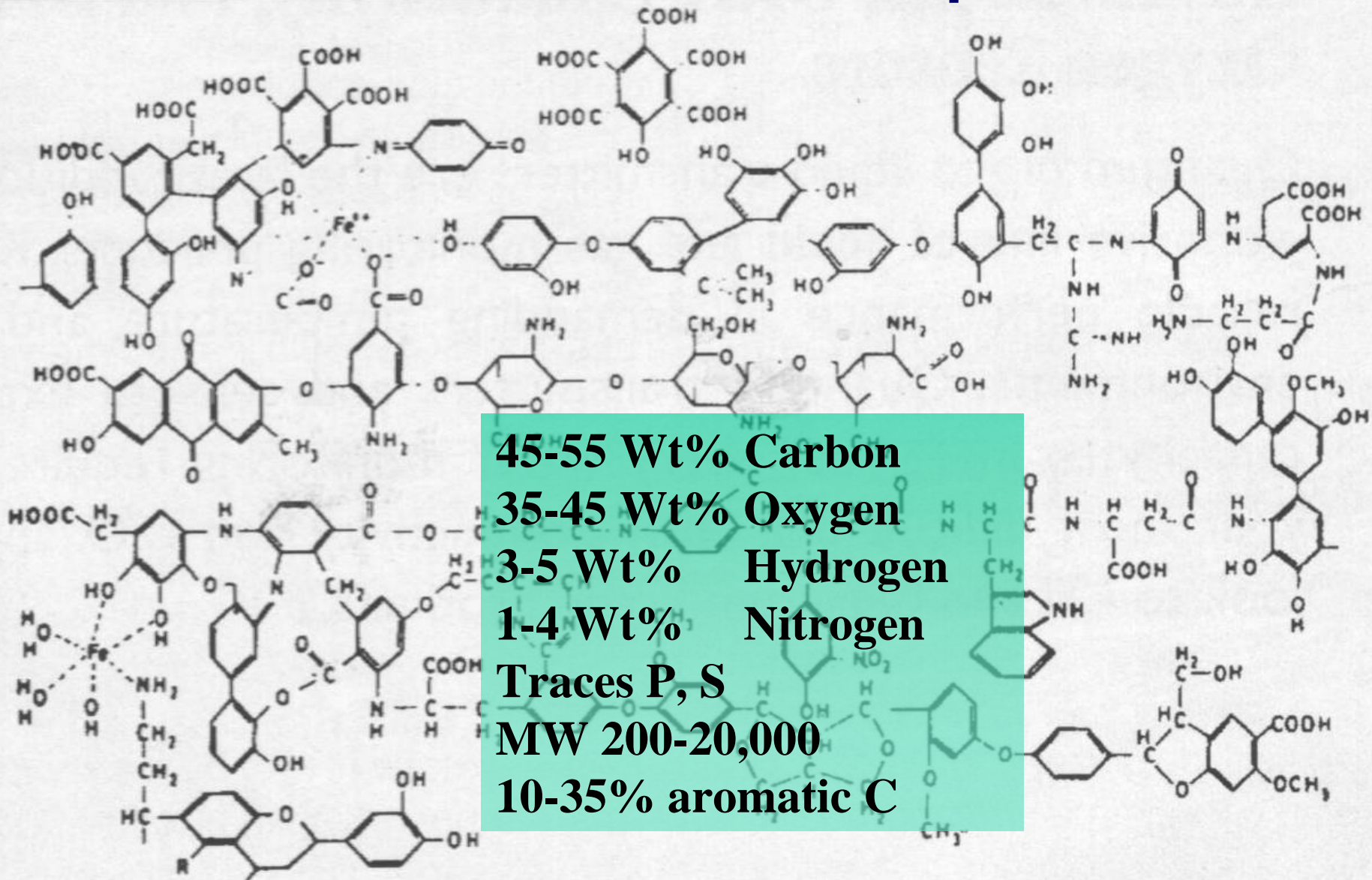
Complex Organics: Naturally Occurring Matter (NOM)

Plant and animal decay products

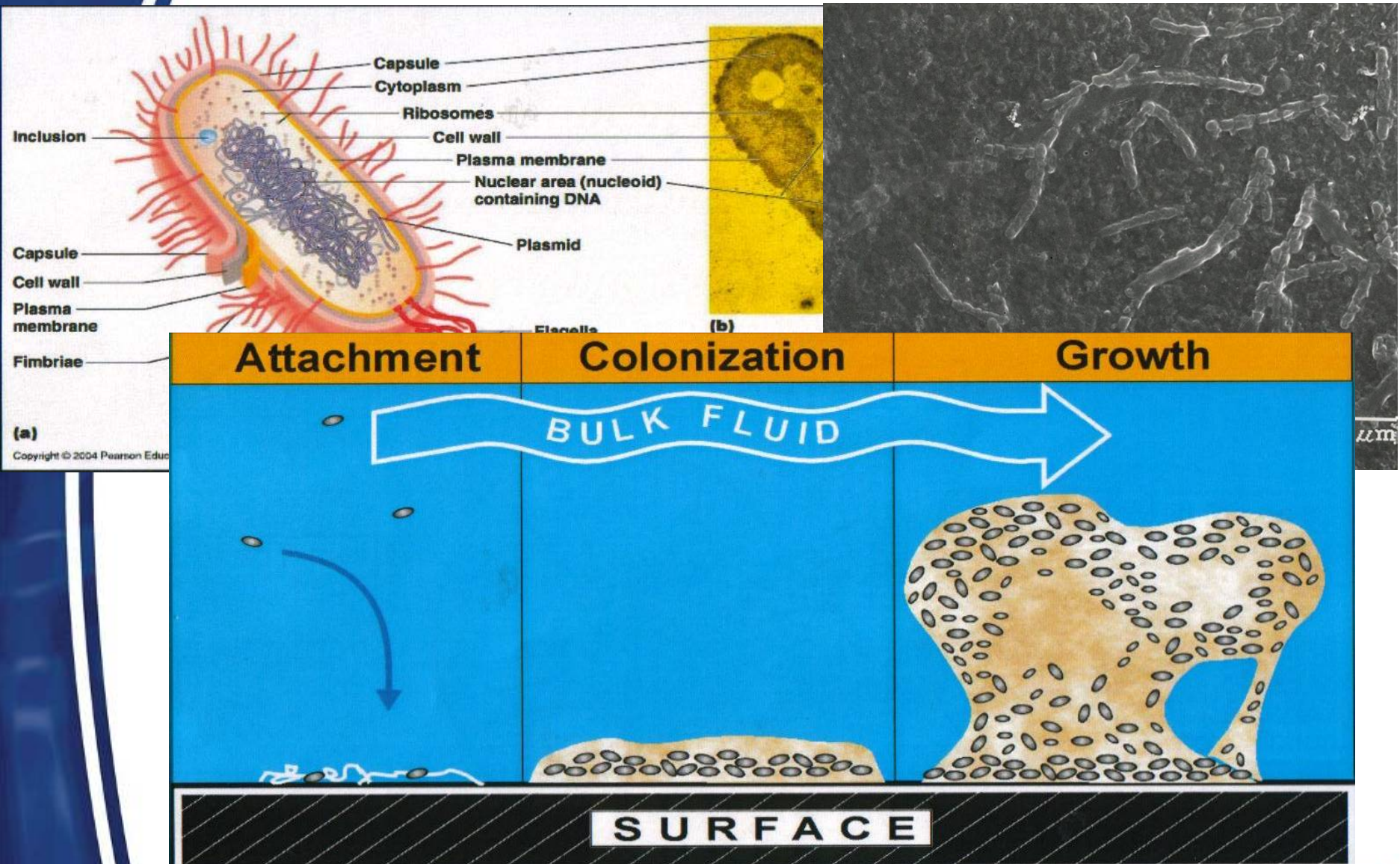
- Terrestrial- woody and herbaceous plants
- Aquatic- algae and macrophytes



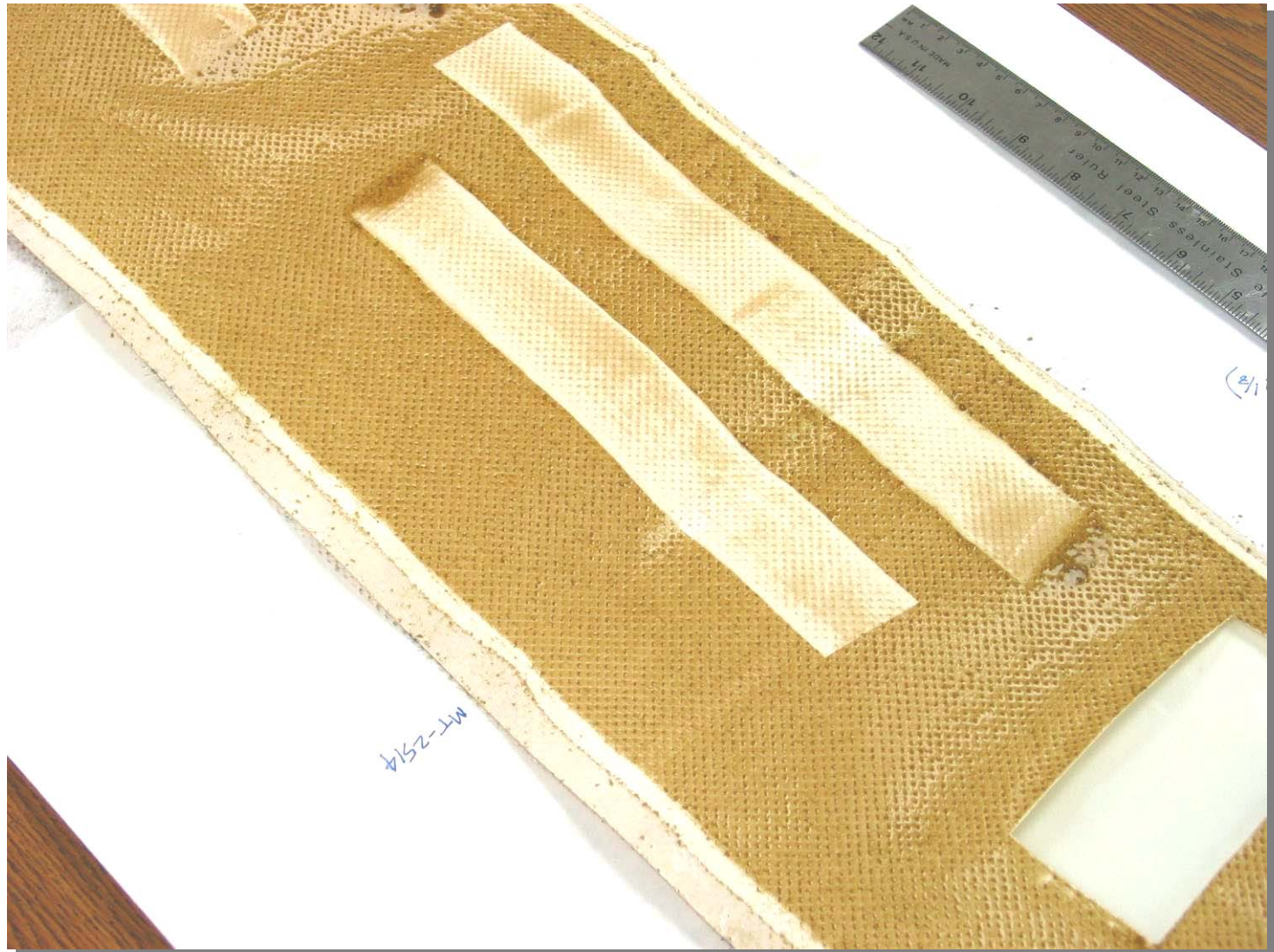
Naturally Occurring Matter(NOM): Structure and Composition



Biofouling



Biofouled Membrane

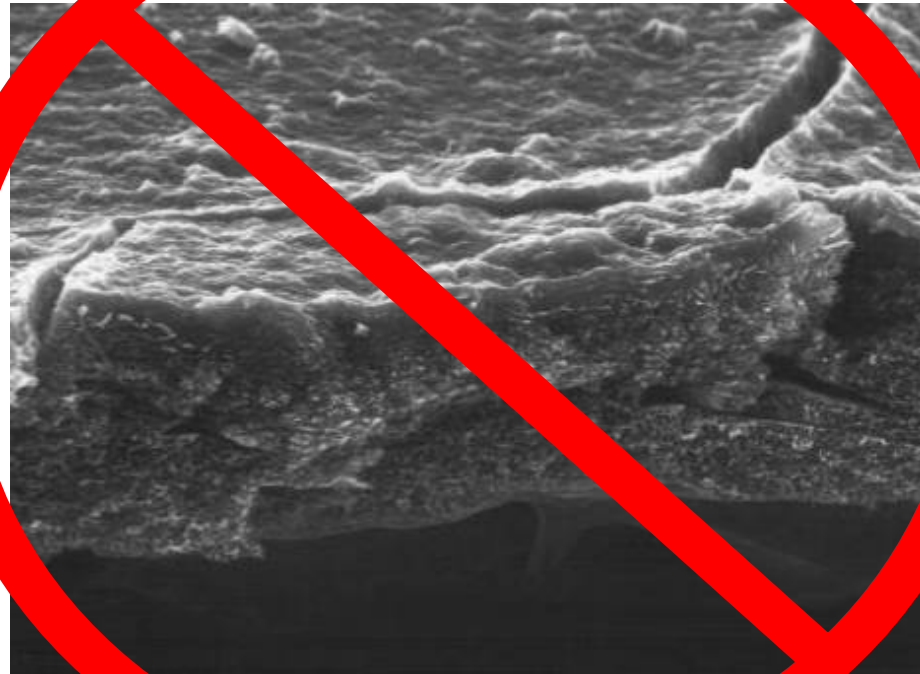


Foulant Distribution in SWRO Pressure Vessel

Position	Ex-fact Rej, %	Ex-fact GPD	Retest Rej, %	Retest GPD	Retest DP, psi
1 (lead)	99.79	5354	99.78	5227	9.5
2	99.70	5126	99.84	5519	5.0
3	99.74	5729	99.81	6013	4.5
4	99.78	5775	99.81	6074	4.0
5	99.80	5427	99.88	5459	3.8
6	99.74	5511	99.83	6171	3.5
7	99.80	6127	99.84	6097	3.0
8 (tail)	99.80	6127	99.85	6074	3.2



Prevention of Fouling



How Can I Prevent Scale Formation in My RO/NF System?

- **Calcium Carbonate**
 - Lower pH, lower recovery, change AS, soften water
- **Calcium Sulfate**
 - Lower recovery, Optimize Antiscalant, soften water
- **Barium Sulfate**
 - Lower Recovery, change AS, soften water
- **Calcium Phosphate**
 - Lower pH, change AS, lower recovery
- **Silica**
 - Increase pH, optimize AS, lower recovery

Recovery Limitations: Scaling

Hydranautics (warnings in program)

Saturation Limits:

CaSO₄ 230 %

SrSO₄ 800 %

BaSO₄ 6000 %

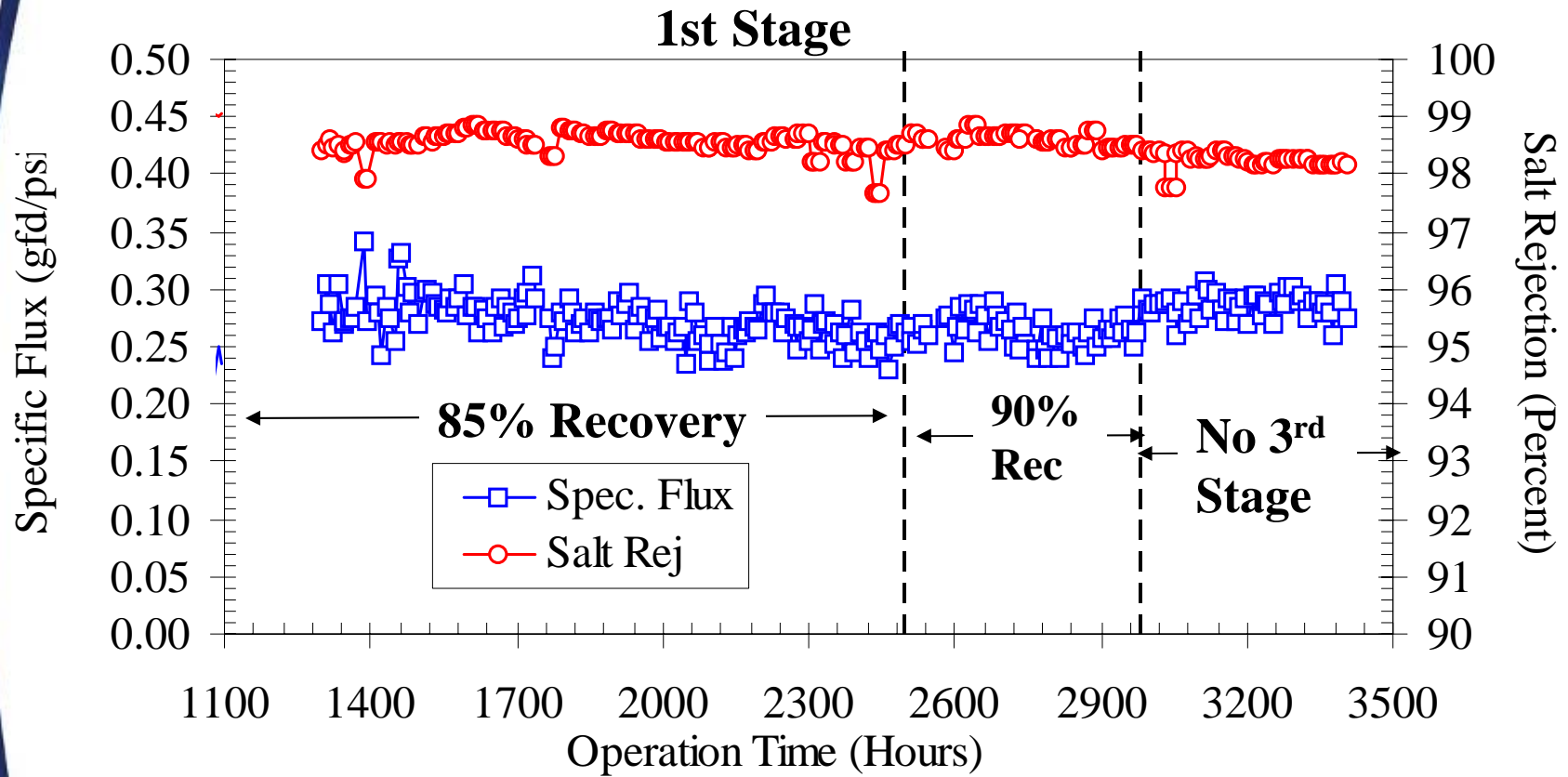
SiO₂ 100 %

LSI <1.8 (Brackish Feed w. Scale Inhibitor)

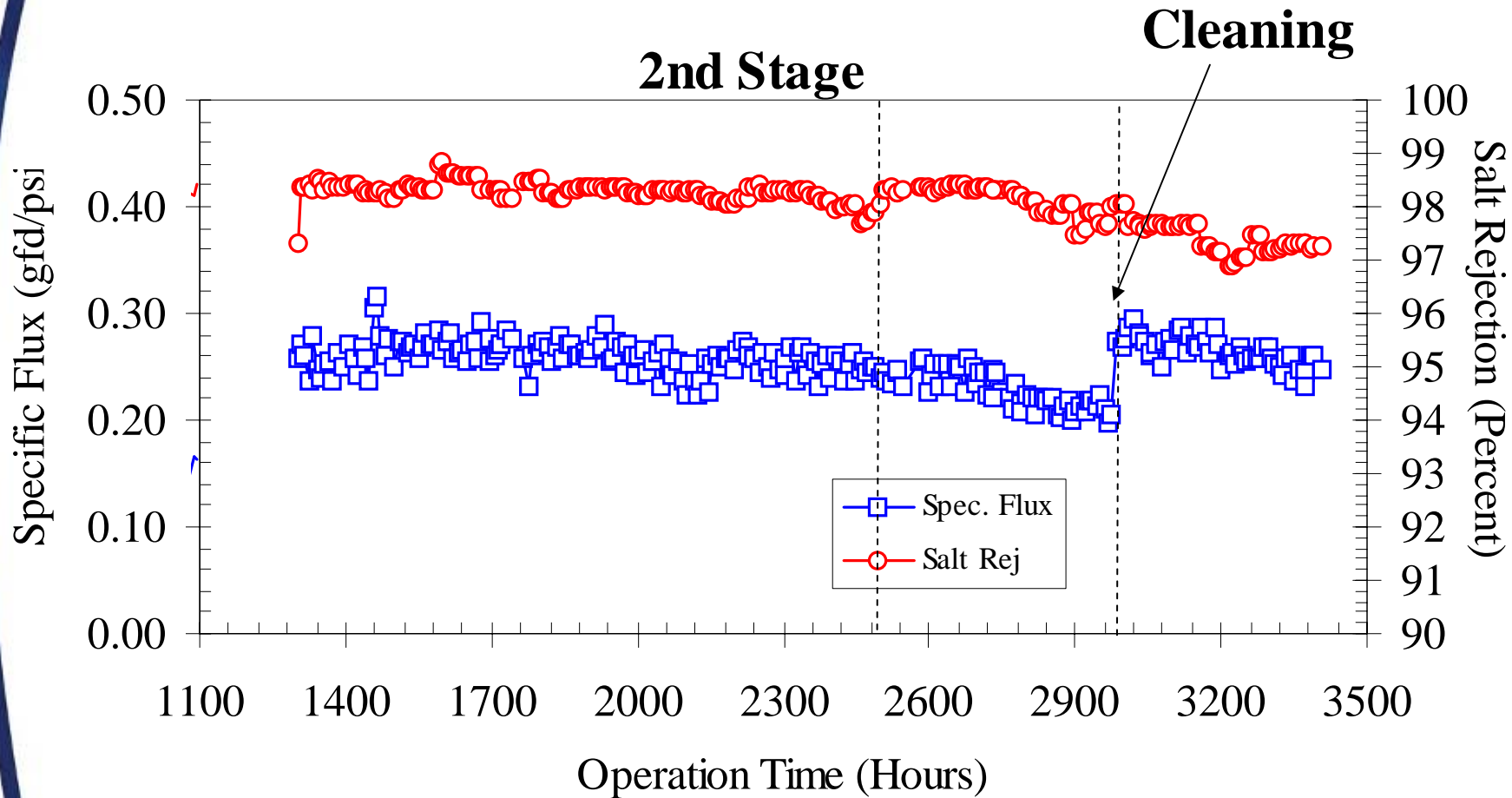
Proprietary Chemicals (software provided by vendors)

	Product A	Product B
CaSO ₄	350%	250%
Sr ₂ SO ₄	3500%	3000%
Ba ₂ SO ₄	10500%	6500%
SiO ₂		240 mg/L
CaCO ₃	L.S.I. 3.0	L.S.I. 2.5
CaF ₂	1300000%	10000%

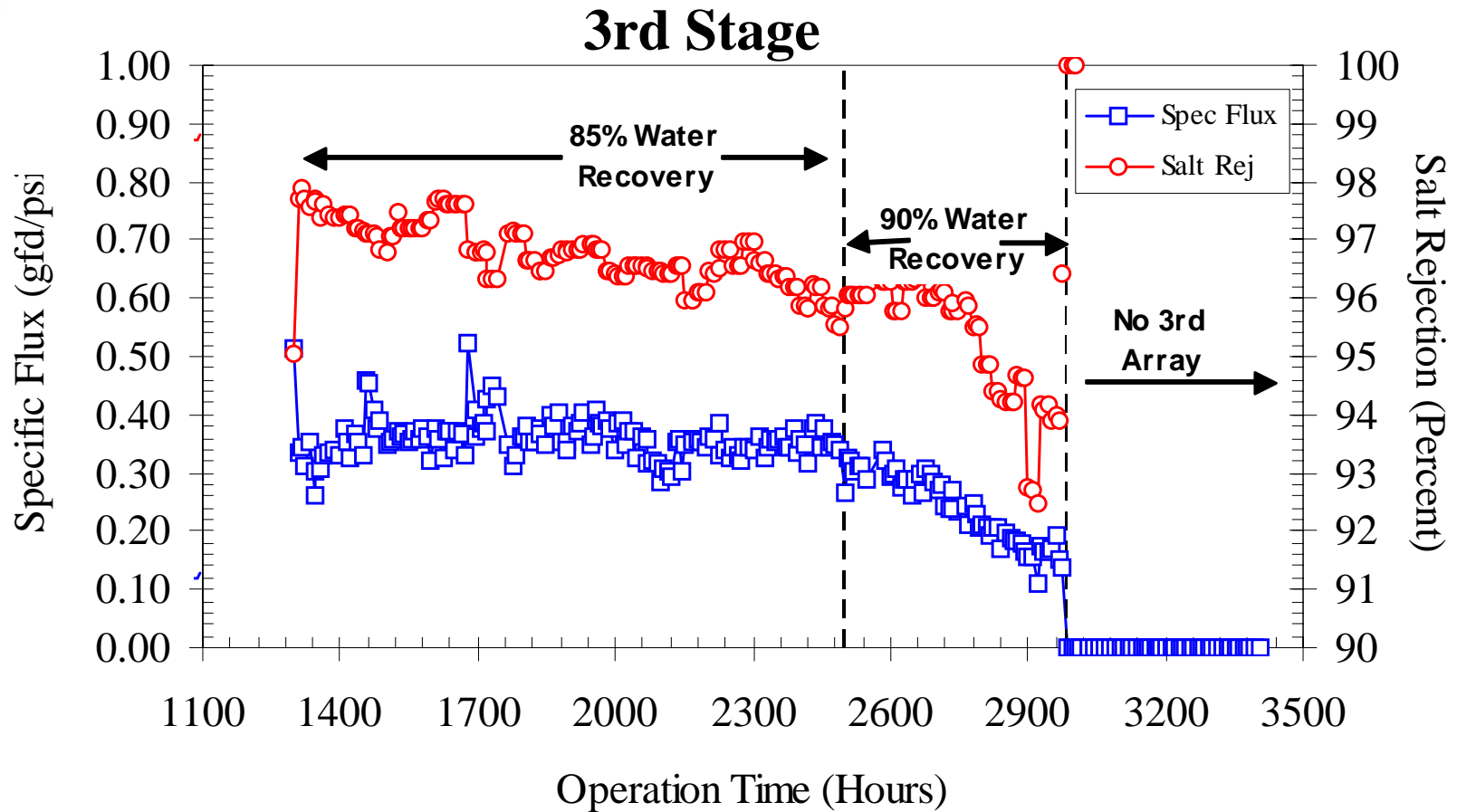
Example of Normalized Data Showing Scale Formation



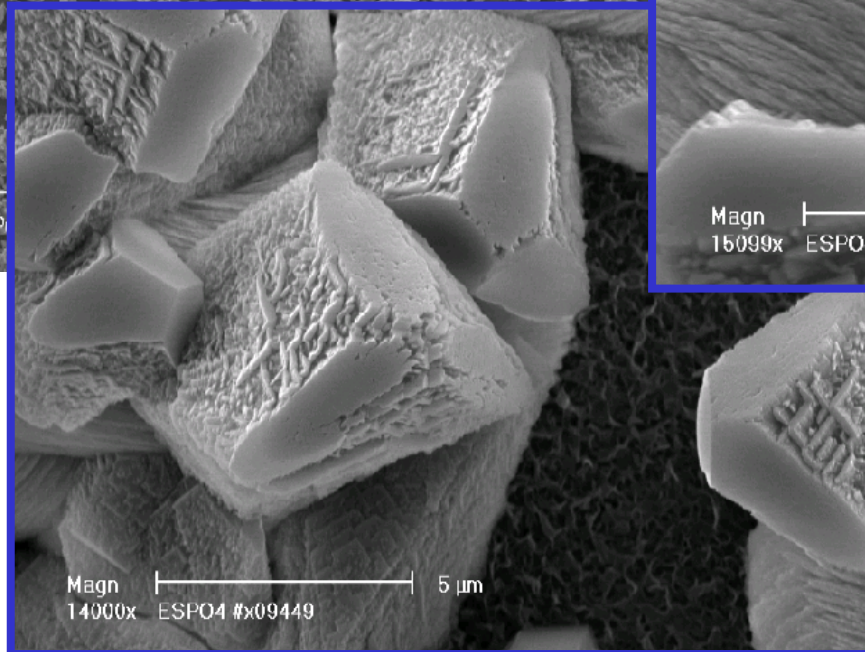
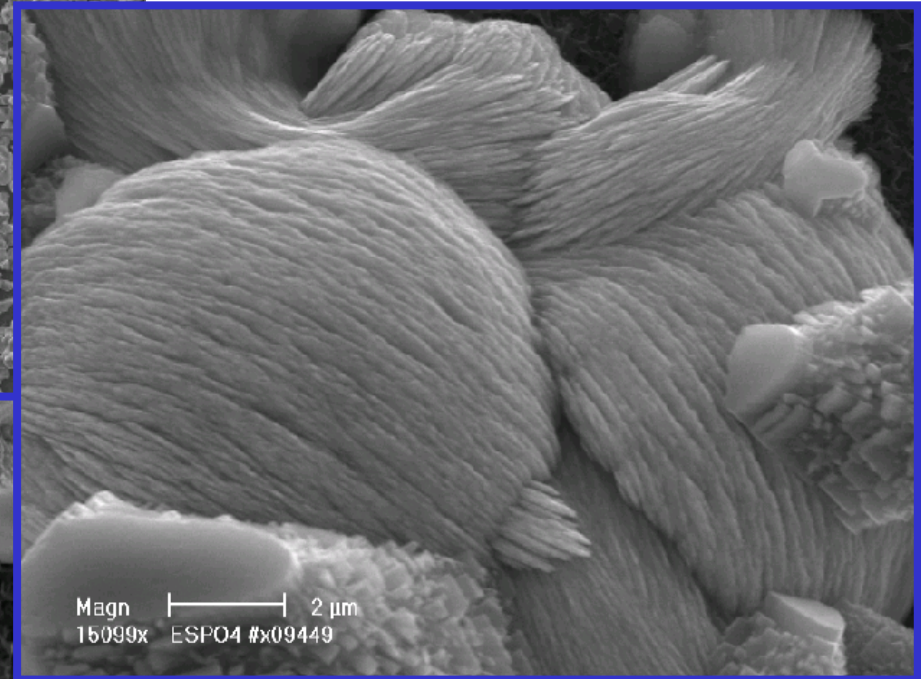
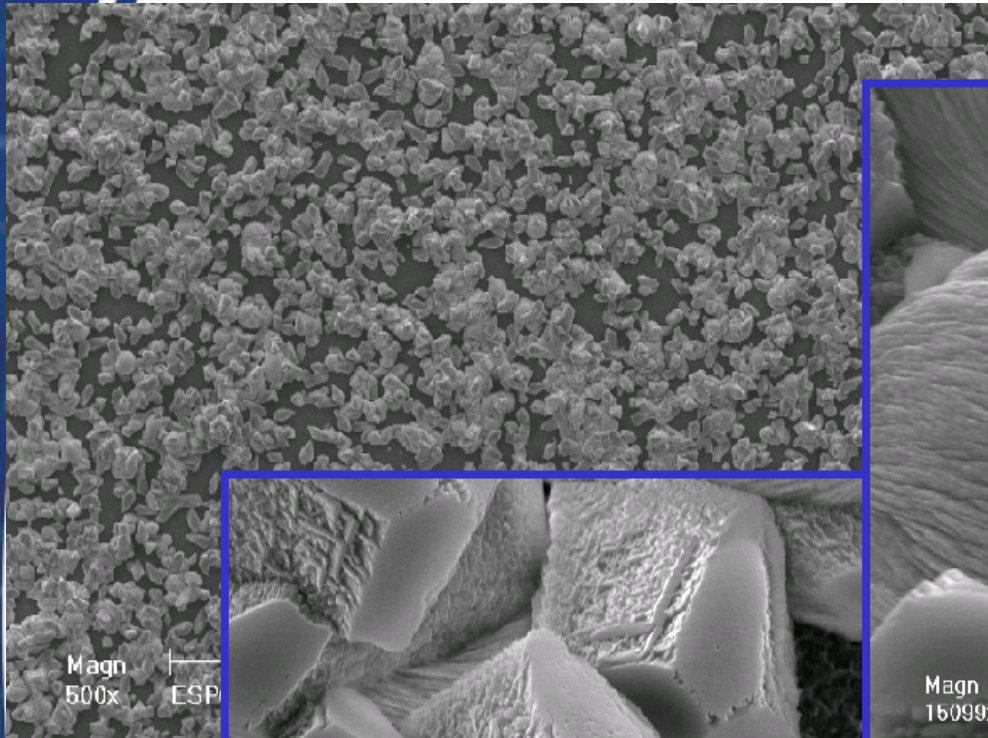
Example of Normalized Data Showing Scale Formation



Example of Normalized Data Showing Scale Formation



Third Stage Element Showing Scale Formation



Barite

Calcite

How Can I Reduce Colloidal Fouling in My RO/NF System?

- **Measurement**
 - **SDI (< 3 preferred)**
 - **Turbidity (< 0.1 preferred)**
 - **Particle Counts (<100 of 2 μm particles/ml)**
- **Optimize Your Media Filtration**
 - **Media Selection**
 - **Coagulant Optimization**
 - **Proper Filtration Velocities**

SDI - Silt Density Index

- Test is used to indicate the quantity of particulate matter in water
- SDI is used to determinate effectiveness of pretreatment
- SDI has been empirically correlated with the fouling tendency of membranes
- SDI will vary with filter paper manufacturer
- SDI may vary with water temperature

SDI formula

$$\text{SDI}_T = (1 - T_0/T_T)/T * 100$$

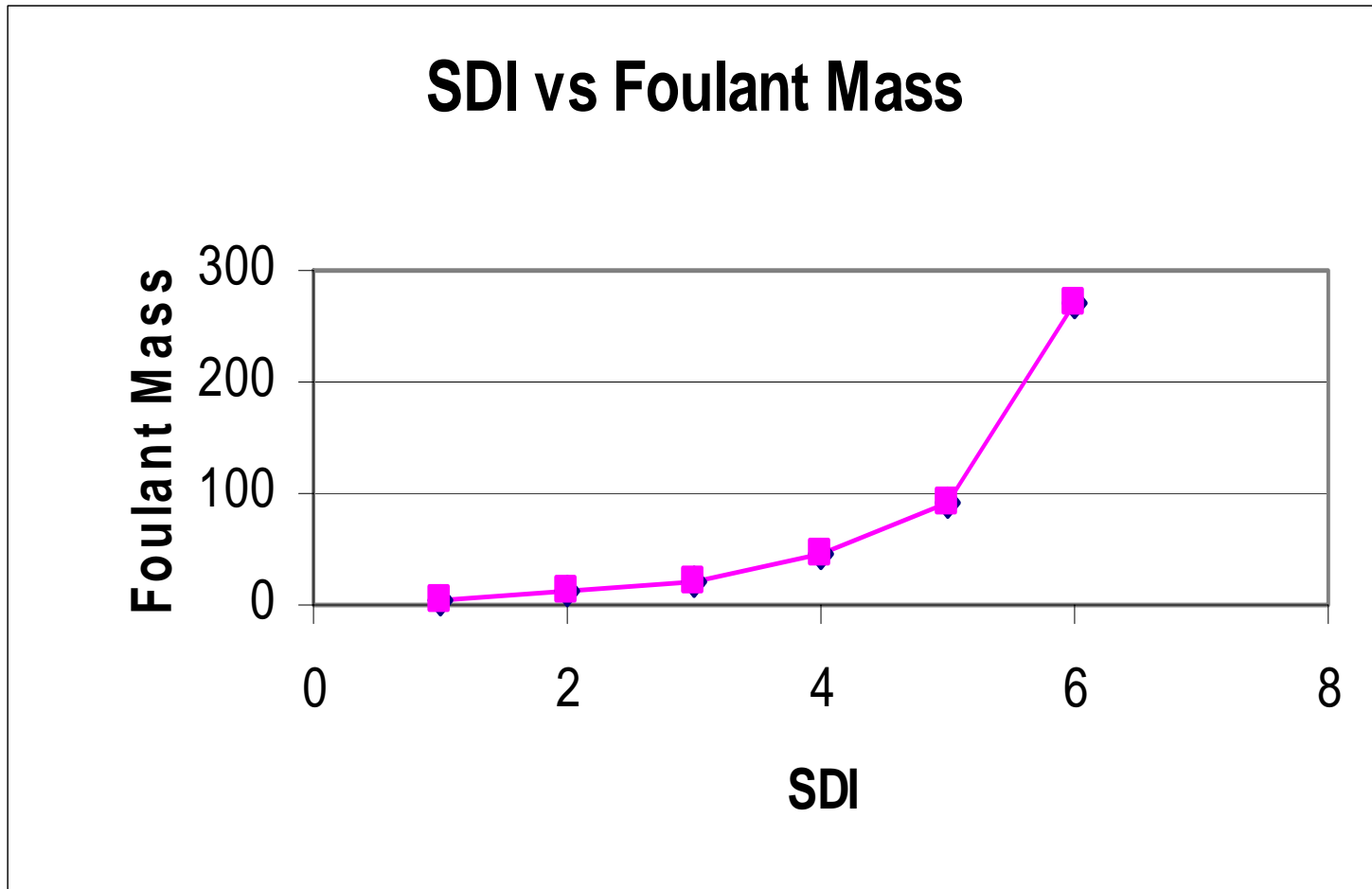
T = Total elapsed flow time (usually 15 min.)

T₀ = Initial time required to collect 500 ml of sample

**T_T = Time required to collect 500 ml of sample after test time T
(usually 15 min.)**

Test should be done at 207 kPa ± 7 kPa

How Can I Reduce Colloidal Fouling in My RO/NF System?



“Silt Density Indices (SDI), percent plugging factor (%PF); their relation to actual foulant deposition”, Seymour S.Kremen, Matt Tanner, Desalination 119 (1998) 259-262

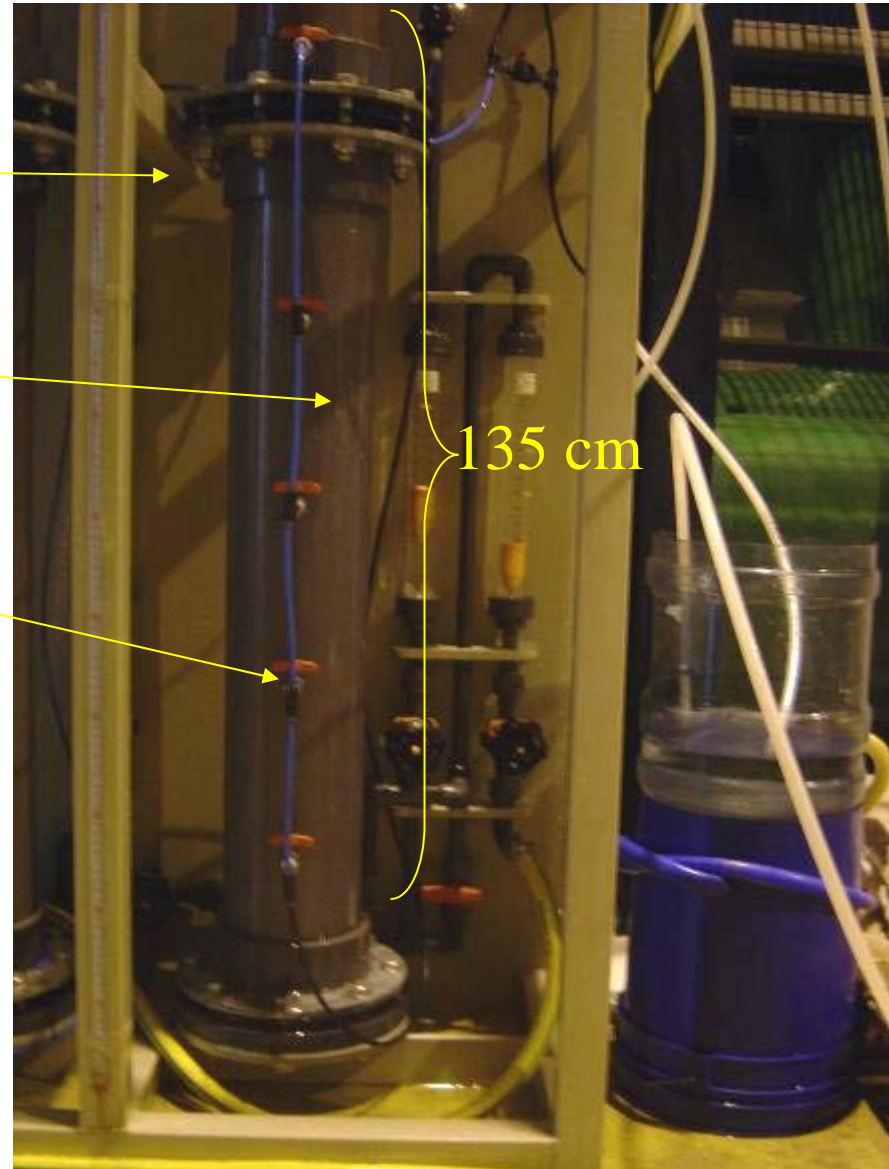
Optimized Media Filtration for Improving Colloidal Foulant Removal

150 mm Tube

Media Layer

Pressure Taps

**Pilot Trials
at the Plant**



Example SWRO Pilot Plant Trials in Mediterranean Region

3 x 150mm columns @ 11 m/hr filtration velocity

- **Col #1: New Media, same coagulant as plant**
- **Col #2: Existing Media, new coagulant**
- **Col #3: New Media, new coagulant**

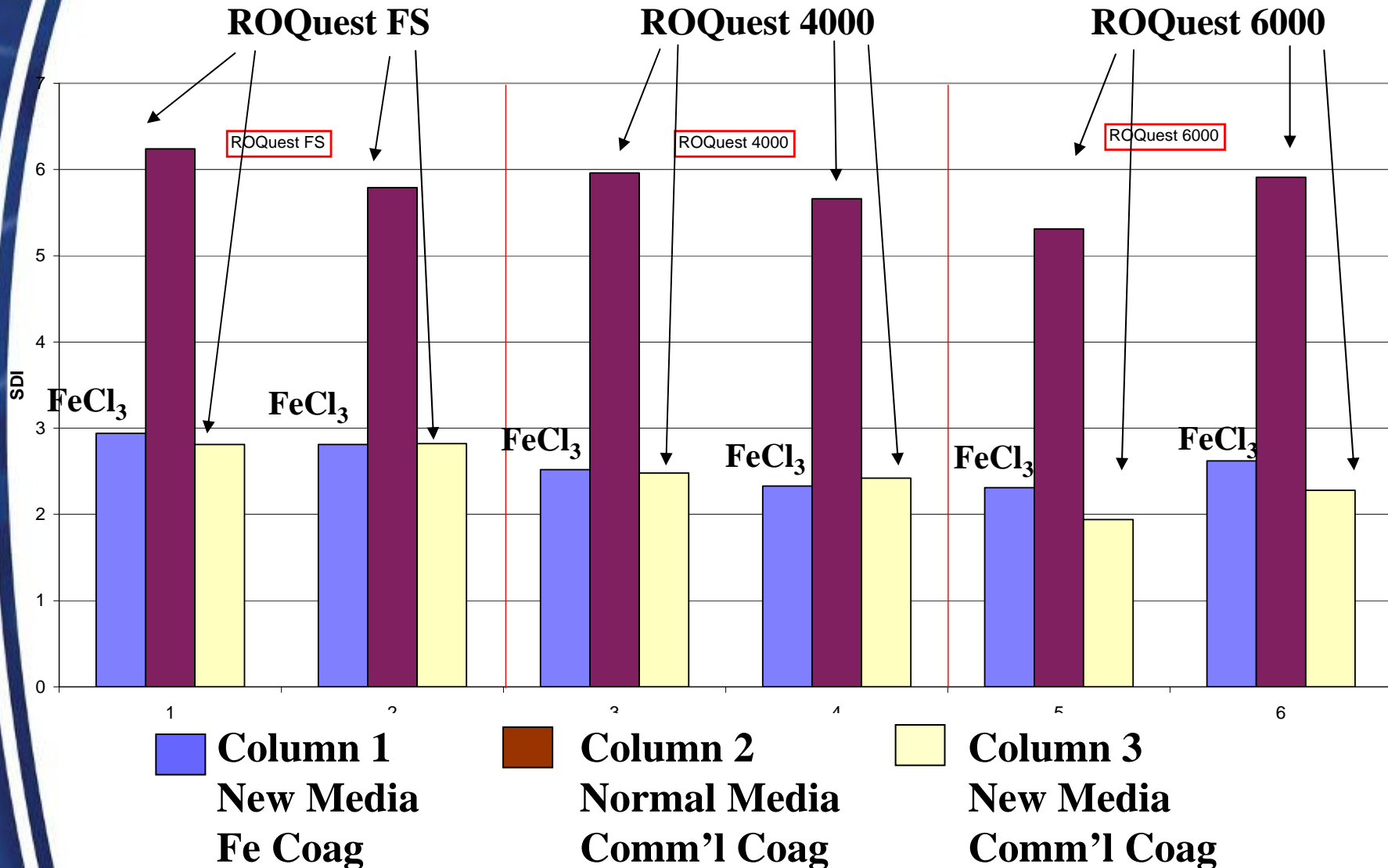
Media

- **Existing Media: 300 mm depth of 2.1 mm grain sand + 1000 mm depth of 0.94 mm grain sand**
- **New Media: 500 mm depth 0.4-0.8mm fine sand + 700 mm porous volcanic pumice**

Coagulant

- **Existing Coagulant: Ferric Chloride**
- **New Coagulant: Avista ROQuest Commercial Coagulant/Flocculant (FS, 4000, 6000)**

Filtration Test : SDI Comparison of Various Coagulants/Flocculants



How Can I Reduce Colloidal Fouling in My RO/NF System?

- Measurement
 - SDI (< 3 preferred)
 - Turbidity (< 0.1 preferred)
 - Particle Counts (<100 of 2 μm particles/ml)
- Optimize Your Media Filtration
 - Media Selection
 - Coagulant Optimization
 - Proper Filtration Velocities
- Membrane Pretreatment
 - MF is generally 0.2 micron
 - UF is generally 0.02 micron

Role of UF Pretreatment: Example Hydracap Capillary UF Membrane

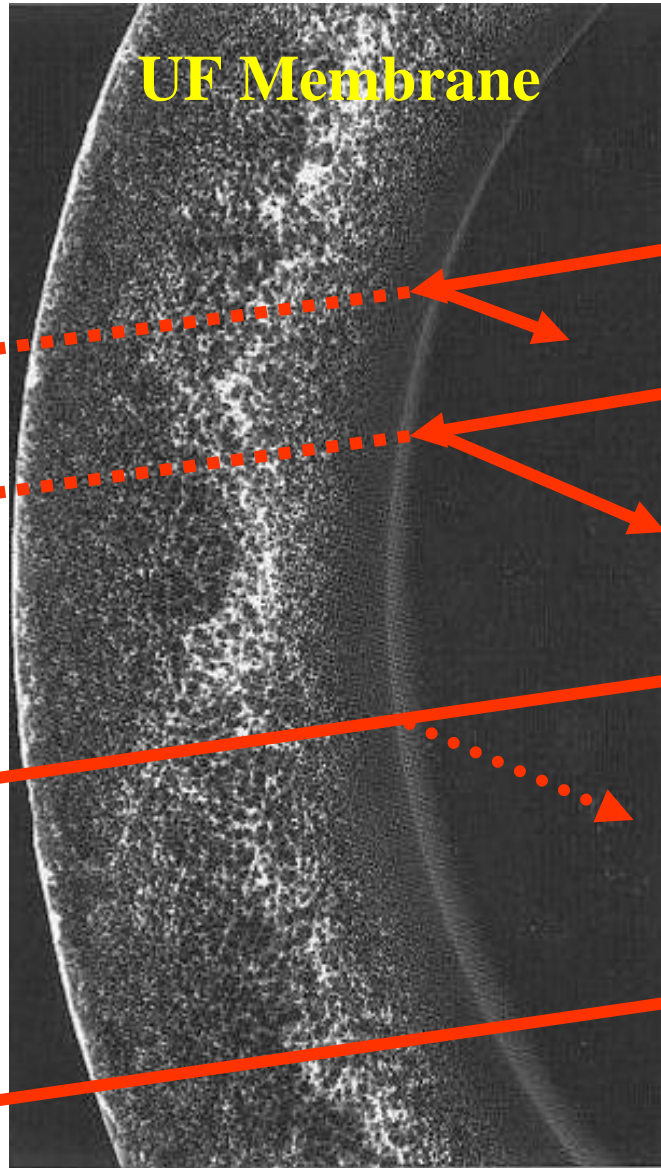
UF FILTRATE

Potential of Bacterial Regrowth
Turbidity < 0.05

RO ←

Food For Bacteria?

Bacterial Contamination?



UF Membrane

Particles **MINIMIZE!**

Sand
Silt

Bacteria/Algae
99.999% Rej

Colloidal
Silica
Silicates
Organic
99.9+% Rej

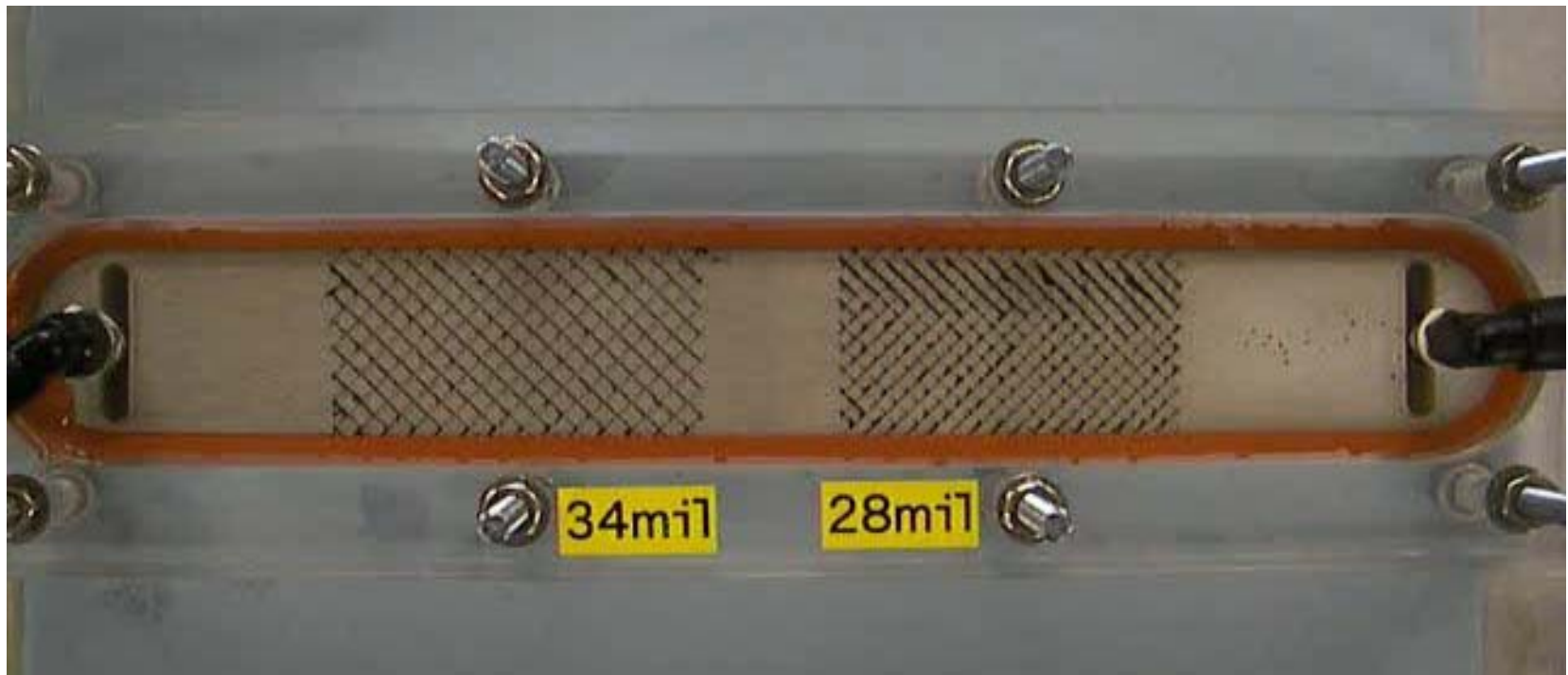
NOM
Humic Acid
Fulvic Acid
Acetic Acid
10-20% Rej

Salts
Scale Forming
Soluble Salts
0% Rej

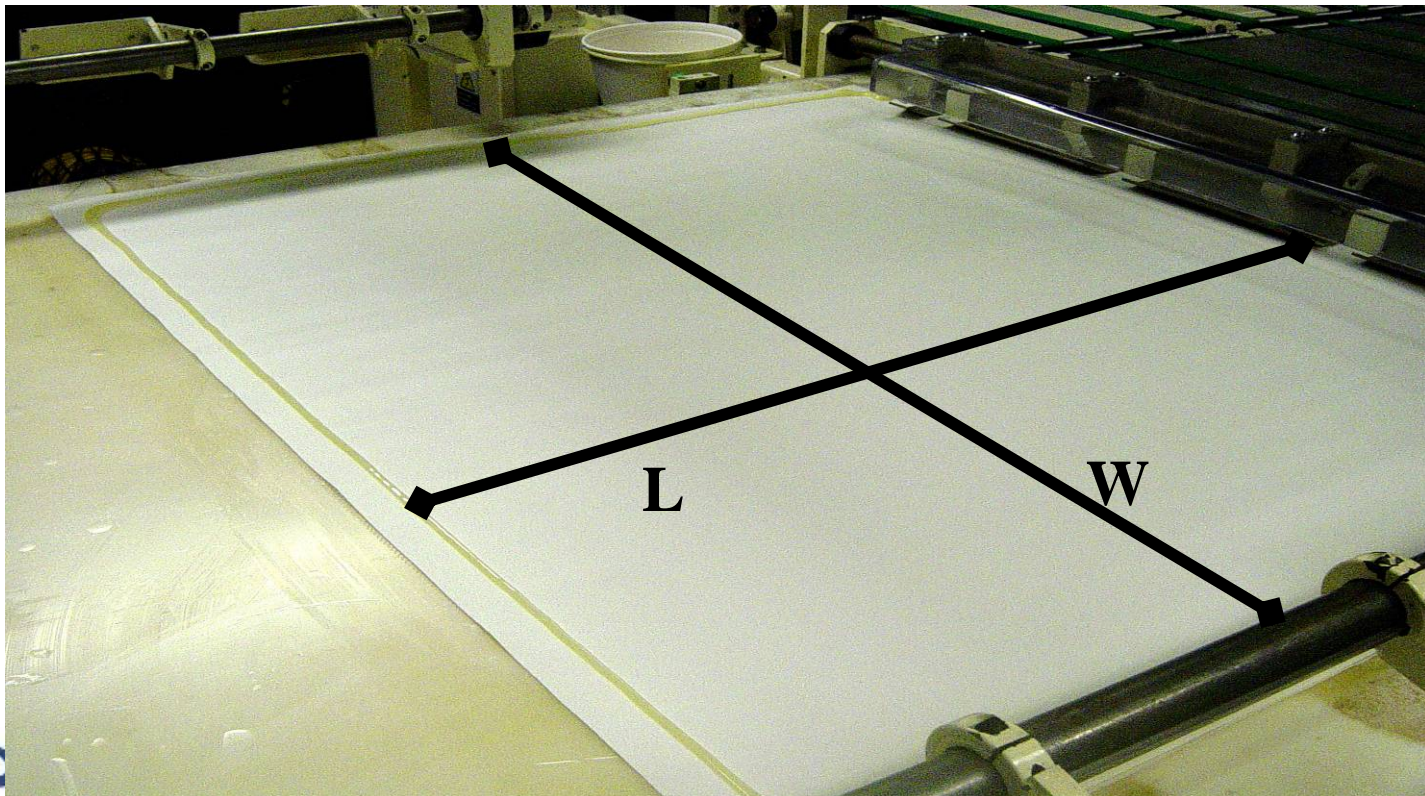
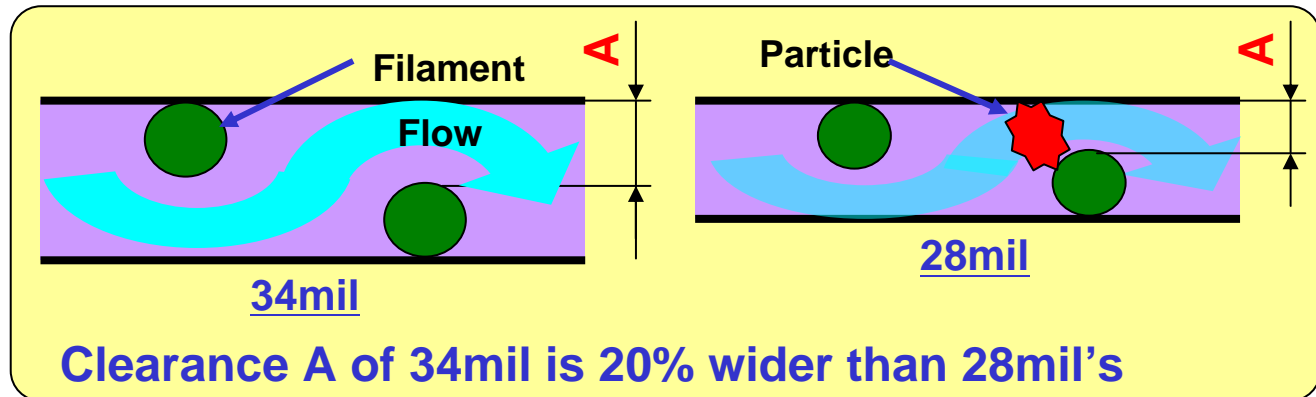
How Can I Reduce Colloidal and Particulate Fouling in My RO/NF System?

- Measurement
 - SDI (< 3 preferred)
 - Turbidity (< 0.1 preferred)
 - Particle Counts (<100 of 2 μm particles/ml)
- Optimize Your Media Filtration
 - Media Selection
 - Coagulant Optimization
 - Filtration Velocities
- Membrane Pretreatment
 - MF is generally 0.2 micron
 - UF is generally 0.02 micron
 - Maintain Fiber Integrity
- Feed/Brine Spacer in the Element
 - Increased thickness minimizes dP Increase

Effect of Feed/Brine Spacer Thickness on Fouling Rates:



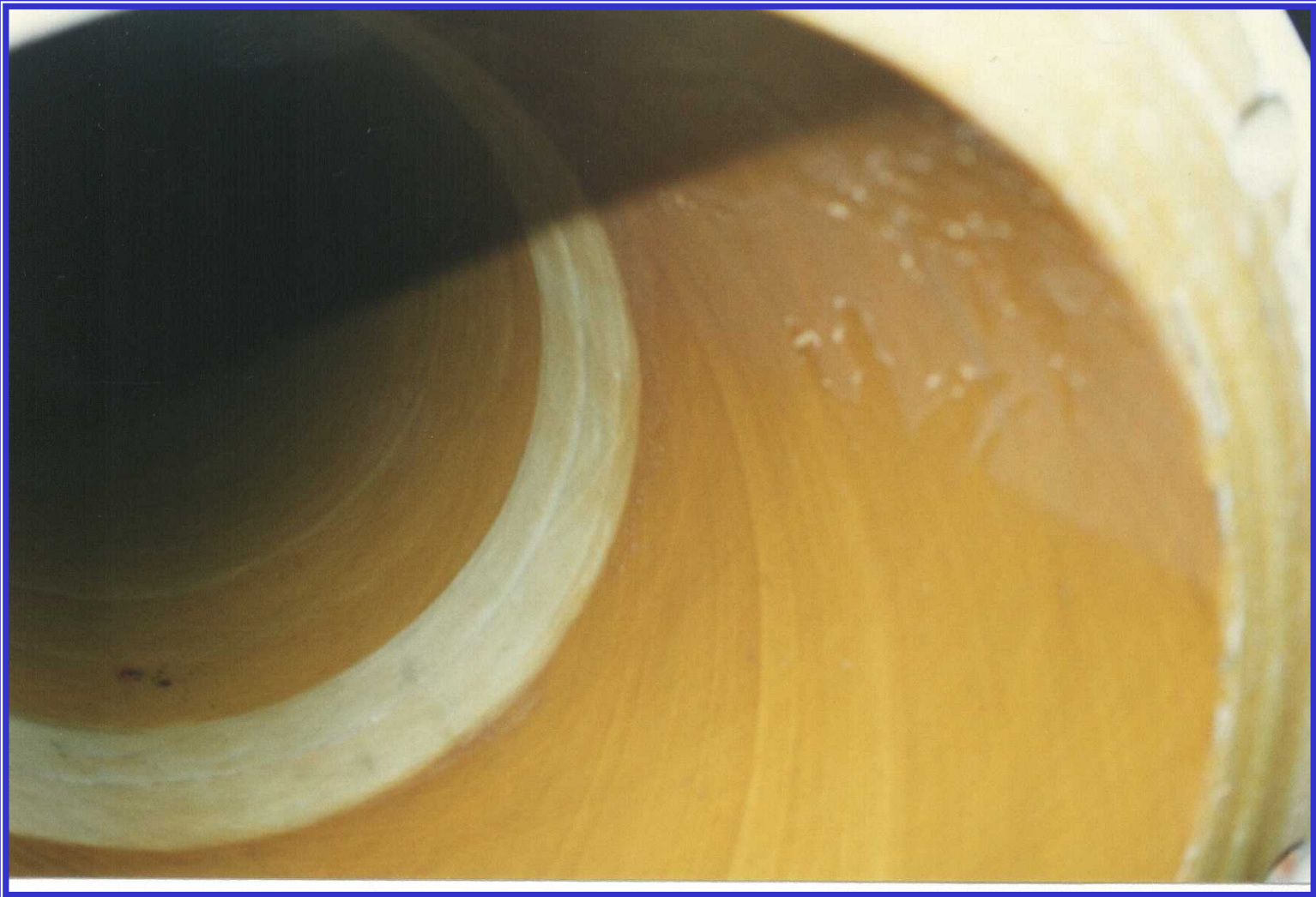
Automated Glue Line Control Allows Thicker Spacers without Membrane Area Loss



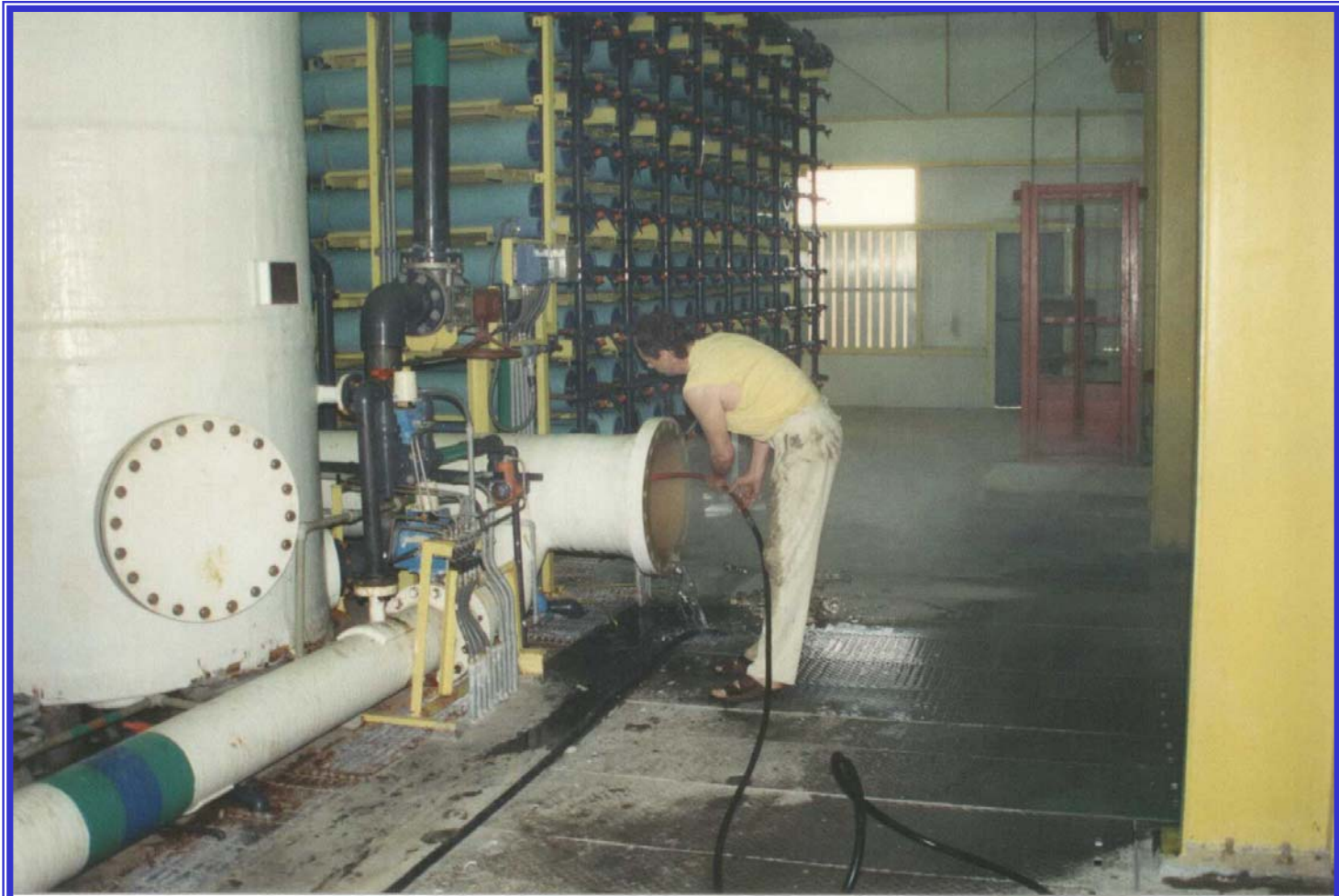
How Can I Reduce Biofouling in My RO/NF System?

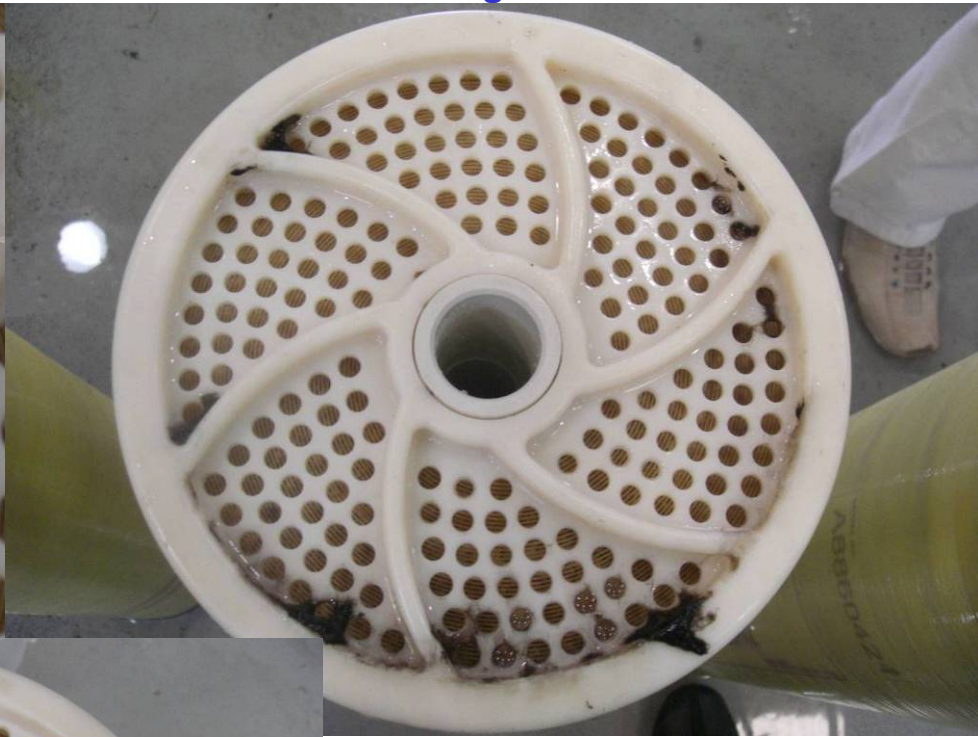
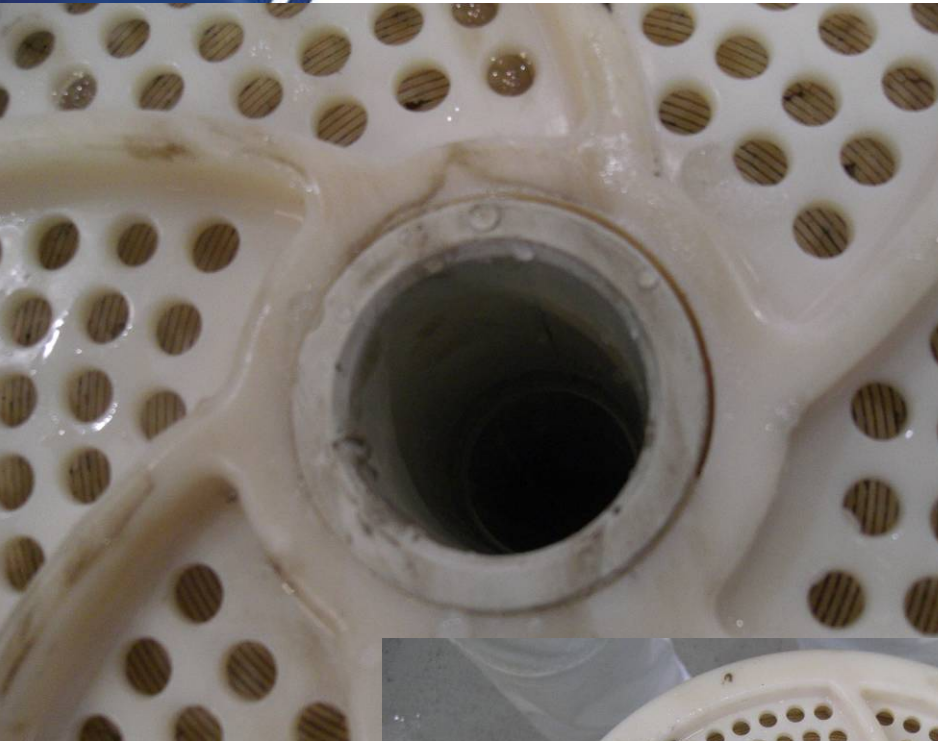
- **Measurement**
 - **Bacteria Counts**
 - **Slime Formation**
 - **Weigh Elements**
- **Prevent Formation of Assimilable Organic Carbon (AOC)**
 - **Avoid Continuous Chlorination/Dechlorination**
 - **Use Shock Dosing as Needed**
 - **Disinfect Lines**
 - **UV Prior to High Pressure pump**
- **Chemical Prevention of Biofouling**
 - **DBNPA Non Oxidizing chemical**
 - **Isothiazolin**
 - **Chloramines (with caution)**

Heavy Biofouling



Biofouling - pipework cleaning





DBNPA Non Oxidizing Biostatic Agent

Slug dosing, 10 – 100 ppm of active ingredient for 30 minutes to 3 hours every 5 days

During slug dosing, the permeate should be dumped if it is for a potable water system.

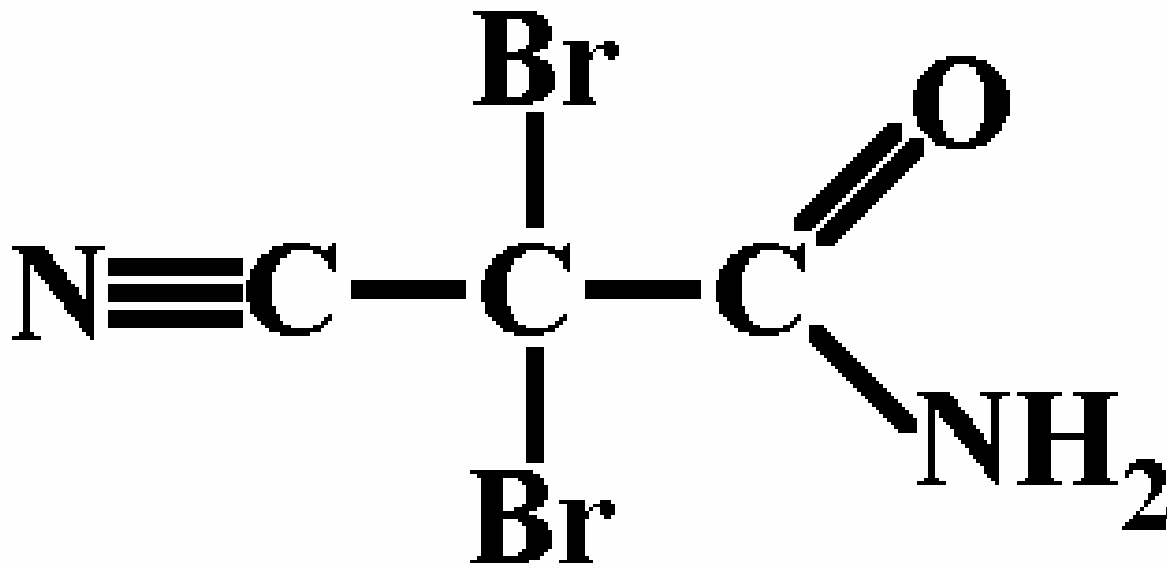
For continuous dosing, between 0.5 to 1 ppm of active ingredient is recommended.

Although DBNPA is non-oxidizing, it will give an ORP reading of about 400 mv when in the range of 0.5 – 3 ppm (for comparison, 1 ppm chlorine typically gives an ORP reading of about 700 mv).

For CIP use, 100-200 ppm of active ingredient for 1 hour would be recommended and keep pH neutral

DBNPA

2,2-dibromo-3-nitrilopropionamide



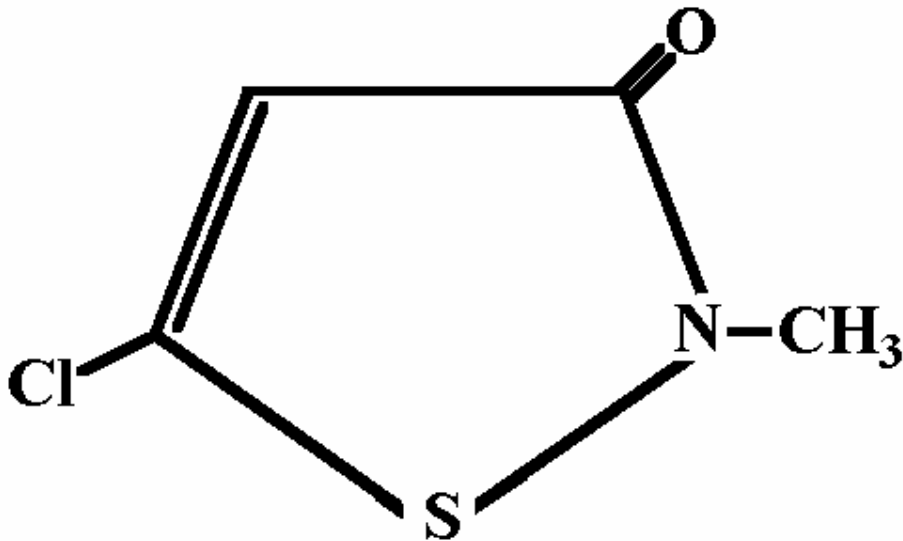
MW = 242

Formula: C₃H₂Br₂N₂O

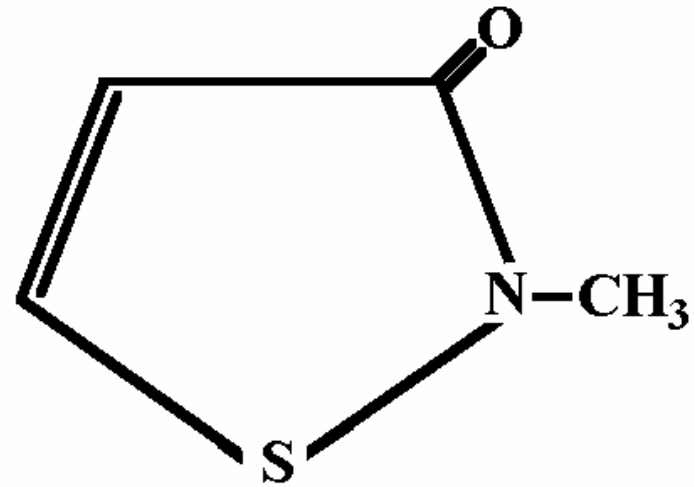
Isothiazolin

- **4 Hours of Contact time (or more)**
- **Broad Spectrum Biostatic Agent**
- **Target: Aerobic and Anaerobic Bacteria, Fungi, and Algae**
- **pH Range: 6 – 9**
- **Known under trade name - Kathon**

Isothiazolin



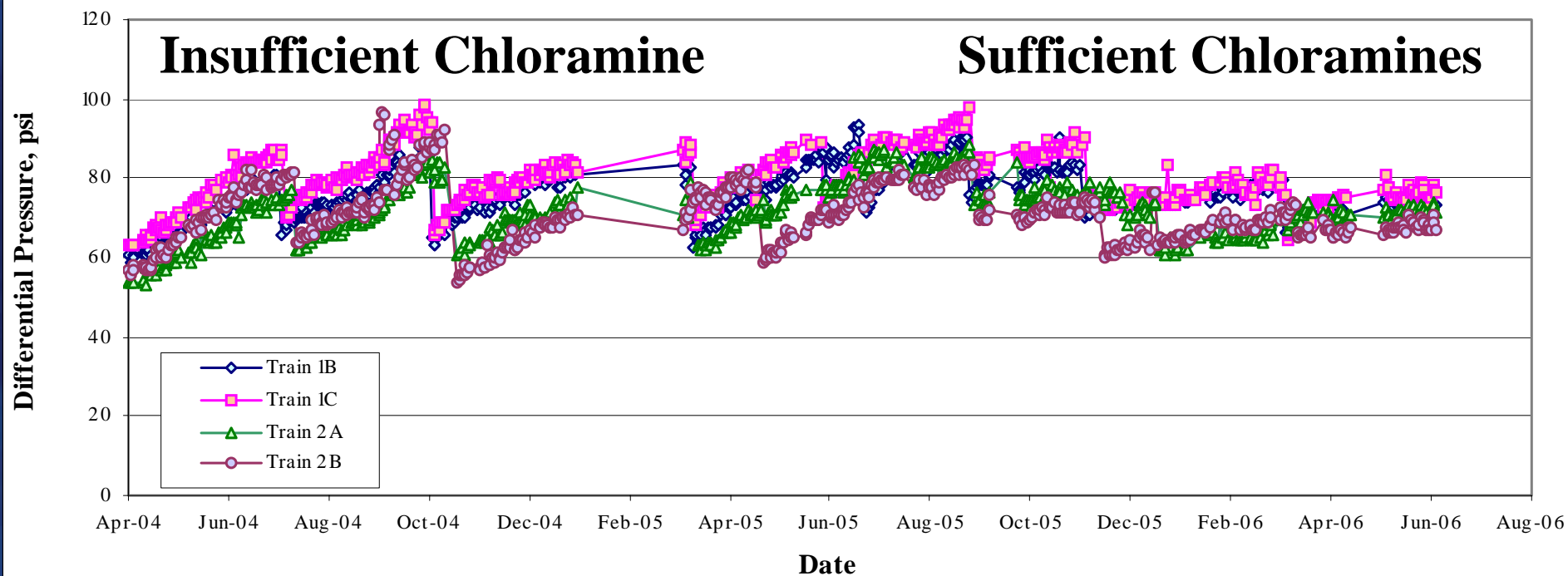
5-chloro-2-methyl-4-isothiazolin-3-one



2-methyl-4-isothiazolin-3-one

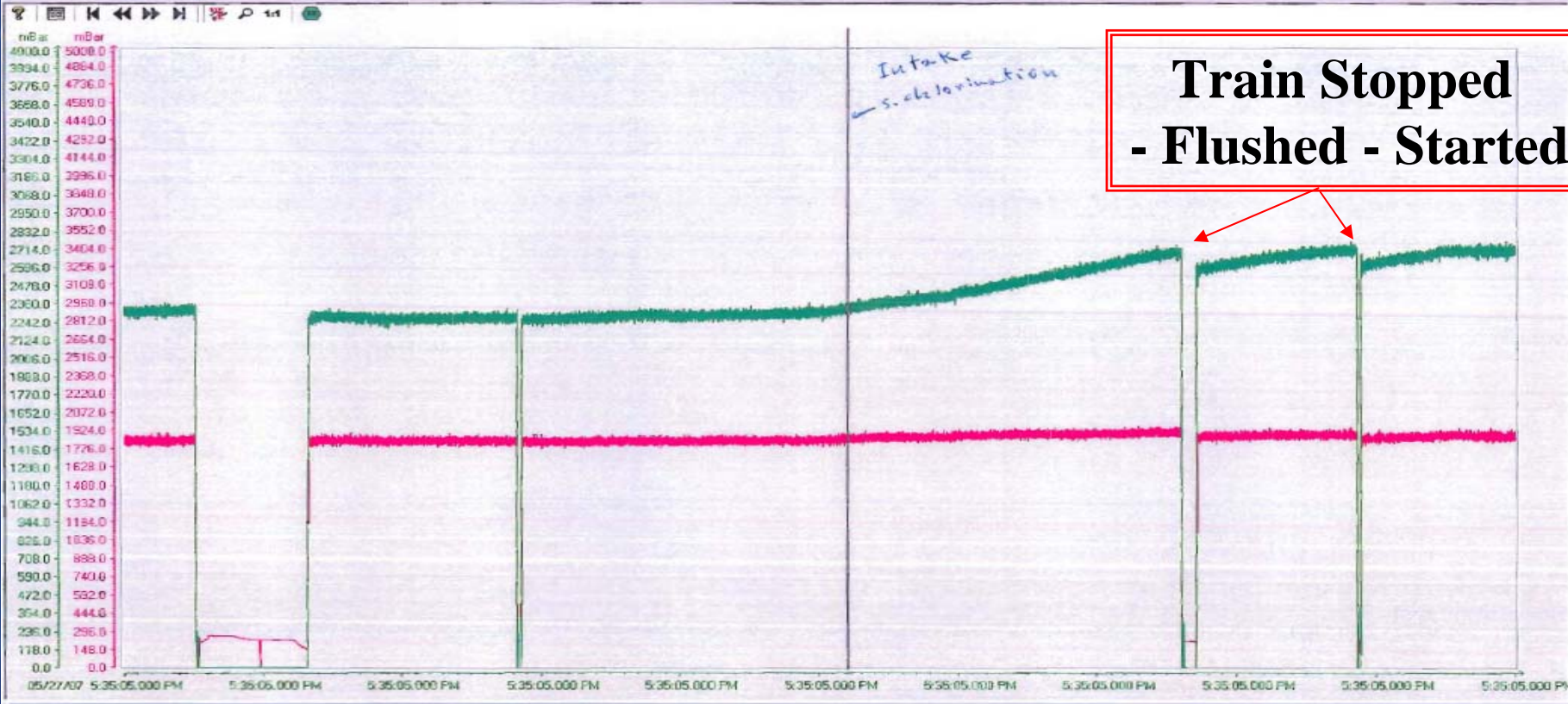
Chloramine Use in Wastewater

GWRS IMF RO - Normalized Differential Pressure



Permeate Flushings

17/06/07		17:35:54.882		UF		20-FIC-003B/PID		UF block B product flow controller High		6/17/2007 5:38:40 PM	
SW INTAKE		SW INTAKE Cont.	W	DUAL MEDIA FILTERS		ULTRAFILTRATION		BACKWASH/FLUSH		KINDASA WATER SERVICES كنداسة للمياه	
RO FEED WATER		RO TRAINS 1st PASS	R	RO TRAINS 2nd PASS	W	PRODUCT LINE	W	FERRIC CHL/ PT CALCIUM HYP		operator	
SW PROD CALCIUM HYP		ACID/ ANTI-SCALANT		SBS/ ALKALI		LINE MAKE-UP/ DOSING	W	RO CLEANING			
UF CLEANING	W	FLUSHING PUMPS		OUTFALL/ NEUTRALISATION	W	AIR COMPRESSOR SYSTEM		SERVICE WATER			
ELECTRICAL SUPPLIES	R	STANDARDS		PRESETS		TIMERS		NORMALISATION			



**Train Stopped
- Flushed - Started**

Trend	Tag Connection	Value	Date/Time
RO Train A 1st Pass 1st Stage Diff. Pressure	SWRO_30A30_PDIT_001A/TRANSMITTER.U	2344.043750	6/17/2007 10:27:08.805 PM
RO Train A 1st Pass 2nd Stage Diff. Pressure	SWRO_30A30_PDIT_002A/TRANSMITTER.U	1877.263794	6/17/2007 10:27:08.805 PM

Trouble-Shooting Guidelines

Possible Cause	Possible Location	Pressure Drop	Feed Pressure	Salt Passage
Metal Oxide Fouling (e.g. Fe, Mn, Cu, Ni, Zn)	1 st stage lead elements	Rapid increase	Rapid increase	Rapid increase
Colloidal Fouling (organic and/or inorganic complexes)	1 st stage lead elements	Gradual increase	Gradual increase	Slight increase
Mineral Scaling (e.g. Ca, Mg, Ba, Sr)	Last stage tail elements	Moderate Increase	Slight increase	Marked increase
Polymerized Silica	Last stage tail elements	Normal to increased	Increased	Normal to increased
Biological Fouling	Any stage, usually lead elements	Marked increase	Marked increase	Normal to increased
Organic Fouling (dissolved NOM)	All stages	Gradual increase	Increased	Decreased
Antiscalant Fouling	2 nd stage most severe	Normal to increased	Increased	Normal to increased

Strategy to Minimize Membrane Fouling

Colloidal Fouling

- Effective Fine Filtration (UF, MF preferred)
- Optimize Coagulation of colloidal particles
- Eliminate Filter Bypass

Biofouling

- No Continuous Chlorination
- Minimize Excess SBS Overdosing
- Sanitize System as Needed
- Remove Bio-food source with effective coagulation/filtration
- Eliminate GAC filters

Scaling

- Prevent oxidation of Fe^{2+}
- Exceeding saturation rarely an issue with 1st pass SWRO
- Calculate correctly saturation in BWRO
- Use proper AS

Organic Adsorption

- Minimize the destabilization of the organics in the water
- Intake protected from free oil, free oil monitors

Cleaning Procedures

System

Planned Regime

Cleaning Sequences

Cleaning Procedure

Manual procedure with necessary interlocks

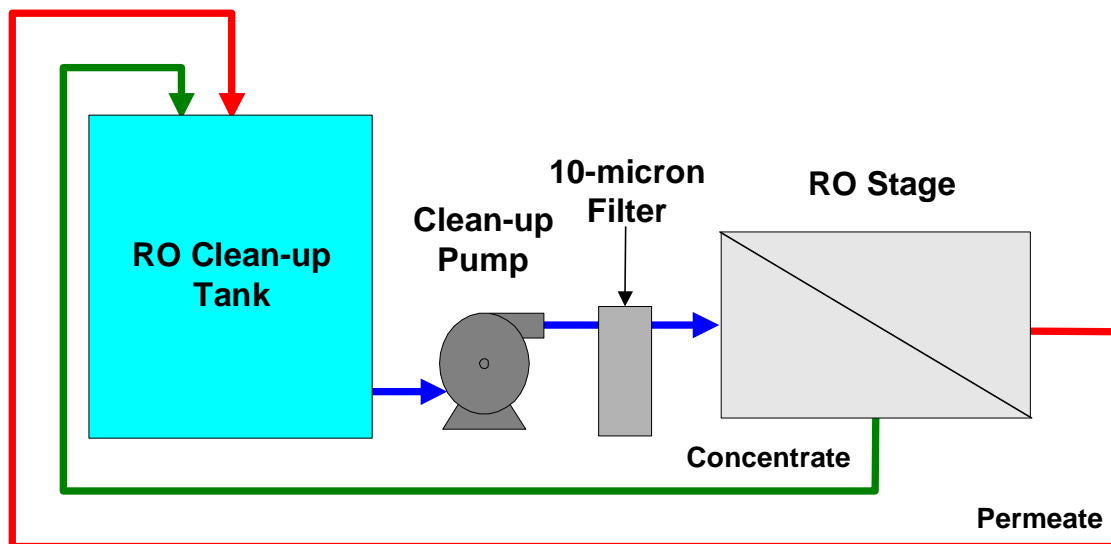
System consists of:

- **cleaning tank**
 - Sizing for retention time ~ 4 minutes
 - Take in consideration losses in the system
 - Eliminate foaming
 - Eliminate suckback
- **cleaning pump – size with flow 10 – 12 m³/h/PV**
- **cleaning heater – important to keep T as high as possible close to 40°C**
- **Cleaning once every 6 months**

Cleaning regime depends on the type of fouling

Cleaning Process

RO Cleanup Skid



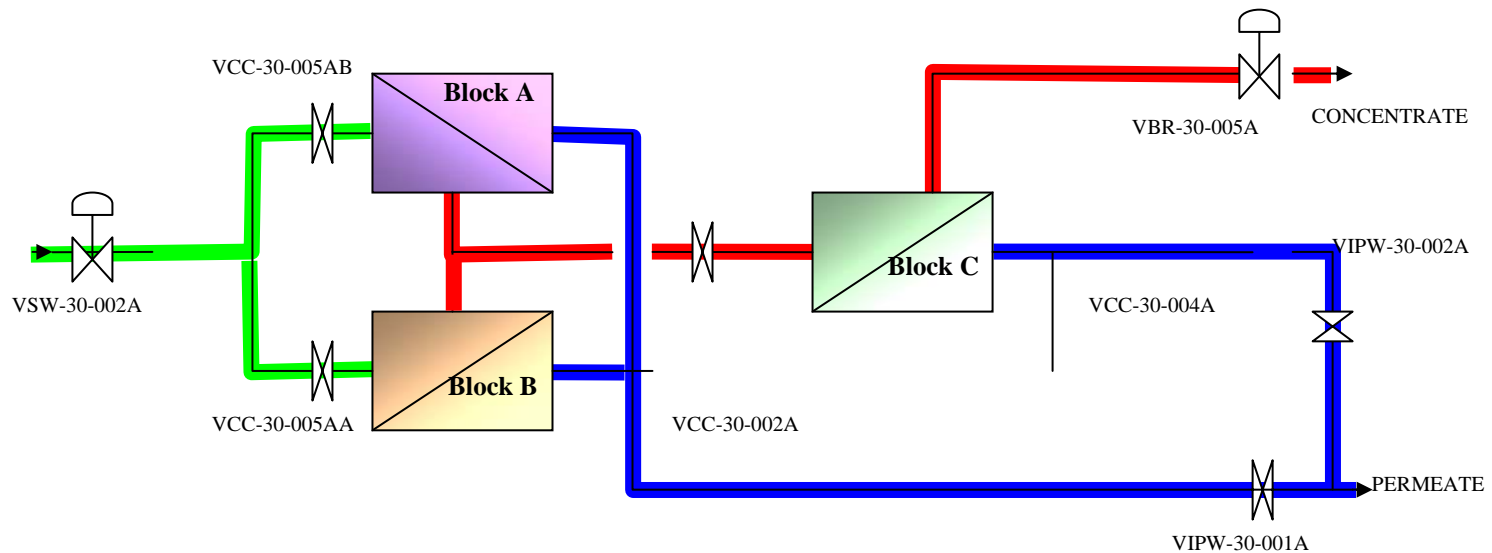
Cleaning Solution Volume Requirement per RO Element

(these volumes do not include volumes required for piping, filters, etc)

(these volumes do not include initial 20% of volume dumped to drain)

Element Size	Normal Fouling (Gallons)	Heavy Fouling (Gallons)	Normal Fouling (Liters)	Heavy Fouling (Liters)
4 x 40 inches	2.5	5	9.5	19
6 x 40 inches	5	10	19	38
8 x 40 inches	9	18	34	68
8.5 x 40 inches	10	20	38	76

Simplified Flow diagram



Cleaning System



Cleaning Process

Alkaline or acidic cleaning

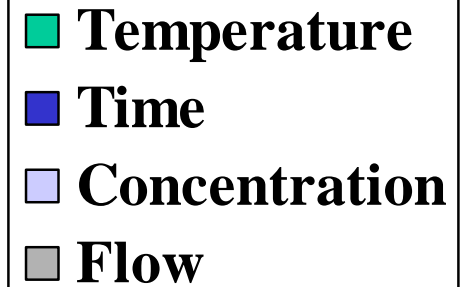
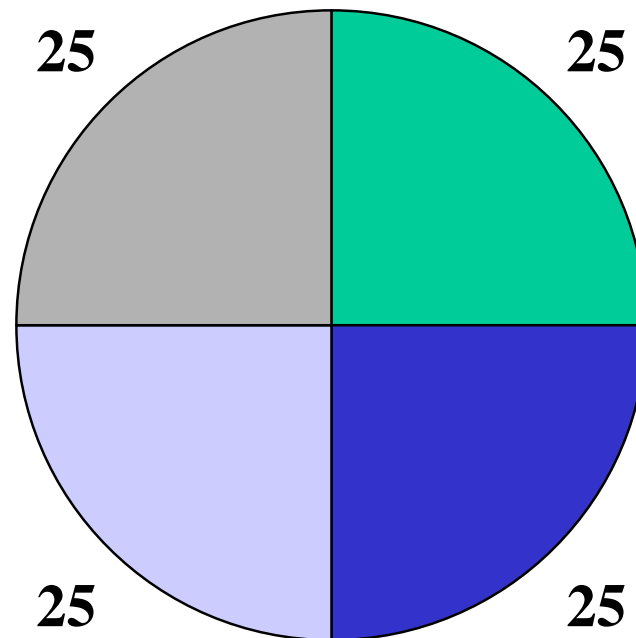
Steps:

- Seawater flush
- Freshwater flush
- Chemical makeup
- Fluid Displacement
- Low flow / high flow recirculation
- Soak
- Drain

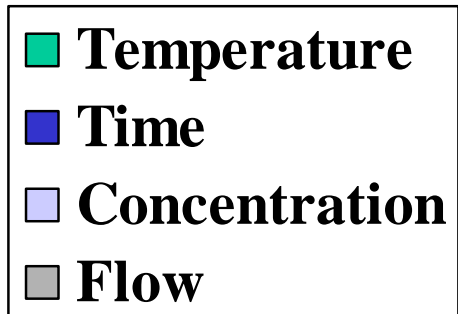
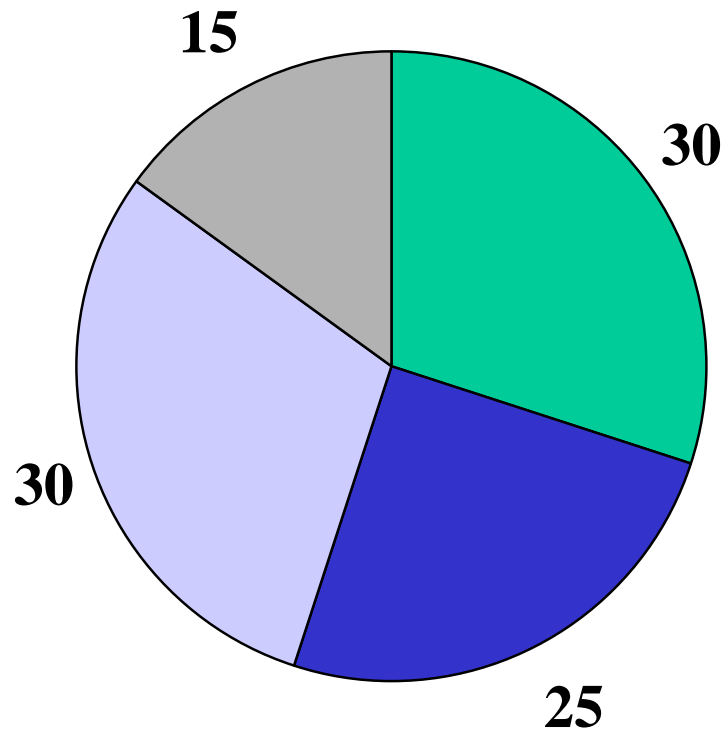
Cleaning Process - Important Points

- **Compatibility of cleaning chemicals with membranes**
- **Correct cleaning temperature - 35° - 45°C**
- **Correct cleaning flows**
- **Do not exceed differential pressure**
- **Soak and recirculate – 30 minutes : 30 minutes**
- **No water hammer or surges must occur!**

Cleaning Procedure - Influencing Factors



Cleaning Procedure - Influencing Factors



Cleaning Solutions – Generic Chemicals

Cleaning Solution	Bulk Ingredients	Quantity	Target pH Adjustment	Target Temp.
1	Citric acid (as 100% powder)	17.0 pounds (7.7 kg)	Adjust to pH 4.0 with ammonium hydroxide.	104 F (40 C)
2	STPP (sodium tripolyphosphate) (as 100% powder) Na-EDTA (Versene 220 or equal) (as 100% powder)	17.0 pounds (7.7 kg) 7.0 pounds (3.18 kg)	Adjust to pH 10.0 with sulfuric or hydrochloric acid.	104 F (40 C)
3	STPP (sodium tripolyphosphate) (as 100% powder) Na-DDBS Na-dodecylbenzene sulfonate	17 pounds (7.7 kg) 2.13 pounds (0.97 kg)	Adjust down to pH 10.0 with sulfuric or hydrochloric acid.	104 F (40 C)

Cleaning Process Conditions

Hydranautics pH and Temperature Limits for Cleaning

(See Table 3 for target pH and temperatures)

Membrane	45 C (113 F)	35 C (95 F)	30 C (86 F)
CPA	2-10	2-12	2-12
ESPA	2-10	2-12	2-12
LFC	2-10	2-12	2-12
SWC	2-10	2-11	2-12
ESNA	3-10	2-12	2-12

Cleaning and Flushing Flow Rates per RO Pressure Tube

(Pressures are not to exceed 60 psi (4 bar) at inlet to tubes.)

Element Diameter	GPM	LPM
4-inches	6 to 10	23 to 38
6-inches	12 to 20	46 to 76
8-inches	24 to 40	91 to 151
8.5-inches	27 to 45	102 to 170

Conclusions

- **Normalize Data at least daily**
- **Analyze your performance indicators to narrow down the type of fouling**
- **Remove elements from the system to analyze the membrane for exact foulant**
- **Follow membrane supplier and chemical supplier guidelines to prevent or alleviate fouling**
- **Clean membranes when normalized permeate flow drops 10-15% or if normalized pressure drop increases by 15-20% or normalized SP increases 10 – 15%**

Preservation

- Preservation should be done if membrane part is stopped for longer than a week
- Typical preservation solution
 - 1% SBS
 - Proprietary chemicals
- Winter preservation solution
 - 1% SBS + 20% glycerine
- pH of solution must not drop below 3
- In case of short time preservation
 - ~1,000 mg/l of SBS is sufficient for 1.5 – 2 months
 - You must regularly analyze residual SBS – it must not drop below 500 mg/l

THANK YOU

