

# Lecture 12

## Introduction to Environmental Engineering

# Effect of Organic Wastes on Stream Ecosystems

## ■ Streeter-Phelps Model – DO sag curve

- ◆ Many equations and computer programs are available today to describe the quality of water in streams, rivers and lakes
- ◆ The most prevalent is the Streeter Phelps equation.
- ◆ Addition of wastewater (BOD) typically causes a slow decrease in  $O_2$ , followed by a gradual increase close to the dissolved oxygen (D.O.) saturation concentration

# Streeter-Phelps Model

- Assumptions of the Model
  - ◆ stream is an ideal plug flow reactor
  - ◆ steady-state flow and BOD and DO reaction conditions
  - ◆ The only reactions of interest are BOD exertion and transfer of oxygen from air to water across air-water interface

# Streeter-Phelps Model

## ■ Mass Balance for the Model

### ◆ Not a Steady-state situation

rate O<sub>2</sub> accum. = rate O<sub>2</sub> in – rate O<sub>2</sub> out + prod. – cons.

rate O<sub>2</sub> accum. = rate O<sub>2</sub> in – 0 + 0 – rate O<sub>2</sub> cons.

### ◆ Both reoxygenation and deoxygenation are 1<sup>st</sup> order

rate of deoxygenation = -k<sub>1</sub>C

k<sub>1</sub> = deoxygenation constant, function of waste type and temperature

# Streeter-Phelps Model

rate of reoxygenation =  $k_2 D$

$D$  = deficit in D.O. or difference between saturation and current D.O.

$k_2$  = reoxygenation constant

$$k_2 = \frac{3.9v^{1/2} \left( [1.025]^{(T-20)} \right)^{1/2}}{H^{3/2}}$$

# Streeter-Phelps Model

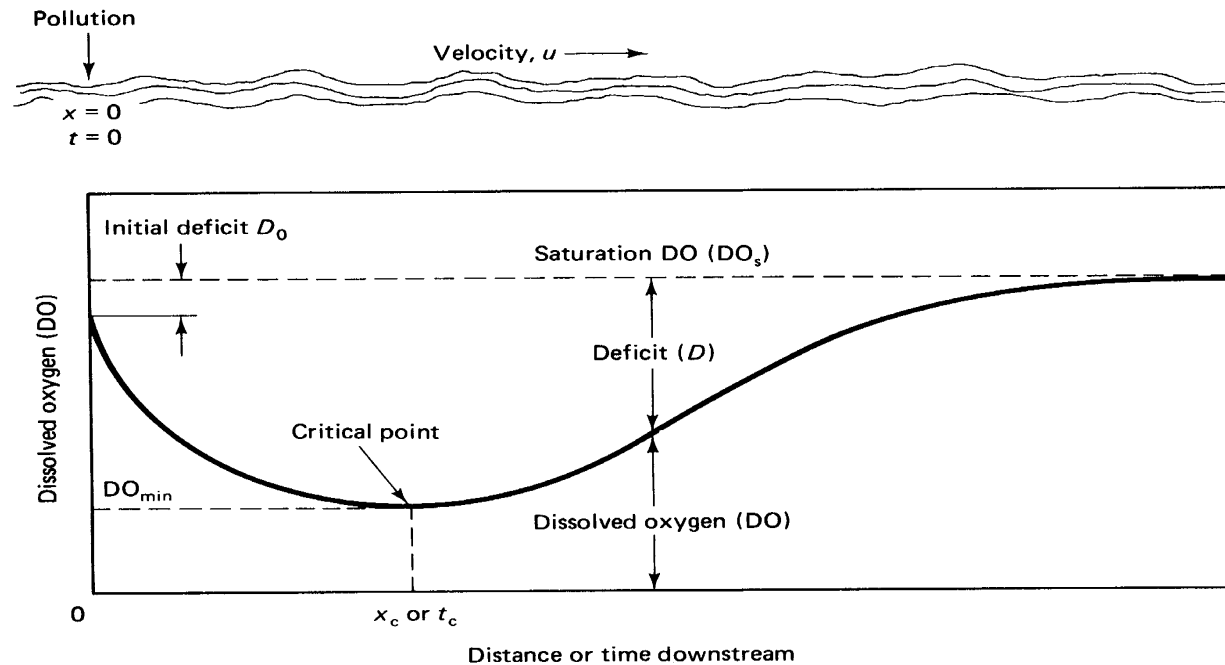
## ■ Where

- ◆  $T$  = temperature of water, °C
- ◆  $H$  = average depth of flow, m
- ◆  $v$  = mean stream velocity, m/s

## ■ Oxygen Deficit

- ◆  $D = S - C$

D.O. deficit = saturation D.O. – D.O. in the water



**Figure 4.7** Streeter-Phelps oxygen-sag curve.

for the critical time yields

$$t_c = \frac{1}{k_r - k_d} \ln \left\{ \frac{k_r}{k_d} \left[ 1 - \frac{D_0(k_r - k_d)}{k_d L_0} \right] \right\} \quad (4.28)$$

The maximum deficit can then be found by substituting the value obtained for the critical time  $t_c$  into (4.24)

# Streeter-Phelps Model

- Deoxygenation rate is equivalent to BOD of waste

- ◆  $r_o = k_1 L_t$

- ◆  $L_t = L_o e^{-Kt}$

$L_o$  or  $L$  = ultimate BOD of the wastewater and stream water mixture



# Streeter-Phelps Model

- In terms of the deficit with time

$$\frac{dD}{dt} = k_1 z - k_2 D$$

$$z = L e^{-k_1 t}$$

# Streeter-Phelps Model

- Substituting and integrating yields the following equations

$$D = \frac{k_1 L_o}{k_2 - k_1} (e^{-k_1 t} - e^{-k_2 t}) + D_o e^{-k_2 t}$$

$$\frac{dD}{dt} = k_1 L_o e^{-k_1 t} - k_2 D = 0$$

$$D_c = \frac{k_1}{k_2} L_o e^{-k_1 t}$$

$$t_c = \frac{1}{k_2 - k_1} \ln \left\{ \frac{k_2}{k_1} \left[ 1 - \frac{D_o (k_2 - k_1)}{k_1 L_o} \right] \right\}$$

# Streeter-Phelps Model

- Example:

- ◆ Wastewater mixes with a river resulting in a

- BOD = 10.9 mg/L,      DO = 7.6 mg/L

- The mixture has a temp. = 20 °C

- Deoxygenation const. = 0.2 day<sup>-1</sup>

- Average flow = 0.3 m/s,      Average depth = 3.0 m

- DO saturated = 9.1 mg/L

- Find the time and distance downstream at which the oxygen deficit is a maximum?
- Find the minimum value of DO?

# Streeter-Phelps Model

- Initial Deficit

$$D_o = 9.1 - 7.6 = 1.5 \text{ mg/L}$$

- Estimate the reaeration constant

$$k_2 = \frac{3.9v^{1/2}}{H^{3/2}} = \frac{3.9(0.3\text{m/s})^{1/2} \left( [1.025]^{(20-20)} \right)^{1/2}}{(3.0\text{m})^{3/2}} = 0.41 \text{ day}^{-1}$$

# Streeter-Phelps Model

- Calculate the time at which the maximum deficit is reached, with  $t_c$ :

$$\begin{aligned} t_c &= \frac{1}{k_2 - k_1} \ln \left\{ \frac{k_2}{k_1} \left[ 1 - \frac{DO_o (k_2 - k_1)}{k_1 L_o} \right] \right\} \\ &= \frac{1}{(0.41 - 0.2)} \ln \left\{ \frac{0.41}{0.2} \left[ 1 - \frac{1.5(0.41 - 0.2)}{0.2 \times 10.9} \right] \right\} \\ &= 2.67 \text{ days} \end{aligned}$$

$$x_c = vt_c = 0.3 \text{ m/s} \times 86,400 \text{ s/day} \times 2.67 \text{ days} = 69,300 \text{ m}$$

# Streeter-Phelps Model

- The maximum DO deficit is:

$$\begin{aligned} D_c &= \frac{k_1}{k_2} L_o e^{-k_1 t} \\ &= \frac{0.2}{0.41} (10.9 \text{ mg/L}) e^{-(0.2 \text{ day}^{-1})(2.67 \text{ days})} \\ &= 3.1 \text{ mg/L} \end{aligned}$$