



Madaba 18 May – 1 June

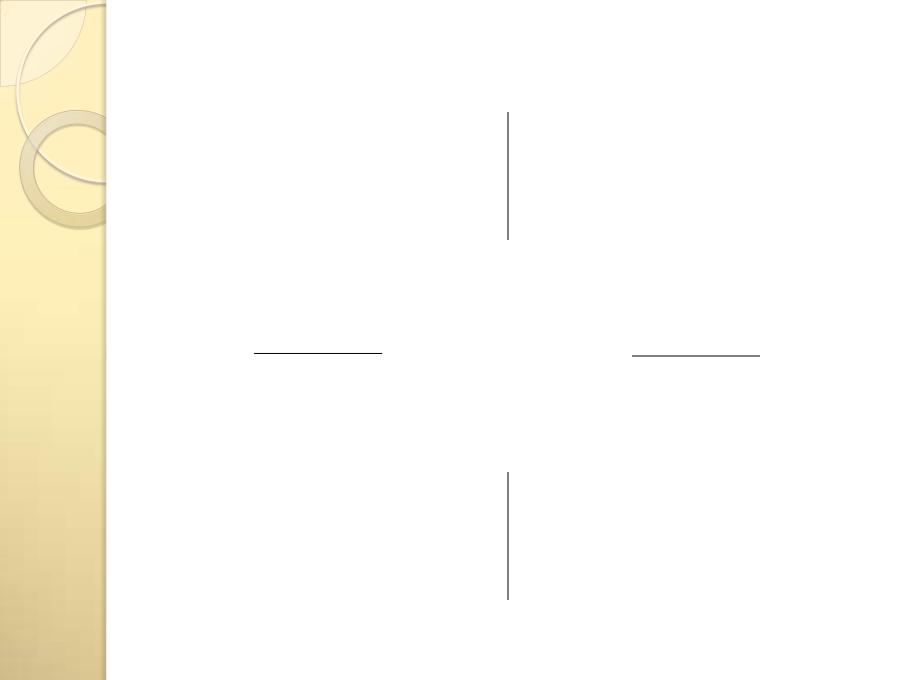
تعريف بالمحاضر

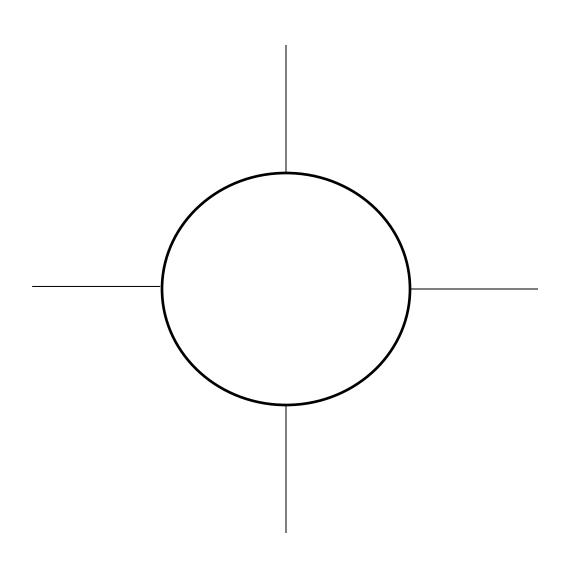
- م <u>محمد زياد الخطيب</u>
- هندسة المياه والبيئة / جامعة البلقاء التطبيقيه
- يعمل كمهندس مشاريع في شركة البحر الابيض لمعالجة المياه
- قدم العديد من الدورات التدريبيه في تخطيط وتصميم محطات تحلية المياه:
 - نقابة المهندسين الاردنيين (عمان ، اربد ، معان ، الزرقاء)
 - الجامعة الاردنيه
- محاضر في يوم علمي ببحثه عن تحلية المياه في الاردن بين المشاكل والحلول ، الواقع والمستقبل
 - عضو جمعية التحلية الاردنيه / لجنة البحث العلمي
 - عضو لجنة المياه والبيئة في نقابة المهندسين

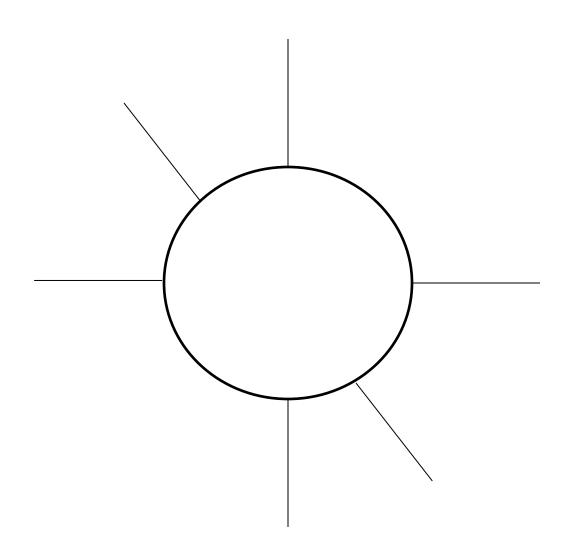
للتواصل M.AlKhateeeb@hotmail.com

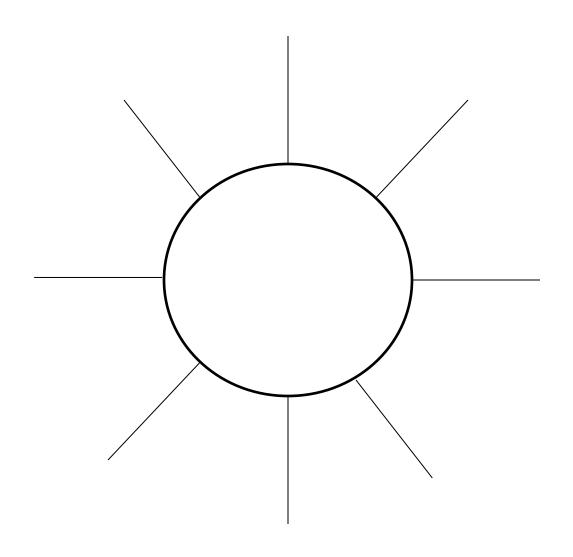
Training Course on Planning of Desalination plant using R.O system

Introduction to Desalination Introduction to Reverse osmosis Pretreatment R.O Post treatment **Others**

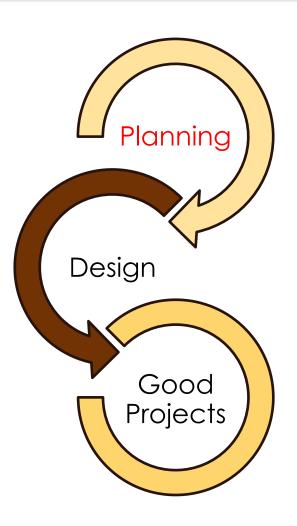




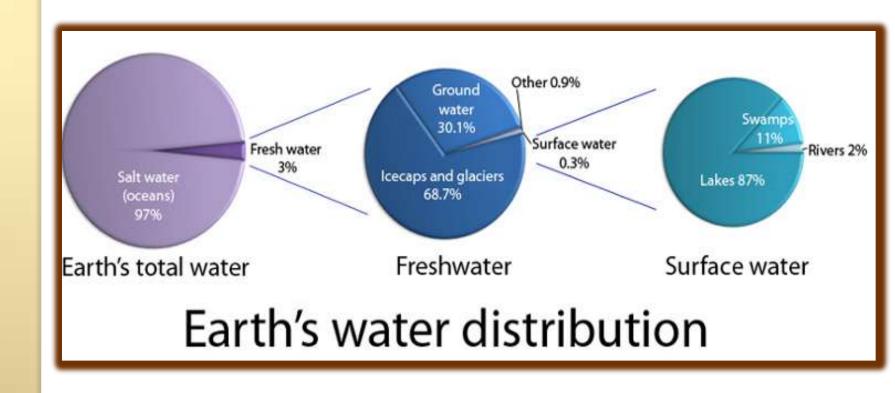






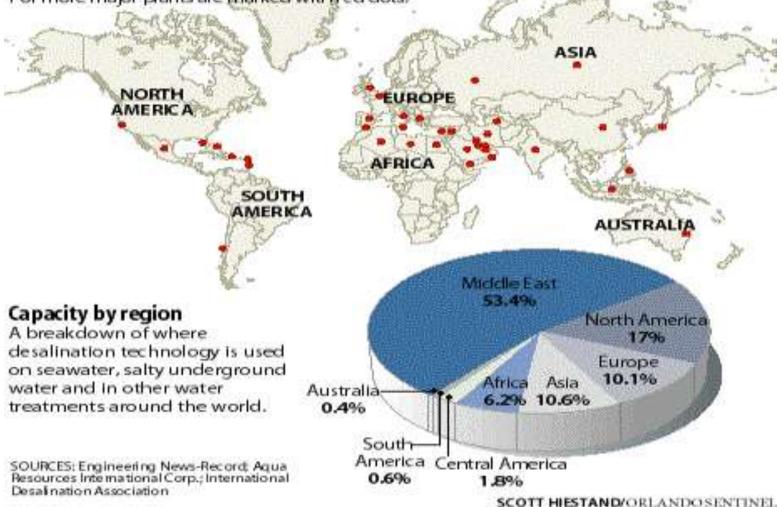


Earth's water distribution



MAJOR DESALINATION PLANTS WORLDWIDE

The United States has 2 major municipal seawater desalination plants — 1 under construction in Tampa and another inactive plant in Santa Barbara, Calif. Other countries with 1 or more major plants are marked with red dots.







What we know about Desalination

• Number of desalination plants in Jordan?

 Amount of permeate water come from desalination plants?

TDS of Brackish water and seawater?

Desalination methods?



Waste water reuse Desalination

□ I 18 MCM / Year
□ 90 MCM / Year

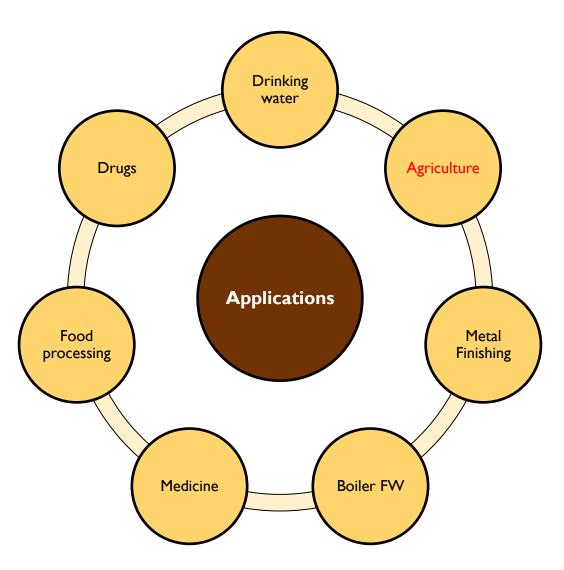
□ More knowledge
□ Less knowledge

□ 27 plants
□ 24 plants

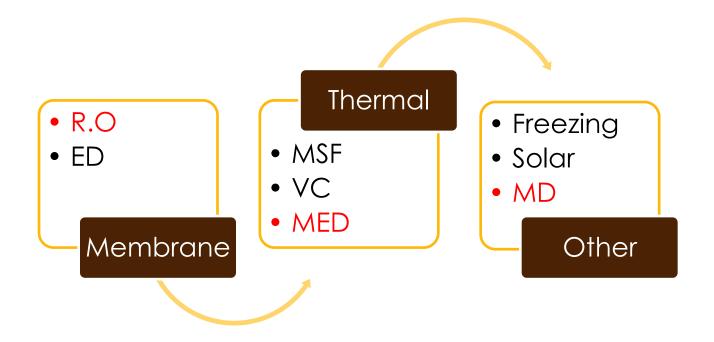
□ AlSamra
□ Zara Ma'in (46 MCM/Yr)

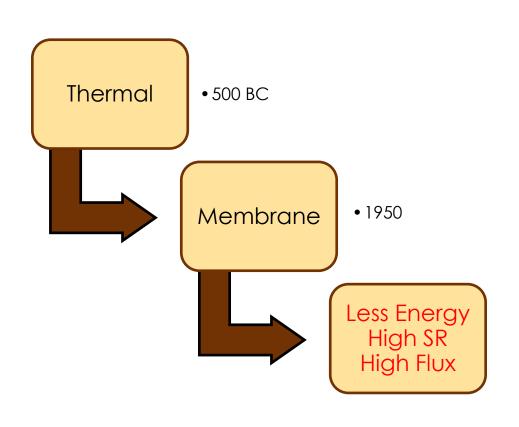
□ 0.08 - 0.55 JD / Cubic meter
□ 0.16 - 0.25 JD / Cubic meter

Applications of Desalination



Desalination Methods





What is Desalination?

Desalination: is a separation process used to reduce the dissolved salt content of saline water to a usable level.

-All desalination processes involve three liquid streams:

the saline feedwater (brackish water or seawater).

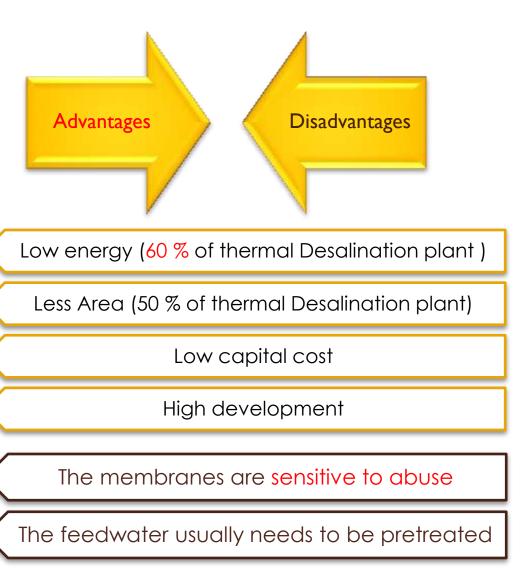
low-salinity product water. (permeate)

very saline concentrate (brine or reject water).

Training Course on Planning of Desalination plant using R.O system

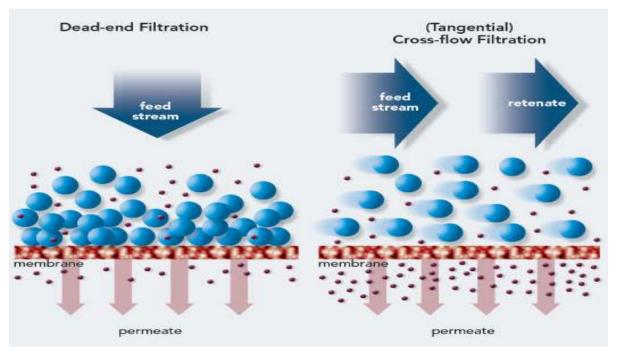
Introduction to Desalination Introduction to Reverse osmosis Pretreatment R.O Post treatment Others

Introduction to R.O.

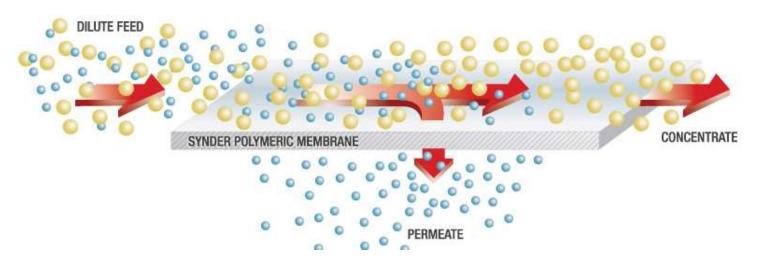


Flow in R.O membrane

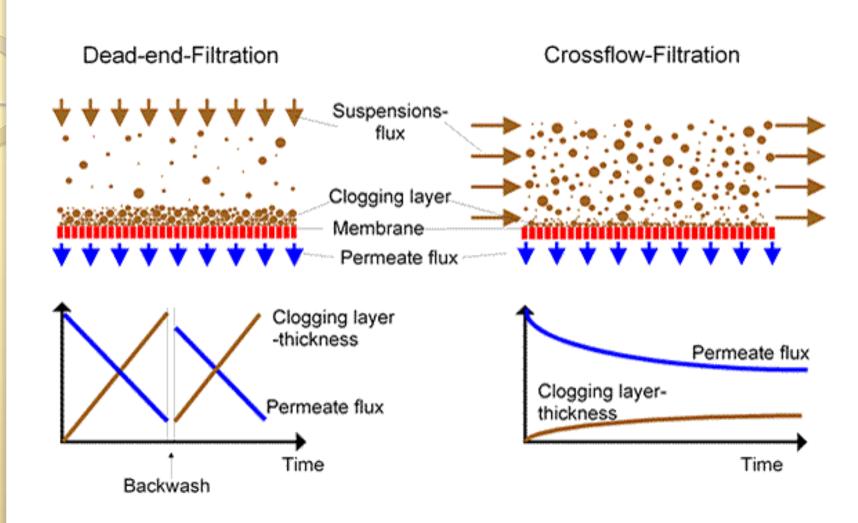




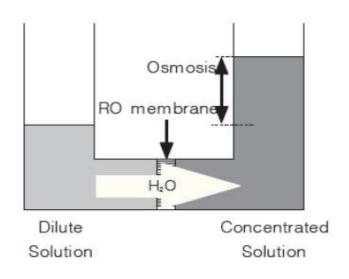
CROSS FLOW PROCESS



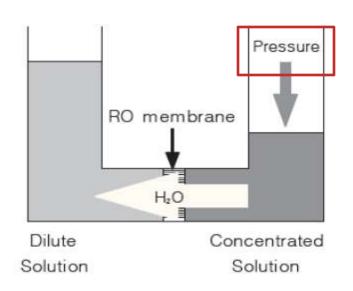
Cross flow Dead End Filtration Flow High permeate flow Low permeate flow Reject flow No reject flow No backwash R.O membrane Sand filter



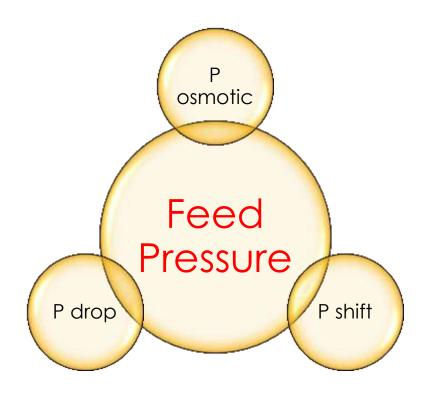
Theory of reverse osmosis



A) Osmatic pressure process



B) Reverse Osmatic pressure process



Feed pressure = P osmotic + P drop + P shift



Major components of reverse osmosis unit



Pre treatment



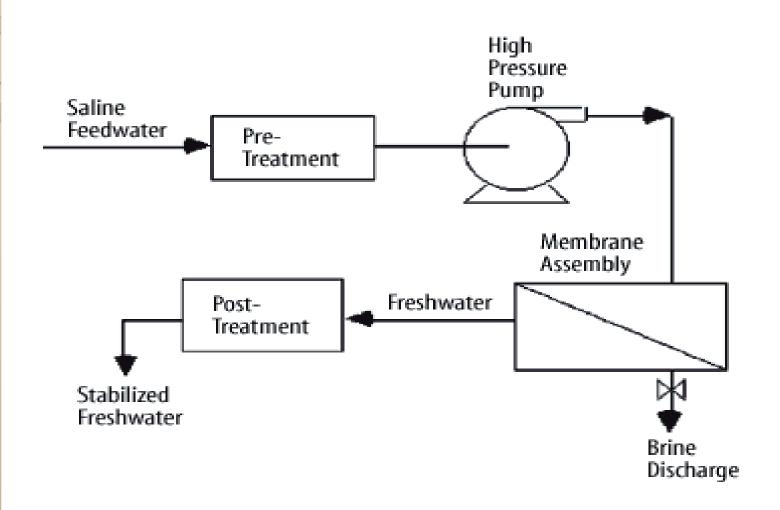
High Pressure pump



Membrane separation



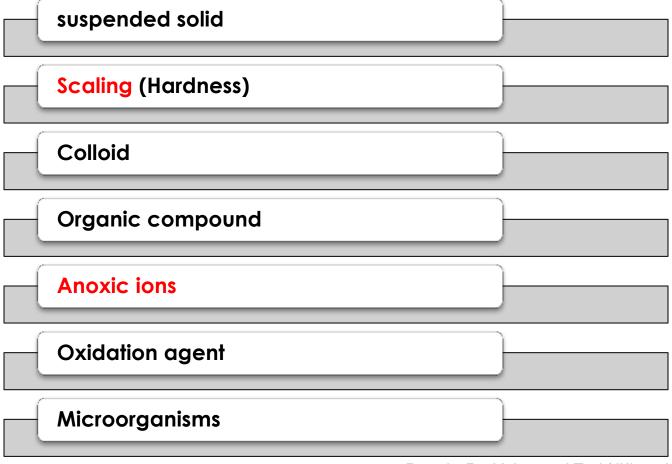
Post treatment

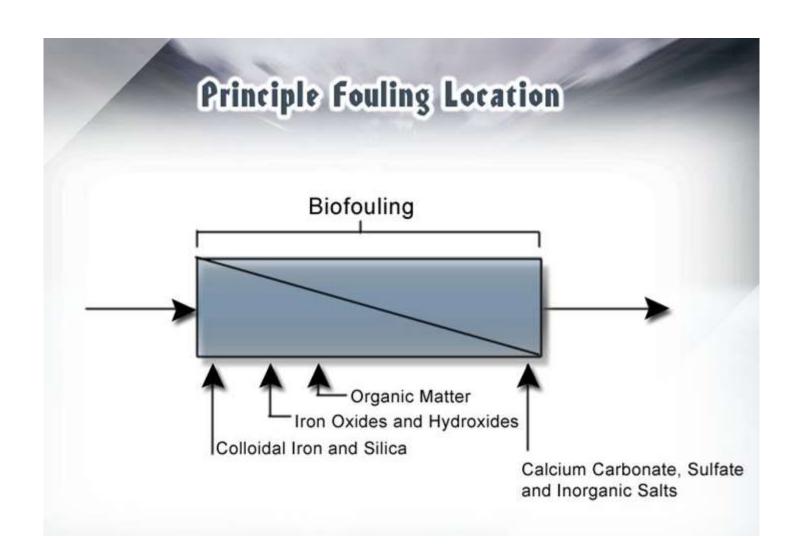


Training Course on Planning of Desalination plant using R.O system

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Major pollutants affects on R.O membrane

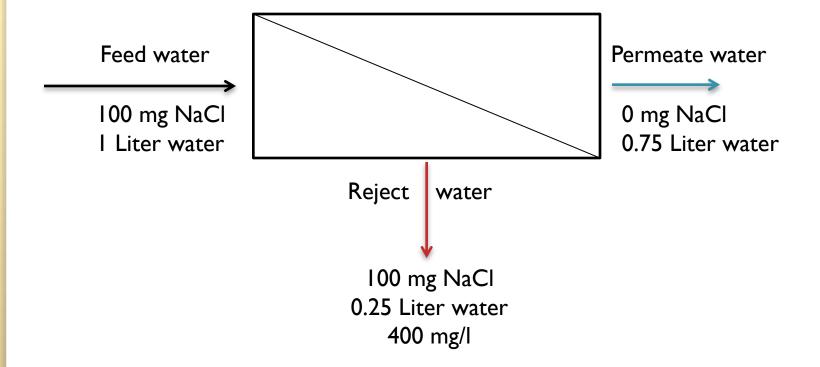








100 mg NaCl in 1 liter water >> Conc = 100 mg/l 100 mg NaCl in 0.25 liter water >> Conc = 400 mg/l



Membrane Data sheet

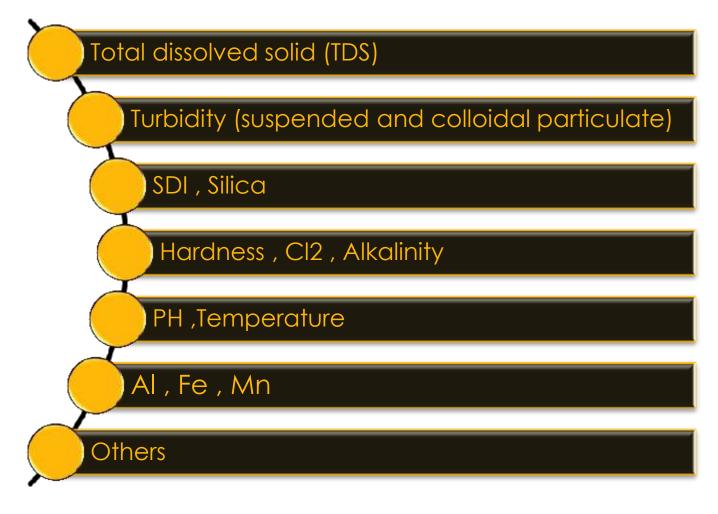
	Membrane Element	ESPA1
Performance:	Permeate Flow: Salt Rejection:	12,000 gpd (45.4 m ³ /d) 99.3 % (99.0 % minimum)
Туре	Configuration: Membrane Polymer: Membrane Active Area:	Spiral Wound Composite Polyamide 400 ft ² (37.1m ²)
Application Data*	Maximum Applied Pressure: Maximum Chlorine Concentration: Maximum Operating Temperature: pH Range, Continuous (Cleaning): Maximum Feedwater Turbidity: Maximum Feedwater SDI (15 mins): Maximum Feed Flow: Minimum Ratio of Concentrate to Permeate Flow for any Element: Maximum Pressure Drop for Each Element:	600 psig (4.16 MPa) < 0.1 PPM 113 °F (45 °C) 2-10 (1-12)* 1.0 NTU 5.0 75 GPM (17.0 m³/h) 5:1 10 psi

Tests

The test considered to be one of the most important factors for the success of any desalination plant, there are so many benefits of the testing, because designing and building of a treatment plant is depending on it. as its reduce the problems that may affect the plant later, and give officials on the station a future predictions for plant efficiency and quality of work and expected problems in the future.



Initial and periodic test



Biological test

1- Culture Techniques

The number of colony forming units (CFU) is a quantitative expression of the number of culturable microorganisms in a water sample. The main advantage of this method is that it can be performed easily without expensive equipment. The test results, however, are only available after up to seven days, and the counted colonies may represent as little as 1-10 % or less of the total bacteria count (TBC).

2- Total Bacteria Count (TBC)

The total bacteria count (TBC) is determined with direct count techniques. These employ filtration of the water sample and counting the retained microorganisms on the filter plate directly under a microscope.

3- Assimilable Organic Carbon (AOC)

The AOC test addresses the growth potential of microorganisms in a given water sample with given nutrients. proposed a standard of 10 μ g/L to prevent biological fouling

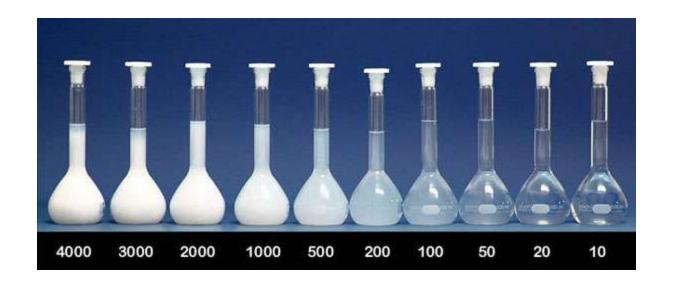
4- Biofilm Formation Rate (BFR)

The BFR value is determined with an online operated biofilm monitor at a continuous flow rate of 0.2 m/s. BFR values exceeding 100 pg/cm2 ATP were observed with severe biofouling, and BFR values of less than 1 pg/cm2 ATP were measured in cases of stable operation without any cleaning needs. The BFR value is most closely correlated with the degree of biofouling in a membrane plant.





Turbidimeter. Three standards are shown-5, 50, and 500 NTRU.

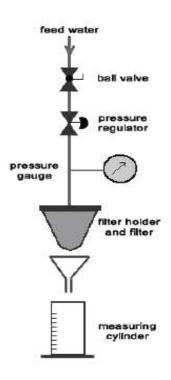


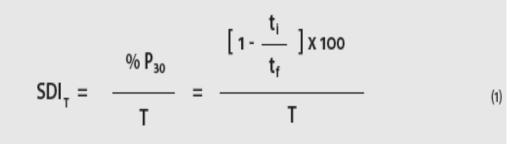
SDI (Silt density index)

SDI stands for silt density index and is a test that attempts to quantify the amount of particulate contamination in a water source. The results are used to estimate how quickly feed water will foul a membrane .It is one of the most important parameter for the design and operation of RO membrane process.

Pressure 2 bar







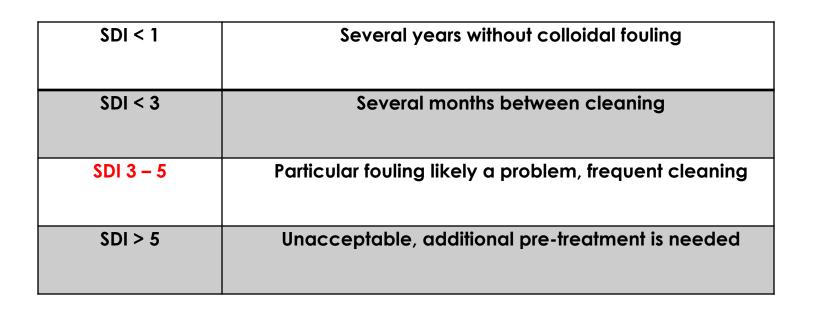
SDIT = Silt Density Index value.

%P30 = plugging filter at 30 psi.

ti = initial filtration time (in seconds),

tf = filtration time (in seconds) after Time,

T = interval between the two readings in most cases 15 minutes



LSI & SDSI Test

The most common method for determining the solubility of calcium carbonate (CaCO3) in water.

LSI (Carrier)	Indication						
-2,0<-0,5	Serious corrosion						
-0,5<0	Slightly corrosion but non-scale forming						
LSI = 0,0	Balanced but pitting corrosion possible						
0,0<0,5	Sligthly scale forming and corrosive						
0,5<2	Scale forming but non corrosive						

Membrane Autopsy

Internal and external visual examination

The Fujiwara test

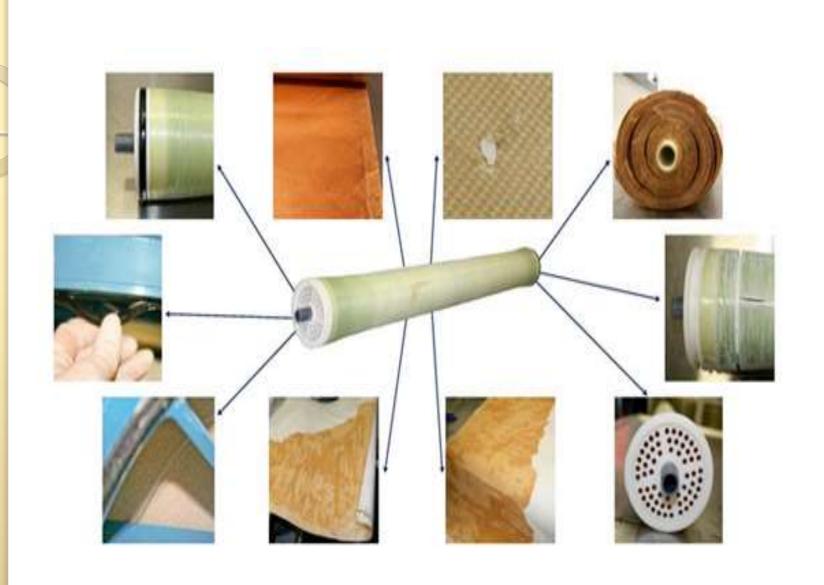
Biological Activity Reaction Test (BART)

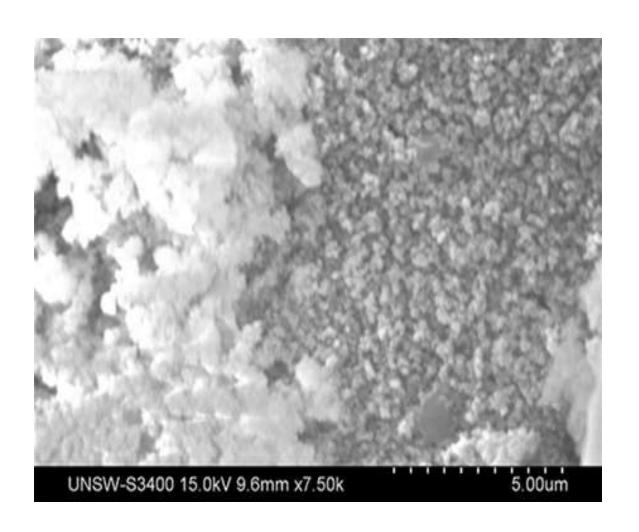
Fourier transform infrared spectrometry (FTIR)

Scanning electron microscopy (SEM)/Energy dispersive x-ray spectroscopy (EDS) Wet testing/Dye integrity testing

Inductively coupled plasma (ICP)/Optical emission spectroscopy (OES/AES)

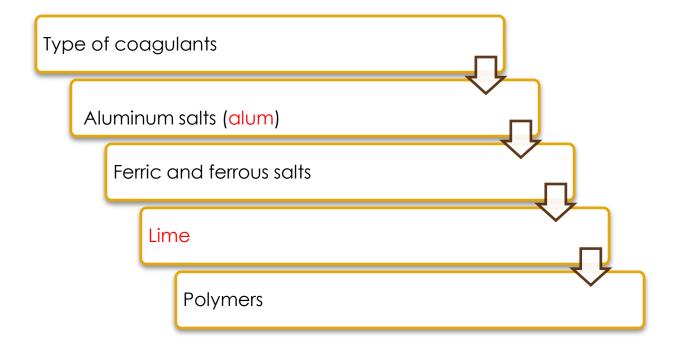
X-Ray Diffraction (XRD)





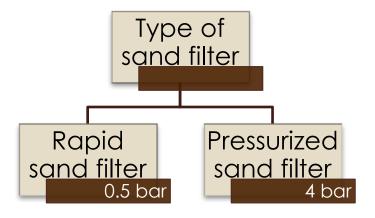


Coagulation & Flocculation



For raw waters containing high concentrations of suspended matter resulting in a high SDI, the classic coagulation-flocculation process is preferred. The hydroxide flocs are allowed to grow and settle in specifically designed reaction chambers. The hydroxide sludge is removed, and the supernatant water is further treated by media filtration.

Sand Filter

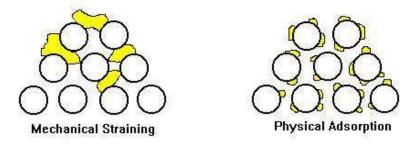


Dual filter media with anthracite over sand permit more penetration of the suspended matter into the filter bed, thus resulting in more efficient filtration and longer runs between cleaning. The design depth of the filter media is a minimum of 31 inches (0.8 m). In the dual filter media, the filters are usually filled with 20 inches (0.5 m) of sand covered with 12 inches (0.3 m) of anthracite.

More details about sand filter



Filtration process



when the differential pressure increase between the inlet and outlet of the pressure filter is (0.3–0.6 bar), and about (0.14 bar) for the gravity filter, the filter is backwashed and rinsed

In-Line filtration (in-line coagulation)

The efficiency of media filtration can be markedly improved if the colloids in the raw water are coagulated and/or flocculated prior to filtration. In-line filtration can be applied to raw waters with a SDI only slightly above 5.

Ferric sulfate and ferric chloride are used to destabilize the negative surface charge of the colloids and to entrap them into the freshly formed ferric hydroxide microflocs. Aluminum coagulants are also effective, but not recommended because of possible fouling problems with residual aluminum. Rapid dispersion and mixing of the coagulant is extremely important. An in-line static mixer or injection on the suction side of a feed pump is recommended. The optimum dosage is usually in the range of 10–30 mg/L, but should be determined case by case.

Cartridge filters

A cartridge filter with an absolute pore size of less than 10 µm is the suggested minimum pretreatment required for every RO system. It is a safety device to protect the membranes and the high pressure pump from suspended particles. Usually it is the last step of a pretreatment sequence. A pore size of 5 µm absolute is recommended



Melt Blown

For SS Removal

Activated Carbon

• For Cl2 removal



Biological Fouling Introduction

All raw waters contain microorganisms such as bacteria, algae, fungi, viruses, and higher organisms. The typical size of bacteria is about 1 µm. The difference between microorganisms and non-living particles, however, is the ability of microorganisms to reproduce and form a biofilm under favorable conditions.

Microorganisms entering a RO/NF system find a large membrane surface where dissolved nutrients from the water are enriched due to concentration polarization, thus creating an ideal environment for the formation of a biofilm. Biological fouling of the membranes may seriously affect the performance of the RO system.

A biofilm is difficult to remove because it protects its microorganisms against the action of shear forces and chemicals. In addition, if not completely removed, remaining parts of a biofilm lead to a rapid regrowth.

Chlorination

Chlorine (Cl2) has been used for many years to treat municipal and industrial water and waste waters to control microorganisms because of its capacity to inactivate most pathogenic microorganisms quickly. The effectiveness of chlorine is dependent on the chlorine concentration, time of exposure, and the pH of the water. A free residual chlorine concentration of 0.5–1.0 mg/L or higher needed.

Chlorine is added continuously at the intake, and a reaction time of 20–30 min should be allowed. A free residual chlorine concentration of 0.5–1.0 mg/L should be maintained through the whole pretreatment line. Dechlorination upstream of the membranes is required, however, to protect the membranes from oxidation.

Chlorine is most commonly available as chlorine gas and the hypochlorites of sodium and calcium. In water, they hydrolyze instantaneously to hypochlorous acid:

CI2 + H2O
$$\rightarrow$$
 HOCI + HCI
NaOCI + H2O \rightarrow HOCI + NaOH
Ca(OCI)2 + 2 H2O \rightarrow 2 HOCI + Ca(OH)2
HOCI \leftrightarrow H+ + OCI-

Dechlorination

When RO or NF membrane is used in the RO/NF process, the feed must be dechlorinated to prevent oxidation of the membrane.

Sodium metabisulfite (SMBS) is commonly used for removal of free chlorine. Other chemical reducing agents exist (e.g., sulfur dioxide), but they are not as cost-effective as SMBS.

When dissolved in water, sodium bisulfite (SBS) is formed from SMBS:

 $Na2S2O5 + H2O \rightarrow 2 NaHSO3$

SBS then reduces hypochlorous acid according to:

 $2NaHSO3 + 2HOCI \rightarrow H2SO4 + 2HCI + Na2SO4$

Shock treatment

Sodium bisulfite can be added into the feed stream (for a limited time period) during normal plant operation. This intermittent application is often referred to as shock treatment. In a typical application, 500–1,000 mg/L NaHSO3 is dosed for 30 minutes.

The treatment can be carried out on every 24 hours or only when biogrowth is suspected.

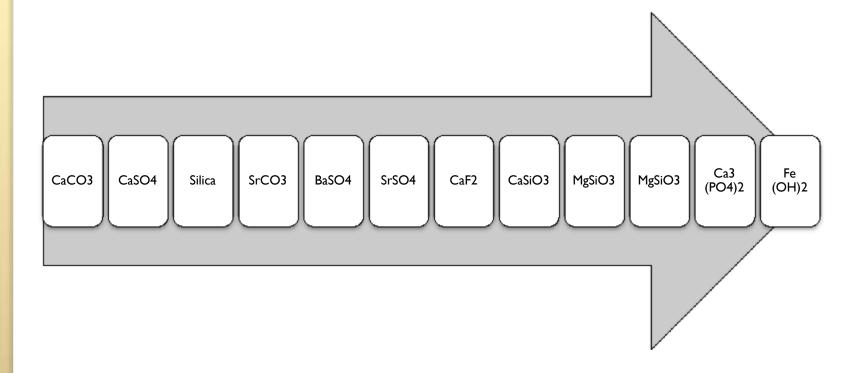
Copper sulfate can be used to control the growth of algae. Typically, copper sulfate is fed continuously at concentrations of 0.1 to 0.5 ppm. pH of the water must be low (to prevent the precipitation of copper hydroxide).

Ozone is an even stronger oxidizing agent than chlorine. However, it decomposes readily. A certain ozone level must be maintained to kill all microorganisms. The resistance of the materials of construction against ozone has to be considered. Usually, stainless steel is employed. Removal of ozone must be performed carefully to protect the membranes



Scaling problem

Scaling of an RO membrane may occur when sparingly soluble salts are concentrated in the RO element beyond their solubility limit.



Acid addition

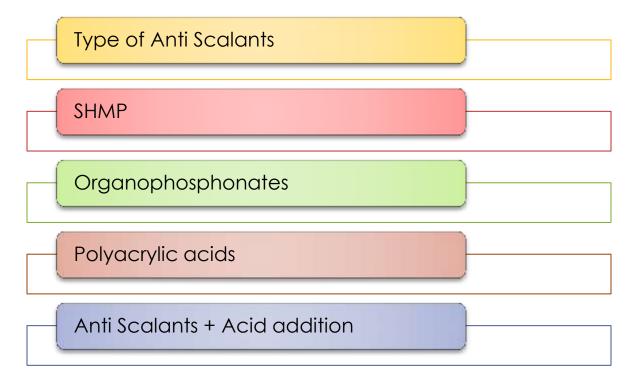
To <u>control</u> calcium carbonate scaling by acid addition alone, the LSI or S&DSI in the concentrate stream must be <u>negative</u>. Acid addition is useful to control carbonate scale only.

For seawater systems, a dosage of typically 10 mg/L sulfuric acid is required to achieve a pH of about 7 and a negative S&DSI in the concentrate.

$$Ca^{2+} + HCO_3 \rightarrow H^+ + CaCO_3$$

Anti Scalants

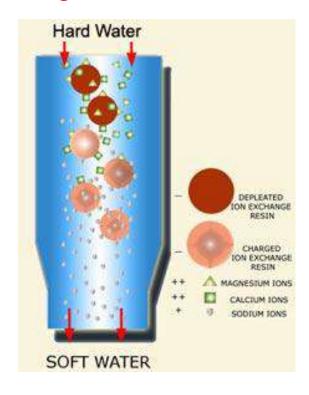
Scale inhibitors (anti-scalants) slow the precipitation process of sparingly soluble salts by being absorbed on the forming salt crystals to prevent the attraction of the supersaturated salt to the crystal surfaces.



scale inhibitor is recommended when operating above a recovery of 35%.

Water Softener with a Strong Acid Cation Exchange Resin

In the ion exchange softening process, the scale-forming cations, such as Ca2+, Ba2+ and Sr2+, are removed and replaced by sodium cations. The resin is regenerated with NaCl at hardness breakthrough. ion exchange resins, the removal efficiency for Ca2+, Ba2+, and Sr2+ is greater than 97%, which usually eliminates any risk of carbonate or sulfate scaling.







Dealkalization with a Weak Acid Cation Exchange Resin

<u>Dealkalization</u> with a weak acid cation exchange resin is used mainly in large brackish water plants for partial softening to minimize the consumption of regeneration chemicals.

In this process, only Ca2+, Ba2+, and Sr2+ associated with bicarbonate alkalinity (temporary hardness) are removed and replaced by H+

The advantages

- For regeneration, acid of not more than 105% of the stoichiometric value is needed.
- High Capacity (2 time of strong acid resin).

The disadvantages are:

- Residual hardness.
- Variable pH of the treated water.

Lime Softening

Lime softening can be used to remove carbonate hardness by adding hydrated lime:

Ca(HCO3)2 + Ca(OH)2
$$\rightarrow$$
 2 CaCO3 + 2 H2O
Mg(HCO3)2 + 2 Ca(OH)2 \rightarrow Mg(OH)2 + 2 CaCO3 + 2H2O

The noncarbonate calcium hardness can be further reduced by adding sodium carbonate (soda ash):

CaCl2 + Na2CO3
$$\rightarrow$$
 2 NaCl + CaCO3
CaSO4 + Na2CO3 \rightarrow CaCO3 + Na2SO4

The lime-soda ash process can also be used to reduce the silica concentration (20 mg/l Max in R.O feedwater) with or without magnesium oxide or sodium aluminate.

silica adsorption onto Mg(OH)2.

Increasing Efficiency

-Hot water 60–70°C -pH 10-11

With lime softening, barium, strontium, and organic substances are also reduced

Greensand filter

Water with divalent iron and manganese, sometimes hydrogen sulfide and ammonium, but no oxygen; therefore, they are also called anoxic water.

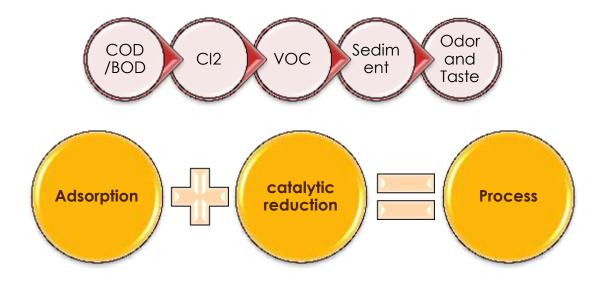
Anoxic water can treated by oxidation and filtration.

Oxidation and filtration can be accomplished in one step by using a filter media with the ability to oxidize divalent iron and manganese by electron transfer. Greensand is such a granular medium, which is a green (when dry) mineral glauconite. It can be regenerated with KMnO4 when its oxidizing capability is exhausted.

Birm filtration has also been used effectively for Fe2+ removal from RO feed water



Carbon Filter



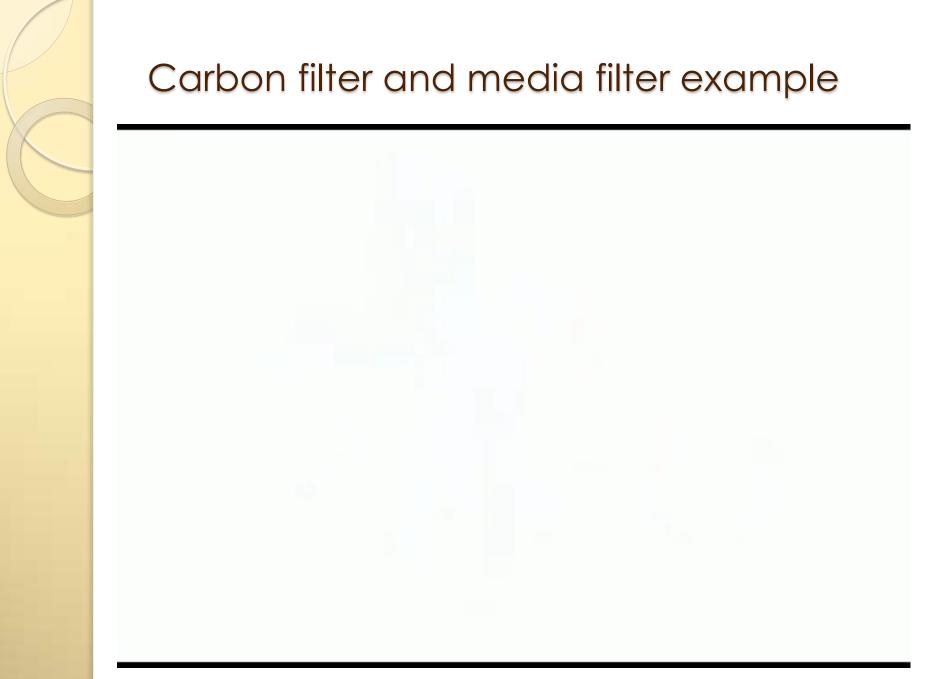
Carbon filtering is a method of filtering that uses a piece of activated carbon to remove contaminants and impurities, utilizing chemical adsorption. Each piece of carbon is designed to provide a large section of surface area, in order to allow contaminants the most possible exposure to the filter media.

An activated carbon bed is very effective in the dechlorination of RO feed water according to following reaction:

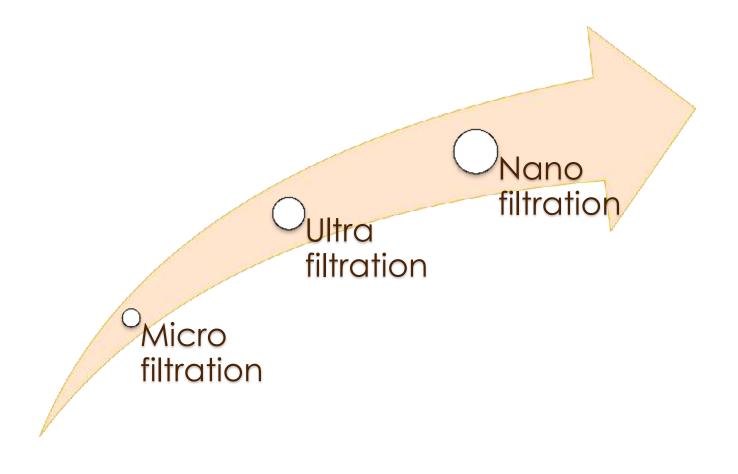
$$C + 2CI2 + 2H2O \rightarrow 4HCI + CO2$$







Membrane as pretreatment



Micro filtration

Remove particles in range 0.1 to 1 micron

Remove SS , large colloid , Bacteria

Low exit turbidity

0.7 bar (Applied pressure)

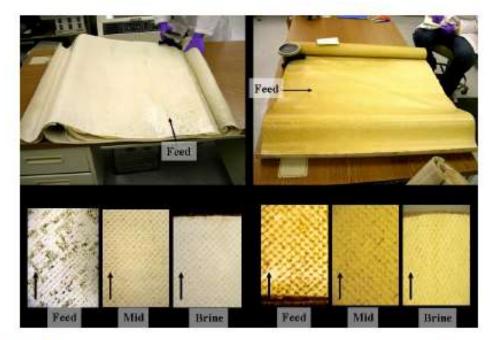
Ultra Filtration

Remove particles in range 2 Nano to 0.1 Micron

colloid , Bacteria , Proteins , microbiological contaminants

Remove biological fouling

1-7 bar (Applied pressure)



MF/UF Pretreatment

Conventional Pretreatment



Nano Filtration

Remove particles in range 1 Nano

Calcium , Sodium , magnesium , color , TOC

Softening , surface water , waste water

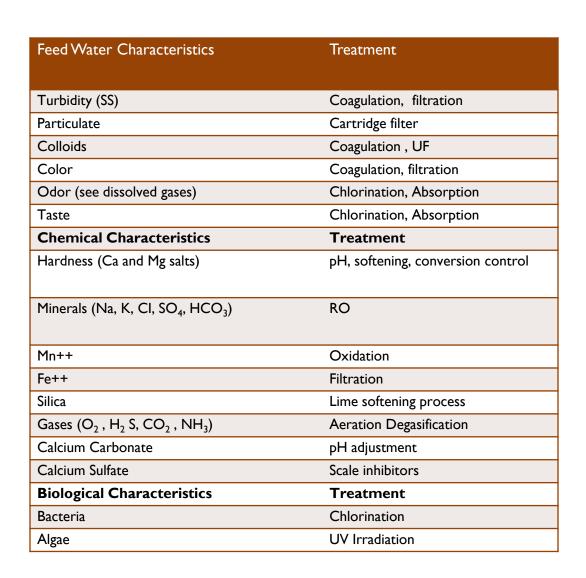
3.5-16 bar (Applied pressure)

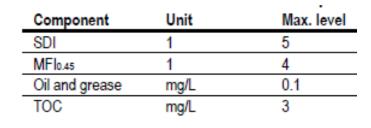
Pre treatment

Table 1

Pretreatment	CaCO,	CaSO ₄	BaSO,	SrSO ₄	CaF ₂	SiO ₂	SDI	Fe	AI	Bacte- ria	Oxid. Agents	Org. Matter
Acid Addition	•			-			-	0				
Scale Inhibitor	0	•	•	•	•	0						
Softening with IX	•	•	•	•	•							
Dealkalization with IX	0	0	0	0	O							
Lime Softening	0	0	0	0	0	0	0	0				0
Preventive Cleaning	0					0	ಂ	0	0	0		0
Adjustment of Operation Parameter		0	0	0	0	•						
Media Filtration						0	ಾ	0	0			
Oxidation-Filtration							0	•				
In-Line-Coagulation							0	0	0			0
Coagulation-Flocculation						0	•	0	0			•
Micro-/ Ultrafiltration						•	•	0	0	0		•
Cartridge Filter						0	0	0	0	0		
Chlorination										•		
Dechlorination											•	
Shock Treatment										0		
Preventive Disinfection										0		
GAC Filtration										0	•	•

O Possible • Very effective





COD	mg/L	10
AOC	μg/l Ac-C	10
BFR	pg/cm ² ATP	5
Free chlorine	mg/L	0.1

Ferrous iron	mg/L	4
Ferric iron	mg/L	0.05
Manganese	mg/L	0.05
Aluminum	mg/L	0.05

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Basic Terms and Definitions

Reverse osmosis system flow rating

• 10 cubic meter / day R.O = 10 cubic meter permeate flow

Recovery rate

Q permeate / Q feed

Rejection

• % Rejection = [(Cf-Cp)/Cf]* 100

Flux

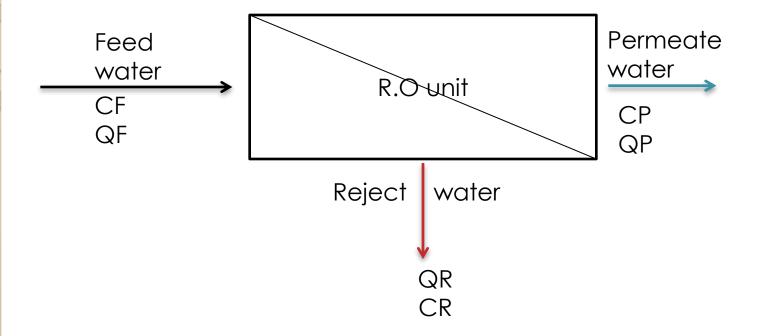
 the volumetric flow rate of a fluid through a given area (Qp / (N *Area of membrane))

Fouling

 suspended solids, organics, or microbes on the surface of the membrane

Scaling

• result of precipitation of saturated salts



CF: Concentration of feed water

QF: Flow rate of feed water

CP: Concentration of permeate water

QP: Flow rate of permeate water

QR: Flow rate of reject water

CR: Concentration of reject water



100 mg/l NaCl I Liter water /S 2 membrane A of membrane = 37 m2

Reject water

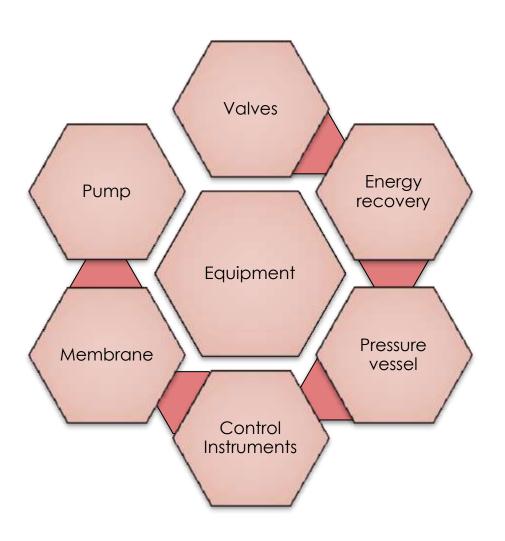
0.25 Liter water /s 400 mg/l

Permeate water

0 mg/l NaCl 0.75 Liter water /s

Recovery rate = 0.75 / I = 0.75 = 75 % Rejection % = (100-0) / 100 = I = 100% Flux = 0.75 *3600 / (2*37) = 36.5 L/m2/hr

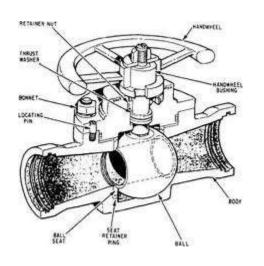
Reverse osmosis unit equipment



Valves

Plug Valve Butterfly and ball Valve **Automated Valve** Needle Valve Check Valve











Pumps

Source intake pump Feed pump High pressure pump Booster pump Dosing pump

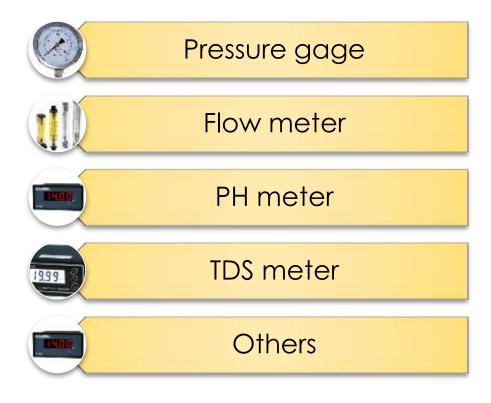


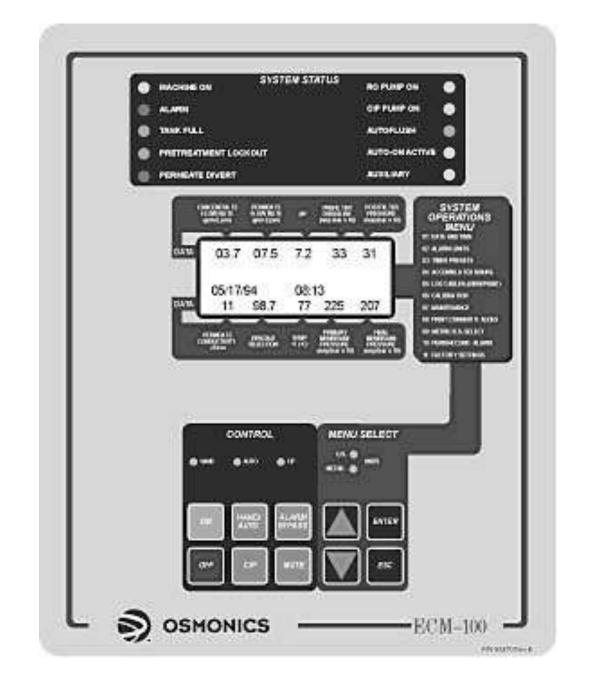






Control instrument





Shutdown switches

Undue operating condition	Provision
Too high feed pressure	High pressure shutdown switch in the feed line
Insufficient feed pressure	Low pressure shutdown switch in the pump suction line
Too high feed temperature	High temperature switch in the feed line
Permeate pressure exceeding feed by more than 0.3 bar (5 psi) pressure	Pressure relief mechanism in the permeate line
Too high concentration of colloidal matter in the feed	Turbidity control in the feed line
Too high concentration of sparingly soluble salts in the feed	Dosing pumps for acid and antiscalant should be electrically interlocked with the feed pump drive
	High pH shutdown switch
Oxidizing agents in the feed	ORP (Oxidation Reduction Potential) control in feed line or chlorine
	detection monitor with automatic shutdown
Oil in the feed	Oil detector in feed line

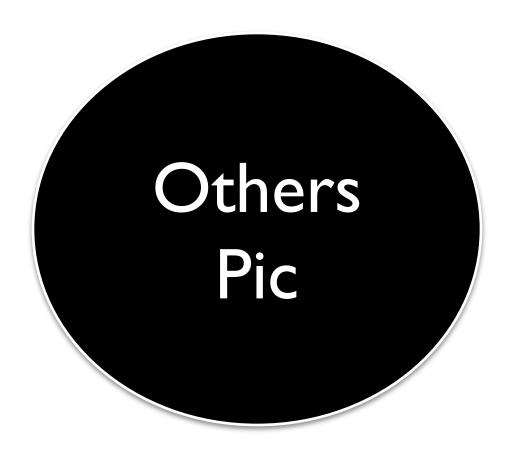










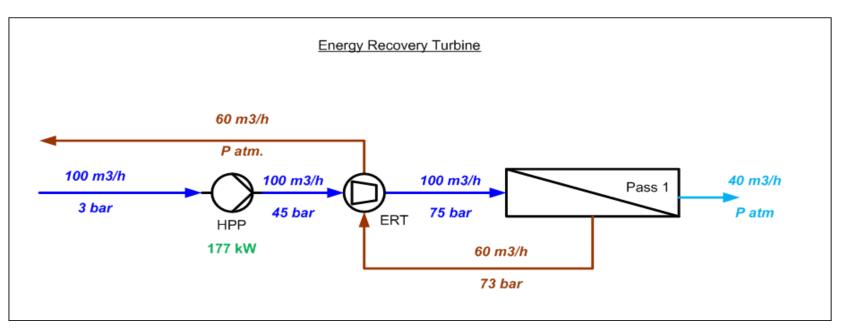


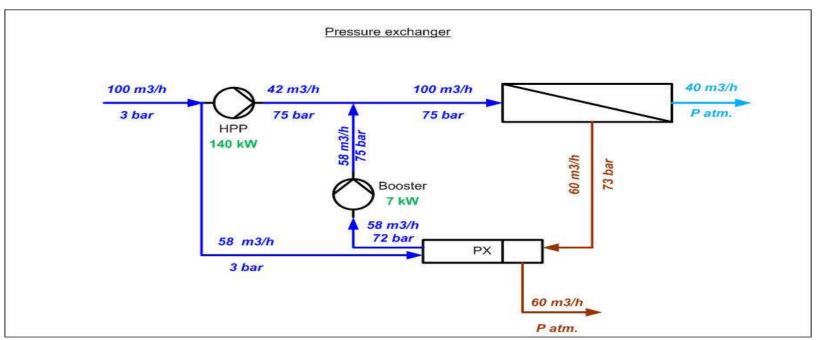
Energy recovery device

Usually pressure drop in Reject flow (1-3 bar)

Turbine (30-40 %)

Pressure exchanger (50-60%)





Pressure exchanger device



Membrane:

The desalination performance of a RO membrane depends largely on the membrane material and the membrane structure.

An industrially useful RO membrane must exhibit several characteristics such as high water flux, high salt rejection, mechanical stability, tolerance to temperature variation, resistance to fouling, and low cost.

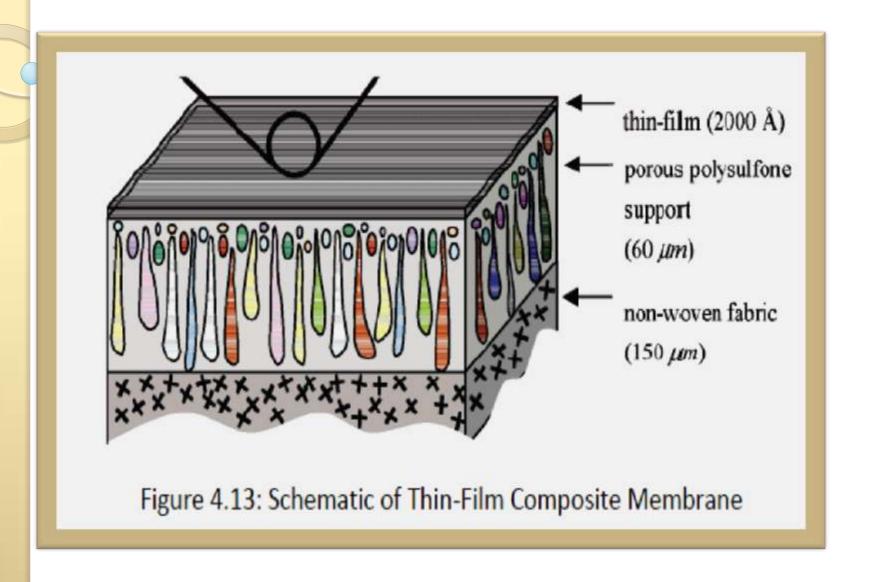
Classification of membrane material

> Cellulose acetate (CA)

A good chlorine resistance, less probability to fouling, PH (4-6)

> Polyamide acetate (PA)

A high flux and very high salt rejection, PH (2-12) >>> (MOST USE)



Membrane Modules (structure)

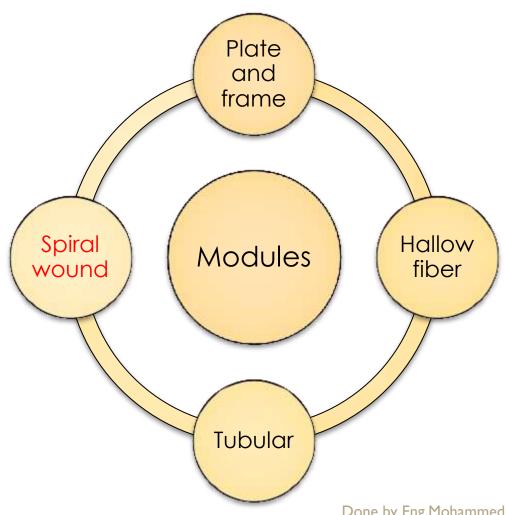




Plate and frame

For Waste water treatment applications

Hallow fiber

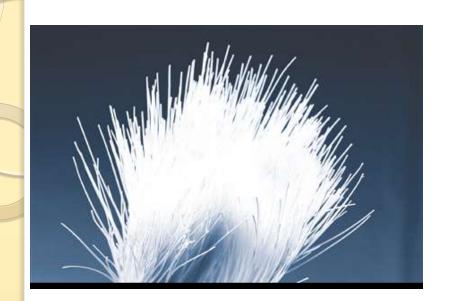
For microfiltration, Ultrafiltration and WWT applications

Tubular

For gas separation

Spiral wound

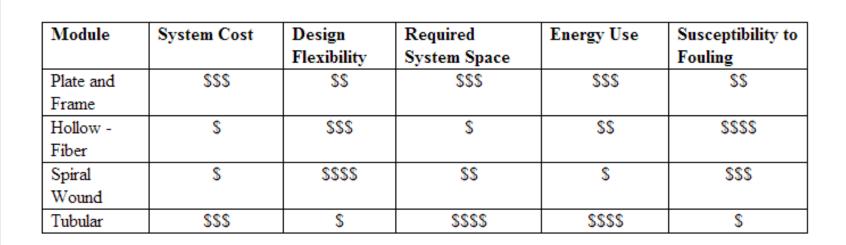
For desalination

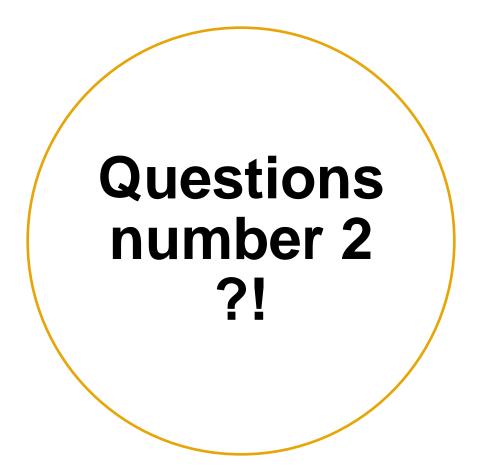






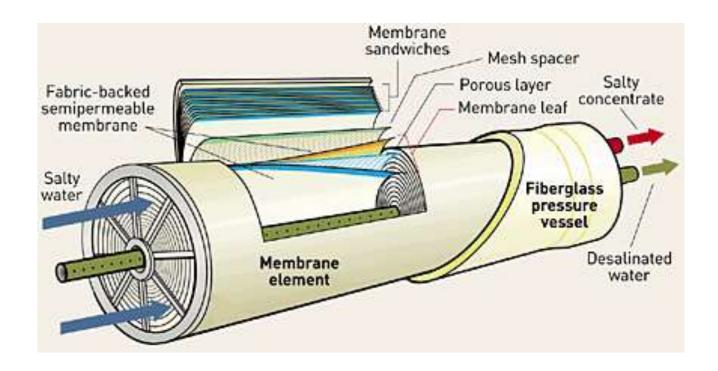




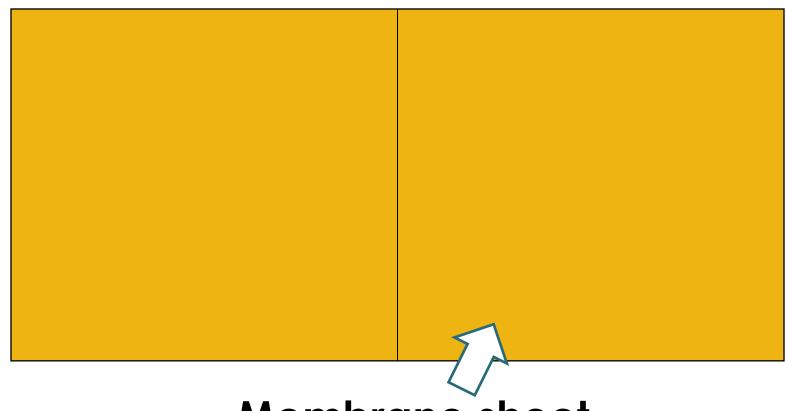


Spiral Wound

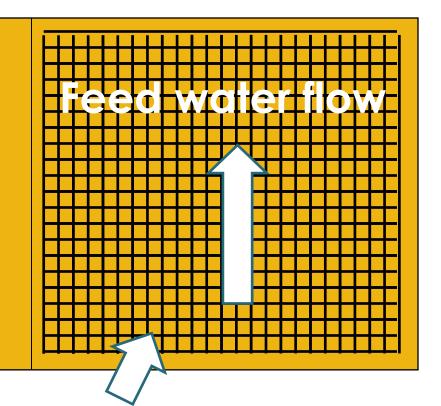
°Spiral wound membrane modules are the most common type of module used for RO today The major advantage of a spiral wound module is that the packing density is fairly high, about 150 - 380 ft2/ft3, higher then for plate and frame or tubular modules.



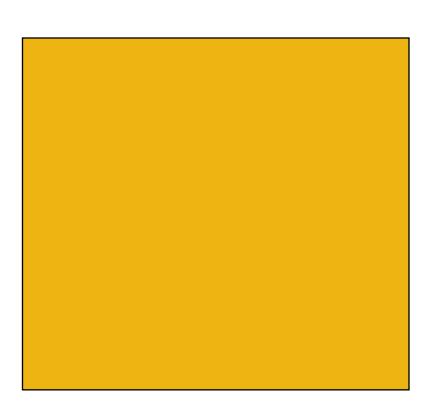
Spiral Wound Modules

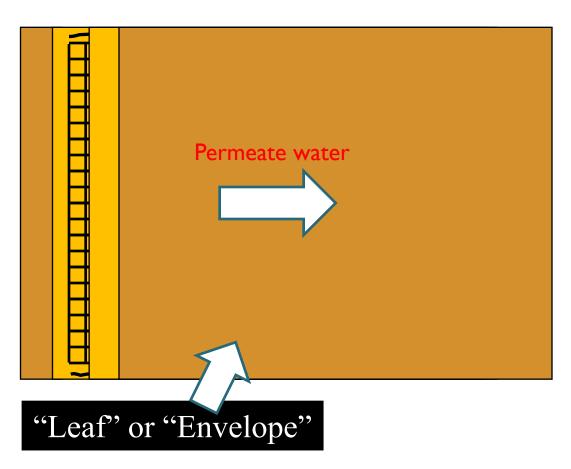


Membrane sheet



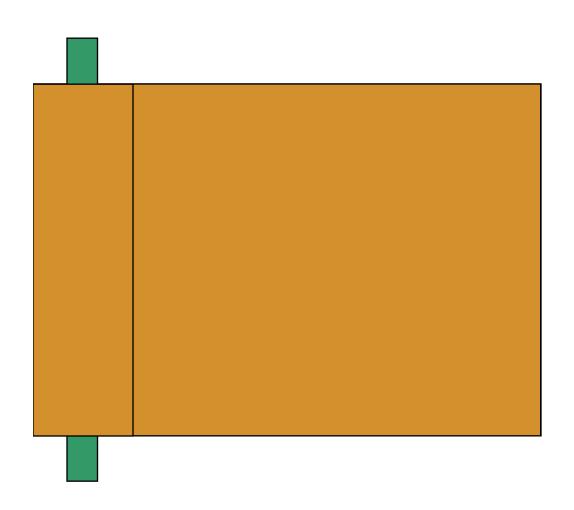
spacer











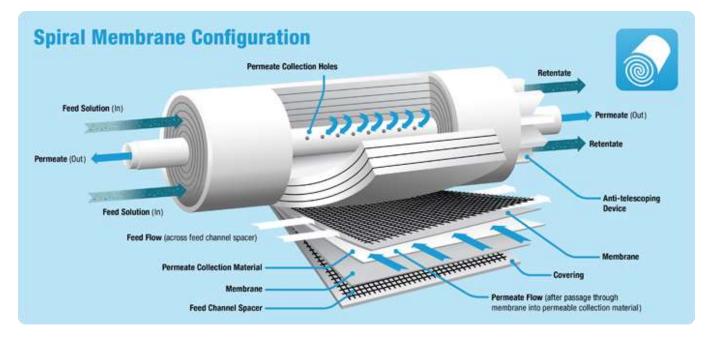
Commercially-Available Membranes

- > Seawater membranes
- > Brackish water membranes
- > Brackish, low-energy membranes
- > Brackish, low-differential pressure membranes
- > Brackish, low-fouling membranes.
- > Other Membrane/Module Types, like Boron rejection membranes.

Spiral wound membrane video







8" element

Membrane area 40m2 (430 ft2) Nominal flow 45 m3/day Avg. field flow 19 m3/day

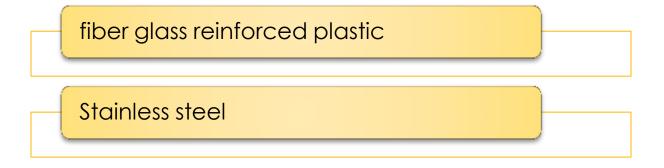
16" element

Membrane area 140 m2 (1,500 ft2) Nominal flow 155 m3/day Avg. field flow 68 m3/day



Membrane vessel(pressure vessel)

Pressure vessel (membrane element housing) is designed for specific pressure applications.



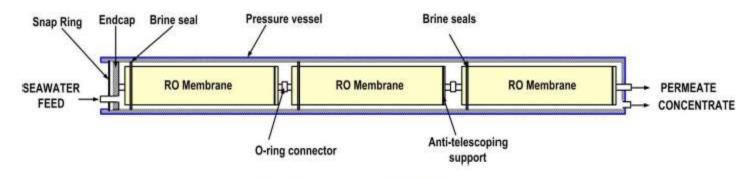
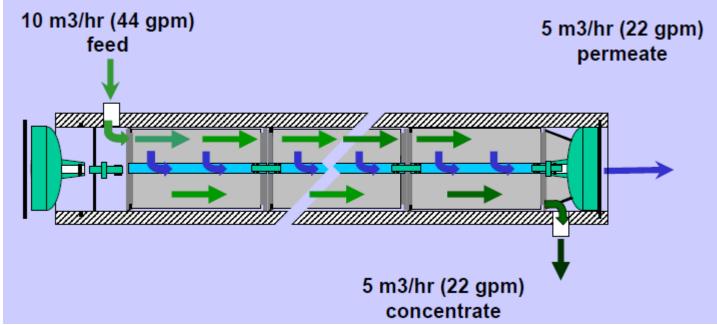


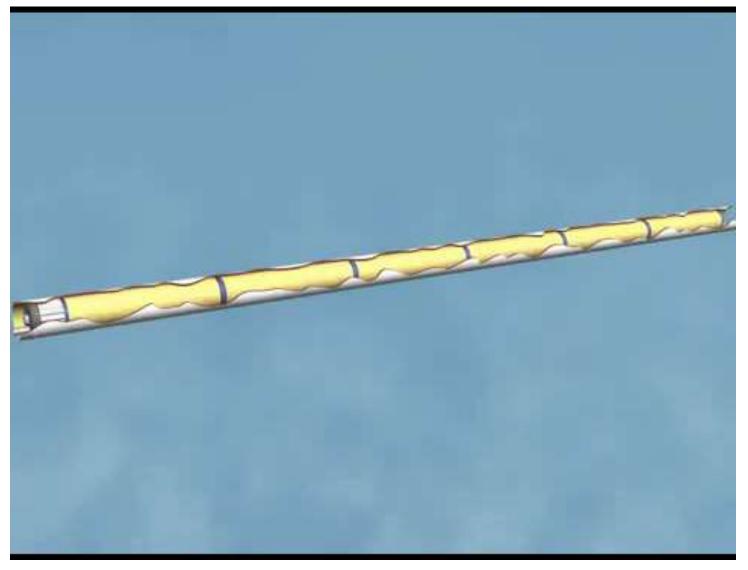
Figure 4.22: cross section Inside the pressure vesselby Eng. Mohammed Ziad AlKhateeb







Membrane Pressure vessel video

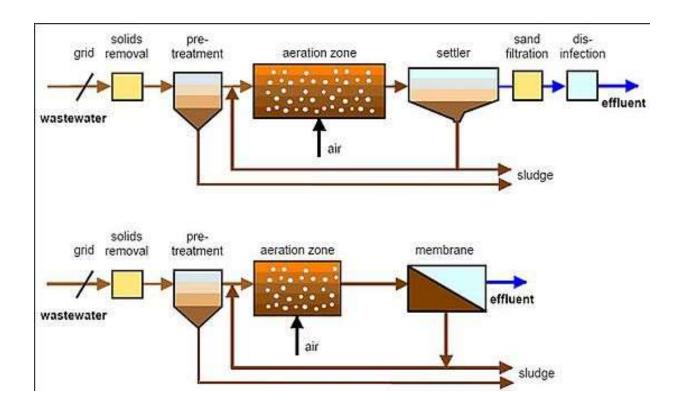


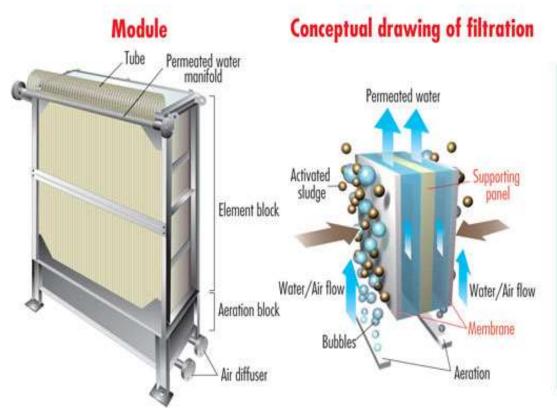
Membrane new technology

Membrane Bioreactor (MBR)

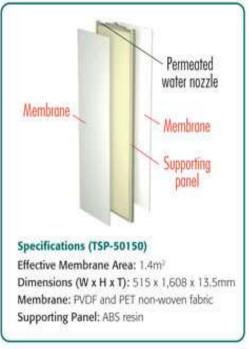
Forward osmosis desalination

Membrane bioreactor





Element





< 2 mg/L

TSS

 $< 0.5 \,\mathrm{mg/L}$

NH3-N

< 0.5 mg/L

TN

< 3 mg/L

TP

 $< 0.05 \, \text{mg/L}$

Turbidity

< 0.2 NTU

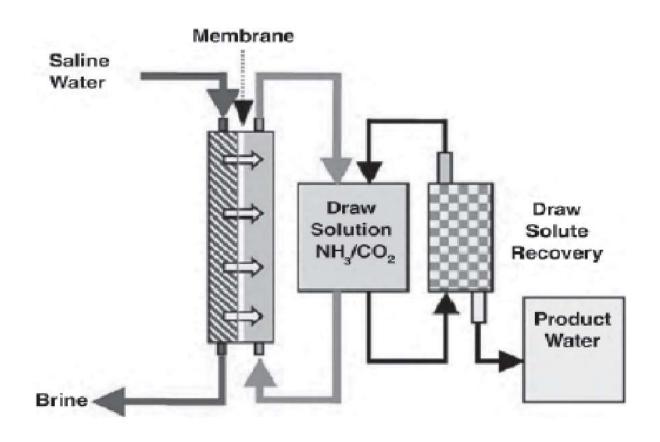
Fecal Coli

< 10 CFU/100 mL

Using R.O without pretreatment



Forward osmosis desalination



Introduction to design



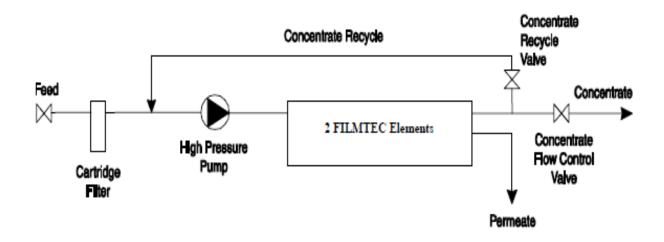
Factors Affecting Reverse Osmosis Performance (Flux & Salt rejection)

- 1-Pressure
- 2-Temperature
- 3-Recovery
- 4-Feed water salt concentration
- 1- Pre treatment
- 2- Operation
- 3- Maintenance

Increasing	Permeate Flow	Salt Passage
Effective pressure	↑	↓
Temperature	↑	1
Recovery	\downarrow	1
Feed salt correction	↓	1

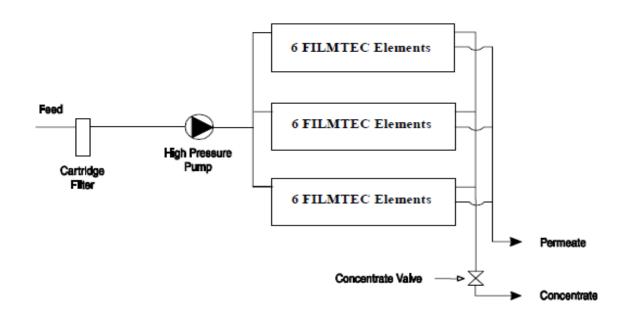
Single-Module System

A module consists of a pressure vessel with up to eight membrane elements, which are connected in series. The concentrate of the first element becomes the feed to the second, and so on. The product tubes of all elements are coupled and connected to the module permeate port. The permeate port may be located on the feed end or on the concentrate end of the module.



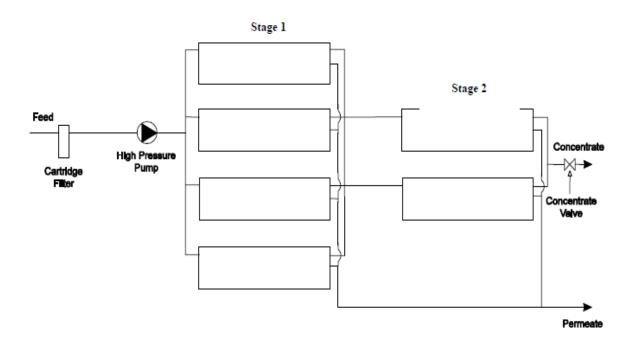
Single-Stage System

In a single-stage system, two or more modules are arranged in parallel. Feed, product and concentrate lines are connected to manifolds. Other aspects of the system are the same as in a single-module system. Single-stage systems are typically used where the system recovery is less than 50%, e.g., in seawater desalination.



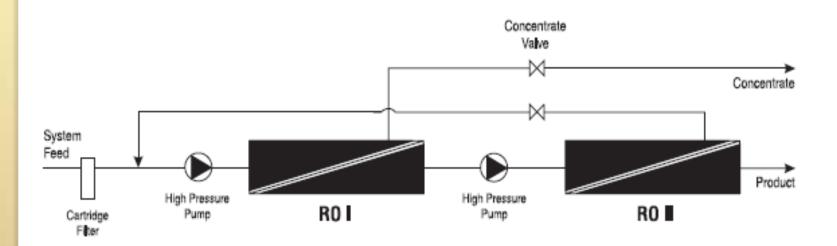
Multi-Stage System

Systems with more than one stage are used for higher system recoveries without exceeding the single element recovery limits. Usually two stages will suffice for recovery up to 75%, and three must be used for higher recovery. These numbers are based on the assumption that standard pressure vessels with six elements are used

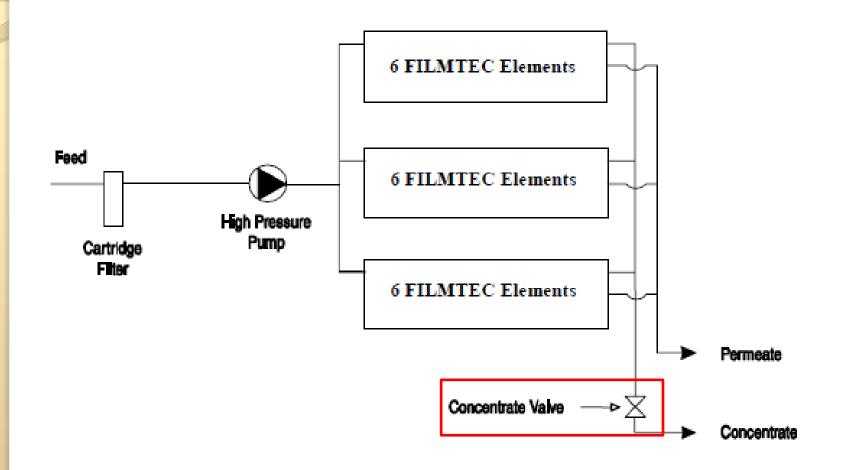


Permeate-Stage System

The production of water for pharmaceutical and medical use is a typical application of permeate staged systems. A permeate staged system is the combination of two conventional RO/NF systems where the permeate of the first system (first pass) becomes the feed for the second system (second pass)



Concentrate flow control & Auto flush valve











Zara Ma'in reverse osmosis desalination plant

One of the 100 largest desalination plant in the world (2005) Total Cost = 125 million \$

Total feed water = 55MCM/Y

Zara Springs 6.4 MCM/Y 11.7%

Wadi Mujib 27.5 MCM/Y 50.0%

Wadi Ma'In 21.1 MCM/Y 38.3%

Recovery rate = 87.5 %

Number of RO trains = 9

Max number of trains in service = 8

Number of stages = 3

Raw water silt density index (SDI) <3 normally Maximum of 5

Membrane type Spiral wound = 400 ft2

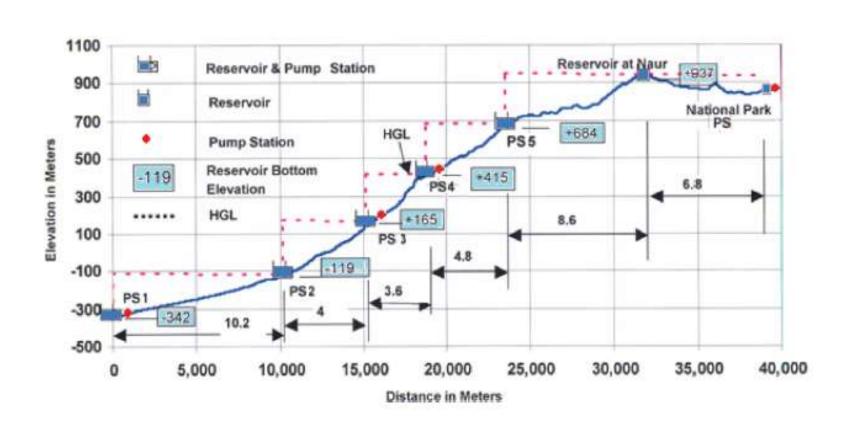
Type 1: BW30-400FR (Dow), Type 2: LE-400 (Dow)

Feed water TDS = 1460 mg/l

Product water TDS = 185 + blending

RO inlet pressure range =14 bar

Differential pressure across the membrane = 2.31 bar



Red Dead Sea projects

The largest desalination plant in the world Total Cost = 13 Billion \$

- -Total feed water = 2150 MCM/Y
- -Total permeate water = 930 MCM/Y (560 MCM/Y in 2035)
- -Total rejected water = 1220 MCM/Y (Dead Sea)

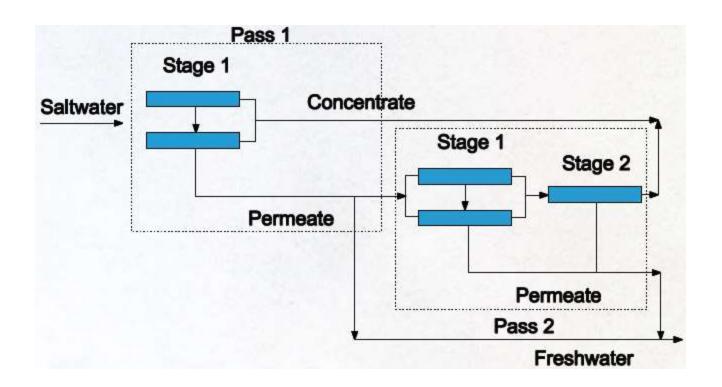
Sea water desalination plant # 1

- phase 1= 50 MCM/Y (Aqaba)
- Phase 2 = 80 MCM/Y

Sea water desalination plant # 2

- phase 1= 150 MCM/Y (Amman)
- Phase 2 = 850 MCM/Y
- Hydropower = 180 mega watts





- 930 MCM/Y total freshwater capacity
- 297 pumps require 194,176 kwh of electrical power
- Plant design recovers 50% of energy used
- Worlds largest desalination facilities

Training Course on Planning of Desalination plant using R.O system

Introduction to Desalination Introduction to Reverse osmosis Pretreatment R.O Post treatment **Others**

Post treatment

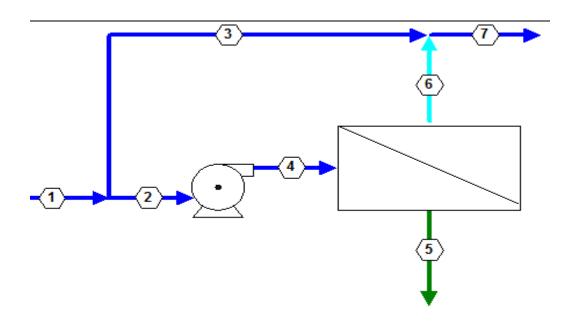
Water from a desalination process is typically void of dissolved solids resulting in finish water with low hardness, low alkalinity and high amount of dissolved gases. As a result, desalinated water without post-treatment is corrosive toward the metal and concrete surfaces of pipelines and other wetted surfaces

The aims of the post treatment:

- 1-Remove corrosion effect of water
- 2-Make sure that the water permeate matching the specifications and standards required, such as pH and demineralization

Blending:

Desalinated waters are commonly blended with small volumes of more mineral-rich waters to improve their acceptability and particularly to reduce their aggressive attack on materials, the major ions added are sodium and chloride.



Neutralization and PH adjustment

Reverse Osmosis (RO) Systems can lower the pH of Process Water by (1 - 2)points, creating slightly acidic water so that can pose a risk to piping and equipment of (RO) unit.

The Adjustment of pH as Pre/post treatment and that depend on the applicable conditions.

Degasification or Decarbonation

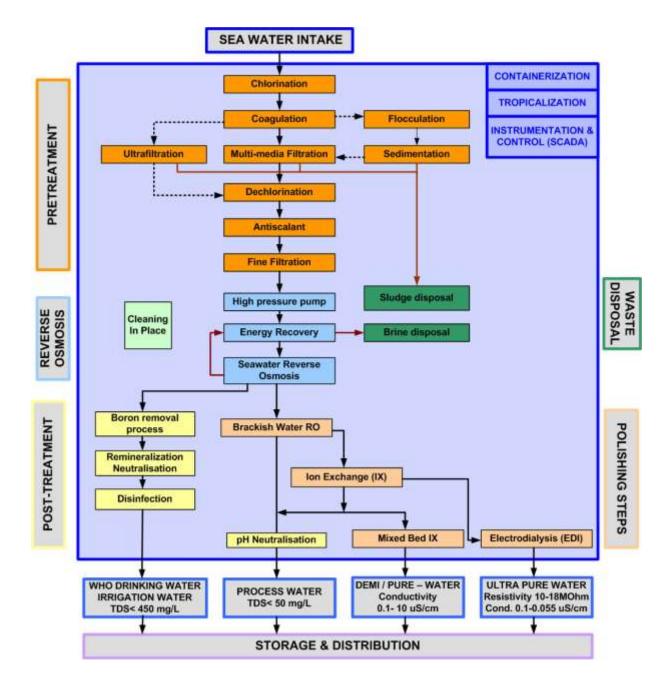
Degasification or Decarbonation is the removal of dissolved gases from liquids, The purpose of the post treatment decarbonation is to remove dissolved carbon dioxide in the reverse osmosis (RO) permeate water in order to increase the pH value.

Corrosion and Corrosion inhibitor

Corrosion is destruction or loss of metal through chemical or electrochemical reaction with its surrounding environment, The solution of Corrosion problem the metal with thin films to prevent free oxygen and water and corrosion inhibitor (Chromate, Zinc and Ortho-phosphate).

Ultra violet (UV)

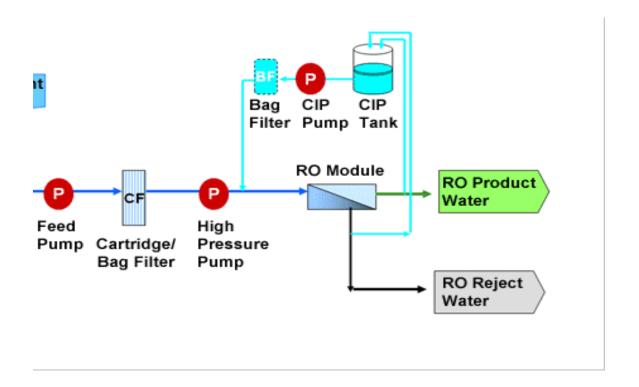
Ultraviolet radiation (254 nm) Uses in pre/post treatment to disinfect water from (M.O), that occur when UV rays penetrate the cells of harmful bacteria and viruses in our drinking water, destroying their ability to reproduce, and same any system UV has Advantages and Disadvantages of using.



Membrane cleaning (CIP)

When:

- permeate flow decrease 10%
- salt passage increases 5 10%
- pressure drop increases 10 15%



Cleaning Procedure

- 1. Make up cleaning solution.
- 2. Low-flow pumping. Use only enough pressure to compensate for the pressure drop from feed to concentrate. The pressure should be low enough that essentially no or little permeate is produced.
- 3. Recycle, for 15 min
- 4. Soak. Turn the pump off and allow the elements to soak. Sometimes a soak period of about 1 hour is sufficient. For difficult fouling an extended soak period is beneficial; soak the elements overnight for 10-15 hours.
- 5. High-flow pumping. for 30-60 minutes.
- 6. Flush out. RO permeate or deionized water is recommended for flushing out the cleaning solution.

Each fouling type need special cleaning solution

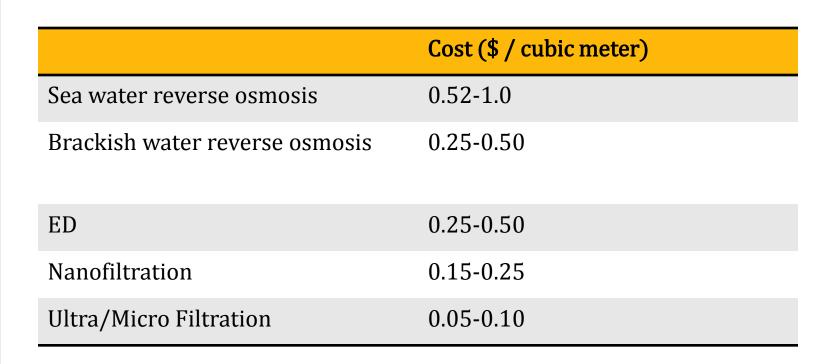
Cost determination

1. The capital cost include:

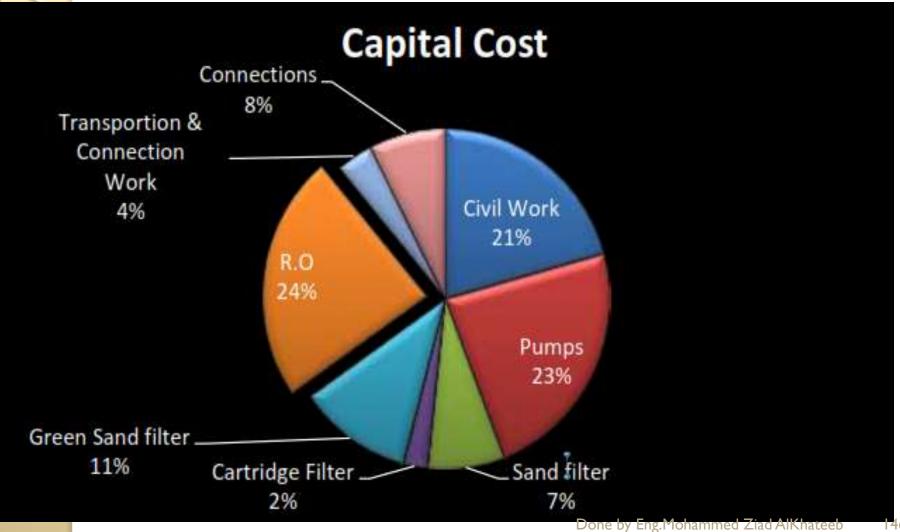
- Civil work cost (water tank, plant cover, concrete ground.....)
- Pumps (feed , dosing ,source)
- Pretreatment (sand, carbon, green sand, cartridge)
- R.O (pressure vessel, membrane, skid, control.....)
- Transportation , Connections work
- Others (pipe, fitting, control instrument)

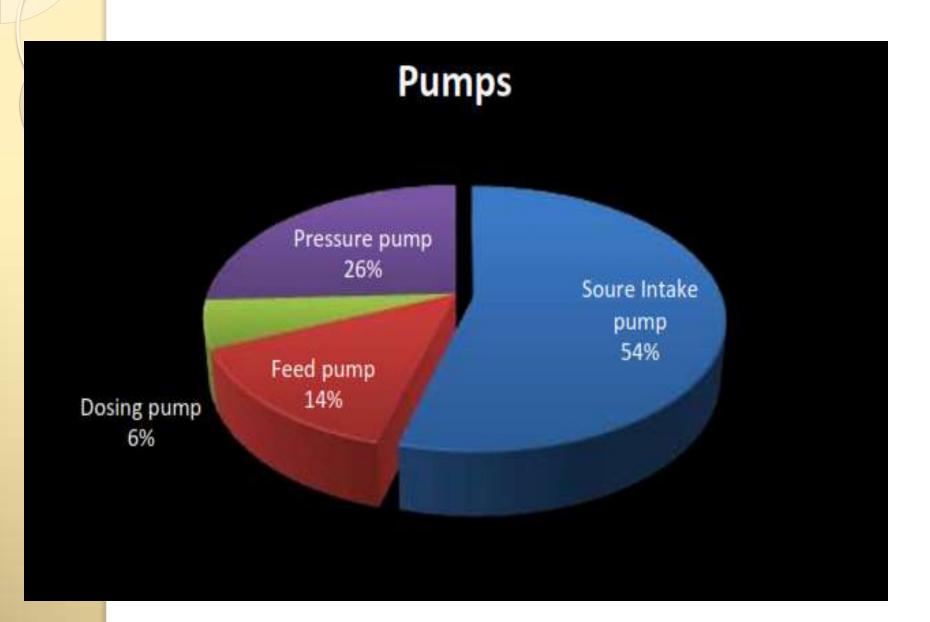
2. The running cost includes:

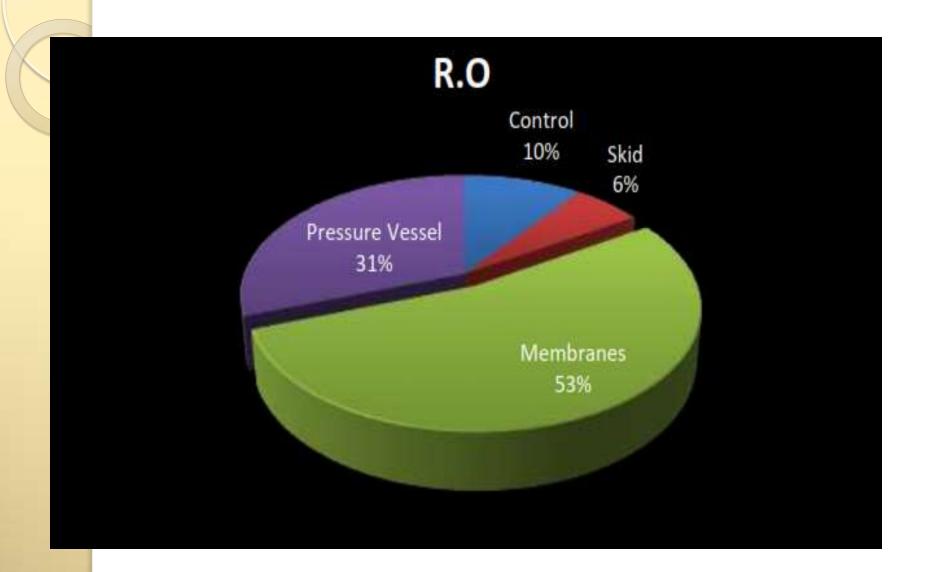
- Chemicals Cost (Anti scalants, chlorine, Acid addition....)
- Staff Cost
- Energy Cost (pump , A/C , lighting ...)
- Maintenance Cost (membrane & filter replacement)



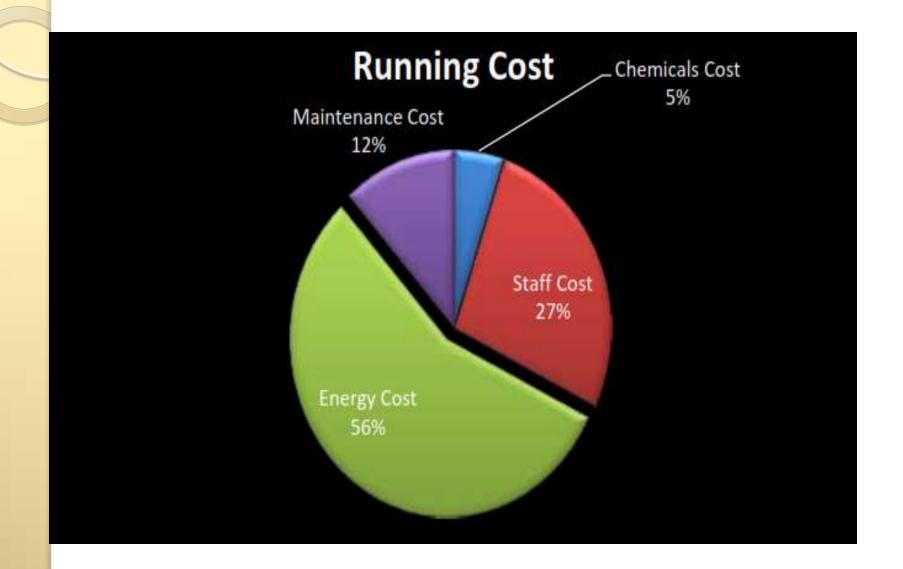
Capital Cost analysis

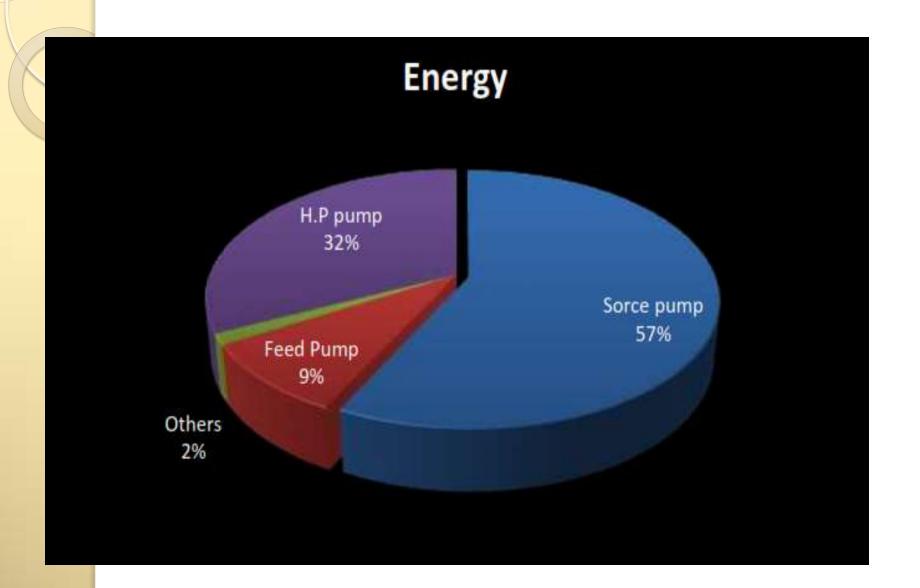


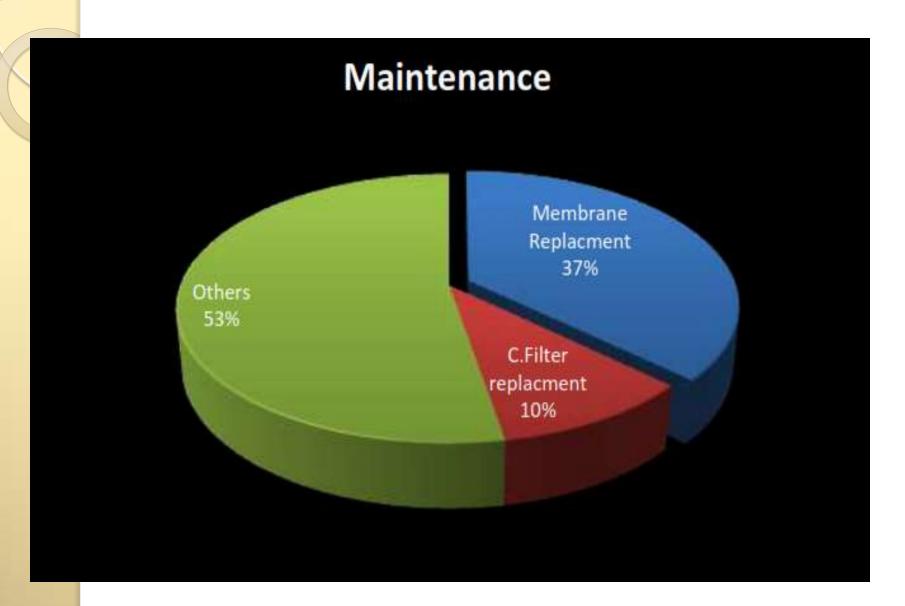




Running Cost analysis







Sea water desalination Cost analysis

