

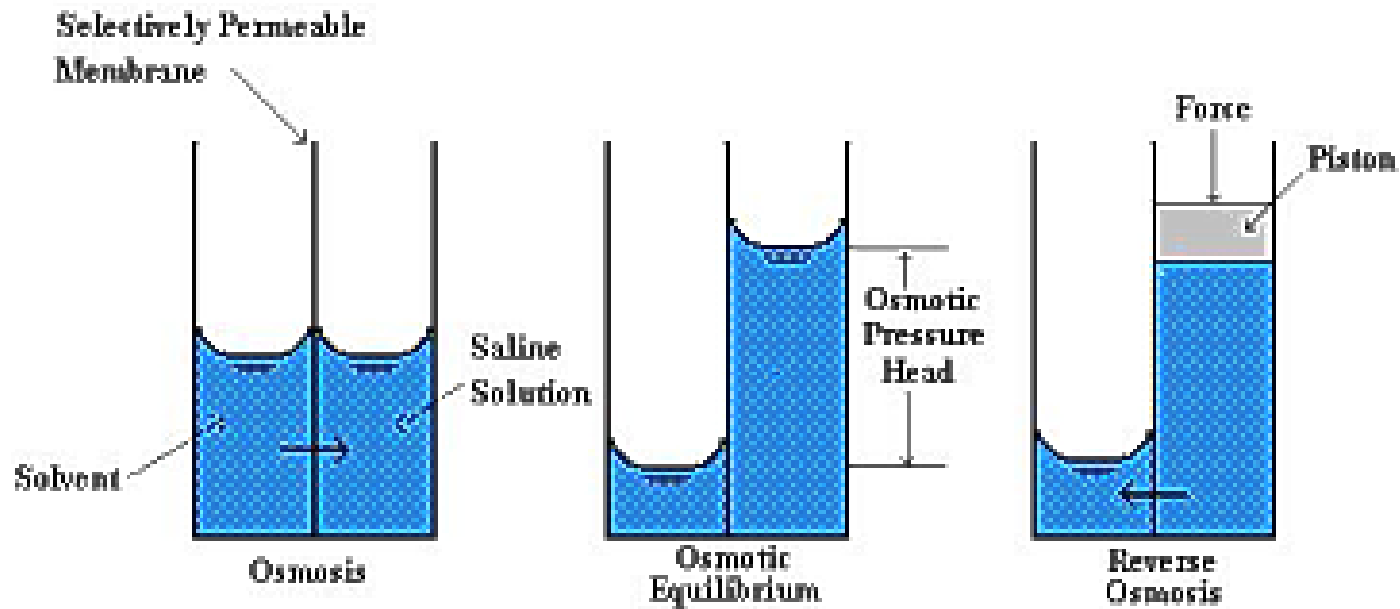
## Membrane Separations

- Reverse Osmosis (RO) [Hyperfiltration]
  - Developed in the 50's.
  - Very popular and used.
  - Ions and low molecular weight species ( $MW < 200$ ).
  - Pore size  $< 1$  nm.
  - Water molecules freely pass (0.2 nm).
  - Driving force: pressure difference.
  - Operating pressure between 10-25 (brackish) and 40-80 bar (seawater).
  - Average flux:  $5-40 \text{ L m}^{-2} \text{ h}^{-1}$ .

## Membrane Separations

- Reverse Osmosis (RO)

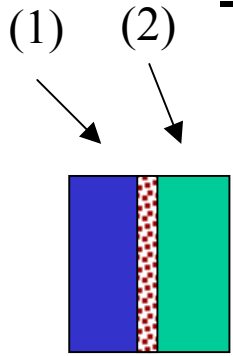
### ILLUSTRATION OF OSMOSIS



## Membrane Separations

### • Reverse Osmosis (RO)

- Why does water cross the membrane ?



$$\mu_i = \mu_i^0(T) + v_i \cdot P + R \cdot T \cdot \ln(\gamma_i \cdot c_i)$$

$$\mu_w^{(1)} = \mu_w^{(2)}$$

$$v_w^{(1)} \cdot P^{(1)} + R \cdot T \cdot \ln a_w^{(1)} = v_w^{(2)} \cdot P^{(2)} + R \cdot T \cdot \ln a_w^{(2)}$$

membrane

$$v_w \cdot (P^{(1)} - P^{(2)}) = R \cdot T \cdot \ln a_w^{(2)} - R \cdot T \cdot \ln a_w^{(1)}$$

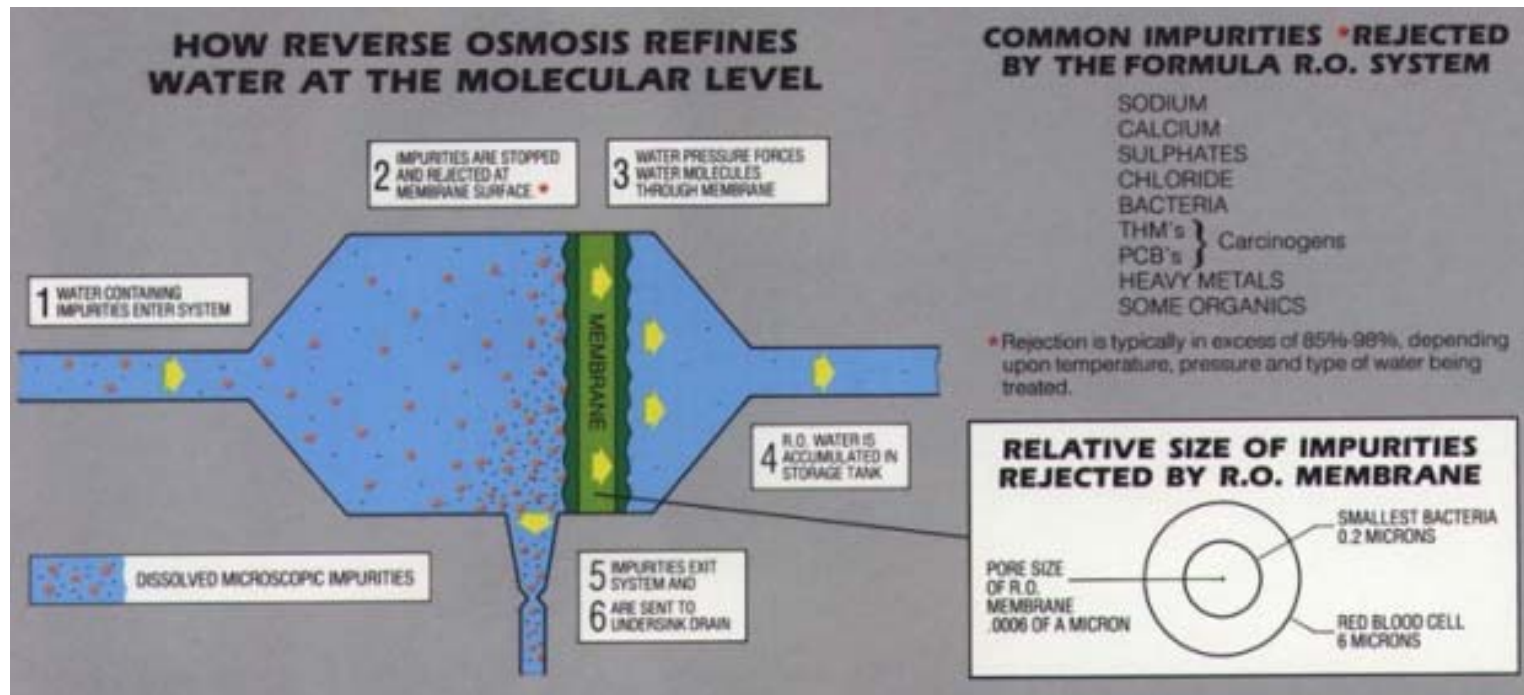
$$\pi = -\frac{R \cdot T}{v_w} \cdot \ln a_w \Rightarrow \Delta P = \Delta \pi$$

**Definition**

$$\pi \approx v \cdot c \cdot R \cdot T \quad \text{if } c_{\text{salts}} \downarrow$$

# Membrane Separations

## • Reverse Osmosis (RO)



## Membrane Separations

- Reverse Osmosis (RO)

- The water freely flows through the membrane due to the pressure difference corrected by the  $\Delta\pi$ .

$$J_w = \frac{Q_w}{A_m} = A \cdot (\Delta P - \Delta\pi)$$

$J_w$ : Solvent volume flux ( $\text{m}^3/\text{s} \cdot \text{m}^2$ )

$Q_w$ : Solvent volume flowrate ( $\text{m}^3/\text{s}$ )

$A_m$ : Membrane area ( $\text{m}^2$ )

$A$ : Permeability ( $\text{m}^3/\text{s} \cdot \text{m}^2 \cdot \text{Pa}$ )\*

$\Delta P$ : Hydraulic pressure difference (Pa)

$\Delta\pi$ : Osmotic pressure difference (Pa)

\*  $A = A_0 \cdot K_t \cdot K_c \cdot K_e$     t: temperature; c: compaction; e: fouling

## Membrane Separations

- Reverse Osmosis (RO)

- Salt flux is due to both diffusion and convective transport.

$$J_s = J_w \cdot C_p = B \cdot (C_m - C_p) + M \cdot J_w \cdot C_m$$

$$C_p = \frac{J_s}{J_w} = \frac{B \cdot (C_m - C_p)}{A \cdot (\Delta P - \Delta \pi)} + M \cdot C_m$$

$J_s$ : Solute mass flux (mol/s·m<sup>2</sup>)

$B_i$ : Permeability (m/s)

$C_m$ : Solute concentration on the membrane surface (mol/m<sup>3</sup>)

$C_p$ : Solute concentration in the permeate (mol/m<sup>3</sup>)

$M$ : Distribution constant (~ 0.005)

## Membrane Separations

- Reverse Osmosis (RO)

- Thus a relation between the local rejection,  $R_1$ , and the permeate flux (or applied pressure) can be found.

$$R_1 = \frac{C_m - C_p}{C_m} = 1 - \frac{C_p}{C_m}$$

$$\frac{1}{R_1} = \frac{1}{(1-M)} + \frac{B}{(1-M)} \cdot \frac{1}{J_w}$$

$$\frac{1}{R_1} = \frac{1}{(1-M)} + \frac{B}{(1-M)} \cdot \frac{1}{A \cdot (\Delta P - \Delta \pi)}$$

## Membrane Separations

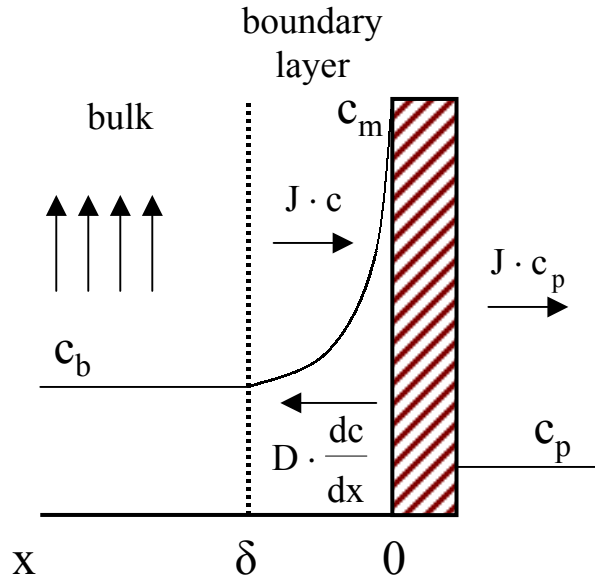
- Reverse Osmosis (RO)
  - Membranes mostly made of CA or PA.
  - Configuration in spiral-wound or hollow fiber.
  - Typical conversion between 10 and 30%.
  - Flow limited by concentration polarisation.
  - Rejection up to 99% (sometimes higher).
  - Fouling problems and cleaning. Pretreatment.



# Membrane Separations

- Reverse Osmosis (RO)

- Concentration polarisation



$$\ln \frac{c_m - c_p}{c_b - c_p} = \frac{J \cdot \delta}{D}$$

$$\frac{c_m - c_p}{c_b - c_p} = \exp\left(\frac{J \cdot \delta}{D}\right)$$

$$\frac{c_m}{c_b} = \frac{\exp\left(\frac{J}{k}\right)}{R_1 + (1 - R_1) \cdot \exp\left(\frac{J}{k}\right)}$$

D: Diffusion coefficient ( $\text{m}^2/\text{s}$ )

$\delta$ : boundary layer thickness (m)

$k=D/\delta$ : mass transfer coefficient (m/s)

## Membrane Separations

- Reverse Osmosis (RO)

- Applications: Drinking water, food industry, wastewater treatment. Examples:

- √ Desalting of process water.

- √ Desalination of brackish or seawater.

- √ Production of ultrapure water for laboratories or electronic industry.

- √ Concentration wastewater in paper pulping.

- √ Concentration of juices, milk or sugar solutions.

- √ Concentration of coffee, te or soups.

- √ Concentration of aminoacids (and other pharmaceutical substances).

## Membrane Separations

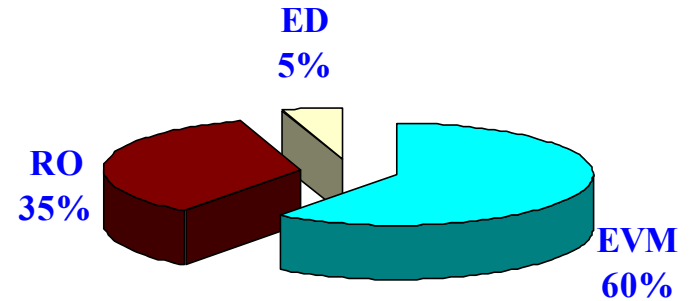
- Reverse Osmosis (RO)

- Seawater desalination:

30-40% market

Very competitive process

Module cascade in series and parallel



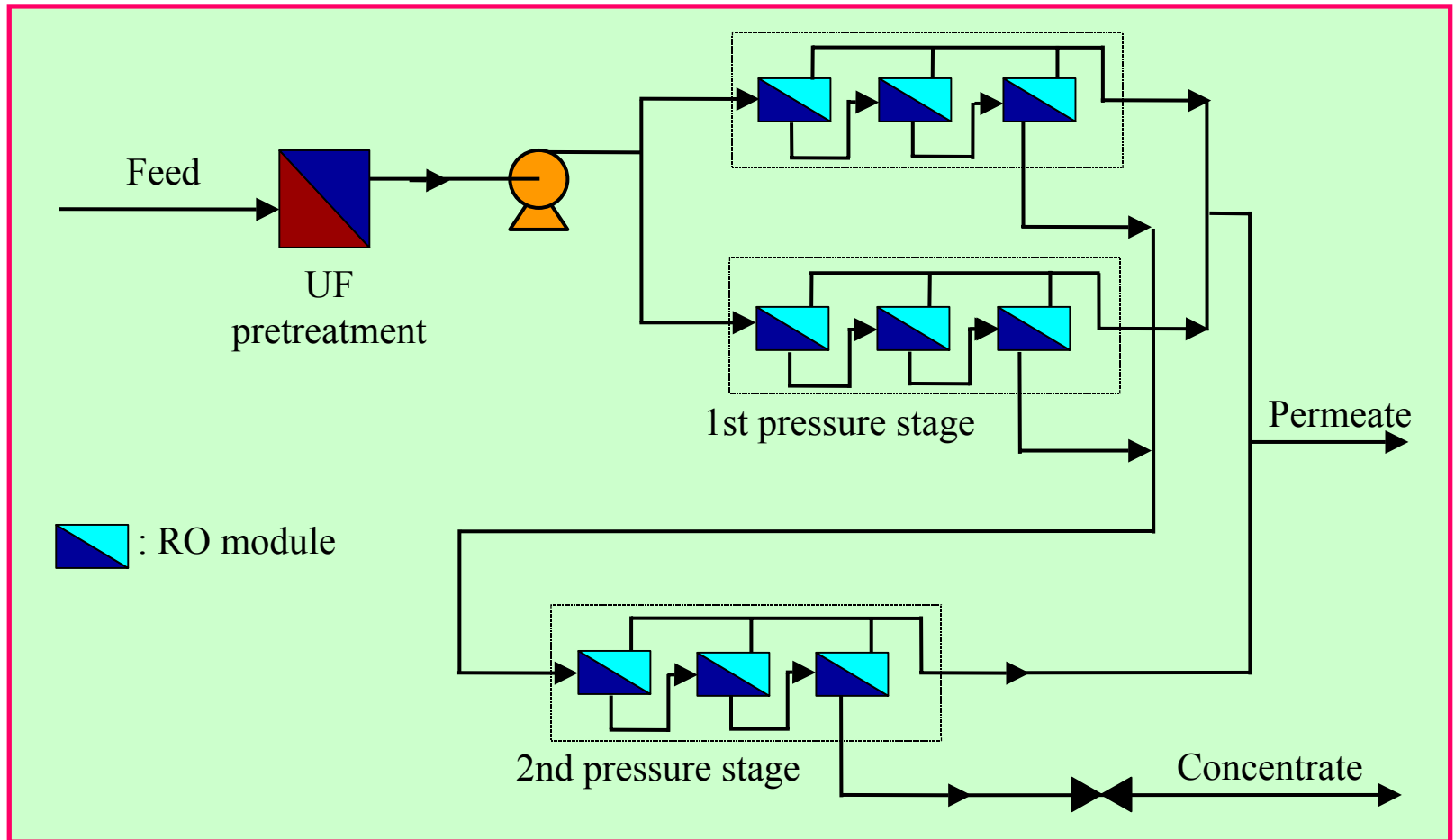
**Desalination technologies (1996)**

### **Comparison between the energetic cost for seawater desalination.**

<b>Process</b>	<b>Energy (kWh/m<sup>3</sup>)</b>
Multiple distillation	15.5
Reverse Osmosis	9
RO with energy recovery	6.5
Electrodialysis	12

# Membrane Separations

- Reverse Osmosis (RO)



**Water production plant based on RO.**

