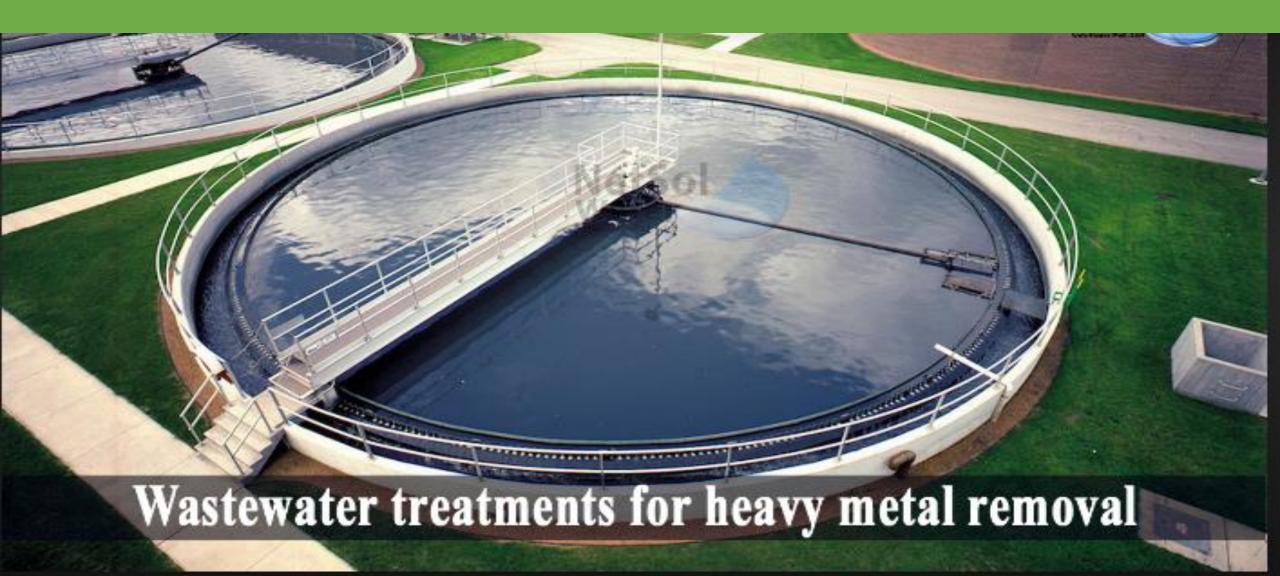
Removal of Heavy metal from wastewater treatment



Definition of Heavy Metals

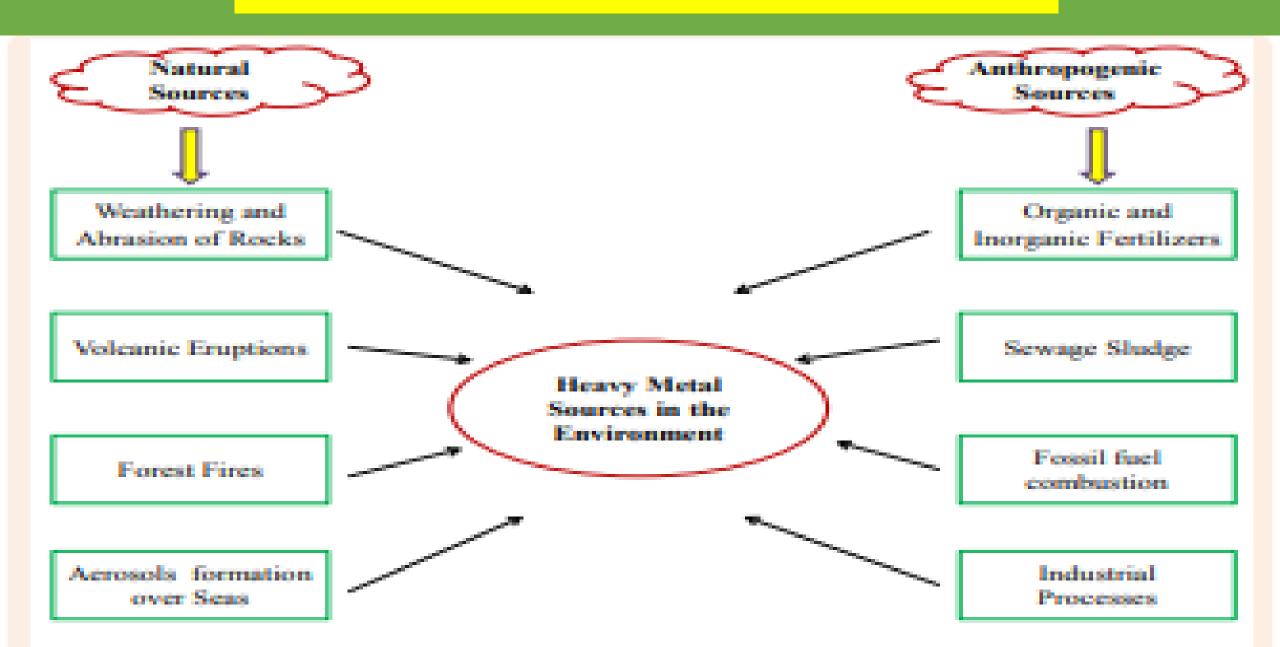
Heavy metals are:

- Any metal and metalloid element with an atomic density ranging from 3.5 to 7 g cm⁻³
- one of the most persistent pollutants and non biodegradable in wastewater
- Includes mercury (Hg), cadmium (Cd), arsenic(As), chromium (Cr), thallium (Tl), zinc (Zn), nickel (Ni), copper (Cu) and lead (Pb).
- Some of them have an essential role in humans and animals metabolism in very trace amounts but their higher concentration may cause toxicity and health hazards.
- Their hazardous nature refer to their bio-accumulative nature in biotic systems.

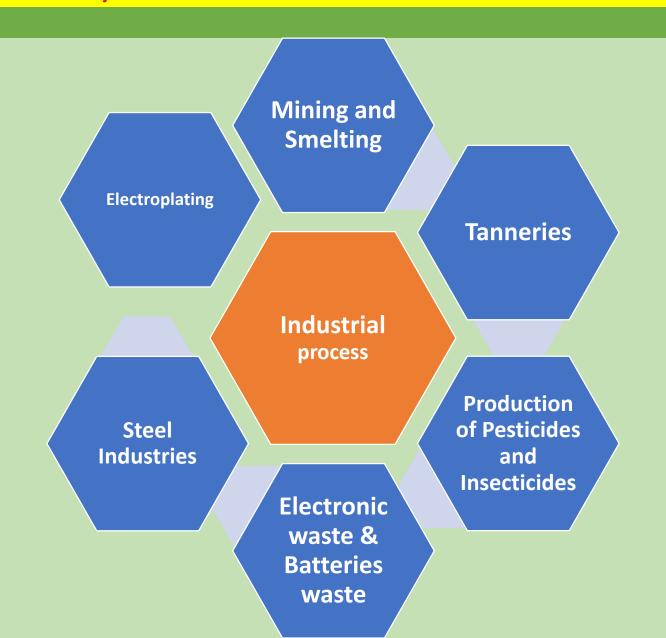
Physical properties of metals



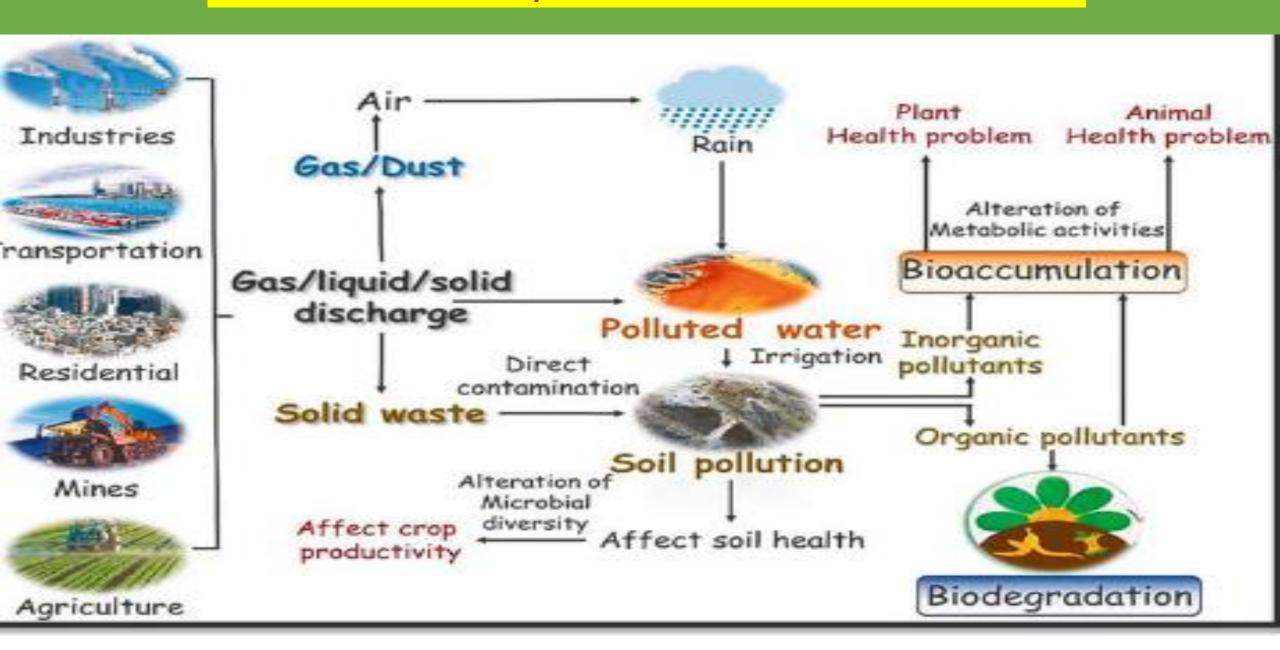
Sources of metals in Environments

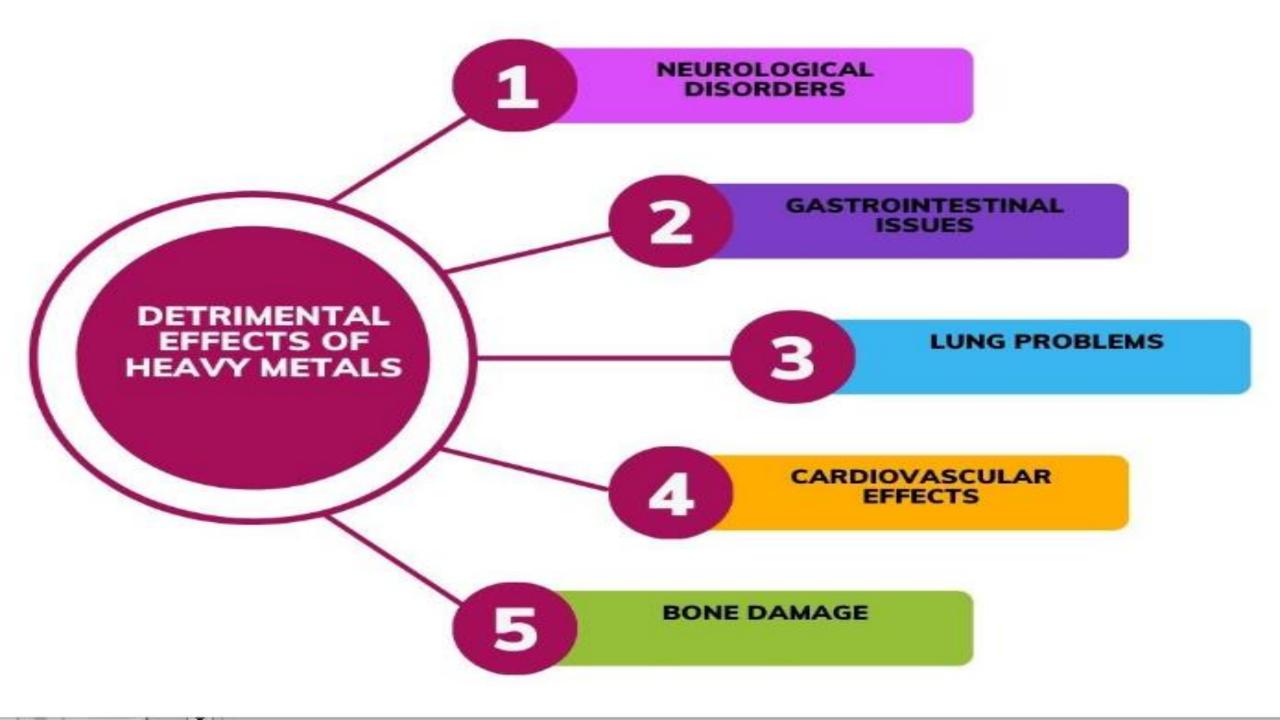


Sources of heavy metals from some Industrial process



Effect of heavy metal in Environment





Cadmium, Mercury attack kidney and liver, hypertention, impaired reproductive functions

Nickel

chronic bronchitis decreased lung function lung cancer

Heavy Metals

Arsenic

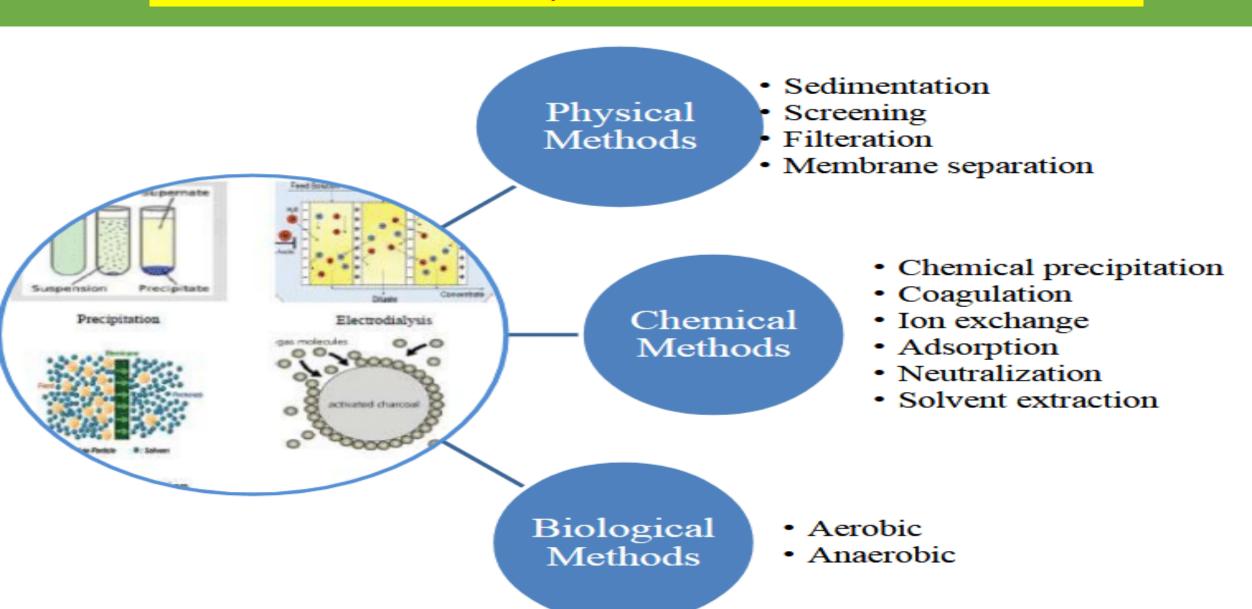
cancer of the skin, lungs, liver and bladder

Lead

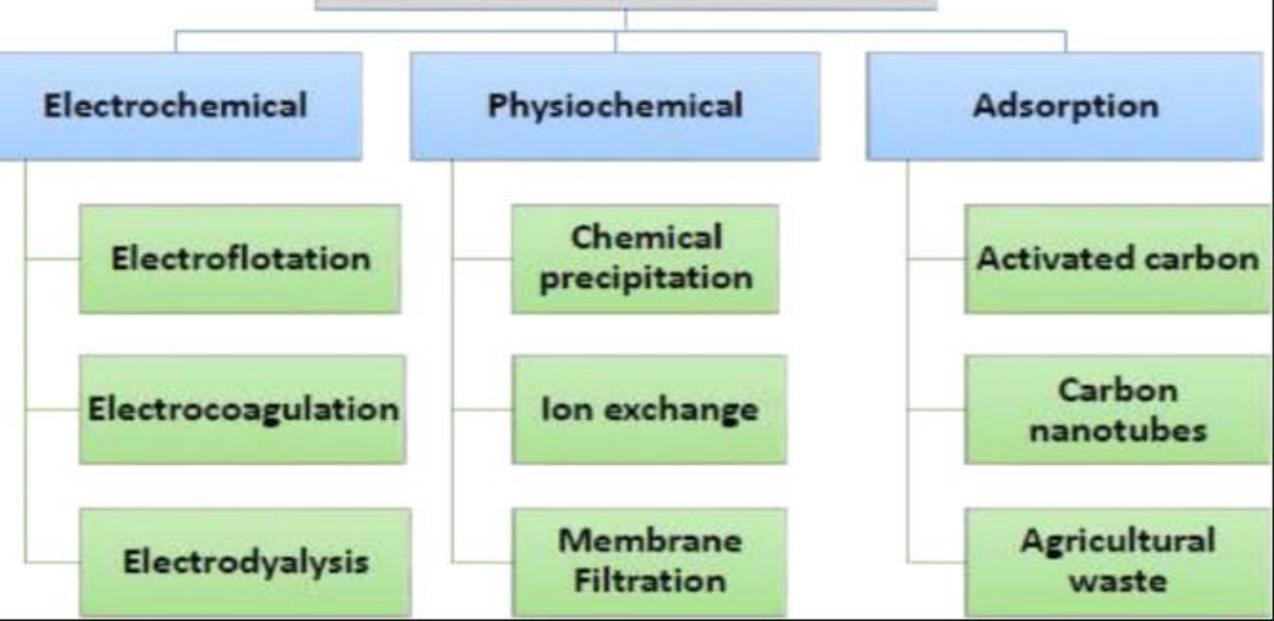
liver damage neurological, hematological disorders kidney failure zinc chromium copper

causes DNA damage, kidney failure

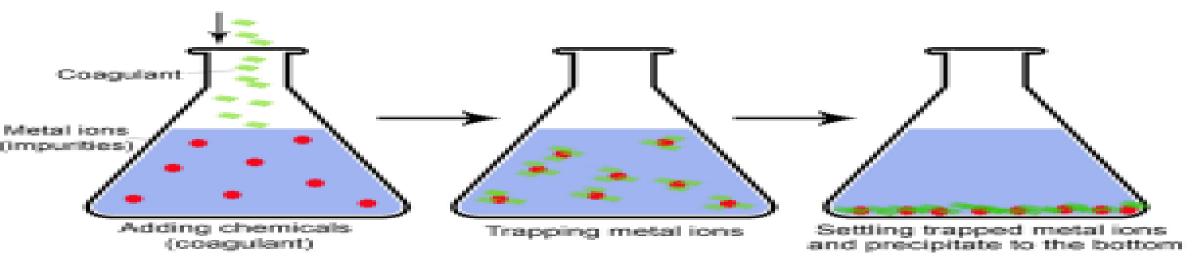
Removal of Heavy metal from wastewater



Conventional methods for the Removal of Heavy Metals



Chemical Precipitation Process



A simple schematic of the chemical precipitation process.

Coagulation and flocculation process

Coegulant Flocoulant Applomeration Treated water Wastewater Flocoulation process Sedimentation process Coagulation process

5 An illustrative schematic of the coagulation-flocculation

Ion Exchange Process

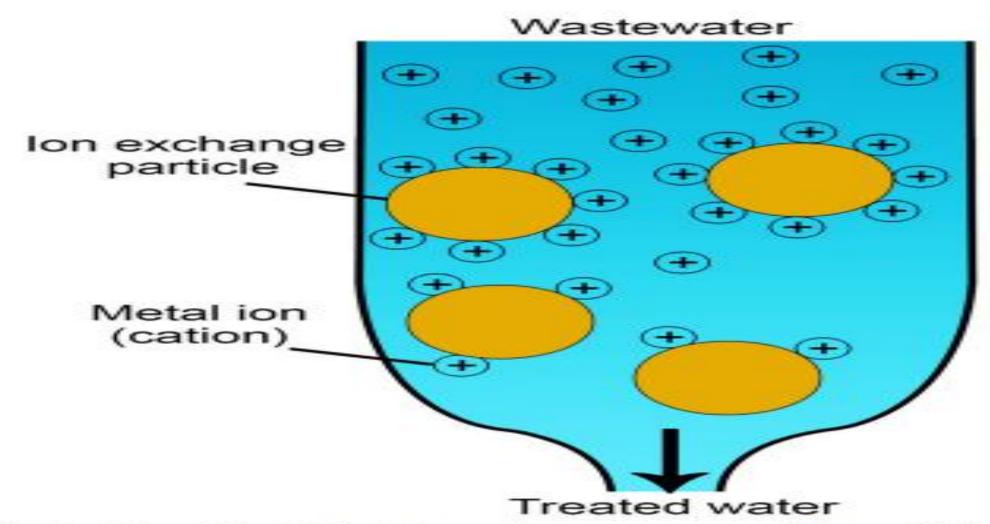
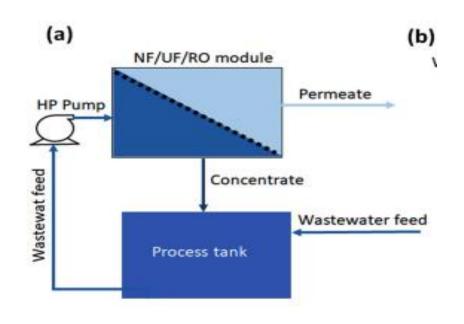


Fig. 8 Schematic of the ion exchange process. The metal ions

Membrane-based Treatments

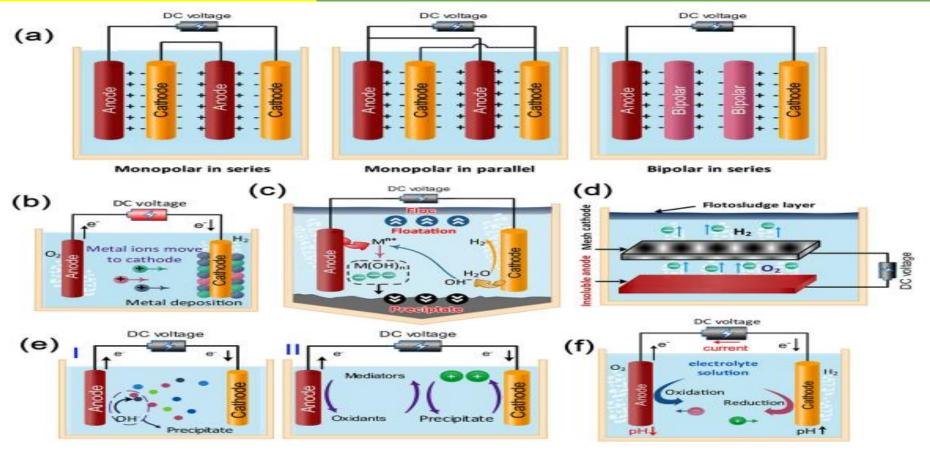


nanofiltration, ultrafiltration, or reverse osmosis method



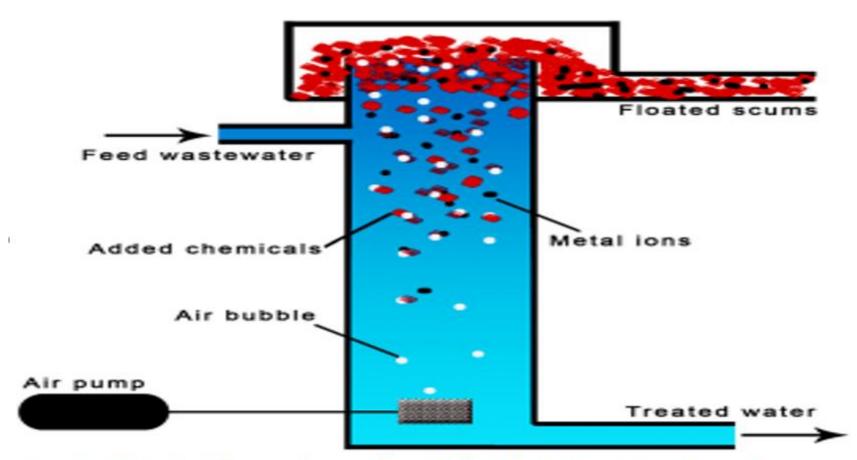
The separation capabilities of different membranes against different pollutants.

Electrochemical methods and arrangements used for heavy metals removal



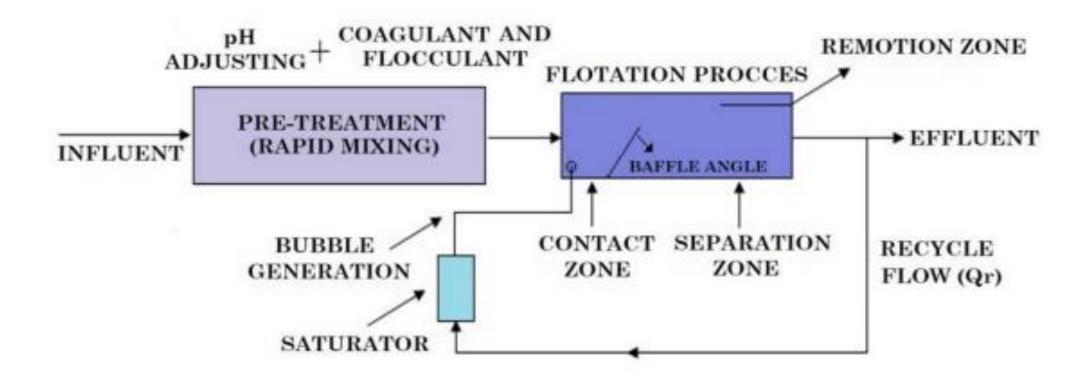
a Different arrangements of electrodes in electrolysis cell (i.e., monopolar electrodes in series (MP-S), monopolar electrodes in parallel (MP-P), and bipolar electrodes in series (BP-S)), b mechanism of the electrochemical reduction method in which positive metal ions are deposited over the cathode, c electrocoagulation process showing the flotation and precipitation of metal ions, d basic electroflotation methods—anions are released from anode to combine with cations (metal ions) and then floating over the water, e electrochemical oxidation (EO) process ((I) direct EO in which the ions react with OH and precipitate or (II) indirect EO in which mediators are used to forming oxidants that interact with the ions to be removed), and d electro-Fenton process in which OH— is released at anode and reacts with metal ions, and hence wastewater is purified due to precipitation

Flotation Treatment Process



An illustrative schematic of the flotation treatment process.

Dissolved Air Flotation

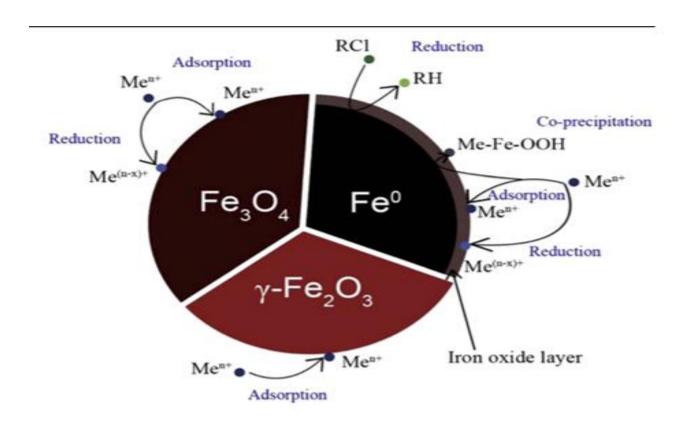


Process diagram of conventional DAF.
Source

Removal of metals by Metal Oxide Nanoparticle

Iron oxides(IO),oxy-hydroxides including amorphous hydrous ferric oxide (FeOOH), maghemite(γ Fe₂O₃),and magnetite(Fe₃O₄).

Having higher adsorption affinity, adsorption capacity, and faster adsorption rate in comparison to many other adsorbents
RCl=chlorinated compounds



Schematic representation of the removal mechanisms of pollutants by iron-based nanoparticles

Conclusion

Methods of Heavy metal depends on Kind of Industry, Type of pollutant and the metal used and its the cost of Removal method

Thank you